SUMMARY OF EXISTING INFORMATION AND IDENTIFICATION OF DATA GAPS TECHNICAL MEMORANDUM

Oakland Bay Shelton, Washington

Prepared for

Washington State Department of Ecology Toxics Cleanup Program Southwest Regional Office

Note:

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Washington State Department of Ecology
Toxics Cleanup Program
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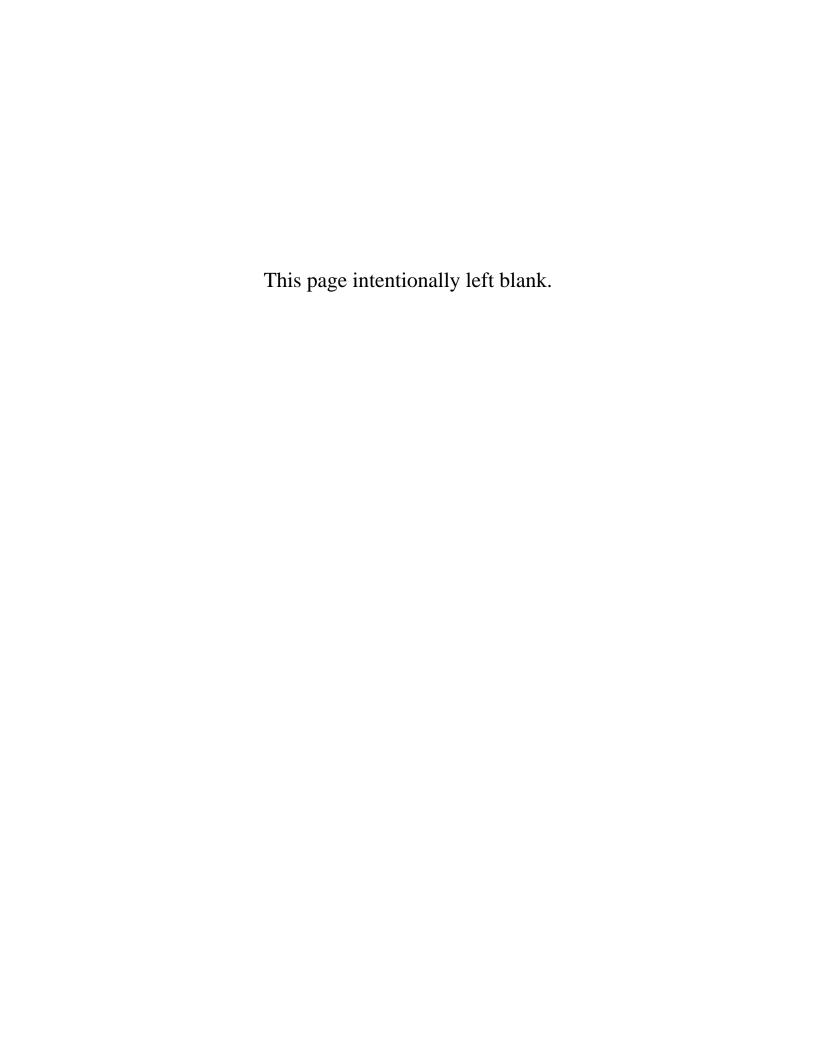
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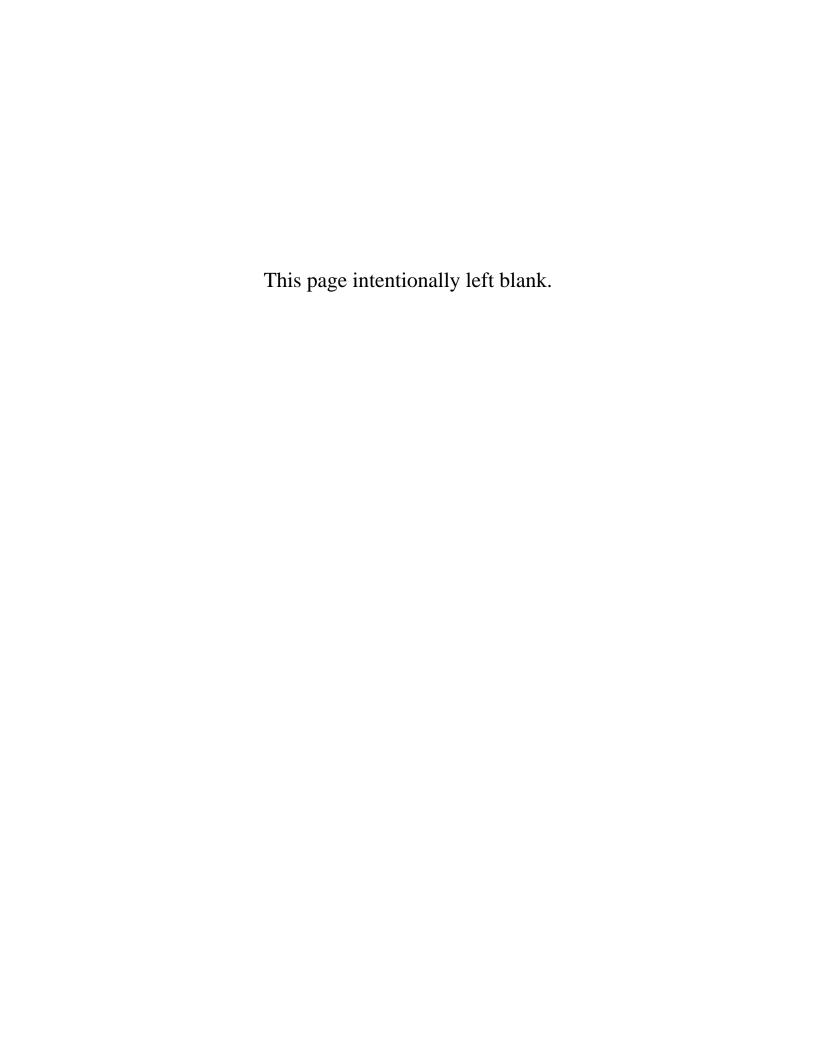
Contents

Abb	previations and	d Acronyms	V
1.0	Introduction.		1
2.0	Oakland Bay and Shelton Harbor		
	2.1. Study A	rea Description	5
	•	rea History	
		ents of Potential Concern (COPCs) in Marine Sediments	
	2.3.1.	Dioxins and Furans	
	2.3.2.	PCBs	
	2.3.3.	Chlorinated Pesticides	13
	2.3.4.	SVOCs	
	2.3.5.	Petroleum Products	14
	2.3.6.	Tributyltin	15
	2.3.7.	Metals	15
	2.3.8.	Wood Waste	16
3.0	Potential Sou	rces of Aquatic Sediment Contamination and Data Gaps	19
	3.1 Former Rayonier Pulp Mill / Manke Lumber Company / ITT Rayonier Research		ı
	Laborato	ory	19
	3.1.1.	Historical Operations – Former Rayonier Pulp Mill	19
	3.1.2.	Historical and Current Operations – Manke Lumber Sort Yard	
	3.1.3.	Historical Operations – ITT Rayonier Research Laboratory	23
	3.1.4.	Previous Investigations	
	3.1.5.	Contaminants of Concern	
	3.1.6.	Potential Contaminant Pathways to the Aquatic Environment	
	3.1.7.	Data Gaps	
	3.2. Simpson	Timber Company	
	3.2.1.	Historic and Current Operations	27
	3.2.2.	Simpson Waterfront Facility	
	3.2.3.	Eagle Point Plywood Facility	
	3.2.4.	Previous Investigations	
	3.2.5.	Contaminants of Concern	
	3.2.6.	Potential Contaminant Pathways to the Aquatic Environment	33
	3.2.7.	Data Gaps	
	3.3. Evergree	en Fuel Company / Former Marine Railway / Shelton Marina	33
	3.3.1.	Evergreen Fuel Company	
	3.3.2.	Former Marine Railway Facility	34
	3.3.3.	Shelton Yacht Club and Marina	35
	3.3.4.	Previous Investigations	
	3.3.5.	Contaminants of Concern	
	3.3.6	Potential Contaminant Pathways to the Aquatic Environment	40

3.3.	/. Data Gaps	40
3.4. Bulk	Fuel Facilities along West Shore of Oakland Bay	41
3.4.		
3.4.	2. Potential Contaminant Pathways to the Aquatic Environment	42
3.4.		
3.5. Wood	d-Treated Marine Lumber	
3.5.	1. Contaminants of Concern	43
3.5.	2. Potential Contaminant Pathways to the Aquatic Environment	43
3.5.		
3.6. Sites	in Downtown Shelton	44
3.6.	1. Contaminants of Concern	44
3.6.	2. Potential Contaminant Pathways to the Aquatic Environment	44
3.6.		
3.7. City (of Shelton Municipal Sewage and Stormwater Systems	44
3.7.	Shelton Wastewater Treatment Plant	44
3.7.	2. Previous Wastewater Treatment Plant	45
3.7.	3. Eagle Point Community Wastewater Treatment Facility	46
3.7.		
3.7	5. Contaminants of Concern	50
3.7.	6. Potential Contaminant Pathways to the Aquatic Environment	50
3.7.	7. Data Gaps	51
3.8. Sites	near Oakland Bay	51
3.8.	1. Johns Prairie Industrial Park	51
3.8.	2. Shelton WSDOT Maintenance Facility	52
3.8.	3. Bayshore Union 76 Gas Station – East 3841 SR 3	52
3.8.		
3.8		
3.8.	6. Sewage Discharge into Malaney Creek	53
3.8.	7. Contaminants of Concern	53
3.8.	8. Potential Contaminant Pathways to the Aquatic Environment	53
3.8.	9. Data Gaps	54
3.9. Oakla	and Bay	54
3.9.	1. Sediment Studies in Oakland Bay	54
3.9.	2. Shellfish Reserves	57
3.9.	3. 2004/2005 Fecal Coliform Bacteria TMDL	57
4.0 Summary	of Data Gaps	59
5.0 Reference	·S	61
Appendix A	ITT Rayonier Laboratory Chemicals	
Appendix B	National Dioxin Study Data	T 1 ·
Appendix C	2004 Survey of Discharge Locations along Oakland Bay and Hammersle	y Inlet
Appendix D	DNR Aquatic Boundaries and Parcels	
Appendix E	Chapman Cove Piling Lease Location Information	

Tables

Table 3-1.	Summary of industrial, stormwater, and sewer discharge points along the Shelton Harbor waterfront
Table 3-2.	Results of 1989 storm drain sediment study in the City of Shelton49
Table 3-3.	2002 metal results (µg/L) in groundwater collected from a monitoring well located at West Franklin and 12th Street, Shelton, Washington50
	Figures
Figure 2-1.	Vicinity map of Oakland Bay in Mason County, Washington7
Figure 2-2.	Site map of Shelton Harbor, Shelton, Washington9
Figure 3-1.	Location of discharge points surrounding Shelton Harbor, Shelton, Washington
Figure 3-2.	Previous investigation sample locations and transformer locations in Shelton Harbor, Shelton, Washington
Figure 3-3.	Site map and sediment sample locations of the former Evergreen Fuel site, Shelton, Washington
Figure 3-4.	Sediment sample locations in Oakland Bay from studies conducted between 1989 and 1999



Abbreviations and Acronyms

AST aboveground storage tank

ATSDR Agency for Toxic Substances and Disease Registry

BTEX benzene, toluene, ethylbenzene, and xylenes

cfu colony forming units

CSL cleanup screening level

COPC constituents of potential concern

CWA Clean Water Act

DDD dichlorodiphenyldichloroethane

DDE dichlorodiphenyldichloroethylene

DDT dichlorodiphenyltrichloroethane

DNR Washington Department of Natural Resources

Ecology Washington State Department of Ecology

HPAH high molecular weight polycyclic aromatic hydrocarbon

kg kilogram

L liter

LPAH low molecular weight polycyclic aromatic hydrocarbon

μg microgram mg milligram

MLLW mean lower low water

MTCA Model Toxics Control Act

NFA No Further Action

NOAA National Oceanic and Atmospheric Administration

NPDES National Pollutant Discharge Elimination System

OC organic carbon normalized

PAH polycyclic aromatic hydrocarbon

PBT persistent, bioaccumulative, toxic chemical

PCB polychlorinated biphenyl

ppb parts per billion

ppm parts per million

ppt parts per trillion

PCP pentachlorophenol

PSDDA Puget Sound Dredged Disposal Analysis

PSEP Puget Sound Estuary Program

RI Remedial Investigation

SAP Sampling and Analysis Plan

SMS Sediment Management Standards

SQS Sediment Quality Standard

SR State Route

SVOC semi-volatile organic compound

SWRO Southwest Regional Office

TBT tributyltin

TCDD 2,3,7,8-tetrachlorodibenzo-p-dioxin

TOC total organic carbon

TPH total petroleum hydrocarbons

TDS total dissolved solids

USEPA U.S. Environmental Protection Agency

UST underground storage tank

UV ultraviolet

WAC Washington Administrative Code

WDFW Washington Department of Fish and Wildlife

WDOH Washington Department of Health

WSDOT Washington State Department of Transportation

WWTP wastewater treatment plant

1.0 Introduction

Oakland Bay is one of seven bays identified as a priority for environmental restoration by the Washington State Department of Ecology (Ecology) as part of the Toxics Cleanup Program's Puget Sound Initiative. Ecology has identified Oakland Bay for focused investigation efforts, sediment cleanup if needed, and source control actions. Earlier environmental investigations throughout the bay, including the Shelton Harbor area, have indicated that chemicals have been generated by historical and current industrial and commercial activities around the bay. Previous sediment sampling in the area has indicated that chemicals in bay sediments and biota may pose a risk to the environment, with data that exceed the Sediments Management Standards (SMS), Washington Administrative Code (WAC) Chapter 173-204, and other established thresholds of environmental concern (USEPA 1998, Long & Morgan 1991, Long et al. 1995).

Ecology is directing sediment investigations, a human health and ecological risk assessment, and a sediment transport study of the bay focusing on the marine environment associated with terrestrial and aquatic sources. This study focuses on contaminated sediment and their toxic effects on the marine environment, not on water quality issues (including bacteria such as fecal coliform). This Summary of Existing Information and Identification of Data Gaps report has been prepared to support the development of a Sampling and Analysis Plan (SAP) for field investigations of the bay. The purpose of this report is to compile readily available information concerning sediment characterization (i.e., data from potential sources and activities relating to chemicals in sediments, results from earlier marine environmental investigations, etc.) and to identify data gaps that will be addressed by the SAP.

Information for the report has been gathered from the following sources:

Relevant reference literature on earlier investigations

Reconnaissance Survey of Inner Shelton Harbor Sediments (Ecology 2000)
Oakland Bay TMDL Quarterly Progress Report (November 2004 through August 2005) (WDOH 2005)
Oakland Bay (Dye and Modeling) Study (Ecology 2004a)
Quality Assurance Project Plan – Fecal Coliform Bacteria TMDL for Oakland Bay-Hammersley Inlet and Tributaries (Ecology

 Regulatory agency database search report obtained using Environmental Data Resources, Inc. (EDR), a commercial database service that searches U.S. Environmental Protection Agency (USEPA) and Ecology records

2004b).

within a three-mile radius centered near Front Street and E. Railroad Avenue (EDR 2008);

- Ecology Southwest Regional Office (SWRO) Central Records relevant Ecology site file information on sites with reported releases of hazardous materials as identified in the EDR database search report (file information dated from 1995 to the present);
- Ecology Environmental Information Management (EIM) Database System (http://apps.ecy.wa.gov/eimreporting/search.asp) relevant sediment quality studies conducted in Shelton Harbor and Oakland Bay as identified in the EIM database including:
 - ☐ 1997-1999 PSAMPNOA Puget Sound Assessment and Monitoring Program and the National Oceanic and Atmospheric Administration (NOAA) National Status and Trends Program.
 - □ 1989-1995 PSAMP HP Puget Sound Assessment and Monitoring Program's historical sediment monitoring.
 - □ 1992 DNRREC92 Washington Department of Natural Resources (DNR) Aquatic Lands Sediment Quality Reconnaissance.
- Washington Southwest Regional State Archives relevant Ecology site file information on sites with reported releases of hazardous materials (file information dated from late 1960s to 1994);
- Ecology Facility/Site Identification System:
 http://apps.ecy.wa.gov/website/facsite/viewer.htm
- USEPA Envirofacts Warehouse: http://www.epa.gov/enviro/index.html
- Ecology Online Publication Archives: http://www.ecy.wa.gov/pubs.shtm
- Ecology Online Shoreline Aerial Photos: http://apps.ecy.wa.gov/shorephotos/index.html
- Washington State Department of Health (WDOH)
 - □ Shellfish Safety Information: http://ww4.doh.wa.gov/scripts/esrimap.dll?name=bioview&Cmd= Map&Step=1
 - □ 2003 Shoreline Survey of Oakland Bay (Berbells 2003).

•	Washir	ngton Department of Fish and Wildlife (WDFW)
		Steve Cuinnell (Fish Program) – fish tissue sampling studies coordinator (Cuinnell 2008)
		Jim West (Fish Program) – fish tissue analysis (West 2008).
•	The Ci	ty of Shelton Public Works Department (City of Shelton 2008)
		Storm Water System Map: http://www.ci.shelton.wa.us/documents/STORMMAP_000.pdf
		Sewer System Map: http://www.ci.shelton.wa.us/documents/SewerMap_000.pdf.
•	Histori	cal Land Use Resources:
		Property characteristics and ownership information obtained from the Mason County Assessor online database (Mason County 2008);
		Sanborn Fire Insurance maps with coverage of the City of Shelton in 1910, 1923, 1929, 1944, and 1961 (UW 2008a);
		Historical parcel maps of Mason County with coverage in 1925, 1935, 1941, 1955, 1969, and 1988 (UW 2008b)
		Historical aerial photographs with coverage from the early 1940s through the early 2000s (UW 2008c);
		Historical topographic maps with coverage in 1952 and 1981 (UW 2008d);
		Historical nautical charts with coverage in 1905, 1934, 1938, 1947, 1957, 1968, 1969, 1970, 1973, 1985, 1989, 1991-1993, 1996, 1998, and 1999 (NOAA 2008);
		Telephone directory and Polk's city directory of Shelton, Washington in 5-year intervals from 1947 to 1996 (UW 2008e);
		Shelton: the first century plus ten (Thomas 1996);
		Mason County Profile 2007 – a supplement to the Shelton-Mason County Journal (MCHS 2008)

Mason County – Thumbnail History; Essay No. 7730
(HistoryLink.org 2008);

☐ Historical real property tax assessment records reviewed from the Washington Southwest Regional State Archives in Olympia, Washington.

4

2.0 Oakland Bay and Shelton Harbor

2.1. Study Area Description

Oakland Bay is a shallow estuary approximately 4 miles long and 3/4 of a mile wide, with water depths ranging between 10 and 35 feet (Figure 2-1). A large area of the foreshore (intertidal area between mean low water and mean high water) is exposed during low tides at the north end of the bay. This intertidal zone is predominately mud flats with narrow deep channels. Due to the restrictive nature of Hammersley Inlet, a long narrow waterway linking the bay to the Puget Sound Basin, the water in Oakland Bay has high refluxing, low flushing, and high retention rates (Ecology 2004a). Nine major creeks discharge into the bay: Deer, Cranberry, Malaney, Uncle John, Campbell, Johns, Shelton, and Goldsborough. The waters of Shelton Harbor and the northern portions of Oakland Bay are currently listed as impaired by the State of Washington under Section 303(d) of the Clean Water Act (CWA) because of fecal coliform bacteria levels (Ecology 2004b).

2.2. Study Area History

Since the mid-1800s, a number of timber industries dominated the Shelton Harbor waterfront, including sawmills and plywood manufacturing, pulp and paper production, and insulation board and fiber board manufacturing. From the mid-1800s to the early 1930s, three sawmills operated along the Shelton waterfront (one operated by the Simpson Timber Company); a shingle mill also operated at Eagle Point (later changed to a plywood plant). A bulkhead was constructed in the mid-1920s to expand operations area along the shoreline. In the early 1940s, Simpson acquired the remaining two sawmills and the shingle mill, and redeveloped the Shelton waterfront into a large sawmill and plywood manufacturing complex that included a fiber and insulating board plant. Simpson installed a second bulkhead during the expansion of their waterfront plant between the early 1940s through the 1960s (Figure 2-2). In 1974, Simpson shifted shingle mill operations at Eagle Point (Olympic Plywood) to the north portion of Simpson waterfront plant.

From 1926 to 1957, a pulp and paper sulfite mill identified as the Rayonier Pulp Mill, operated on the south side of Shelton Harbor. Wood chips used to produce pulp were either supplied by the adjacent sawmills or brought in by barges or trucks. Sulfite waste liquor was a by-product generated during the pulping process. Initially, the liquor was released into the harbor during outgoing tides or conveyed through a five-mile long pipeline that ran east along Hammersley Inlet to be discharged at Mill Creek. Environmental impacts to shellfish in the bay and inlet occurred soon after release of the pulp mill liquor. Rayonier then redirected the waste to Goose Lake, located three miles inland to the west (early 1930s to the mid-1940s), or evaporated the liquor into a thick syrup that was then sprayed on roads in Shelton for dust control (1934 to the late 1930s). In 1945, Rayonier constructed a facility located on the hillside above the pulp mill

to dispose of waste liquor by burning it. The pulp mill and burn plant continued to operate until both facilities were shut down in 1957; both facilities were demolished by the mid-1960s.

From 1936 to 1995, the ITT Rayonier central chemical laboratory operated west of and adjacent to the Rayonier pulp mill to support production of rayon fibers and other specialty wood products, including cellophane, cellulose acetate, nitrocellulose used for explosives, and rayon tire cords (Figure 2-2).

The Manke Lumber Company purchased the pulp mill property in the 1960s and converted it into a sorting yard for logs brought in by trucks, which were then shipped out in rafts to Tacoma for processing (Figure 2-2).

Pilings had been installed in the harbor to support three over-water railroad spurs used for unloading logs from trains directly into the water; other pilings were installed across the harbor for stabilizing log rafts. Logs were stored in the water prior to processing. By the early 1970s, Simpson abandoned most log boom storage, and developed a dry log sort yard located inland about three miles west of Shelton. Some log rafting operations continued with log truck unloading shifted to the north end of the harbor next to the vessel haul-out and marine railway facility (Figure 2-2).

A bulk fuel storage marine facility operated at the north end of Shelton Harbor (Standard Oil/Evergreen Fuel) from the early 1900s to late 2005. In addition to Evergreen Fuel, three other bulk fuel storage facilities also operated about a half mile northeast of Shelton along the west shore of Oakland Bay (Union Oil, Shell, and ARCO), from the early 1930s to the mid-1980s.

Other businesses that developed within the city of Shelton west of and adjacent to the Simpson plant since the late 1920s include gasoline service stations; automotive, truck, and heavy lumber equipment maintenance and repair services; dry cleaning facilities; and other urban businesses.

The Shelton Yacht Club marina has operated at the northeast corner of the inner harbor since 1974.

The Squaxin Island Tribe has harvested shellfish from Oakland Bay for several centuries. Commercial shellfish harvesting has been important since the 1880s (Kenny 2007). Pollution associated with sulfite waste liquor discharged by the Rayonier pulp mill is believed to have lead to declining commercial oyster production by the mid-1940s (Shaffer 2003). Re-populating with oysters and other shellfish in the northern portions of Oakland Bay, including Chapman Cove, and developing a second-generation of shellfish production began in the late 1960s. Over the last 40 years, water quality impacts to Oakland Bay appear to have shifted from industrial effluent to non-point source pollution, including low dissolved oxygen (DO), chlorine from sewage outfalls, sedimentation and siltation, and herbicides, which may have contributed to recent closures of shellfish harvesting in portions of the bay (Shaffer 2003). Fecal coliform contamination from excessive inflow and infiltration to the city's aging sewer and stormwater collection systems, onsite septic systems, and surface water runoff from small farms, may also have contributed to recent closures of shellfish harvesting in portions of the bay.



Figure 2-1. Vicinity map of Oakland Bay in Mason County, Washington.

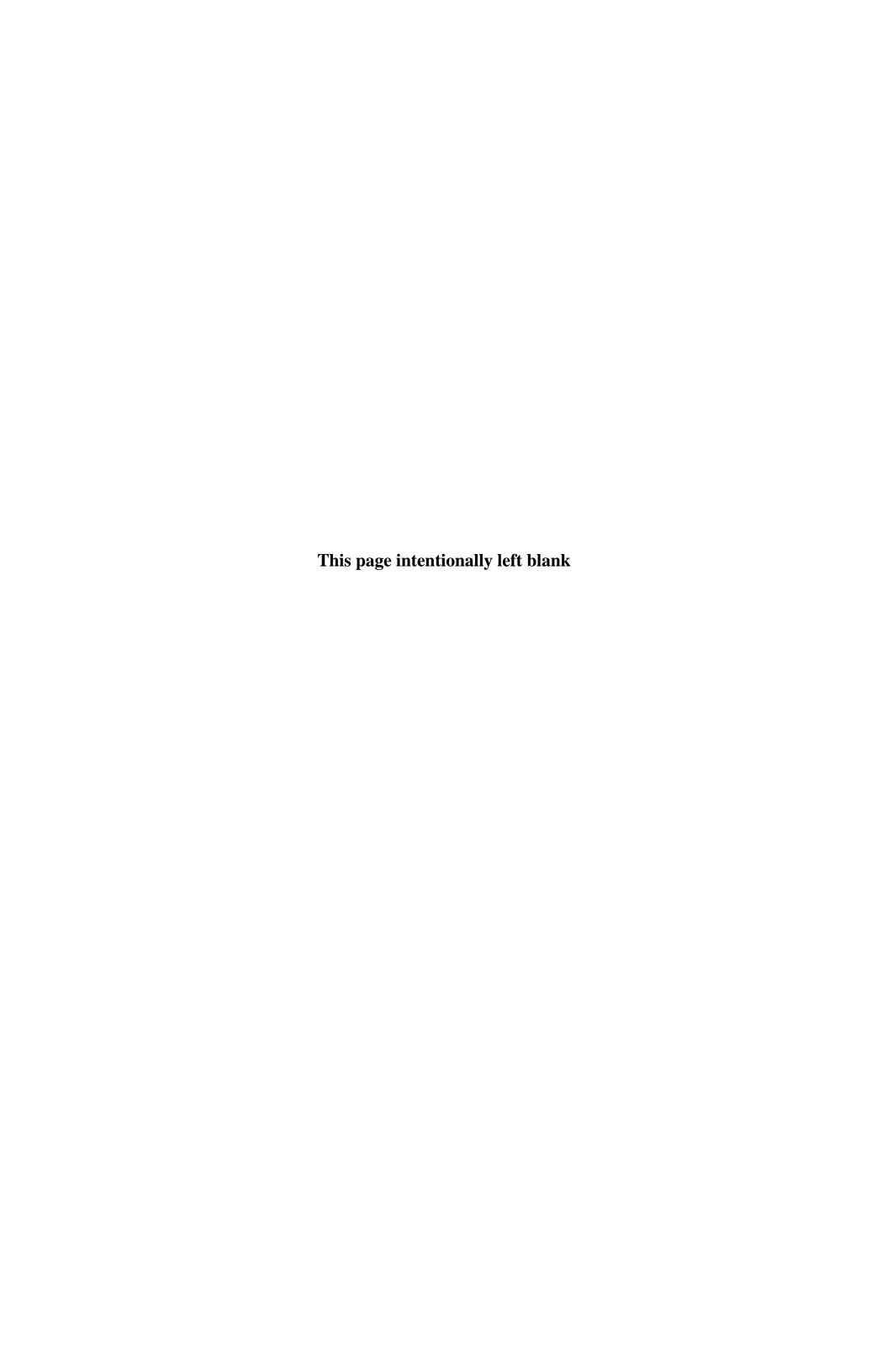
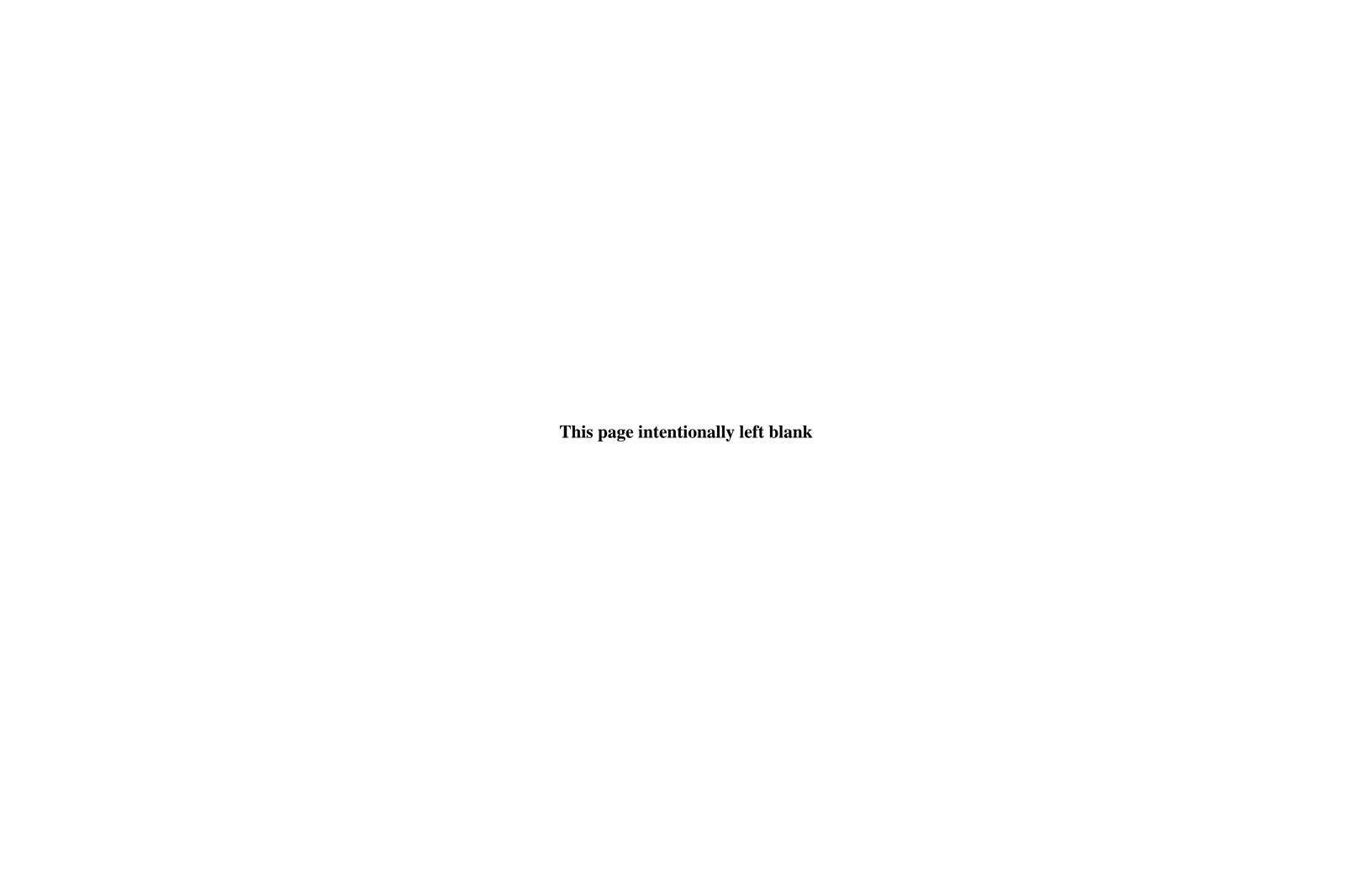




Figure 2-2. Site map of Shelton Harbor, Shelton, Washington.



As of 2007, the City of Shelton has an estimated population of 8,895 people (MRSC 2008). Shelton Harbor and Oakland Bay have received discharges from the city sewer and stormwater collection systems, wastewater treatment plant (WWTP) discharge outfalls, urban and industrial stormwater outfalls associated with timber industries, septic systems in various stages of maintenance outside Shelton City limits, and non-point source runoff from stormwater (City of Shelton 2008, Brown & Caldwell 1988, Michaud 1988). Organic wastes remaining in sediments in the cove from the sources discussed above may be contributing to low levels of DO, in addition to the natural oceanographic characteristics of the bay. The first sewer lines were installed in downtown Shelton in the early 1900s using vitrified clay pipe intentionally left open to drain away groundwater and rainwater induced infiltration. In 1950, the city constructed a WWTP on Pine Street next to the railroad embankment north of the Simpson plant to replace the old sewage disposal pump station that stood on Front Street near Goldsborough Creek (Thomas 1996). This plant provided primary treatment, including grit removal, sedimentation, singlestage digestion, and chlorination (Shelton-Mason County Journal 1950) and discharged into Oakland Bay. In 1979, the Pine Street plant was replaced by a secondary WWTP constructed at Eagle Point that discharges effluent through a 1,250-foot long outfall extending into Hammersley Inlet. The end of the discharge pipe is positioned approximately 45 feet below mean sea level to ensure dispersal of the effluent.

Land use throughout the study area adjacent to Oakland Bay consists of rural residential, commercial forest, and agricultural, with some industrial and commercial development along the west and south sides of the bay. Approximately 100 small farms are located within the watershed (Berbells 2003). Septic systems are used to treat waste throughout most of the study area.

2.3. Constituents of Potential Concern (COPCs) in Marine Sediments

COPCs to Oakland Bay sediments and biota were identified based on known chemical associations with historic and current land uses and activities, as well as from earlier sediment investigations conducted within the study area. The following chemicals are identified as COPCs, some of which have Washington State SMS criteria:

- Polychlorinated dibenzo-p-dioxins (dioxins) and polychlorinated dibenzofurans (furans)
- Polychlorinated biphenyls (PCBs)
- Chlorinated pesticides
- Semi-volatile organic compounds (SVOCs), including polycyclic aromatic hydrocarbons (PAHs), phenols, pentachlorophenol (PCP), cresols, and phthalates

- Petroleum products (gasoline-, diesel-, and lube oil-range hydrocarbons)
- Tributylin (TBT)
- Heavy metals
- Wood waste (resin acids, guaiacols, ammonia, sulfides).

Many of these chemicals are known to be persistent in the environment, potentially bioaccumulative, and toxic (PBTs), including dioxins/furans, PCBs, and PAHs. The following sections describe the type of processes that produce each of the COPCs listed above.

2.3.1. Dioxins and Furans

Dioxins and furans are byproducts produced from the combustion of organic compounds with chloride present and from pulp bleaching processes. Combustion sources include the incineration of municipal and medical wastes, boilers and industrial furnaces, diesel heavy-duty trucks, sintering plants, automobiles using leaded gasoline, oil-fired utilities, lightweight aggregate kilns that combust hazardous waste, petroleum refining, crematoria, and drum reclamation (USEPA 2006). Dioxin source assessments conducted in Washington indicate that incinerators, hog fuel (wood waste) boilers, bleached pulp and paper mills, cement kilns, kraft black liquor boilers, tire combustion, and sewage sludge incineration are other potential sources of dioxin production (Ecology 1998). Burning salt-laden hog fuel (wood waste from logs rafted on saltwater) has been implicated in the production of dioxins (Ecology 1998). Because PCP is typically contaminated with low concentrations of dioxins, PCP wood treatment facilities are also a concern (Ecology 1998).

2.3.2. PCBs

PCBs belong to a broad family of man-made organic chemicals known as chlorinated hydrocarbons. PCBs were domestically manufactured from 1929 until their manufacture was banned in 1979. They have a range of toxicity and vary in consistency from thin, light-colored liquids to yellow or black waxy solids. Due to their non-flammability, chemical stability, high boiling point, and electrical insulating properties, PCBs were used in hundreds of industrial and commercial applications including:

- Transformers and capacitors
- Other electrical equipment including voltage regulators, switches, reclosers, bushings, and electromagnets
- Oil used in motors and hydraulic systems
- Old electrical devices or appliances containing PCB capacitors

- Fluorescent light ballasts
- Cable insulation
- Thermal insulation material including fiberglass, felt, foam, and cork
- Adhesives and tapes
- Oil-based paint
- Caulking
- Plastics
- Carbonless copy paper

PCBs have been demonstrated to cause cancer, as well as a variety of other adverse health effects on the immune system, reproductive system, nervous system, and endocrine system (ATSDR 2001c).

2.3.3. Chlorinated Pesticides

Organochlorine insecticides were commonly used in the past, but many have been removed from the market due to their health and environmental effects and their persistence. Dichloro-diphenyl-trichloroethane (DDT) is an organochlorine insecticide once widely used in the U.S. before being banned in 1972. Dichlordiphenyldichloroethylene (DDE) and dichlorodiphenyldichloroethane (DDD) are breakdown byproducts of DDT that contaminate commercial DDT preparations. All three compounds are highly persistent and have similar chemical and physical properties; these compounds together are known as total DDT. DDT, DDE and DDD magnify through the food chain, with apex predators such as raptors having a higher concentration of these chemicals stored mainly in body fat than in other animals sharing the same environment. DDT is also highly toxic to aquatic species, including sea shrimp, crustaceans, and many species of fish. In addition to acute toxic effects, DDT may bioaccumulate significantly in fish and aquatic species, leading to long-term exposure to high concentrations (ATSDR 2002).

2.3.4. **SVOCs**

SVOCs are a class of compounds that include PAHs, phenols, PCP, creosols, and phthalates.

PAHs are a group of over 100 chemicals that primarily form by incomplete combustion of carbon-containing fuels, including wood, coal, and gas, garbage, and other organic substances. PAHs are usually found as a mixture consisting of two or more chemicals. They are found in coal tar, crude oil, creosote, roofing tar, and products used to make dyes, plastics, and pesticides (ATSDR 1995c).

Phenols are a class of widely distributed chemicals that are both manufactured and occur naturally. Phenols are used primarily in the production of phenolic resins, the manufacture of synthetic fibers, disinfectants, antiseptic products, algaecides, and fungicides (ATSDR 2006b).

PCP is a manufactured phenolic chemical that does not occur naturally. It has been widely used as a pesticide and wood preservative and is still used industrially as a wood preservative for utility poles, railroad ties, and wharf pilings (ATSDR 2001b). Creosols are methylphenols and are one of the chemicals that, along with PAHs, are in creosote, which is created from the high temperature treatment of wood, coal, or from the resin of the creosote bush. Creosote has been used as a wood preservative in marine lumber applications (e.g., dolphins, pilings) for over 100 years. Creosote-treated pilings and remnants have been identified as a continuous source of marine pollution, as they leach creosols and PAHs to marine waters and sediments (MRC 2008).

Phthalates are widely-distributed synthetic compounds, primarily used as a plasticizer in the production of flexible polyvinyl chloride (PVC) products, in ethyl cellulose and nitrocellulose lacquers, resin solvent, paper coatings, adhesives, as a solvent and fixative in perfumes, and in insecticides (ATSDR 2001a).

2.3.5. Petroleum Products

Petroleum products, such as gasoline, fuel oil (including diesel fuel), and mineral-based crankcase motor oil, are distilled from crude oil and are refined to meet specifications for each use.

Gasoline is a mixture of over 150 compounds, including benzene, toluene, ethyl benzene, and xylenes. Organic lead compounds were added to gasoline as anti-knock agents prior to the mid-1980s. Gasoline is used exclusively for internal combustion engines in automobiles and other vehicles (ATSDR 1995b).

Fuel oils are mixtures of aliphatic and aromatic petroleum hydrocarbons, and may also contain small amounts of nitrogen, sulfur, and other elements as additives. Six types of fuel oil include:

- Fuel oil No. 1 kerosene, range oil, coal oil, and jet fuel
- Fuel oil No. 1-D diesel fuel
- Fuel oil No. 2 home heating oil, No. 2 burner oil, and gas oil
- Fuel oil No. 2-D No. 2 diesel
- Fuel oil No. 4 heavy residual fuel oil, marine diesel fuel, and diesel fuel oil No. 4
- Fuel oil Nos. 5 and 6 Bunker C fuel oil

Fuel oils have many commercial and military uses, including jet fuel; home heating oil; diesel fuel for diesel engines in trucks and heavy machinery; as a vehicle for insecticides and fungicides; space and water heating; road oils; pipeline pumping; and gas compression (ATSDR 1995a).

Mineral-based crankcase motor oil consists of aliphatic and aromatic hydrocarbons, or PAHs, that are distilled from crude oil. Various additives may be included in motor oil to improve performance. Metals such as aluminum, chromium, copper, iron, lead, manganese, nickel, silicon, and tin, are found in used motor oil derived from engine parts as they wear down. Motor oil is used as fuel in boat engines, furnaces and oil burners for domestic and industrial power plants, industrial steam boilers, municipal incinerators, and rotary cement kilns (ATSDR 1997).

2.3.6. Tributyltin

TBT is a highly toxic compound used as an anti-fouling agent in marine paints applied to the bottom of boats and can be released to marine sediments during the practice of scraping vessel hulls. NOAA's Mussel Watch Program, a long-term status and trends program that monitors contaminants in sediments and mussels, includes TBT as an important monitored contaminant (NOAA 2007).

2.3.7. Metals

Metals such as inorganic arsenic, cadmium, chromium, copper, lead, mercury and zinc occur naturally from geologic processes, and are also used extensively in manmade products. Common sources of these metals from processes include:

- Arsenic wood preservative (chromated copper arsenate or CCA) in utility poles, building lumber, and pilings; herbicides and pesticides (ATSDR 2007a; Lewis 1997)
- Cadmium nickel-cadmium batteries, pigments used in plastics, ceramics, and glasses; stabilizers for PVC; coatings on steel and some nonferrous metals; components in various specialized alloys; and fungicides (ATSDR 1999; Lewis 1997)
- Chromium alloying and plating element on metal and plastic substrates for corrosion resistance including high temperature industrial furnaces and cooling towers; pigments; and wood preservative (ATSDR 2000; Lewis 1997)
- Copper electroplated protective coatings; anti-fouling paints; car brake dust; incineration; chemical and pharmaceutical machinery; corrosiveresistant piping; insecticides; and electrical wiring, plumbing, heating, roofing, and building construction materials (ATSDR 2004b; Lewis 1997)

15

- Lead lead-acid batteries; gasoline additive; lead alloys used in bearings, brass and bronze, and some solders; radiation shielding; cable covering; chemical resistant linings; ammunition; and pigments in glass making, ceramic glazes, plastic stabilizers, caulk, and paints (ATSDR 2007b; Lewis 1997)
- Mercury cathodes for production of chlorine and caustic soda; catalysts for manufacture of certain polyurethanes; electrical apparatus; instruments (thermometers, barometers, etc.); amalgam; mercury vapor lamps; mirror coating; boilers; and fungicide in paint (banned since 1990) (ATSDR 1999d; Lewis 1997)
- Zinc alloys; galvanizing iron and other metals; white pigment; fertilizers and animal feed as trace element and disease-control agent; manufacture of rayon (as a crenulating agent), in paper bleaching and glue; wood preservative; catalyst; waterproofing agent; and in fungicides (ATSDR 2005b; Lewis 1997).

2.3.8. Wood Waste

Wood waste is created by deposition of bark, wood chunks, wood chips, and sawdust within the marine environment. These wood products decay over time and can have a variety of physical and chemical adverse impacts on aquatic life, including:

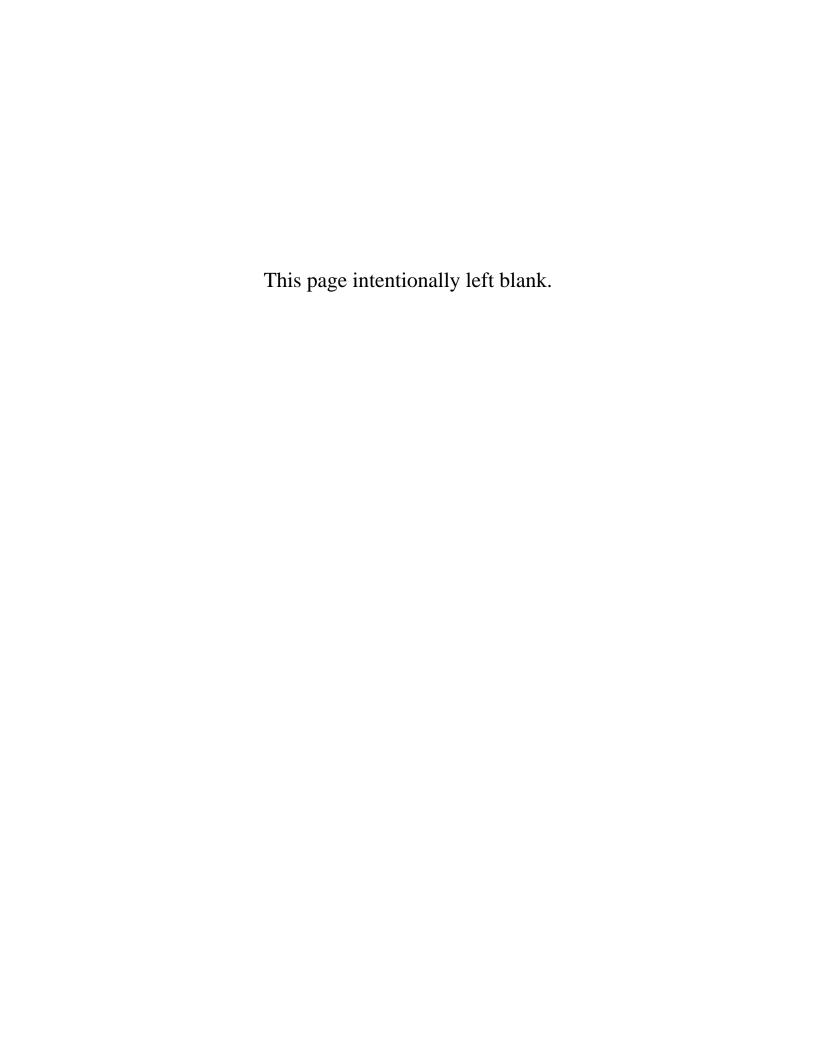
- Organic enrichment of sediments
- Oxygen depletion in the water column
- Alteration of benthic communities to more pollution tolerant species
- Leaching of toxic chemicals such as phenols, methylated phenols, benzoic acid, benzyl alcohol, terpenes, and tropolones
- Physical alteration of the benthic substrate.

The severity of wood waste toxicity depends on the physical form (i.e., size), degree of water flushing, and type of wood it is generated from.

COPCs associated with wood waste include resin acids, guaiacols, ammonia, and hydrogen sulfide. Resin acids and guaiacols are naturally occurring organic chemicals found in wood, hardwood tar, and pulp and paper mill processes. Guaiacols also may be derived from creosote and are present in wood smoke, resulting from the chemical decomposition of lignin (Malcolm Pirnie 2006).

Ammonia occurs naturally throughout the environment in air, soil, water, and in plants and animals. It is an important source of nitrogen that is needed by plants and animals. The largest and most significant use of ammonia and ammonia compounds is the agricultural applications of fertilizers. The small proportion of commercially produced ammonia not incorporated into fertilizers is used as a corrosion inhibitor, in the purification of water supplies, as a component of household cleaners, and as a refrigerant. It is also used in the pulp and paper, metallurgy, rubber, food and beverage, textile, and leather industries (ATSDR 2004a).

Hydrogen sulfide is a poisonous, flammable, colorless gas with a characteristic odor of rotten eggs. It occurs both naturally and from man-made processes. Hydrogen sulfide is a component of gases associated with volcanoes, sulfur springs, swamps, stagnant bodies of water, and in crude oil and natural gas. It is also associated with municipal sewers and WWTPs, manure-handling operations, and pulp and paper operations. Hydrogen sulfide is released primarily as a gas and disperses in the air; however, in some instances, it may be released in the liquid waste of an industrial facility or as the result of a natural event. It can change into sulfur dioxide and sulfuric acid, and is soluble in water (ATSDR 2006a).



3.0 Potential Sources of Aquatic Sediment Contamination and Data Gaps

3.1 Former Rayonier Pulp Mill / Manke Lumber Company / ITT Rayonier Research Laboratory

3.1.1. Historical Operations – Former Rayonier Pulp Mill

The former Rayonier sulfite pulp mill occupied over 10 acres of land along the southwest end of the Shelton Harbor. The mill produced pulp and paper from 1926 to 1957. A former burn plant occupied approximately six acres of land on the hillside above the mill, which burned sulfite liquor waste from 1945 until 1957. Both sites are currently owned by the Manke Lumber Company, who has operated a log sorting yard across the former mill site and a headquarters office on the former burn plant since the late 1960s (Figure 3-1).

3.1.1.1. Pulp Production

Limited operations information indicates that the acid sulfite process was used at the Shelton pulp mill to produce pulp from wood chips. Information on this process was obtained from the process description for a similar Rayonier pulp mill in Port Angeles (Malcolm Pirnie 2006). Both mills were owned and operated by the same company during the same time period using the sulfite pulp production process. This pulp process required a large amount of wood chips, acid digestion tanks, blow pits, screening tanks, hog fuel boilers and stacks, landfill disposal staging areas, and a storage facility for new and spent digestion chemicals. Wood chips for the Shelton mill were obtained either from the adjacent sawmills or brought in on barges, railcars, or trucks. Barges loaded with wood chips were moored in the harbor adjacent to the pulp mill. The Shelton pulp mill started generating its own wood chips onsite in the early 1950s (Thomas 1996).

3.1.1.2. Bleaching and Paper Production

Based on available information, paper production occurred from 1926 to 1930, and to a limited extent after 1930 when the mill began producing pulp for manufacturing rayon and other specialty wood products. In general, the bleaching process consisted of subsequent treatments of pulp using chlorine, caustic soda, sodium hypochlorite, and chlorine dioxide, with multiple washing stages. Bleached pulp was then transported to a drying machine where water was removed. The pulp was then pressed and dried, and the resulting material made into large rolls.

3.1.1.3. Disposal of Sulfite Waste Liquor

The main difference between the Shelton and Port Angeles pulp mills was the handling and disposal of the sulfite waste liquor, the end liquid by-product of the acid digestion process for producing pulp. The liquor consisted of dissolved lignin, sugars, and other organic constituents. Starting in 1926 through the early 1930s, the spent liquor was either released to the harbor during

outgoing tides or it was conveyed in a five-mile long pipeline that ran east along Hammersley Inlet to be discharged at Mill Creek (Thomas 1996). The locations of discharge points into the harbor from the mill are unknown. Environmental impacts to the shellfish industry in Oakland Bay and Hammersley Inlet occurred soon after the pulp mill began production. To mitigate the impacts, Rayonier constructed a three-mile long pipeline that ran inland from the mill to Goose Lake located near the airport (Thomas 1996). The liquor was pumped and discharged into the lake to settle and eventually evaporate; it was later discharged nearby to two diked disposal ponds. At the pulp mill site, liquor was temporarily stored in five aboveground storage tanks (ASTs) located east of and adjacent to the mill before being pumped to the lake.

In the mid- to late 1930s, Rayonier attempted other methods for disposing the waste liquor, including treatment of roads in Shelton for dust control and producing other specialty products such as linoleum paste and plywood adhesives.

By the early 1940s, discharging the liquor to Goose Lake was discontinued due to evidence that it had seeped into groundwater and a nearby creek (Thomas 1996). In 1945, Rayonier constructed a burn plant with ovens and a 325-foot high stack on the hillside above the pulp mill to incinerate the liquor. The burn plant continued operations until the pulp mill shut down in 1957.

3.1.1.4. Aboveground Storage Tanks

A 15,000-gallon fuel oil AST surrounded by a three-foot high earth dike was present east of and adjacent to the pulp mill facility, along with the five ASTs containing spent liquor (discussed above). Four acid-containing ASTs used in pulp production were situated immediately west of the pulp mill facility. Both the fuel oil and the acid ASTs were present from the mid-1940s to the early 1960s, based on Sanborn fire insurance maps.

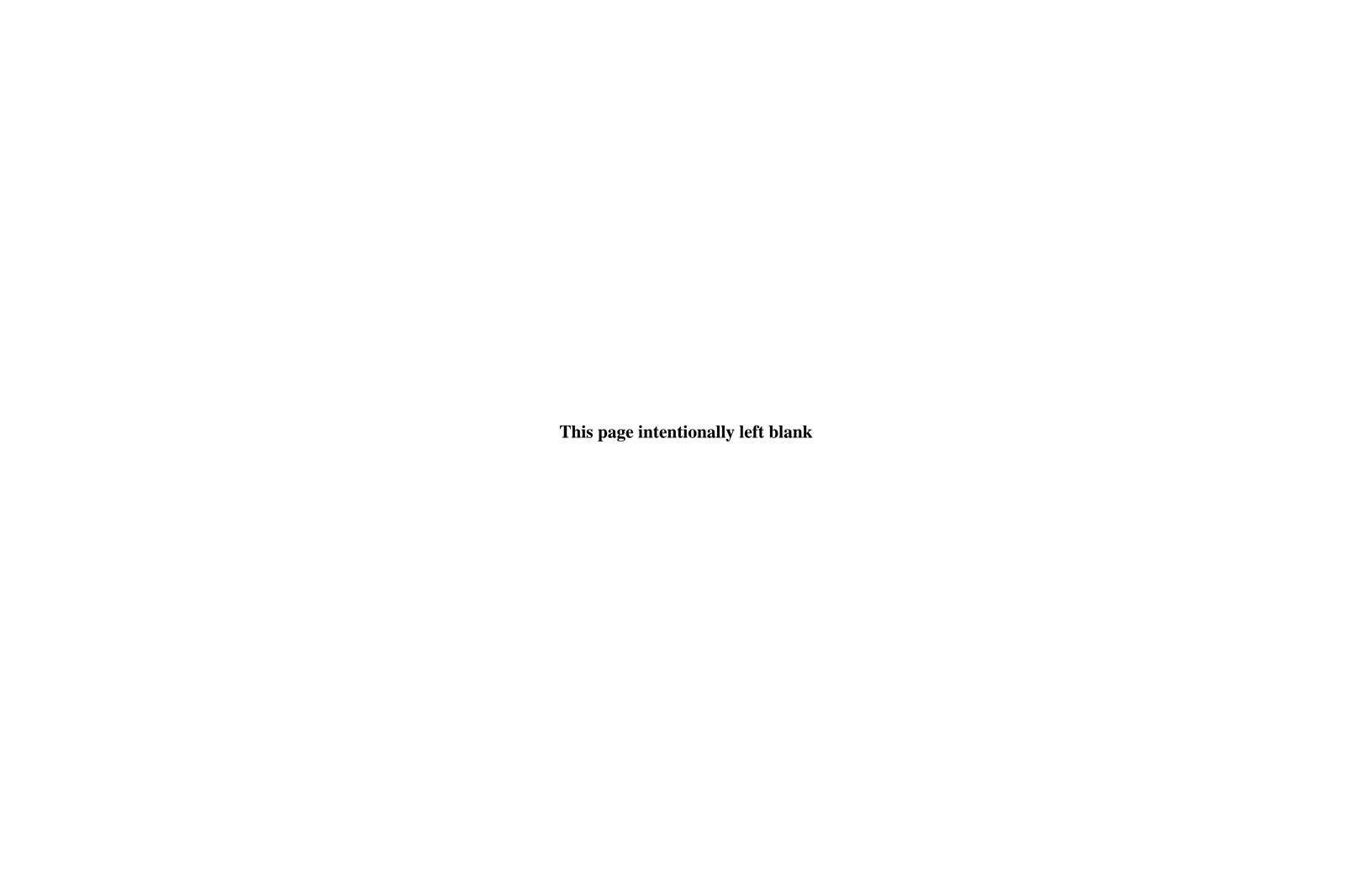
3.1.2. Historical and Current Operations – Manke Lumber Sort Yard

The Manke Lumber Company has operated an unpaved log sorting yard across the entire former pulp mill property since the late 1960s (Figure 3-1). A truck wash facility and a truck and heavy logging equipment maintenance facility exist on the property. No information was available regarding the use, storage, and disposal of hazardous materials by the log sorting company from the late 1960s through the mid-1980s. In 1985, oil reportedly discharged into a hand dug ditch in the vicinity of the maintenance facility, according to an Ecology water quality file. A substantial oil spill had occurred at the site (spill location not reported) in the previous year. A 660-gallon diesel fuel AST with no secondary containment was identified in the Spill Prevention, Control, and Countermeasure (SPCC) plan for the site. Floatable oil, solvent, and emulsified oil were reportedly used at the site.

Information regarding the Manke property during an Ecology bacterial investigation indicated that runoff had been very turbid with visible sheen on the water surface from oils and grease (Michaud 1988). The poor quality runoff water emanated from the unpaved log sort yard and truck wash facility. The company had installed a closed-loop truck wash system prior to 1988 to control some of the runoff; however, overland flow continued to discharge into the harbor east of



Figure 3-1. Location of discharge points surrounding Shelton Harbor, Shelton, Washington.



the main building. Two stormwater runoff samples had fecal coliform bacteria concentrations exceeding discharge permit limits (Michaud 1988).

In 2000, Manke's oil/water separator that collects stormwater at the northeast corner of the building reportedly overflowed and discharged oily liquid onto the ground surface, which eventually plugged a neighboring drainfield. Some work was performed on the drainfield and best management practices implemented; Ecology concluded that no further action was required. During a 2007 inspection, Ecology noted floating oil in the final chamber of the oil/water separator. The 2007 inspection notes also referred to three ponds situated in a series that eventually discharge through an outfall structure into the Harbor when enough stormwater is generated at the site (no discharge was occurring at the time).

3.1.3. Historical Operations – ITT Rayonier Research Laboratory

The former ITT Rayonier Research Laboratory occupied approximately three acres of land located to the west of and adjacent to the former pulp mill at 409 East Harvard Avenue from 1936 to 1995 (Figure 3-1). The laboratory specialized in cellulose chemistry and silvichemicals produced from pulp and other wood pulp by-products. The laboratory initially tested samples of pulp to be spun into rayon for product quality control. Between the late 1930s and the mid-1940s, the laboratory tested pulp for manufacturing other specialty wood products, including cellophane film, cellulose acetate (a plastic used in fabrics or solid castings), nitrocellulose used for explosives, and rayon tire cords. Currently, the former laboratory building and property are used by the school district and other private entities (Mason County 2008).

Laboratory waste reportedly consisted of cellulose in caustic slurry and other unknown chemicals; no information was available regarding specific chemicals used. Waste generated in the late 1930s and 1940s was reportedly disposed of at Goose Lake, along with waste sulfite liquor from the pulp mill. No information was available regarding handling and disposal of laboratory wastes after the mid-1940s, when industrial wastewater from both the mill and the laboratory could no longer be discharged to lake.

In March 1987, the ITT Rayonier laboratory director compiled a list of chemicals and estimated the loss of these chemicals during normal laboratory operations as part of their National Pollutant Discharge Elimination System (NPDES) effluent discharge permit requirements. A letter sent by the ITT Rayonier laboratory director to Ecology listing chemicals used by the laboratory during normal operations and analytical results of sampled effluent reported on discharge permit forms are provided in Appendix A. An effluent sample was collected from the laboratory outfall pipe located adjacent to the former pulp mill and Manke log sort yard (Figure 3-1). A high fecal coliform concentration (27,000 colony forming units [cfu] per 100 milliliters) was found to exceed the permit discharge limit. Other constituents detected in the effluent that tested within discharge permit limits included: fluoride, phosphorus, sulfate, aluminum, barium, boron, iron, magnesium, molybdenum, manganese, copper, zinc, total phenols, methylene chloride, and dinoctylphthalate. Further sampling of laboratory effluent for fecal coliform conducted in December 1987 and March 1988 found that concentrations significantly decreased due to bypassing an equalization tank in the discharge line (Michaud 1988).

According to a 1986 underground storage tank (UST) notification form, a single-walled 1,000-gallon diesel fuel tank was installed on the site sometime between 1966 and 1970. The tank reportedly was last used in 1986 and was closed in-place, filled with an inert material. No information was available regarding the location of the tank.

3.1.4. Previous Investigations

3.1.4.1. Reconnaissance Survey of Inner Shelton Harbor Sediments (Ecology 2000)

The objectives of this screening survey of Shelton Harbor sediment focused on evaluating metals and organic contaminant levels near 10 potential sources and estimating the distribution of wood waste in subtidal sediments. Three sediment samples were collected adjacent to the former pulp mill, existing Manke log sorting yard, and a stormwater drainage outfall associated with the former ITT Rayonier laboratory (Figure 3-2). Wood waste samples also collected in the area are discussed in the next section that addresses the Simpson Timber Company. Sediment samples were collected within the intertidal zone at the east end of the south shore sort yard (SCS-1), below the Manke machine shed (SCS-2), and below the south shore marine lab drainage outfall (SCS-3). Benzoic acid exceeded the cleanup screening level (CSL) criterion below the Manke machine shed; no other exceedances were found. Phenol was found at SCS-1 at a concentration above the sediment quality standard (SQS) criterion. Both benzoic acid and phenol are likely associated with decay of wood debris found in the area.

3.1.5. Contaminants of Concern

Based on the results of the above activities, the following contaminants of concern have been identified for the marine environment at and near the former Rayonier Pulp Mill / Manke Lumber Company / ITT Rayonier Research Laboratory:

- Heavy metals from the laboratory
- SVOCs (PAHs, phenols, phthalates) from the laboratory, produced from machinery and trucks associated with lumber storage, and also released during spills/leaks of petroleum-based substances such as hydraulic fluid and fuel
- Dioxins/furans associated with chlorine bleaching operations and the use of PCP
- Benzoic acid, phenol, sulfides, ammonia and TOC produced from wood waste degradation.

3.1.6. Potential Contaminant Pathways to the Aquatic Environment

Historic nearshore outfalls and overland flow provide the major pathways for contaminants to move from upland source areas. Wood waste has been deposited based on direct input from logs

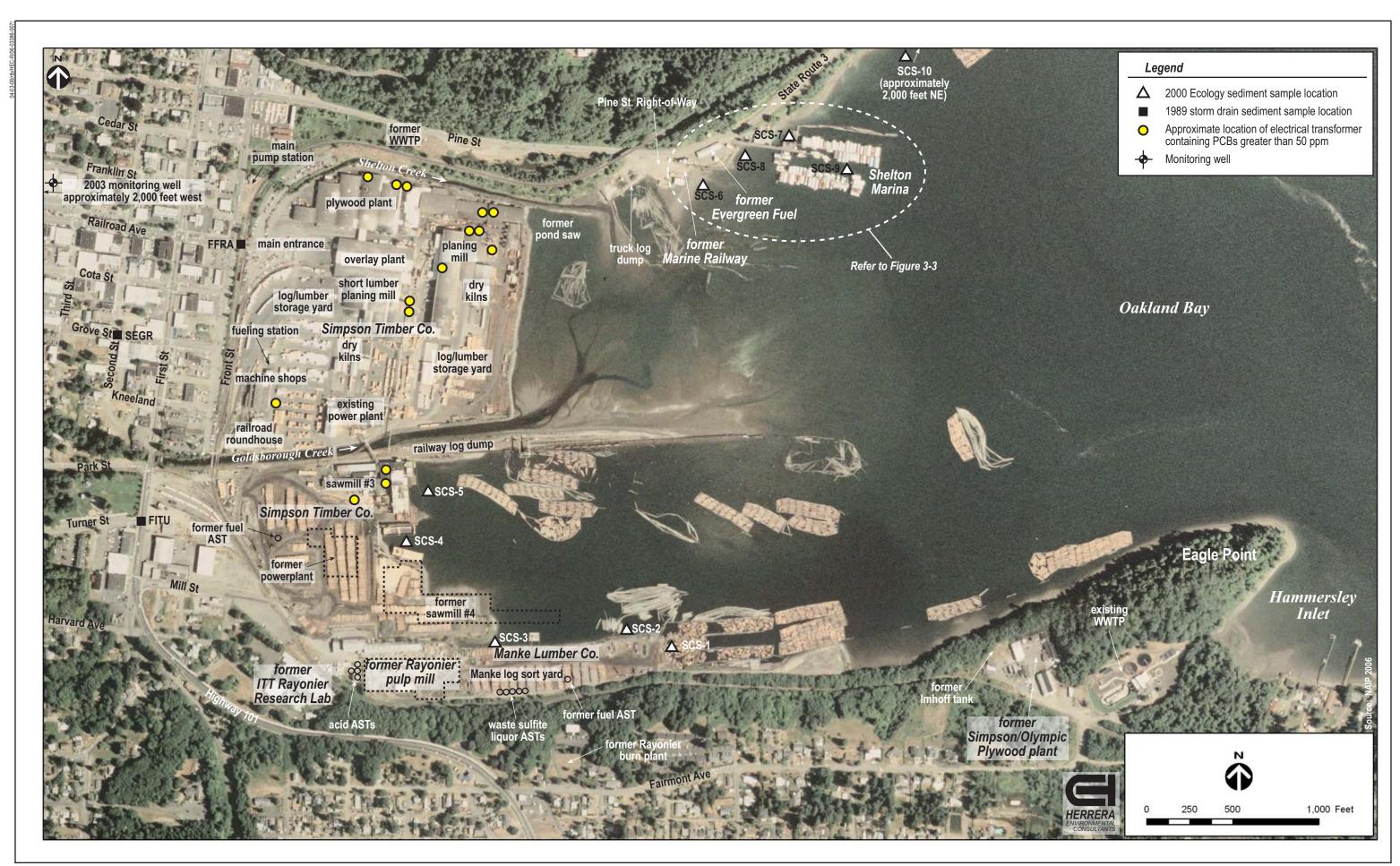
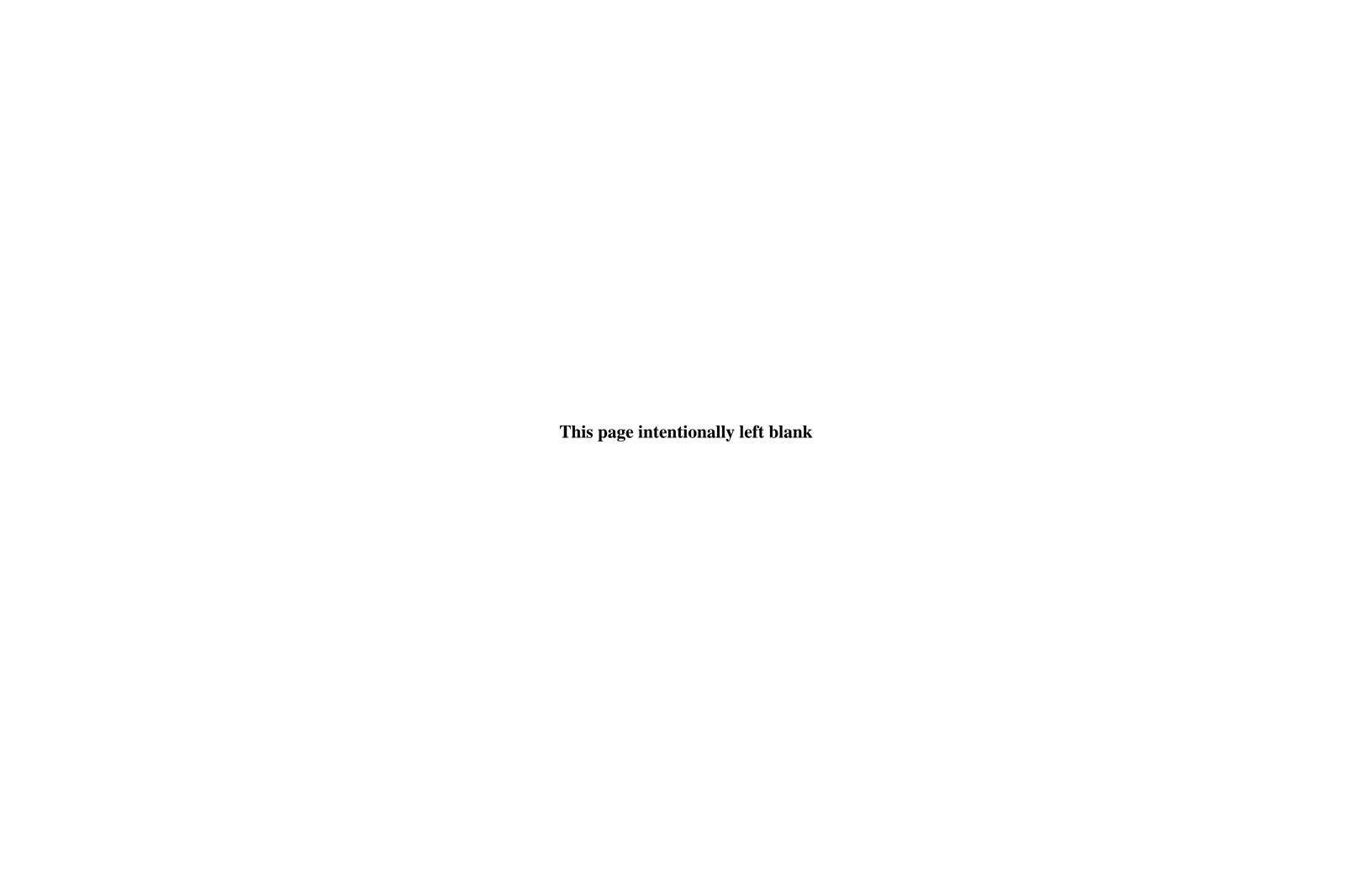


Figure 3-2. Previous investigation sample locations and transformer locations in Shelton Harbor, Shelton, Washington.



floating on the water surface, as well as from debris and sawdust associated with nearshore activities. Stack emissions from boiler and incineration activities also can be expected to contribute contaminants to the bay.

3.1.7. Data Gaps

Three sediment samples are known to have been collected along the south shore of Shelton Harbor. SMS exceedances for benzoic acid and phenol were identified at two of the three locations. Further sampling should be conducted at identified discharge points (discussed later in this report). Analyses should include the broad range of COPCs discussed for marine sediments in general, including bioassays to evaluate the effects of conventionals and contaminants without SMS chemical criteria.

3.2. Simpson Timber Company

3.2.1. Historic and Current Operations

The Simpson Timber Company occupied two properties in Shelton: the main waterfront facility and the Eagle Point plywood facility (Figure 2-2). The main Simpson sawmills and plywood manufacturing plant covers over 80 acres of the Shelton waterfront (Mason County 2008); the address to its main entrance is 421 South Front Street. The plant currently operates a sawmill (Sawmill #3), a particle board overlay plant, planing mills, an existing power plant with a steam generating boiler plant, a log yard, and rail and barge loading facilities (ORCAA 2005). Forest products currently produced by Simpson include dimensional green and kiln-dried lumber, wood residuals, and wood chips. A former Insulating Board plant, located where the current overlay plant is situated, manufactured fiberboard from the mid-1940s until 1974. In 1963, the plant began manufacturing water and fire resistant products. Simpson formerly operated the plywood mill at the north end of the waterfront facility, which has been owned and operated by Olympic Panel Products since 2003.

The Eagle Point facility housed the former Olympic Plywood plant at 1698-1700 Fairmount Avenue. Plywood was manufactured at this location between 1941 and 1974, when operations moved to the north end of the Simpson waterfront complex (Figure 2-2). It is unknown what activities took place on this property from the mid- to late-1970s; Simpson leased the property to Shelton Structures, a glue laminate beam fabrication plant from the late 1970s to the early 1990s. Currently, the Port of Shelton owns the property.

In the early 1920s, pilings were driven across the waterfront south of Goldsborough Creek to create a bulkhead and dredged sediment from the harbor was used as backfill to create approximately 30 acres of industrial land for two sawmills (existing Sawmill #3 and former Sawmill #4). The railroad log dump was constructed at this time, along with over 4,000 pilings driven in the harbor, many of which to provide moorings for log rafts. In the mid-1940s, Simpson developed the central tideflats across the waterfront north of Goldsborough Creek and built the Insulating Board plant, the plywood plant, the railroad roundhouse, and the machine

shops. From the 1950s to the early 1960s, Simpson expanded their waterfront plant further east on the north side of Goldsborough Creek to its present day bulkhead, and built the planing mill, the short lumber planing mill, and dry kilns.

3.2.2. Simpson Waterfront Facility

3.2.2.1. Former Hog Fuel Boilers and Power Plant

In 1926, a power plant was built between the sawmills south of Goldsborough Creek burning wood waste (hog fuel) to supply power for both sawmills and the former Rayonier pulp and paper mill (Figure 3-1).

The wood-fired power plant, located southwest of Sawmill #3, operated from 1926 until it was decommissioned and replaced with a new, more efficient salt-free wood burning power plant in 1986 (Figure 3-1; shown north of Goldsborough Creek as the existing power plant). Bark removed from logs and rejected wood chips and debris generated from the sawmills were burned in the hog fuel boilers; when necessary, fuel oil was used to supplement fuel demands. Air emissions were regularly discharged from the mill boilers as part of normal operations.

The old power plant contained seven boilers and two emission stacks, all of which operated without air emission controls between 1926 and 1976. In 1976, Simpson installed two banks of baghouses, including one between a high-pressure boiler and a short emission stack (125-foot high) and the other between a bank of six low-pressure boilers and a tall emission stack (244-foot high). Approximately 3,000 pounds per day of air emission residue from both stacks was produced when all boilers were operating. Residue consisted of 1/3 unburned and charred wood and 2/3 salts. From July to September 1976, September to December 1978, and July 1981 to February 1982, the baghouses were bypassed. A fire disabled the tall stack in September 1984 and the boiler to the tall stack was shut down in 1986; the short stack was shut down in 1986.

Residues from both baghouses were mixed into slurries and discharged to both the former WWTP on Pine Street (1976 to 1979) and to the existing plant at Eagle Point (1979 to 1984). Solids that had settled out at each WWTP were disposed of at two public landfills (the Shelton "C Street" Landfill and the Mason County Landfill) and at the Dayton wood waste landfill owned by Simpson. After 1984, all residues generated at the Simpson plant were disposed of at their Dayton wood waste landfill.

In 1986, Simpson began using the salt-free wood burning power plant (Thomas 1996). According to the 1998 state dioxin source assessment, hog fuel used at the Simpson plant consisted of 20 percent salt-laden hog fuel, with tire chips fuel used as an alternative fuel source (Ecology 1998).

3.2.2.2. Other Features and Operations

A septic system and drainfield, located at the extreme east end of the mill (location unknown), was no longer functioning in 1969. Simpson used a new sewer pipeline installed to discharge

sewage to the city WWTP. In 1993, Simpson obtained a Stormwater Baseline General Permit for its private storm-sewer system present onsite. No information was available as to whether this system is still in operation.

Non-contact cooling water associated with Simpson's power plant was discharged to the harbor, according to the 1974 - 1976 NPDES waste discharge permits issued for the power plant. Water from the blow pits was mixed with baghouse residues and discharged to the city WWTP (see the Hog Fuel Boilers and Power Plant section above). Handling of process wastewater that consisted of phenolic resin and wood chips was reportedly boiled down before going to the boiler plant to be burned in the early 1990s.

Four USTs associated with a fueling station located near the machine shops were removed in 1991, according to an Ecology UST site check form (Figure 3-1). Two tanks previously contained gasoline and the other two diesel fuel. It was reported on the UST site check form that no petroleum contamination was present during tank removal. No additional information was available regarding these USTs or site conditions.

A total of 29 electrical transformers were identified during a 1988 TSCA inspection conducted by USEPA. Of the 29 transformers, 13 contained PCBs at concentrations greater than 500 parts per million (ppm) and two contained concentrations between 50 and 499 ppm. The locations of the 15 transformers with PCB contents greater than 50 ppm are shown in Figure 3-2.

In 1991, Bunker C fuel was discovered in soil and groundwater during Simpson work on a drainage trench located between the Sawmill #3 building and Goldsborough Creek. The source of contamination was attributed to leaks and spills associated with two 13,000-gallon ASTs present from the 1920s to the mid-1940s. Simpson removed approximately 10 cubic yards of visibly contaminated soil from the source area; however, soil contamination was left in-place in proximity to a bulkhead along Goldsborough Creek, railroad tracks, and a metal frame tower. In 2001, Simpson reported to Ecology that no petroleum odors or contaminated soil was present during installation of a sheet pile bulkhead in the vicinity of the previous contamination. In 2007, Ecology concluded that further site characterization and removal of residual soil contamination was warranted.

Ecology visited the Olympic Panel Products site (occupied the plywood plant) for the Industrial stormwater NPDES general permit compliance inspection in February 2007. Two outfalls are being monitored, including the log yard and OF #11 (locations of these outfalls are unknown). Results of stormwater discharge samples collected from both outfalls in September 2005 and January 2007 indicated that BOD, turbidity, zinc, and copper exceeded their permitted discharge limits.

3.2.2.3. Process Chemicals and Reported Spills

Process chemicals used at the Simpson waterfront plywood and sawmill facilities, as reported to Ecology since the early 1980s, have included phenolic and urethane resins, urea/formaldehyde glues, fungicides, and waterproofing wax. Tanks containing resin and caustic chemicals

associated with the plywood plant were located at the northwest corner of the building near Shelton Creek.

A wood preservative dip tank (location unknown) was referred to in a 1981 Ecology file letter stating that approximately 9,400 gallons of dilute Permatox 200 wood preservative was removed and disposed of by spraying it across the Dayton dry log sort yard. According to the material safety data sheet (MSDS), the preservative contained chlorinated phenols and PCP.

Between 1980 and 2004, numerous spill incidents reportedly occurred, including accidental spills and discharges into the harbor and to Shelton and Goldsborough Creeks, as well as across upland portions of the facility. Most of the reported spills were petroleum products, including hydraulic oil, soluble or biodegradable lube oil, gear oil, and diesel. Other chemical spills included the following:

- Resin (1987 incident) and veneer waste water (1988 incident) were reportedly spilled and discharged to Shelton Creek from the plywood plant.
- In 1984, a PCB-contaminated oil spill to soil occurred within the electrical switchyard located west of and adjacent to the railroad roundhouse (Figure 3-2).
- In 1990, Simpson discovered that waste oil stored in two ASTs located within a diked area adjacent to the northwest corner of the plywood plant contained high concentrations of PCBs. The ASTs were located approximately 60 feet southeast of Shelton Creek (Figure 3-2). Ecology inspected the site and noted PCB-contaminated oil had pooled within the diked area surrounding the ASTs. The amount of oil spilled within the containment was not reported. Cleanup was conducted, including removing and disposing of PCB-contaminated oil from both tanks, as well as removing the concrete sump, asphalt floor, and plywood walls above the dike.

3.2.3. Eagle Point Plywood Facility

No information was available regarding process chemicals used or spills that may have occurred at the former Simpson/Olympic Plywood plant at Eagle Point, which operated between 1941 and 1974. Process chemicals used by Shelton Structures, who operated at the former plywood plant from the late 1970s through the early 1990s, included urea/formaldehyde glue and phenolformaldehyde resin.

In 1980, Shelton Structures was fined by Ecology for illegally discharging phenolic wastewater to the harbor. Before 1985, Shelton Structures disposed of wash water contaminated with phenolic glues and caustic compounds generated from daily cleanup of equipment by

discharging it to a floor drain into a series of three tanks for settling solids. Untreated wash water was then drained to a storm drain that discharges to the harbor. After 1985, the facility recycled washwater in their manufacturing process and no longer discharged it to the harbor.

In 1991, Ecology responded to a complaint filed against Shelton Structures regarding possible soil contamination and runoff from water used to control burn piles containing beams with phenolic resins and formaldehyde. It was reported that no contamination was detected in soil samples collected from the burn piles.

In 1994, an abandoned 750-gallon gasoline UST was removed, along with 70 cubic yards of contaminated soil, from around the tank. The location of the tank is unknown.

3.2.4. Previous Investigations

3.2.4.1. 1984-1988 USEPA National Dioxin Study

Between 1984 and 1988, the USEPA conducted a National Dioxin Study to determine the extent of dioxin contamination in the United States and the associated risk to humans and the environment (CH2M Hill 1987). Simpson volunteered to have their wood-fired power plant tested as part of this study. The Simpson power plant was the only wood-fired boiler plant operating at the time of this study. Other wood-fired power plants and associated stacks that previously operated along the south shore of the harbor included the former Rayonier pulp and paper mill; the former Rayonier burn plant above the pulp mill, and the former Simpson/Olympic Plywood plant (Figure 3-2).

In 1986, samples were collected from Simpson facilities for dioxin and furan analyses, including the two baghouses and two boiler systems from the former power plant, and a multicyclone unit at the existing power plant. Soil samples from residential neighborhoods to the south (Hillcrest), to the north (Capitol Hill), and east across the bay (Shorecrest) were collected as likely areas of deposition from stack discharges. Composites of five discrete sediment samples were collected in the vicinity of the outfalls of both WWTPs. Soils from the three landfills accepting WWTP sludge and baghouse residues also were collected as part of the study. Maps showing the sample locations as described above, and tables summarizing the reported concentrations of dioxins and furans detected in each sample are provided in Appendix B.

Results of the dioxin study indicated:

- Baghouse residue contained up to 4,200 parts per trillion (ppt) of 2,3,7,8-tetrachlorodibenzo p-dioxin (TCDD).
- Low concentrations of dioxins/furans were detected in the multicyclone of the existing power plant. None of the more toxic isomers of chlorinated dioxins/furans were detected. The detected dioxin concentrations were similar to amounts reported for woodstove and fireplace ash.

31

- Dioxins and furans were either not detected or found at less than 1,000 ppt in soil collected from the three residential areas.
- No chlorinated dioxins/furans were detected in sediment samples collected from either WWTP outfall.
- A concentration of 170 ppt 2,3,7,8-TCDD was found, along with other dioxins and furans compounds, at the Shelton "C Street" Landfill (the landfill received sludge from the former WWTP containing residues generated from 1976 through 1979).

3.2.4.2. Reconnaissance Survey of Inner Shelton Harbor Sediments (Ecology 2000)

The objectives of this screening survey of Shelton Harbor sediment focused on evaluating metals and organic contaminant levels near 10 potential sources and estimating the distribution of wood waste in subtidal sediments. Two samples addressing potential contaminant source inputs to the harbor were collected adjacent to Simpson operations south of Goldsborough Creek (SCS-4 and SCS-5; Figure 3-2). CSL exceedances for benzoic acid, phenol, 4-methylphenol, and 2,4-dimethylphenol were identified, all of which can be associated with wood waste. Both samples exhibited mercury levels exceeding the SQS criterion; PCP exceeded the SQS criterion at SQS-5.

Since Simpson managed log storage across the harbor on a grand scale, all wood waste sample results are discussed here. The harbor was divided into nine areas (strata) from which 37 composite surface sediment samples were collected and evaluated for the presence of wood waste. Strata were defined based on log handling practices (e.g., rafting, storing, dumping) and water depth. Results indicated that mean wood content on a weight basis ranged from 5.1 to 18.5 percent. Wood waste was most predominant across the southern half of the harbor, with the highest concentrations near the Simpson and Manke property shorelines.

Chemicals associated with wood waste that exceeded SMS were found in three samples representing the south harbor and one sample near the marina, including benzoic acid, phenol, and 4-methylphenol.

3.2.5. Contaminants of Concern

Based on the results of the above activities, the following contaminants of concern have been identified for the marine environment at and near the Simpson waterfront facility:

- Heavy metals from the non-contact cooling water and as noted in recent NPDES monitoring; mercury, as found in both sediment samples exceeding SQS
- SVOCs (PAHs, phenols, phthalates) from plywood and laminate production, produced from machinery and trucks associated with lumber

storage, and also released during spills/leaks of petroleum-based substances such as hydraulic fluid and fuel

- PCBs associated with transformers located across the site
- PCP associated with wood preservation
- Dioxins/furans associated with chlorine bleaching operations and the use of PCP
- Benzoic acid, phenol, sulfides, ammonia and TOC produced from wood waste degradation.

Based on the results of the identified activities, the following contaminants of concern have been identified for the marine environment at and near the Simpson Eagle Point facility:

 SVOCs (PAHs, phenols, phthalates) from plywood and laminate production.

3.2.6. Potential Contaminant Pathways to the Aquatic Environment

Historic nearshore outfalls and overland flow provide the major pathways for contaminants to move from upland source areas. Both Goldsborough and Shelton Creeks have offered additional opportunities for wide-spread contamination across the harbor, due to their relatively high rate of flow. Wood waste has been deposited based on direct input from logs floating on the water surface, as well as from debris and sawdust associated with nearshore activities. Burning of hog fuel and supplemental fuel materials has resulted in wide-spread distribution of air contaminants with sources at both the waterfront facility and the burn plant up on the hill to the south.

3.2.7. Data Gaps

Two sediment samples are known to have been collected along the east shore of Shelton Harbor adjacent to the Simpson facility. SMS exceedances for benzoic acid, phenol, 4-methylphenol, 2,4-dimethylphenol, mercury and PCP were identified. Further sampling should be conducted at identified discharge points (discussed later in this report). Analyses should include the broad range of COPCs discussed for marine sediments in general.

3.3. Evergreen Fuel Company / Former Marine Railway / Shelton Marina

3.3.1. Evergreen Fuel Company

The former Evergreen Fuel Company petroleum bulk plant historically operated at the northeast corner of Shelton Harbor at 661 East Pine Street from 1913 until it was shut down in 2005

(Figure 2-2). The bulk plant consisted of an upland area and a tidelands/subtidal area. The upland area consisted of a small office building and warehouse surrounded by a gravel lot, nine ASTs, an overhead tanker truck fueling station, an oil/water separator, an over-water pipeline that ran from the bulk fuel plant to the dock, and a barge off-loading area. The upland area was used for storage, distribution, and sales of gasoline, diesel, stove/heating oil, kerosene, and other miscellaneous petroleum products. The north tank farm, located north across State Route (SR) 3 from the main facility, had four ASTs; the west tank farm located west of the main facility and above the shoreline had five ASTs. A wooden bulkhead separated the upland from the tidelands. The tidelands area consisted of eastward-sloping intertidal and marine mud and gravel, with deeper waters toward the east edge of the property. No information was found regarding use of the tidelands area by Evergreen Fuel Company.

Adjacent properties include the City Pine Street right-of-way, a defunct dock and tidal grid (accessed only from Evergreen Fuel property), and a former marine railway owned by Simpson to the south; SR 3 roadway and forested hillslope and railroad right-of-way owned by Burlington Northern Santa Fe to the west; the Shelton Yacht Club and Marina, owned by the Port of Shelton to the north; and aquatic tidelands owned by the Washington State DNR to the east.

Between 1973 and 2003, several petroleum spills to soil, groundwater and surface water into the harbor were reported. In 2003, approximately 400 gallons of fuel spilled into an AST containment area, releasing between 50 and 70 gallons into the harbor. An initial cleanup was conducted that consisted of removing approximately 150 tons of petroleum-contaminated upland soil from the tank site.

Between 2005 and 2007, a remedial investigation (RI) and remedial cleanup action were conducted pursuant to an Agreed Order. Cleanup activities conducted from December 2006 to June 2007 included:

- Removal of approximately 7,500 tons of soil contaminated with gasoline;
 diesel; benzene, toluene, ethylbenzene, and xylenes (BTEX); and cPAHs
- Groundwater treatment through enhanced aerobic bioremediation
- Groundwater monitoring.

Total petroleum hydrocarbon (TPH) contamination in soil was left in-place beneath the SR 3 roadway, beneath utilities located adjacent to SR 3, and near the bulkhead retaining walls. Residual TPH contamination in groundwater is still present, but TPH concentrations have decreased since cleanup activities were completed and compliance groundwater monitoring began in April 2007, based on results from the October 2007 monitoring event.

3.3.2. Former Marine Railway Facility

Southwest of and adjacent to the Evergreen Fuel site is the City Pine Street right-of-way. Features of the right-of-way include a short abandoned dock, a tidal boat grid, and an abandoned

sanitary sewer outflow pipe associated with the former Shelton WWTP located on Pine Street (this outflow pipe is discussed further below in Section 3.7.2 - Previous Wastewater Treatment Plant).

Located further southwest beyond the Pine Street right-of-way is a former marine railway property located at 655 East Pine Street, which is owned by the Simpson Timber Company. The Shelton Yacht Club operates the facility to haul boats out of the water for maintenance and repair activities. The former Simpson marine railway facility includes a vacant single-story wood-framed building supported by pilings, a small shed, and a boat cradle with two sets of launching rails that extend east from the building into the harbor. Simpson previously used the facility for sandblasting and cleaning log-boom boat bottoms (Ecology 2006). ASTs previously occupied an area west of the marine railway facility. No information was available regarding the number, capacity, or type of product stored in these tanks, or whether a site assessment was conducted to determine if a release had occurred.

A complaint was filed with Ecology in 1987 regarding Simpson personnel sandblasting their boats over water near the boat launching area. From the 1990s through 2005, a small log towing company (identified as Smith & Sons Log Boom Towing) operated at this facility, lifting logs from trucks or the platform using an electric hoist and lowering them into the bay.

3.3.3. Shelton Yacht Club and Marina

The Shelton Yacht Club and Marina are located east-northeast of and adjacent to the former Evergreen Fuel site at 659 East Pine Street (Figure 2-2). The marina is owned and operated by the Port of Shelton; the yacht club leases portions of the property from the Port. The marina, developed in 1974, includes docks and boathouses that extend over tidelands owned by CC Cole & Sons (Evergreen Fuel) and the DNR. Features at the marina include a former timber tidal boat grid located south of and adjacent to the main dock, timber docks with covered and uncovered slips for moorage of marine vessels, a single-story wood-framed building supported by pilings, and a gravel parking area. According to a NPDES boat yard permit issued for the marina in December 2005, no industrial vessels are moored at the marina. Services provided at the marina include vessel haul-out, vessel repair, pressure washing, and paint removal and application. Currently, all wastewater generated during pressure washing is collected and recycled back to a concrete wash pad. Storm water is discharged to the bay.

A complaint filed with Ecology in 1993 reported discharge of gray water and sewage waste from boats occupied by live-aboards at the marina. It was reported that the marina did not have any septic system for boaters at the time. No information was available in the site file as to whether a septic system or other means of disposal of sewage has been implemented at the marina.

Discharge monitoring is required for the NPDES boat yard permit. The results of a sample collected at the haul-out discharge point in January 2007 had a copper concentration (250 micrograms per liter [μ g/L]) that exceeded the permit limit of 229 μ g/L. Copper concentrations in samples collected since January 2007 were below the permit limit. No boats

reportedly had been hauled out or cleaned in the vicinity of the vessel haul-out area since October 2006.

Historically, the Port formerly owned and operated two USTs for fueling marine vessels at the docks. The tanks were located at the north side of the marina building. The total reported capacity of both tanks was 48 gallons, with one tank containing gasoline and the other diesel fuel. No site file information was available as to whether a site assessment was conducted to determine if a release had occurred from either tank.

3.3.4. Previous Investigations

3.3.4.1. Reconnaissance Survey of Inner Shelton Harbor Sediments (Ecology 2000)

Four surface sediment samples were collected in the vicinity of the Evergreen Fuel/former marine railway/Shelton Marine area (Figure 3-2). Wood waste samples also collected in the area are discussed in the previous section addressing the Simpson Timber Company. Sediment samples were collected from within the intertidal zone near the base of the launching rails at the vessel haul-out area (SCS-6), within the intertidal zone adjacent to a boat crib north of the marina (SCS-7), within the intertidal zone east of the former Evergreen Fuel site (SCS-8), and within the marina, approximately 600 feet east from the shoreline, and 14 feet mean lower low water (MLLW) (SCS-9).

Results of these samples indicated copper exceeding the CSL criterion near the vessel haul-out area (SCS-6) and adjacent to the boat crib (SCS-7) and benzoic acid exceeding the CSL criterion east of Evergreen Fuel (SCS-8). Other SQS exceedances were identified for bis(2-ethylhexyl)phthalate, fluoranthene, chrysene, PCP, and high molecular weight polycyclic aromatic hydrocarbons (HPAH).

Butyltin concentrations were generally low in sediment samples, except for SCS-6 collected near the base of the launching rails at the vessel haul-out area. Individual concentrations of mono- $(1,300 \text{ micrograms per kilogram } [\mu g/kg])$, di- $(4,100 \mu g/kg)$, and tri- $(1,500 \mu g/kg)$ butyltin chloride all exceeded $1,000 \mu g/kg$.

3.3.4.2. Remedial Investigation of the Evergreen Fuel Site (Farallon 2005, 2007)

Sampling was performed in both upland and sediment areas (sampling location map is provided in Figure 3-3). Ten surface sediment samples were collected for chemical analysis from the following locations:

- EFS-01 Within the intertidal zone near the base of the launching rails of the former Simpson marine railway facility
- EFS-02 East of the Pine Street right-of-way, within the intertidal zone near a former tidal boat grid and vessel haul-out area

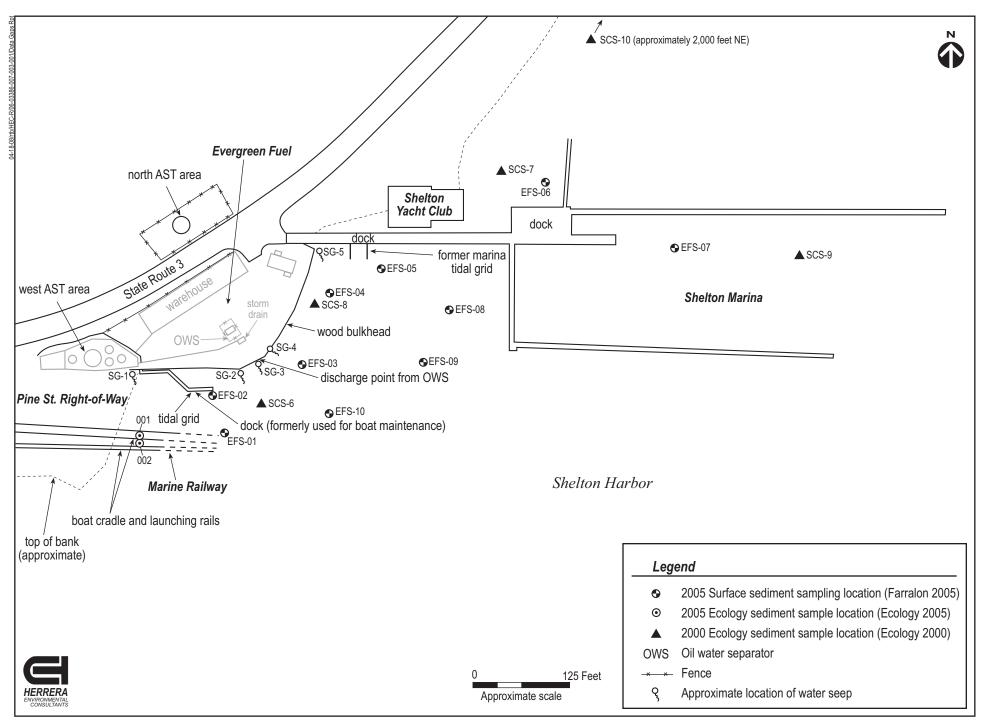


Figure 3-3. Site map of former Evergreen Fuel site, Shelton Yacht Club and Marina, and former marine railway, Shelton, Washington.

- EFS-03 Within the intertidal zone near an oil/water separator discharge pipe.
- EFS-04 Within the intertidal zone east of the Evergreen Fuel office and warehouse
- EFS-05 Within the intertidal zone in the vicinity of the marina's remnant tidal boat grid
- EFS-06 Within the intertidal zone north of the marina dock
- EFS-07 In the subtidal zone within the marina to characterize former fuel loading
- EFS-08 and EFS-09 Approximately 200 feet east and offshore the Evergreen Fuel site within the marina
- EFS-10 Approximately 130 feet east and offshore from the Pine Street right-of-way, near Ecology (2000) sediment sample location SCS-6.

Results of these sediment samples indicated the following:

- A phenanthrene concentration that exceeded the SQS criterion was found at EFS-02SS collected within the Pine Street right-of-way.
- Concentrations of diesel- and/or lube oil-range hydrocarbons were detected in nine of the 10 sediment samples collected (no hydrocarbons were detected in EFS-09). Further evaluation of chromatograms from each of the nine samples indicated that TPH concentrations consisted entirely of PAHs that elute as TPH in the analyses (Farallon 2005). The PAH concentrations in all nine sediment samples (with the exception of phenanthrene in EFS-02) were below SMS criteria; the PAHs in these samples were interpreted by Ecology as likely associated with residual creosote from nearby treated timbers of the bulkheads and docks.
- No PCBs were detected above practical quantitation limits in either soil or groundwater samples collected during the RI from the upland portion of the site. However, low concentrations of total PCBs were detected at all ten sediment samples ranging from 0.84 to 4.77 mg/kg organic carbon normalized (OC); all concentrations were below SMS criteria.
- TBT in interstitial water was detected in two of the sediment samples: 0.888 μg/L at EFS-02 and 0.0326 μg/L at EFS-03. The PSDDA interstitial porewater criterion of 0.15 μg/L was used as a screening level

- of concern for the site. No TBT was detected above practical quantitation limits in the remaining eight sediment samples.
- Two samples (EFS-08 and EFS-09) exceeded the SQS criterion for silver because the laboratory reporting limit was slightly above the criterion. However, silver was not detected at the method detection limit of 0.114 mg/kg, well below the SQS criterion of 6.1 mg/kg. All ten samples contained detected concentrations of arsenic, chromium, copper, lead, mercury, and zinc below their respective SMS criteria. Low concentrations of cadmium below the SMS criteria were detected in nine of the ten samples (no cadmium in EFS-01).

In addition to sediment samples, water samples at five seep locations in the intertidal area adjacent to the Evergreen Fuel site were sampled (Figure 3-3; SG-1 through SG-5). Based on results for specific conductance and other geochemical results, it appears that seep SG-1 is groundwater discharging from the site, while the other two seeps tested (SG-2 and SG-4) appear to be salt water that recharges into the site during high tides and discharges back to the bay as the tide goes out (Mercuri 2005). No COPCs associated with the Evergreen Fuel site investigation were found at concentrations above regulatory cleanup criteria in samples SG-3 through SG-5. Results for samples SG-1 and SG-2 indicated the following:

- Reddish-brown stained soil was observed above the mean higher high water line near the discharge point for seep location SG-1, located southeast and adjacent to the west AST area. Ecology expressed concern that the stained soil may be associated with petroleum product. Since cleanup of the site in 2007, no seeps have been identified.
- The seep sample collected at SG-2 near the bulkhead indicated the presence of PAHs that likely can be attributed to contact with creosote-treated pilings.

Ecology determined that photo-activated PAH compounds were present in sediment at three locations at concentrations warranting bioassay testing. In January 2005, Ecology conducted bioassay tests on sediment sample EFS-03 (split sample) using both the standard method (guidelines for conducted lab bioassays on Puget Sound sediments – Puget Sound Estuary Program [PSEP] 1995) and the ultraviolet (UV) exposure method (Ecology 2008a). The guidance for the UV exposure method cites several studies that suggest that photo-activation of certain PAH compounds can result in increased toxicity to organisms. Results indicated no substantial differences between the standard method and the UV exposure method in the 10-day and 20-day tests. In the 48-hour test, the mean individual growth rate for mussels using the UV method was substantially lower than that of the standard method. No further documentation was found in the site file.

3.3.4.3. 2005 Ecology Initial Investigation of the Simpson Marine Railway (Ecology 2005)

Ecology conducted further investigation of the former Simpson marine railway to follow-up on TBT found in sediment in the Farallon RI. Two surface sediment samples were collected and analyzed for the presence of TBT, metals, total organic carbon (TOC), SVOCs, and diesel- and lube oil-range hydrocarbons. Both samples contained mono- (180 and 580 μ g/kg), di- (510 and 1,800 μ g/kg), and tri- (1,100 and 2,400 μ g/kg) butyltin chloride. To compare tributyltin chloride results to the PSDDA screening level, multiply the chloride results by 0.89 (Michelsen et al. 1996); this results in 979 and 2,136 μ g/kg as TBT for the two samples, both exceeding the PSDDA screening level of 73 μ g/kg. No information was available concerning other analytes of concern.

3.3.5. Contaminants of Concern

The following are potential chemicals of concern to the Harbor:

- PCBs, SVOCs, TBT, and heavy metals associated with industrial stormwater and boat cleaning/repair processes. TBT specifically found in boat hull paint.
- TPH/PAHs associated with gasoline and diesel spills, potentially leaking storage tanks, creosote pilings, and boat exhaust.

3.3.6. Potential Contaminant Pathways to the Aquatic Environment

The primary source of TBT in the marine environment is from sandblasting and use of antifouling paints on boats. Sandblasting and cleaning of log-boom boat bottoms occurred directly over water in at least one incident reported in 1987. Anything that was scraped or dropped into the water during boat maintenance and repair activities becomes a potential sediment contaminant source.

Boat exhaust, accidental fuel leaks, and spills from the fueling station and ASTs directly to the water surface are potential pathways for TPH and PAH contamination of the marine environment.

3.3.7. Data Gaps

Upland area activities can generate potential industrial stormwater runoff impacts. Improperly cleaned areas or improperly stored hazardous materials increase the chance of spills/leaks and loading of chemicals into stormwater. Contaminated groundwater discharging to surface water is another potential pathway.

A total of 16 surface sediment samples are known to have been collected from intertidal and subtidal areas adjacent to the former Evergreen Fuel, Shelton Marina, and former Simpson marine railway. TBT concentrations exceeding SMS criteria were detected in samples collected in the vicinity of the marine railway and in one subtidal sample collected within the marina. Other SMS exceedances detected in one or more samples included copper, benzoic acid, bis(2-ethylhexyl)phthalate, PAHs, and PCP. Low concentrations of PCBs below SMS criteria were also detected. Further sampling should be conducted to 1) further assess the extent of TBT in the intertidal area adjacent to the former Simpson marine railway and Evergreen Fuel site; 2) assess the vertical extent of COPCs in sediment in the vicinity of the discharge point for the oil/water separator; and 3) collect a tissue sample to assess effects on biota.

3.4. Bulk Fuel Facilities along West Shore of Oakland Bay

In addition to the Evergreen Fuel bulk fuel plant, three other petroleum bulk storage facilities have historically operated about 1/2 mile northeast of Shelton on SR 3 along the west shore of Oakland Bay (Figure 2-1). These facilities generally consisted of large ASTs containing petroleum products located on the west, uphill side of SR 3. Fuel was dispensed either by overhead tanker truck fueling stations or through underground piping that ran from the tanks to a barge off-loading dock on the shore. Based on Ecology Bulk Oil Handling Facility Inspection Reports compiled between 1973 and 1978, the following facilities were identified:

- Union Oil Company bulk plant (B&R Oil) operated from the early 1930s to the mid-1980s. The facility had five ASTs with a total capacity of 2,274 gallons; petroleum products included gasoline, diesel fuel, and stove/heating oil. Oil pollution hazards identified during inspections included inadequate containment structure surrounding the ASTs and additional catch basins required. No spills were reported for the site.
- Shell Oil Company (Gott Oil) operated from the early 1930s to the mid1980s. The facility had five ASTs with a total capacity of 1,952 gallons;
 petroleum products included gasoline and diesel fuel. Oil pollution
 hazards identified during inspections included valve leakage, no
 containment structures surrounding the tanks, vulnerable storm drains, and
 inadequate catch basins. Approximately 294 gallons of diesel fuel
 reportedly spilled at the site in 1973 (spill location not reported).
 Approximately 50 gallons of lube oil or diesel fuel spilled at the site in
 1978 (spill location not reported).
- ARCO (Evans) operated from the mid-1960s through the late 1970s. This facility consisted of four ASTs with a total capacity of 1,904 gallons; petroleum products stored at this site included gasoline, diesel, and stove/heating oil. Oil pollution hazards identified during inspections

included inadequate containment structures surrounding the tanks and vulnerable storm drains. No spills were reported for this site.

None of these facilities are listed on state or federal regulatory databases as a documented or suspected release site. No site assessments have been conducted to determine whether petroleum releases have occurred at any of the three former bulk fuel storage facilities. Manke Family Resources LLC has owned all three former bulk plant properties since the mid-1980s. Manke constructed a gravel barge loading and staging facility just north of the bulk plant sites for transporting finished aggregate by barge in 1997.

3.4.1. Contaminants of Concern

SVOCs (including TPH, low molecular weight polycyclic aromatic hydrocarbons [LPAH], and HPAH) and heavy metals from crude and refined petroleum products are the primary chemicals of concern associated with petroleum bulk plants and major fuel spills.

3.4.2. Potential Contaminant Pathways to the Aquatic Environment

Spills reported at the Shell Oil facility suggest that soil and groundwater are likely impacted by COPCs, with surface runoff and groundwater as potential routes of entry into Oakland Bay.

3.4.3. Data Gaps

No sampling has been conducted in sediment near any of the bulk fuel storage facilities. Initial screening of surface and subsurface sediment should be conducted to determine the need for potential further assessment. Analyses should include the broad range of COPCs discussed for marine sediments in general.

3.5. Wood-Treated Marine Lumber

Wood pilings and other wooden marine structures have been used in waters of the U.S. for over a hundred years. These structures are usually employed as structural supports, for anchoring log rafts, for vessel mooring stations (i.e., dolphins), and for bulkheads. Marine-use wood has been preserved with chemicals to prevent deterioration from wood-degrading marine organisms (e.g., boring clams and crustaceans). Creosote is the most commonly used chemical wood preservative worldwide (Stratus 2006). PCP is not resistant to marine borers, and therefore, used for pilings in freshwater or in saltwater splash zones (Stratus 2006).

Numerous support pilings for docks and log storage areas, mooring dolphins, and bulkheads have been driven throughout Shelton Harbor since timber industries began operating along the waterfront in the mid-1920s. It is assumed that the majority of marine wood structures within the harbor were treated with creosote. However, there is a possibility that PCP-treated wood

pilings and structures were also used in the harbor, based on low to moderate concentrations of PCP detected in nine of the ten chemical screening stations and in all wood waste strata composites collected within the harbor (Ecology 2000).

Other areas in Oakland Bay where support pilings and other marine structures had been used since the late 1930s (NOAA 2008) include the barge off-loading docks associated with the three former bulk fuel storage facilities, along the west shore of Oakland Bay, two large log-boom areas including one area west of Chapman Cove and the other area west of and adjacent to Munson Point, and four small and separated sets of pilings associated with the oyster reserves within the northern portion of Oakland Bay including three sets north of Bayshore Point and one set near the mouth of Deer Creek.

The condition of pilings and other marine structures throughout Shelton Harbor and Oakland Bay varies. Three large log raft-boom areas (a total of 92.7 acres) that Simpson leased from DNR (two areas of which Rayonier previously controlled) were reportedly cleared of pilings and sunken logs in 1973. Two of the three areas were located west of and adjacent to Munson Point (identified as Maple Beach 1 and 2; 61.1 acres); the location of the third area was not reported, but consisted of 31.6 acres of aquatic land.

3.5.1. Contaminants of Concern

Creosote is a distillate of coal tar and its compounds are derived from several chemical classes, including PAHs, alkyl-PAHs, tar acids/phenolics, O-heterocyclics/furans (e.g., benzofurans), and aromatic amines (e.g., aniline). The types of PAH compounds found in creosote include phenanthrene, naphthalene, acenaphthene, fluorene, 2-methylnapthalene, pyrene, anthracene, and chrysene. PCP is within the class of phenolic compounds.

3.5.2. Potential Contaminant Pathways to the Aquatic Environment

PAHs can leach from wood into both sediment and water. The rate at which PAHs leach from treated wood is a complex function of many factors, including water chemistry, temperature, and salinity, as well as wood type and age. PAHs in the aquatic environment are present in both dissolved form and adsorbed to particulate materials. Low molecular weight compounds dissolve more readily in water than the heavier PAHs. The higher molecular weight PAH compounds, creosote, and PCP tend to accumulate in sediment (Stratus 2006).

3.5.3. Data Gaps

Based on a review of available data, there is a general lack of information on the extent of creosote- and PCP- treated timbers present in Oakland Bay and their possible influence on sediments and biota.

3.6. Sites in Downtown Shelton

A total of 43 sites located within the first three city blocks west of Front Street are identified as historically or currently handling hazardous materials. These sites are located less than a 1/2 mile from the Shelton waterfront. Of the 43 sites identified, 39 are petroleum-related businesses, including gasoline stations, automotive repair, and fuel distribution card-lock stations. Only six of the 39 sites have had documented releases to soil and/or groundwater, four of which have been cleaned up. The four non-petroleum sites include one historic and two current dry cleaning facilities, and a historic sheet metal workshop. Based on their locations, these 43 sites are not considered significant sources of contamination to sediment in Shelton Harbor and Oakland Bay.

3.6.1. Contaminants of Concern

The two types of contaminants released from downtown sites include petroleum products and dry cleaning solvents.

3.6.2. Potential Contaminant Pathways to the Aquatic Environment

Releases to the environment from downtown sources appear generally to be into groundwater. Contaminant plumes identified at a few of the sites are not extensive; however, there is the possibility that affected groundwater may enter storm drain piping that discharges to the harbor, as discussed in the next section. This does not appear to be an important pathway, due to the volatile nature of the contaminants and the relatively low concentrations possible.

3.6.3. Data Gaps

No sampling has been performed to address the possibility of contaminant inputs from downtown sources; however, loading to the harbor is likely insignificant.

3.7. City of Shelton Municipal Sewage and Stormwater Systems

3.7.1. Shelton Wastewater Treatment Plant

Since 1979, the City of Shelton owns and operates a WWTP located on Eagle Point at 1700 East Fairmount Avenue. It discharges secondary treated and disinfected municipal wastewater through a deepwater outfall pipe that extends into Hammersley Inlet (Figure 3-1). A solids digester and other solids handling improvements were made at the facility in 2004. Treatment is accomplished through screening, grit removal, oxidation ditches, final clarification, and disinfection with chlorine (Ecology 2008b).

The existing wastewater collection system extends throughout the majority of the current city limits. The system consists of 4- through 24-inch diameter gravity sewer lines, three small cul-

de-sac pump stations and one main pump station, and a one and one-half mile force main (Ecology 2008b). Most of the existing sewer lines are very old; pipes within the downtown core were installed in the 1910s and 1920s; pipes in the southern part of the city were installed in the 1940s and 1950s; and pipes in the northern part of the city were installed in the 1950s through 1970s (Ecology 2008b). The collection system is constructed of wood stave, vitrified clay, and concrete (Brown & Caldwell 1988).

Based on discharge monitoring reports between 2002 and 2007, discharge of treated effluent exceeded permit required limits for one or more of the following parameters (the latest exceedances):

- Biochemical oxygen demand (BOD) exceeded the limit of 2,200 pounds per day in October 2002; August through December 2006; and February through June, and November 2007 (2,419 pounds per day)
- Total dissolved solids (TDS) exceeded the limit of 810 pounds per day in December 2006, January 2007, and March 2007 (1,083 pounds per day)
- Chlorine exceeded the limit for average (0.27 milligrams per liter [mg/L]) and/or maximum discharge (0.39 mg/L) in September 2006, April 2006, May, June, and October 2007 (maximum discharge of 0.45 mg/L).

The collection system experiences excessive inflow and infiltration (I&I) during periods of heavy rainfall. Most of the I&I is associated with shallow groundwater that leaks into pipe cracks. Other sources of I&I include surface waters from pipes connecting roof or basement drains and flow into manholes (Ecology 2008b). The collection system overflowed several times in recent years, with wastewater surcharging through manholes located on Mill, Park, and First Streets. Fecal coliform results of surcharged wastewater during two separate events in December 2001 were too numerous to count. These overflows generally flow into Goldsborough Creek. The WWTP has also exceeded their design flow limit during periods of rain. Excessive I&I dilutes influent wastewater and reduces the efficiency of treatment processes to remove pollutants (Ecology 2008b).

Because of the excessive I&I, Ecology issued an administrative order (DE 97WQ-S182) to the City in 1998, requiring replacement of the collection system within the downtown area and installing flow monitoring equipment. Portions of the sewer lines within the downtown area had been replaced in 1998; work on other portions of the downtown area have been completed recently or are scheduled to start in 2008.

3.7.2. Previous Wastewater Treatment Plant

Between 1950 and 1979, the City operated a WWTP located on Pine Street; the plant was designed for primary treatment of sewage including grit removal, sedimentation, single-stage digestion, and chlorination (Shelton-Mason County Journal 1950). Disinfected effluent

discharged to an outfall pipe located at the east end of the Pine Street right-of-way on the north side of the harbor (Figure 3-1). Before 1950, sewer pipes conveyed untreated sewage directly into the harbor (Shelton 2006).

3.7.3. Eagle Point Community Wastewater Treatment Facility

A former Imhoff septic tank and drainfield was located near Eagle Point, west of and adjacent to the former Simpson/Olympic Plywood plant at 1698 East Fairmount Avenue. This system was designed to treat residential waste for approximately 500 nearby residences. The septic system consisted of an Imhoff tank adjacent to an abandoned sludge bed, four drainfields, and two water wells. The system was used from 1889 until 1992, when the city dismantled it. Residences previously connected to this system switched either to individual on-site septic systems or connected to the existing WWTP at Eagle Point. No industrial waste discharged into this system.

3.7.4. Sewer Outfalls and Stormwater Discharge Points

Various historical and current discharge points, their descriptions, and type of discharge (if known) identified along the Shelton Harbor waterfront are summarized in Table 3-1 and shown in Figure 3-1.

The City stormwater collection system was installed throughout the downtown area sometime between the late 1800s and early 1900s. According to the City's public work map of the system, untreated stormwater runoff flows through buried pipes and outfalls directly into Goldsborough and Shelton Creeks, and eventually into the harbor.

In 1989, a storm drain sediment study was conducted to establish a stormwater priority pollutant concentration baseline (Ecology 1990). Fine-grained sediment was collected at six locations in representative areas of the city. Samples were analyzed for TOC, percent solids, grain size, SVOCs (USEPA method 8270), pesticides and PCBs (USEPA method 8080), and metals (USEPA method 200 series). Because storm drains were cleaned each summer, data reflected accumulations of less than one year.

The results of this study indicated that three of the six samples collected had contaminant concentrations exceeding SQS criteria. The locations of these samples and contaminants of concern detected are summarized in Table 3-2.

Table 3-1. Summary of industrial, stormwater, and sewer discharge points along the Shelton Harbor waterfront.

Discharge Point Map ID	Description	Source
1	Former city WWTP outfall to Oakland Bay (in operation from 1950 to 1979)	Michaud 1988; Brown & Caldwell 1988
2	Former 1-foot diameter concrete outfall pipe for dryer cleaning effluent containing treatment chemicals to Shelton Creek from the plywood plant	Archives Ecology 1969- 1988 Simpson Plywood 1972 map
3	Stormwater discharge outfall to Shelton Creek from overlay plant	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
4	Stormwater discharge outfall to Shelton Creek from west end of the plywood plant.	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
5	A 1-foot inside diameter concrete pipe for discharging compressor cooling water either from the plywood plant or the planing mill.	Archives Ecology 1969- 1988 Simpson Plywood 1972 map
6	Former 1-foot diameter concrete outfall pipe of screened effluent from a pond saw.	Archives Ecology 1969- 1988 Simpson Plywood 1972 map
7	Stormwater discharge outfall discharge to harbor from east log storage yard.	Archives Ecology undated map, assumed 1987
8	Discharge from the shops, the boiler plant, and the west side of the overlay plant (a clarifier may also be connected).	Michaud 1988; Archives Ecology undated map, assumed 1987
9	Stormwater discharge outfall on north side of Goldsborough Creek from the planing mill and dry kiln area.	Archives Ecology 1969- 1988 Simpson Plywood undated, assumed 1987
10	Stormwater discharge outfall on north side of Goldsborough Creek from boiler plant.	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
11	Stormwater discharge outfall on north side of Goldsborough Creek from west log storage yard and parking lot at main entrance	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
12	Stormwater discharge outfall on north side of Goldsborough Creek from railroad roundhouse.	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
13	Stormwater discharge outfall on south side of Goldsborough Creek from Simpson sawmill log storage yard.	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987

Table 3-1 (continued). Summary of industrial, stormwater, and sewer discharge points along the Shelton Harbor waterfront.

Discharge Point Map ID	Description	Source
14	Stormwater discharge outfall on south side of Goldsborough Creek from west end of Sawmill #3	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
15	Industrial stormwater discharge below Sawmill #3	Michaud 1988; Archives Ecology 1969- 1988 Simpson Plywood undated, assumed 1987
16	Non-contact cooling water discharge plus stormwater discharge below Sawmill #3	Michaud 1988; Archives Ecology 1969- 1988 Simpson Plywood undated, assumed 1987
17	City of Shelton 54-inch stormwater discharge culvert	Michaud 1988; Archives Ecology 1969- 1988 Simpson Plywood undated, assumed 1987
18	Industrial stormwater discharge pipe below former Sawmill #4	Michaud 1988; Archives Ecology 1969- 1988 Simpson Plywood undated, assumed 1987
19	Stormwater discharge outfall to harbor from north side of the former Sawmill #4	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
20	Stormwater discharge outfall to harbor from south side of the former Sawmill #4	Archives Ecology 1969- 1988 Simpson Plywood undated map, assumed 1987
21	Former ITT Rayonier Research Laboratory effluent outfall pipe	Michaud 1988; Archives Ecology 1969- 1988 Simpson Plywood undated, assumed 1987
22	Overland runoff from Manke Lumber Company property (former Rayonier pulp mill)	
23, 24	Two outfalls west adjacent to the former Simpson/Olympic Plywood plant (possibly one outfall from the former Shelton Imhoff Tank treatment plant and the other outfall a stormwater discharge outfall). Unknown which pipe is which.	
25	Outfall connected to the former Simpson/Olympic Plywood plant	Archives Ecology 1969- 1988 Simpson Plywood, undated map assumed 1987
26	Existing City of Shelton WWTP (in operation since 1979)	Ecology 2008b

Refer to Figure 3-1 for the locations of these discharge points.

Table 3-2. Results of 1989 storm drain sediment study in the City of Shelton.

Sample Station ID	Location	Contaminants of Concern Exceeding SQS Criteria		SQS Criteria
SEGR	Intersection of Second	Chromium	185 and 284 mg/kg	260mg/kg
	and Grove Streets	Lead	1,010, to 1,350 mg/kg	450 mg/kg
		Zinc	434 to 473 mg/kg	410 mg/kg
		Acenaphthene	21 mg/kg, OC	16 mg/kg, OC
		Fluorene	35 mg/kg, OC	23 mg/kg, OC
		Phenanthrene	158 and 204 mg/kg, OC	100 mg/kg, OC
		Indeno(1,2,3-cd)pyrene	58 mg/kg, OC	34 mg/kg, OC
		Benzo(g,h,i)pyrene	55 mg/kg, OC	31 mg/kg, OC
		Butylbenzylphthalate	28 to 284 mg/kg, OC	5 mg/kg, OC
		Total PCBs	3 to 13 mg/kg, OC	12 mg/kg, OC
		4-Methylphenol	22 mg/kg, OC	0.67 mg/kg, OC
FFRA	Intersection of Front	Zinc	489 mg/kg	410 mg/kg
	and Railroad Streets at the main entrance to Simpson Timber	Butylbenzylphthalate	20 mg/kg, OC	5 mg/kg, OC
FITU	Just north of the intersection of First and Turner Streets, across from Kneeland Park	Butylbenzylphthalate	8 mg/kg, OC	5 mg/kg, OC

Note: SEGR sample was analyzed three times, as an environmental sample, a duplicate sample, and a replicate sample.

3.7.4.1. 2002 Geotechnical Study for City of Shelton Basin II Project

A 2002 geotechnical study was conducted in support of the city's sewer replacement lines through portions of downtown Shelton. A groundwater sample was collected from a monitoring well installed at the southwest corner of West Franklin and 12th Street, approximately 1 mile west from the Shelton waterfront. The sample was collected to evaluate disposal options should dewatering of groundwater be required. The sample was analyzed for the presence of total and dissolved metals (arsenic, cadmium, chromium, copper, mercury, nickel, lead, silver, and zinc) using USEPA method 6000/7000 series, and fats, oil, and grease using USEPA method 413.2. The results indicated that the total and dissolved concentrations of copper, mercury, and zinc exceeded state surface water criteria (Table 3-3); none of the remaining six metals (total and dissolved) exceeded their respective surface water criteria and no fats, oil, or grease were detected.

The monitoring well is located in a residential area; the source of these metals is unknown. Because shallow groundwater likely enters the sewer and stormwater collection systems throughout the downtown area, there is a possibility that groundwater exhibiting these characteristics represent a potential source of contamination for the harbor.

Table 3-3. 2002 metal results (μg/L) in groundwater collected from a monitoring well located at West Franklin and 12th Street, Shelton, Washington.

	State Surfac	e Water Criteria ^a		
Metal Constituent	Marine Criteria	Freshwater Criteria b	Total Metal Results	Dissolved Metal Results
Copper	3.1	6.3	7.54	28.1
Mercury	0.025	0.012	0.825	0.075
Zinc	81	58	79.5	386

Notes:

Metal values are reported in micrograms per liter ($\mu g/L$).

Only the metals results detected that exceeded marine and/or freshwater criterion are shown in this table.

3.7.5. Contaminants of Concern

Based on the results of various activities associated with the city's stormwater and sewer collection systems, including the Simpson waterfront facility, the following contaminants of concern have been identified for the marine environment near the shoreline:

- Heavy metals, potentially from the non-contact cooling water and noted in recent NPDES monitoring; mercury, found in two Ecology 2000 sediment samples exceeding SQS; and chromium, lead, and zinc, found in storm drain sediment exceeding SQS
- SVOCs (PAHs, phenols, phthalates) from plywood and laminate production; from machinery and trucks associated with lumber storage; from spills/leaks of petroleum-based substances such as hydraulic fluid and fuel; and from sediment and surface water runoff from roadways
- PCBs associated with transformers located across the Simpson waterfront facility, as well as throughout downtown Shelton
- PCP associated with wood preservation
- Benzoic acid, 4-methylphenol, and 2,4 dimethylphenol produced from wood waste degradation.

Based on the fact that some industrial waste was discharged to both the Pine Street and Eagle Point WWTPs, there is a potential for:

SVOCs (PAHs, phenols, phthalates) and dioxins/furans from hog fuel boiler baghouse waste.

3.7.6. Potential Contaminant Pathways to the Aquatic Environment

Historic nearshore outfalls and overland flow provide the major pathways for contaminants to move from upland source areas. Numerous storm drains have discharged to both Goldsborough

Water Quality Standards for Surface Waters of the State of Washington, Chapter 173-201A WAC.

Criteria calculated assuming a receiving water hardness of 50 mg/L.

and Shelton Creeks, offering additional opportunities for wide-spread contamination across the harbor. Both WWTPs have discharged effluent to Oakland Bay and Hammersley Inlet.

3.7.7. Data Gaps

Two sediment samples are known to have been collected along the east shore of Shelton Harbor adjacent to the Simpson facility and south of Goldsborough Creek. SMS exceedances for mercury, PCP, benzoic acid, phenol, 4-methylphenol, and 2,4-dimethylphenol were identified at these two locations. Further sampling should be conducted at identified discharge points. Analyses should include the broad range of COPCs discussed for marine sediments in general.

3.8. Sites near Oakland Bay

This section summarizes several sites identified as potential sources of contamination located distant from Shelton Harbor and near Oakland Bay or adjacent to creeks that drain into the bay (see Figure 2-2).

3.8.1. Johns Prairie Industrial Park

Johns Prairie Industrial Park consists of 270 acres of developed land used for over 40 years by forest product businesses, such as sawmills, wood chipping companies, wood-treatment facilities, and utility pole manufacturers (Port of Shelton 2008). The industrial park complex is situated on top of a large prairie approximately 3/4 miles northwest of Oakland Bay (Figure 2-1). Johns Creek is the nearest surface water stream that flows west to east toward Oakland Bay, and between 850 and 1,550 feet north of the industrial park.

The U.S. Navy historically used the land that is now the industrial park as a Naval supply depot in the 1940s. In 1965, a creosote wood treatment facility began operating in the area (Thomas 1996). The former D.H. Knudsen Pole Company manufactured wood pilings within the industrial park between the late 1960s through the early 1990s.

The former Olympic Wood Products Company treated wood in the north central portion of the industrial park from 1981 to 2003. From 1971 to 1981, Pacific Forest Products operated on the site. Permatox-1 (containing PCP), PQ-8 (a copper wood preservative), fungicides, and other wood treatment chemicals were used in a dip tank from the late 1970s to the mid-1990s. During a 2003 site investigation, soil and groundwater samples were collected and analyzed for the presence of petroleum products and PCP. Based on soil results, cleanup activities at this site were completed in 2004, which included removal of 452 tons of contaminated soil and removal of the dip tank. No groundwater contamination was found.

3.8.2. Shelton WSDOT Maintenance Facility

The Shelton Washington State Department of Transportation (WSDOT) maintenance facility is located west of State Route 3 on Johns Prairie County Road, approximately 1,800 feet west-northwest of Oakland Bay (Figure 2-1). Two USTs were decommissioned and removed, along with an oil house, in 1995. The tanks, installed in 1962, previously contained diesel and unleaded gasoline. During tank removal, oil contamination was present surrounding the tanks, as well as in an east-west drainage swale used to capture runoff with petroleum contaminants from pressure-washing trucks. Approximately 80 cubic yards of contaminated soil from the tank area and the drainage swale was removed from the site.

No groundwater was encountered during excavation and removal of the tanks. Depth to groundwater from a nearby well measured approximately 117 feet below ground surface, with assumed flow direction to the east-southeast toward Oakland Bay.

3.8.3. Bayshore Union 76 Gas Station – East 3841 SR 3

The Bayshore Union 76 gas station is located adjacent to Johns Creek at East 3841 SR 3, west of and adjacent to Bayshore Point (Figure 2-1). Johns Creek flows east and south through Bayshore Point and discharges into Oakland Bay. This gasoline station has operated at this location since the early 1970s. The station is listed in the Ecology UST database. One gasoline tank was closed in-place in 1989 (location unknown); three gasoline tanks are currently in operation (locations unknown). These tanks were installed in 1986 and upgraded with cathodic protection in 2006.

3.8.4. Calvin J. Moran Property – East 4483-4487 SR 3

The Calvin J. Moran property is located at East 4483-4487 SR 3 less than 300 feet of Oakland Bay (Figure 2-1). The Moran property is listed in the Ecology UST database. A concrete UST with a total tank capacity ranging between 600 and 750 gallons existed on the property, according to a 1986 UST notification form. The tank was used to preserve fence posts with Osmos; indicated on a form by the property owner to contain a high concentration of arsenic. The tank was reportedly between 21 and 30 years old and last used in 1960. It was also reported that the concrete tank overflowed during periods of heavy rainfall. The location of this tank is unknown.

3.8.5. Deer Creek Store – East 5881 SR 3

Deer Creek Store is a convenience store and gasoline service station located at East 5881 SR 3, approximately 2,300 feet northeast of the Oakland Bay shoreline (Figure 2-1). The gasoline station has operated at this location since 1970. In 1990, three USTs located in the south-central portion of the site were closed in-place, and three additional gasoline USTs were installed. In 1995, gasoline-contaminated soil was present during the removal of two of the three closed in-place tanks for installing a pump island canopy. Approximately 15 tons of contaminated soil

was removed from the site, but no confirmation soil samples were collected. Results from a 2002 site assessment involving probe sampling indicated gasoline, diesel, and BTEX at concentrations above MTCA cleanup criteria in soil and groundwater. Four groundwater monitoring wells were installed and monitored conducted from 2002 until March 2007, when Ecology issued a No Further Action (NFA) designation for the site based on groundwater results.

3.8.6. Sewage Discharge into Malaney Creek

In January 2004, the northern portion of Oakland Bay was closed for shellfish harvesting due to massive amounts of sewage solids illegally disposed of by a septic tanker on his property located adjacent to Malaney Creek (Figure 2-1), approximately 1.7 miles east-northeast of Oakland Bay on East 80 Halberg Road. Biosolids hauled from three Mason County-owned WWTPs were spread across the property over several months. It was also reported that biosolids had been disposed of on the property prior to 2002.

Water quality samples were collected on the property, in Malaney Creek, and in Oakland Bay by the Mason County Health Department, the Squaxin Island Tribe, and WDOH. Results of samples collected on the property indicated fecal coliform contamination ranging from 8 to over 1,600 cfu/100 ml. The result of a sample collected at the mouth of Malaney Creek had a concentration of 30 cfu/100 ml. Results of previous sampling of Malaney Creek and Oakland Bay conducted by the various agencies were also reviewed to determine if there was any correlation between various septic hauling dates and fecal coliform results. Based on quantity, location, and site observations during the inspection visit, WDOH concluded that any runoff from this site could provide an impact to the Oakland Bay shellfish growing area. Further monitoring was warranted.

3.8.7. Contaminants of Concern

- SVOCs associated with petroleum products associated with the Bayshore Union 76 gas station and Deer Creek store
- Arsenic (and possibly PCP) associated with the wood preservative concrete dip tank at the Moran property.

3.8.8. Potential Contaminant Pathways to the Aquatic Environment

Most of the sites identified with hazardous substance releases, or potential releases, are located far enough away from Oakland Bay so as to not pose a major concern from the perspective of overland flow. However, the bacterial impacts associated with the Malaney Creek release show that surface water is a viable transport pathway at distances of over a mile. There is the possibility that groundwater also can be a transport mechanism, although contaminants typically do not travel further than a few hundred feet.

3.8.9. Data Gaps

No samples have been collected and analyzed for hazardous substances near any of the identified discharge points along Oakland Bay. Sampling should be conducted at identified discharge points. Analyses should include the broad range of COPCs discussed for marine sediments in general.

3.9. Oakland Bay

This section covers three sediment quality studies conducted in Oakland Bay between 1989 to 1999, information regarding ownership of shellfish reserves in the northern portion of the bay, and bacterial contamination studies conducted in 1987/1988 and 2004/2005.

3.9.1. Sediment Studies in Oakland Bay

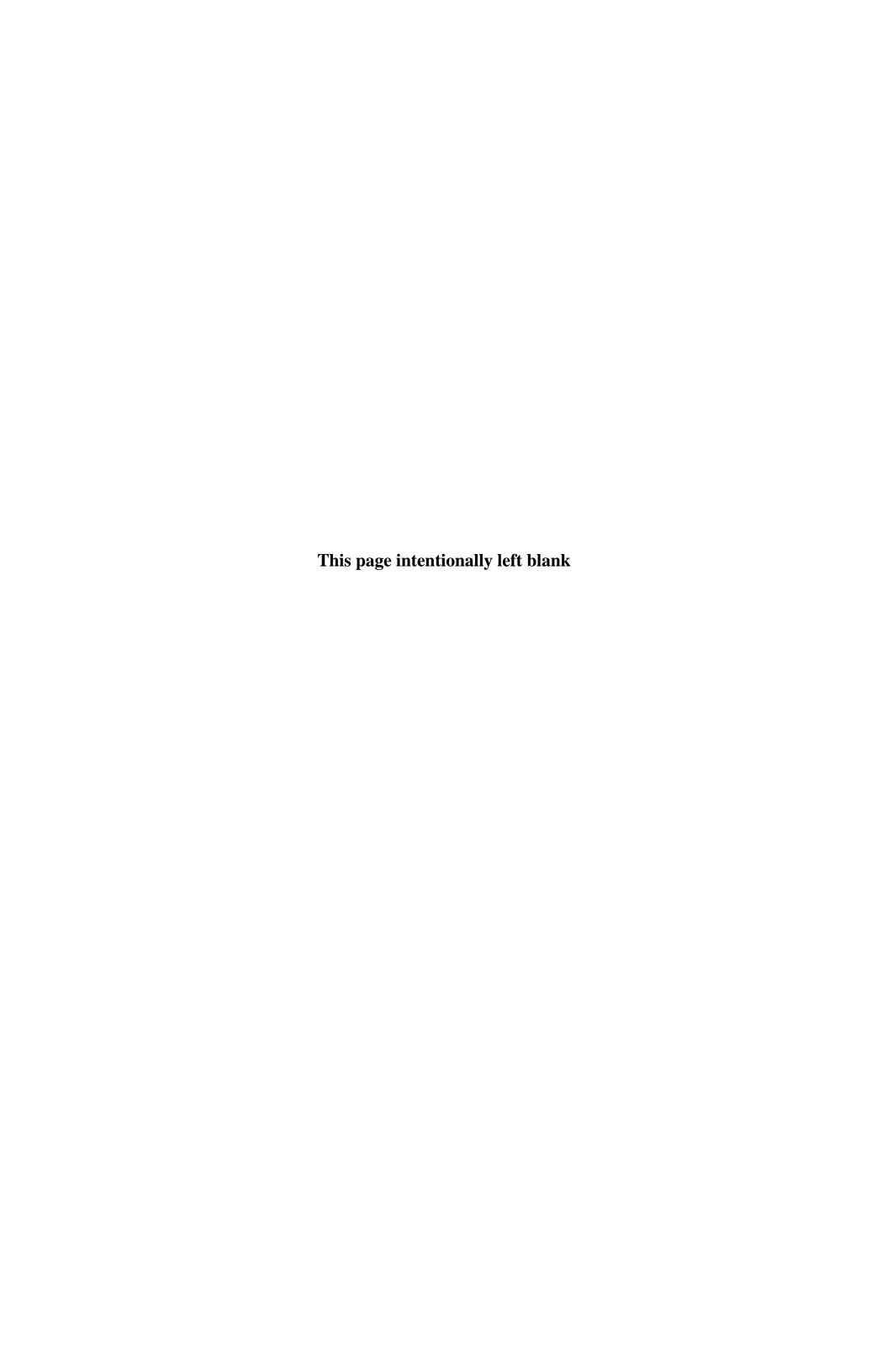
Three sediment quality studies conducted in Oakland Bay were identified in the Ecology Environmental Information Management (EIM) database. The locations of sediment samples collected from these studies are shown in Figure 3-4; the name of the study conducted and samples collected in Oakland Bay include the following:

- 1997-1999 PSAMPNOA Puget Sound Assessment and Monitoring Program and the NOAA National Status and Trends Program. Six sediment samples were collected in Oakland Bay near Shelton Harbor and adjacent to Munson Point. The samples include PSAMP/NOAA-227, PSAMP/NOAA-228, PSAMP/NOAA-229, PSAMP/NOAA-230, PSAMP/NOAA-231, and PSAMP/NOAA-232.
- 1989-1995 PSAMP HP Puget Sound Assessment and Monitoring Program's historical sediment monitoring program. This program was part of the Puget Sound Ambient Monitoring Program; the sediment component of this program was implemented in 1989 by Ecology. Four sediment samples were collected in Oakland Bay as part of this study, including two samples collected outside of Shelton Harbor and one sample near Chapman Cove (same area was sampled twice). The samples include PSAMP HP-50 (in 1989), PSAMP HP-70 (in 1992), and PSAMP HP-101R (in 1990 and 1993).
- 1992 DNRREC92 Washington DNR Aquatic Lands Sediment Quality Reconnaissance. Two sediment samples were collected in Shelton Harbor, including DNREC92SHLT01XX and DNREC92SHLT02XX.

Analyses conducted on the sediment samples collected in the three studies included grain size, TOC, SVOCs, metals, PCBs, and pesticides (DNR samples were not analyzed for PCBs or pesticides). No samples exceeded SMS criteria.



Figure 3-4. Sediment sample locations in Oakland Bay from studies conducted between 1989 and 1999.



3.9.2. Shellfish Reserves

Privately-owned portions of Oakland Bay include the following:

- Most of the northern portion of Oakland Bay north of Bayshore Point
- South of and adjacent to Bayshore Point, and extending south from Bayshore Point along the western shoreline of Oakland Bay
- Chapman Cove and extending south from Chapman Cove along the eastern shoreline of Oakland Bay
- Shelton Harbor and east of Eagle Point
- Southeast of Munson Point along the northern shoreline of Hammersley Inlet.

The remaining portions of Oakland Bay are owned by DNR (Appendix D).

Rayonier was identified as the owner of several large aquatic land parcels located within the northern portion of Oakland Bay in 1952. By the early 1970s, Taylor Shellfish Company acquired these lands to re-establish shellfish reserves.

In 1994, Barbare Brothers Inc. applied for a shoreline permit to operate a shellfish reserve within a former log boom area previously owned by Rayonier and Simpson. The former log boom area covers approximately 91 acres of aquatic land and is located in Oakland Bay west of Chapman Cove. Wood pilings associated with the log boom area were present during a 1991 survey of the area (Appendix E). The permit reviewer reported that sea lions and harbor seals had been seen within the former log boom area, and suggested that fecal coliform contamination may derive, in part, from the presence of marine animals.

3.9.3. 2004/2005 Fecal Coliform Bacteria TMDL

This study, conducted by the WDOH, involved bi-monthly sampling at 27 sites on 10 major tributaries draining to Oakland Bay and Hammersley Inlet, including: Mill, Goldsborough, Coffee, Shelton, John's, Cranberry, Deer, Malaney, Uncle John's, and Campbell Creeks. The water quality standard for fecal coliform bacteria in these Class A streams are that the geometric mean of samples cannot exceed 100 cfu/100 mL of water and the 90th percentile of samples cannot exceed 200 cfu/100 mL of water. In August 2005, Shelton Creek, Deer Creek, and Uncle John's Creek had sites exceeding the 90th percentile portion of the standard, but not the geometric mean.

All of Hammersley Inlet and most of Oakland Bay are classified as Class A waterbodies. The marine water quality standard for fecal coliform bacteria in these waterbodies are that the

geometric mean of samples cannot exceed 14 cfu/100 mL of water and the 90th percentile of samples taken cannot exceed 43 cfu/100 mL of water. Shelton Harbor is classified as Class B, with a water quality standard of 100 cfu/100mL for the geometric mean and 200 cfu/100mL for the 90th percentile.

In Shelton Harbor, two samples exceeded the 90th percentile portion of the water quality standard. In Oakland Bay, only at the mouth of Chapman Cove was the 90th percentile portion of the standard exceeded, with a 90th percentile value of 209 cfu/100ml. All Hammersley Inlet sites met the water quality standard.

In September 2004, the shorelines of Oakland Bay and Hammersley Inlet were surveyed for small drainages and/or discharge points in addition to the major tributaries that may contribute pollution to the marine water bodies (Appendix C). The 279 identified locations included storm water culverts, unnamed tributaries, and direct runoff points from agricultural activities. In February 2005, out of the 142 drainage points sampled (those conveying water), only 10 had fecal coliform concentrations greater than 100 cfu/100 ml. Two sites were located in the upper portion of Oakland Bay, one on the northwest shore and one near the mouth of Deer Creek; one site was located just south of Chapman Cove; and the remaining sites were located on the southern shoreline of Shelton Harbor and the southern shoreline of Hammersley Inlet.

4.0 Summary of Data Gaps

Analysis of existing information for Oakland Bay and Shelton Harbor has included a review of the historical and current industrial, commercial, municipal, and residential activities prevalent around the waterfront. This review has also used data from prior marine sediment and biological investigations throughout the area. Based on the available information, the most important data gaps and areas of concern include the following:

- The spatial extent and contaminant concentration of wood waste debris and by-products found in specific areas of Shelton Harbor have been identified by the Ecology 2000 study. These areas of wood waste are directly associated with pulp, paper, and lumber mill activities, and COPCs include dioxins/furans, resin and fatty acids, guaiacols, and TOC. Sediments, bioassays, and tissue analyses are needed. Areas extending into Oakland Bay need to be sampled as well to characterize potential sediment contamination associated with log rafting activities.
- Petroleum contamination of sediment needs to be assessed in likely areas of concern that have been associated with petroleum-based industry, machinery and vehicles associated with timber processing, boating, and stormwater runoff from roads. Areas of concern are along the entire Shelton Harbor shoreline, areas of sedimentation from Goldsborough and Shelton Creeks, and at the bulk fuel storage facilities north of Shelton Harbor on Oakland Bay.
- Further characterization and delineation of COPC concentrations in sediment and biota near discharge points along the Shelton Harbor shoreline, including the city's sewer and stormwater discharge points, is needed. Specifically, dioxins/furans need further spatial and vertical extent delineation across the harbor, while other COPC characterization needs to occur at specific discharge locations.
- Sediment and biota contaminant characterization and delineation of areas of Shelton Harbor that receive input from the two creeks These areas receive point and non-point source influences that may carry COPCs (SVOCs, pesticides, PCBs, metals, dioxins/furans) into the harbor.
- TBT characterization in sediments and biota from areas of concern directly influenced by the marina.
- Characterization of sediments in association with dense areas of intact, degrading, and/or submerged creosote pilings. PAHs and creosols are of particular concern with regard to creosote pilings.

Sediment and biota characterization is needed where significant intertidal and shallow subtidal beaches are present, and where contact with human populations is increased.

The data gaps identified in this report will help guide the development of the SAP for the Oakland Bay Sediment Characterization Study.

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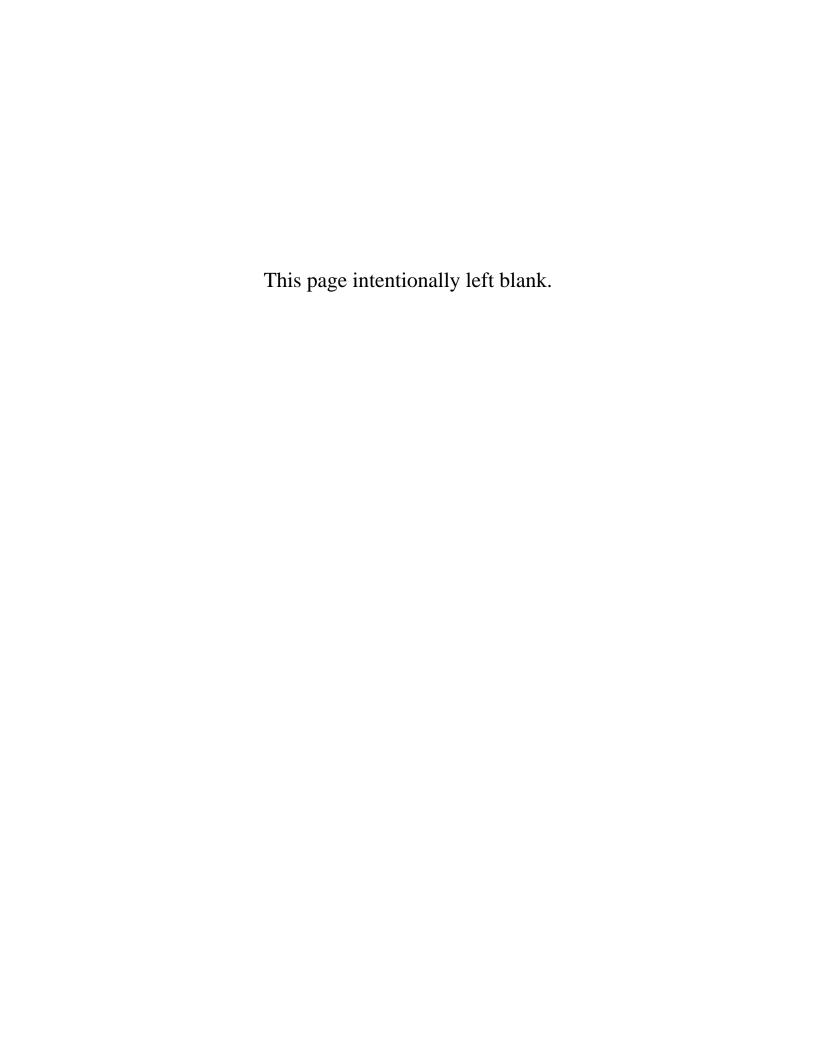
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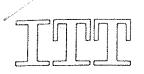
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APPENDIX A

ITT Rayonier Laboratory Chemicals





ITT Rayonier Inc.

Research Center 409 E. Harvard Avenue Shelton, Washington 98584 Telephone (206) 426-4461 March 30, 1987

Mr. Greg Cloud
Department of Ecology
Southwest Regional Office
Olympia Airport LU-11
Olympia, WA 98504-6811

SUBJECT: ESTIMATED CHEMICAL LOSS FROM ANALYTICAL TESTING

As you requested at our March 16, 1987 meeting I have compiled the estimated loss of chemicals to our NPDES effluent discharge as a consequence of analytical testing here at the Rayonier Research Center. Since the testing is all batch, each test was evaluated as to the quantity of chemicals used. From this, and the average number of tests run during normal laboratory operations an average daily quantity was calculated. The calculated effluent concentration [in part per million (mg/L)] was determined by dividing the weight of chemical loss by the average volume of water discharged during 9 hours (1,230,000 liters). The result of this exercise is contained in the attached Table I.

I have discussed the feasability of diverting these discharges to the local POTW. Our plant engineer believes it would be cost prohibitive. The laboratory installation never included any provision for such circumstances. You will note the chemicals in use are small in both number and quantity. The resulting concentrations are very low and in some instances chemically neutralize one another.

The results of our fecal coliform testing to date are:

		Disc	harge				"Ambient"	t
Date	M-1	FC	M-	TEC	M-F	C.	M-T	EC
	F.C.	Kleb.	E. coli	Kleb.	F.C.	Kleb.	E. coli	Kleb.
March 4 10 18	NT 2,300* 1,000	NT NT 20,000	0 50* 0	12,000 6,200* 15,000	NT 2,100 100	NT NT 4,700	900 1,500 0	9,600 700 5,200

*Includes some "ambient."

Sincerely yours,

ITT RAYONIER INCORPORATED

Research Center

Philip A. Hamlin

tar Attachment

cc: R. L. Casebier, Director

	Compound	Average Quantity (grams)/day	Average 2 Concentration (ppm, mg/L)
P058 F003	Acetic Acid EHW(B) Acetone Ammonium Hydroxide (30%) Cupriethyenediamine [Copper Content] [Ethylenediamine] Carbon Black Clay Cleaning Solution [Chromere] [Initially potassium dichromeminantly chromemine] sulfuric acid.	13,580 1,011 4 250 16 32 <2 <2 24 mate/sulfuric acid: fate, potassium sulfate	0.82 0.003 0.20 0.013 0.026 <0.002 <0.002 0.02 Spent is predo-
FOO	Ferrous Ammonium Solfate Gypsum (Calcium Solfate) Hydrochloric Acid Isopropyl Acetate Magnesium Sulfate Methanol Methylene Chloride Nitric Acid Perchloric Acid Sodium Chloride Sodium Hyrdoxide 10% solution 18% solution 50% solution Sodium Sulfate Sulfuric Acid Vanillin Miscellaneous: Photo Developer Film Developer Fixer (Sodium Thiosulfate)	25 <100 15 200 60 10 60 70 16 20 440 60 100 15 20 125 10	0.02 <0.08 0.01 0.16 0.05 0.008 0.05 0.06 0.01 0.02 0.36 0.049 0.08 0.012 0.016 0.102 0.008 0.002 0.008

- 1. Summation of chemical losses resulting from chemical (analytical) testing and photography.
- 2. Concentration was calculated by dividing the weight of chemical loss by the volume of water discharged during a 9 hour laboratory day. [Effluent volume would be about 865,000/24 x 9 = 324,375 gal/9 hr (1.23 x 10^6 liters).]

Also perhaps Osmin Tetroside it doing electron microscopy works

U080 F00€ EHW (C+H)

3 Cylouseud

no map showing location of stations in sile

25 Feb 87 -	Oakland Bay (Inne	Harbor and Stream Sampling)
Station		OBKES
ITE!	7 90,000 (heavy bland growth have 72,000 may have theodown)	0
IT EZ	72,000 may have bleed ever)	
5A	8 (heavy bkgrd)	\mathcal{O}^{-1}
58	9	0
Hormwater 4A Chyc Boat ng Fair h G	900	8%
48	240	<i>○</i> %
10 A Point Sources	18	
10 B	//	Q
13	< <i>/</i>	
7	11	25%
Shel STPA	20	
hel STPB	69	
TTIA	900	0
T 18.	1200 (heavy bright)	<i>O</i> -
72 Inner Harbor	3	6
24	. 1	0
25	/	< I
SI A	23	
18	71	4
Gold M	3	production of the second secon
old 2 Streams	5	· · · · · · · · · · · · · · · · · · ·
Shel 1A	15	0
Shel 1B	15	
hel M	$\frac{f}{f}$	

the ITT Rayonier lab

EPA I.D. NUMBER (copy from Item 1 of Form 1)

W0130102

OUTFALL NO

Form Approved OMB No. 2000-0059 Approval expires 12-31-85

PLEASE PRINT OR TYPE IN THE UNSHADED AREAS ONLY. You may report some or all of this information on separate sheets (use the same format) instead of completing these pages. SEE INSTRUCTIONS.

V. INTAKE AND EFFLUENT CHARACTERISTICS (continued from page 3 of Form 2-C)

PART A - You must provide the results of at least one analysis for every pollutant in this table. Complete one table for each outfall. See instructions for additional details.

PARI A - You m	TGST PAGE TOO THE	, , , , , , , , , , , , , , , , , , , ,		EFFLUENT				3. UN (specify if			TAKE (optiona	ıl)
1. POLLUTANT	a. MAXIMUM	DAILY VALUE	b. MAXIMUM 3	DAY VALUE	c.LONG TERM A	VRG. VALUE		a, CONCEN-	b. MASS	a LONG AVERAGI		b, NO. OF
	(1)	(z) MASS	(1) CONCENTRATION	(2) MASS	(1) CONCENTRATION	(2) MASS	ANALYSES	TRATION	D. MASS	CONCENTRATION	(2) MASS	
a. Biochemical Oxygen Demand (BOD)	6.2	16.8					4	mg/L	pounds			M 20 00 - 1 21 - 1 A 70 1 A 7 - 1 1 A 70
b, Chemical Oxygen Demand (COD)	ND			No state and the Second Se		A1 100 11 11 11 11 11 11 11 11 11 11 11 1	1					
c. Organic Car (TOC)	5.5	14.9				<u></u>	1	mg/L	pounds			
d. Total Suspended Solids (TSS)	0.35	0.9					4	mg/L	pounds			
e. Ammonia (as N)	0.35	0.9					1	mg/L	pounds	VALUE		
_	VALUE		VALUE		VALUE					VACUE		
f. Flow	0.822				0.726			MGPD	<u> </u>	VALUE		
g. Temperature	VALUE		VALUE		VALUE			°c	:	VALUE		
(winter)	19.0				14.5		Cont.					
h, Temperature	VALUE		VALUE		VALUE			ەر	-	VALUE		
(summer)	18.4				16.0		Cont.					
i. pH	MINIMUM 3.0	11.4	3.0	11.4			continuo	S STANDAR	D UNITS			

PART B - Mark "X" in column 2-a for each pollutant you know or have reason to believe is present. Mark "X" in column 2-b for each pollutant you believe to be absent. If you mark column 2a for any pollutant which is limited either directly, or indirectly but expressly, in an effluent limitations guideline, you must provide the results of at least one analysis for that pollutant. For other pollutants for which you mark column 2a, you must provide quantitative data or an explanation of their presence in your discharge. Complete one table for each outfall. See the instructions for additional details and requirements.

	T				3 (EFFLUENT		· · · · · · · · · · · · · · · · · · ·		4. UI	STIV		AKE (optional))
1. LUT- ANT AND	a. BE-			DAILY VALUE	b. MAXIMUM 3	O DAY VALUE	c.LONG TERM (if ava	AVRG. VALUE	d NO. OF	a. LONCEN-	b. MASS	a. LONG AVERAG	TERM E VALUE	D. NO. OF
CAS NO. (if available)	PRE- SENT	SENT	CONCENTRATION	(2) MASS	(1) CONCENTRATION	(2) MASS	(1) CONCENTRATION	[2] MASS	YSES	"RATION	B. 81A33	(1) CONCENTRATION	(Z) MASS	YSES
a, Bromide (24959-67-9)			ND	-					1	A SUMMER PROPERTY STOPPARTY				
b. Chlorine, Total Residual			ND						2					
c. Calar			ND		PARTY				1					
d, Fecal Coliform			27,000	-	10,274		1605		36	CFU/ 100 ml				
e, Fluoride (16984-48-8)			0.5 ррм	(1.2)	7				1	mg/L	pounds			
f. Nitrate- Nitrite (as N)			ND		Samuel Marie Control				1					

1. POLLUT-		TARK 'X'			EFFLUENT	•			4. U	NITS	5, INT	AKE (optional	1)	
ANT AND CAS NO.	A. DE- LIEVED PRE- SENT	D.BE-	a, MAXIMUM D	AILY VALUE	b. MAXIMUM 3	O DAY VALUE	c.LONG TERM	AVRG. VALUE ilable)	d. NO. OF	a. CONCEN-	b. 44.5.5	A LONG	TERM E VALUE	D. NO. OF
(if available)	PRE- SENT	SENT	(1) CONCENTRATION	(2) MASS	CONCENTRATION	(2) MASS	(1) CONCENTRATION	(2) MASS	YSES	a, CONCEN- TRATION	b. MASS	CONCENTRATION	(2) MASS	YSES
g. Nitrogen, Total Organic (as N)			ND						1					
h. Oil and Grease			ND						1					
i. Phosphorus (as P), Total (7723-14-0)			0.32	0.9					1	mg/L	pounds			
j. Radioactivity	11													
(1) Aipha, Total		Х												
(2) Beta, Total		Х						1000000			· · · · · · · · · · · · · · · · · · ·			
(3) Radium, Total		Х												
(4) im 226, tal		Х												
k. Sulfate (as SO ₄) (14808-79-8)			75.0	2.03		-			1	mg/L	pounds			
l. Sulfide (as 3)			ND						1.					
m, Sulfite (as SO3) (14265-45-3)			ND						1 1					
n. Surfactants			ND						1					
o. Aluminum, Total (7429-90-5)			0.025	0.031					1	μg/L	grams			
p. Barlum, Total (7440-39-3)			0.003	0.004					1	μg/L	grams	Arman		
q. Boron, Total (7 440 -42-8)			0.05	0.061	. <u></u>				1	μg/L	grams			
r. Cobelt, Tot (74-4)			ND	MALE WAS WAS AND A SECOND OF THE PARTY OF TH					1					
s. Iron, Total (7439-89-6)			0.071	0.087					l	μg/L	grams			
t. Magnesium, Total (7439-95-4)			1.4	1.722					1	μg/L	grams			
i. Moly bdenum, Total (7439-98-7)			0.001	0.001					1.	μg/L	grams			
v. Manganese, Total (7439-96-5)			0.011	0.013				William Waller	1	μg/L	grams			
w. Tin, Total (7440-31-5)			ND						1					
x. Titanium, Total (7440-32-6)			ND						1					

Form Approved. OMB No. 2000-0059 Approval expires 12-31-85

If C - If you are a primary industry and this outfall contains process wastewater, refer to Table 2c-2 in the instructions to determine which of the GC/MS fractions you must test for. Mark "X" in column 2-a for all such GC/MS fractions that apply to your industry and for ALL toxic metals, cyanides, and total phenols. If you are not required to mark column 2-a (secondary industries, nonprocess wastewater outfalls, and nonrequired GC/MS fractions), mark "X" in column 2-b for each pollutant you know or have reason to believe is present. Mark "X" in column 2-c for each pollutant you believe is absent. If you mark column 2a for any pollutant, you must provide the results of at least one analysis for that pollutant. If you mark column 2b for any pollutant, you must provide the results of at least one analysis for that pollutant if you know or have reason to believe it will be discharged in concentrations of 10 ppb or greater. If you mark column 2b for acrolling, and introphenol, or 2-methyl-4, 6 dinitrophenol, you must provide the results of at least one analysis or each of these pollutants which you know or have reason to believe that you discharge in concentrations of 100 ppb or greater. Otherwise, for pollutants for which you mark column 2b, you must either submit at least one analysis or briefly describe the reasons the pollutant is expected to be discharged. Note that there are 7 pages to this part; please review each carefully. Complete one table (all 7 pages) for each outfall. See instructions for additional details and requirements.

I. POLLUTANT	2. 1	MARK 'X'			3.	EFFLUENT				4, UI	VITS	5. IN 1	AKE (optic	ənalj
	a.rest	b. 8 E - C. 1	a. MAXIMUM	DAILY VALUE	b. MAXIMUM	30 DAY VALUE	C.LONG TERM	AVRG. VALUE	d. NO. OF	a. CONCEN-		8. LONG AVERAG	TERM E VALUE	b. NO. C
(if available)	RE-	D. BE- C. I	NT CONCENTRATIO	(2) MASS	(1)	(2) MASS	(1)	(z) MASS	ANAL- YSES	TRATION	b, MA55	(1) CONCEN- TRATION	(2) MASS	YSE5
METALS, CYANID	E, AND	TOTAL	PHENOLS											
M. Antimony, (197440-36-0)			ND						1					
2M. Arsenic, Total 7440-38-2)			ND			10			1.					
BM, Beryllium, Fotal, 7440-41-7)			ND	7.38	·e)	Key Igan			1					
IM, Cadmium, Fotal (7440-43-9)			ND		Mye" Je"				1			75		
5M. Chromium, Fotal (7440-47-3)			ND	(csp"	IN 16 "				1					
5M. Copper, Total 7440-50-8)			6.0	7.38					1	μg/L	grams			
7M. Lead, Total 7439-92-1}			ND						1					
M. Mercury, Total 7439-97-6)		-	ND						1					
M. Nickel, Total 7440-02-0)			ND						1					
0N			ND					The state of the s	L					
11M. Silver, Total 7440-22-4)			ND						1					
2M. Thallium, Total (7440-28-0)			ND						1		,			
3M. Zinc, Total 7440-66-6)			12.0	14.76					1	μg/L	grams			
4M. Cyanide, Fotal (57-12-5)	***************************************		ND						1		**************************************			
I5M. Phenols, Fotal			8.0	9.84					1	μg/L	grams			

2,3,7,8-Tetra DESCRIBE RESULTS chlorodibenzo-P-Diox in (1764-01-6)

I. POLLUTANT	2.	MARK	, X ,				EFFLUENT				4. UI	NITS	5, INT	AKE (optic	onal)
AND CAS NUMBER	ATEST	b. se-	C BE.	8, MAXIMUM I	DAILY VALUE	b. MAXIMUM 3	O DAY VALUE	c.LONG TERM (if ava	AVRG. VALUE	d, NO. OF	a. CONCEN-		a LONG AVERAGI		b. NO. 01
(if available)					(2) MASS	(1) CONCENTRATION	(2) MASS	(1) CONCENTRATION	(2) MASS	ANAL- YSES	TRATION	b. MASS	(I) CONCEN-	(z) MASS	YSES
GC/MS FRACTION	- VO	LATIL	E COM	IPOUNDS									1831108		
1V. Acrolein (107-02-8)				ND						1					
2V. Acrylonitrile (107-13-1)				ND			***************************************			1					
3V. Benzene (71-43-2)				ND		·		-		1		- OWALLOW I I I			
4V. Bis (Chloro- methyl) Ether (542-88-1)			X				***************************************								
5V. Bromoform (75-25-2)				ND			-			1					
6V. Carbon Tetrachioride (56				ND		÷				1					
7V,5 orobenzene (108-90-7)				ND			-			1					
8V. Chlorodi- bromomethane (124-48-1)				ND						1					
9V. Chloroethane (75-00-3)				DM						1					
10V, 2-Chloro- ethylvinyl Ether (110-75-8)				ND						1					
11V, Chloroform (67-66-3)				ND						1		T- 5.00 0			
12V. Dichloro- bromomethane (75-27-4)				ND						1					
13V, Dichloro- difluoromethane (75-71-8)			Х												
14V. 1,1-Dichloro- ethane (75-34-3)				ND						1					
15% Dichloro- ethane (107-06-2)				ND						1					······································
16V. 1,1-Dichloro- ethylene (75-35-4)				ND					an community of a fisher annual and a fisher a	1					
17V. 1,2-Dichlero- propene (78-87-5)				ND						1					
18V. 1,3-Dichloro- propylene (542-75-6)				ND						1					
19V. Ethylbenzene (100-41-4)				ND						1					
20V. Methyl Bramide (74-83-9)				ND						1					
21V, Methyl Chloride (74-87-3)				ND					, , , , , , , , , , , , , , , , , , ,	1					
3 02 3510-2C	(Rev	2-851			······································		PAG	E V-4	1			I was a second s	CAN	TINUE ON	DACEN

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ANT	2. M	ARK 'X				b. MAXIMUM 3	O DAY VALUE	CLONG TERM	AVRG. VALUE	d NO OF	a. CONCEN-		a LONG AVERAG	TERM EVALUE	b, NO.OF ANAL
MBER (available)	arest t ing ci ae- guia-	D. BE- C. IEVED LIE PRE- /	: V = H_	a. MAXIMUM D	(2) MASS	(1) ava	(2) MASS	(1) CONCENTRATION	(2) MASS	ANAL. YSES	TRATION	b. MASS	(1) CONCEN- TRATION	(z) MASS	VSES
C/MS FRACTION				CONCLATRATION	ued)	CONCLUTION	, 40	CONCENTRATION						.,-,	
	- VOL.						lev Cul			1	-107 /T	arame			
V. Methylene iloride (75-09-2)		ļ 		17.0	20.91 <i>G</i>	TTC	ans out			1	μg/L	grams			
V. 1,1,2,2-Tetra- loroethane 9-34-5)				ND		+60 9	alevita	<u> </u>		1				v	
V. Tetrachloro- hylene (127-18-4)				ND	Analysis and a second of the s					1				manan , r r r r r r r g y y a a a a a a a a a a a a a a a a a	
V. Toluene 08-88-3)				ND						1	**************************************	-			
3V. 1,2-Trans- ichtoroethylene 56-60-				ND		-				1		••••			
rV. 1, in li- iloroethane 1-55-6)				ND						1		many recent code to a code			
3V. 1,1,2-Tri- iloroethane '9-00-5)			,	ND						1	100-101-00-00-00-00-00-00-00-00-00-00-00				
3V. Trichloro- hylene (79-01-6)				ND					-	1					
OV. Trichloro- uoromethane '5-69-4)			Х	·					-	1				Mr	
1V. Vinyl hloride (75-01-4)		-		ND		And the state of t				1			and the same of th	and the state of t	
C/MS FRACTION	ı – ACII	D COMP	OUN	DS		<u> </u>									-
A. 2-Chloropheno				ND						1					
A, 2,4-Dichloro- henol (120-83-2)				ND						1			500 - 100 -		
A. 2,4-Dimethyl- henol (2,467-9)				ND						1					
A. 4.6-Dimitro-O- resol (534-52-1)				ND	Marian Marian		.,			1					
A. 2,4-Dinitro- henol (51-28-5)				ND						1			g/s		-
A. 2-Nitrophenol 38-75-5)				ND						1			,		
A. 4-Nitrophenol 100-02-7)				ND						1					
:A. P-Chloro-M- cresol (59-50-7)				ND	·					1					
IA. Pentachloro- sheno! (87-86-5)				ND						1					
10A, Phenoi 108-95-2)				ND						1					
11A. 2,4,6-Tri- chlorophenol (88-06-2)				ND				3F V-5		1	Marie of Confession Supplemental Science Science Supplemental Science Sci	LVA W T LEWIS FOR MEN WAS AND A STATE OF THE	/epitylete entre e	NITIMITE ON	REVERSE

TINUED FROM						3, E	FFLUENT		MOC VALUE		4. UN		A LONG	TERM	b. NO. OF
LLUTANT ND CAS		MARK		a. MAXIMUM D	ALLY VALUE	b, MAXIMUM 3 (if avai	DAY VALUE	c.LONG TERM	lable)	d NO.OF	a. CONCEN- TRATION	b. MASS	(I CONCEN-	(z) MASS	ANAL- YSES
UMBER	ATEST	D, BE- LIEVED PRE- SENT	C' SPE-	a. MAXIMUM D	(z) MASS	(1) CONCENTRATION	(z) MASS	(1)	(/) MASS	YSES	TRATION		TRATION	127	
f available)	AE. QUIR- EQ	SENT	BENT	CONCENTRATION	(2) MASS	CONCENTRATION		Curcenting							+
S FRACTION	– BA	SE/NE	JTRAL	COMPOUNDS						1					
Acenaphthene 32-9)				ND						1					
Acenaphtylene 3-96-8)				ND			100								
Anthracene 0-12-7)		-		ND	A A A A A A A A A A A A A A A A A A A			-		1					
Benzidine -87-5)				ND						1				44,7,7,444,7,7,7	
Benzo (a) thracene				ND						1					
.55-3) Berg(2) ene (2-8)	-		-	ND						<u> </u>					
3,4-Benzo- oranthene				ND						1.				7	
)5-99-2) . Benzo (ghi) -ylene	-			ND						1					
1-24-2) Benzo (k) Joranthene			1	ND						1					
07-08-9) B. Bis (2-Chlor 10xy) Methane	0-			ND						1					
11-91-1) B. Bis (2-Chlor Ivl) Ether				ND						1.					
11-44-4) 3. Bis <i>(2-Chloroiso-</i> pyl) Ether (102-60-	1)		<u> </u>	ND						1					
BB. Bis (2-Ethy exyl) Phthalate	1-			4.0*						1	μg/L				
17-81-7)				ND						1.					
henyl Phenyl ther (1995-3)	yΙ			ND		¢6				1.					
hthalate (85-68 6B. 2-Chloro- aphthalene				ND						1.					
91-58-7) 7B. 4-Chloro-				ND						1					
henyl Phenyl ther (7005-72- 8B. Chrysene 218-01-9)	3)			ND						1					
9B. Dibenzo (d	i,h)			ND						1					
53-70-3) 0B. 1,2-Dichlo genzene (95-50	ro- -1)			ND						1					
				ND						1				CONTINUE	ON PAG
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CEONT CONTROL OF THE PROPERTY		MARK	,х,			3.1	EFFLUENT				4. UN	IITS		AKE (optio	nal)
學 / 整	a + 1, s + 1	b, se-	C. BE-	a. MAXIMUM I	DAILY VALUE	b. MAXIMUM 3 (i/ ava	O DAY VALUE	CLONG TERM	AVRG. VALUE	d, NO, OF	a. CONCEN-		a. LONG AVERAG	TERM EVALUE	b. NO. OF
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5 6 FRACTION	RW:	oc/NE	JIKAL	. CONIFOUNDS	continuea.										
FRACTION 1,4-Dichloro- mozene (106-46-7)				ND						1			\$1,		
238. 3,3'-Dichloro- benzidine (91-94-1)				ND						1					Sin. Maria
24B. Diethyl Phthalate (84-66-2)				ND						1	·				
25B. Dimethyl Phthalate (131-11-3)				ND						1					
26B. Di-N-Butyl Phthalate (84-74				ND			\$\$\$\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\			1		*,,,,,=	**************************************		
27B. 2,4 (nitro- toluene (121-14-2)				ND						1	-				
28B. 2,6-Dinitro- toluene (606-20-2)				ND						1					
29B. Di-N-Octyl Phthalate (117-84-0)				8.0	9.84					1	μg/L	grams			
308. 1,2-Diphenyl- hydrazine (as Azo- benzene) (122-66-7)				ND		***************************************				1					
31B. Fluoranthene (206-44-0)				ND			,			1	Manual II	*********			
328. Fluorene (86-73-7)				ND	و سور سند و در سرد و در	menendra for the first of the f			10 No. 30	1					
33B. Hexachlorobenzene (118-74-11				ND						1					
34B. Hexa- chlorobutadiene (87-68-3)		Norman de plus must as omnive		ND						1					
35B. He loro- cyclope. Jiene (77-47-4)				ND						1					
36B. Hexachloro- ethane (67-72-1)				ND						1					
378, Indeno (1,2,3-cd) Pyrene (193-39-5)				ND			a HAR DOTA	, Mark In		1					The second secon
38B. Isophorone (78-59-1)				ND						1		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			
39B. Naphthalene (91-20-3)				ND						1					. , , , , , , , , , , , , , , , , , , ,
40B. Nitrobenzene (98-95-3)				ND						1					
41B. N-Nitro- sodimethylamine (62-75-9)				ND				Proceedings and the second		1					
42B. N-Nitrosodi- N-Propylamine (621-64-7)				ND						1					a pra primeto de constituido de cons

CONTINUED FROM THE FRONT 1. POLLUTANT 4. UNITS 5. INTAKE (optional) 3. EFFLUENT AND CAS b. MAXIMUM 30 DAY VALUE | C.LONG TERM AVRG. VALUE (if available) ILONG TERM AYCHAGE VALUL A TI ST D. BE- C. BE-ING LIEVEDLIEVED BE- PHL- AB-QUIN SENT SENT b Missor a, MAXIMUM DAILY VALUE d NO. OF NUMBER a. CONCEN b. MASS AHAL YSES ANAL TRATION (I) (I) CONCENT (if available) (1) CONCENTRATION (1) CONCENTRATION YSES (2) MASS (2) MASS [2] MASS (2) MASS GC/MS FRACTION - BASE/NEUTRAL COMPOUNDS (continued) 13B, N-Nitro-1 odiphenylamine 86-30-6) ND14B. Phenanthrene ND 85-01-8) ISB. Pyrene ND 1 129.00 16B. 1,2% hlorobenzene ND 1 120-82-1) **IC/MS FRACTION - PESTICIDES** P. Aldrin ND 1 309-00-2) P. a. BHC 319-84-6} ND 1 ₽. β-внс ND 319-85-71 1 P. Y-BHC ND1 58-89-9) P. δ -BHC ND 1 319 86 8) P. Chlordane 57-74-9) ND 1 P. 4.4' 5 ND 1 50-29-3 P. 4,4'-DDE 72 55-91 ND 1 P. 4,4'-DDD ND 72-54-8) 1 OP, Dieldrin ND30-57-1) 1 1P. a-Endosulfan 115-29-7) ND1 2P. β -Endosulfan ND1 115-29-7) 13P. Endosulfan Sulfate (1031-07-8) ND 1 ND1

ND

Form: Appa oved. OMB No. 2000-0059 Approval expires 12-31-85

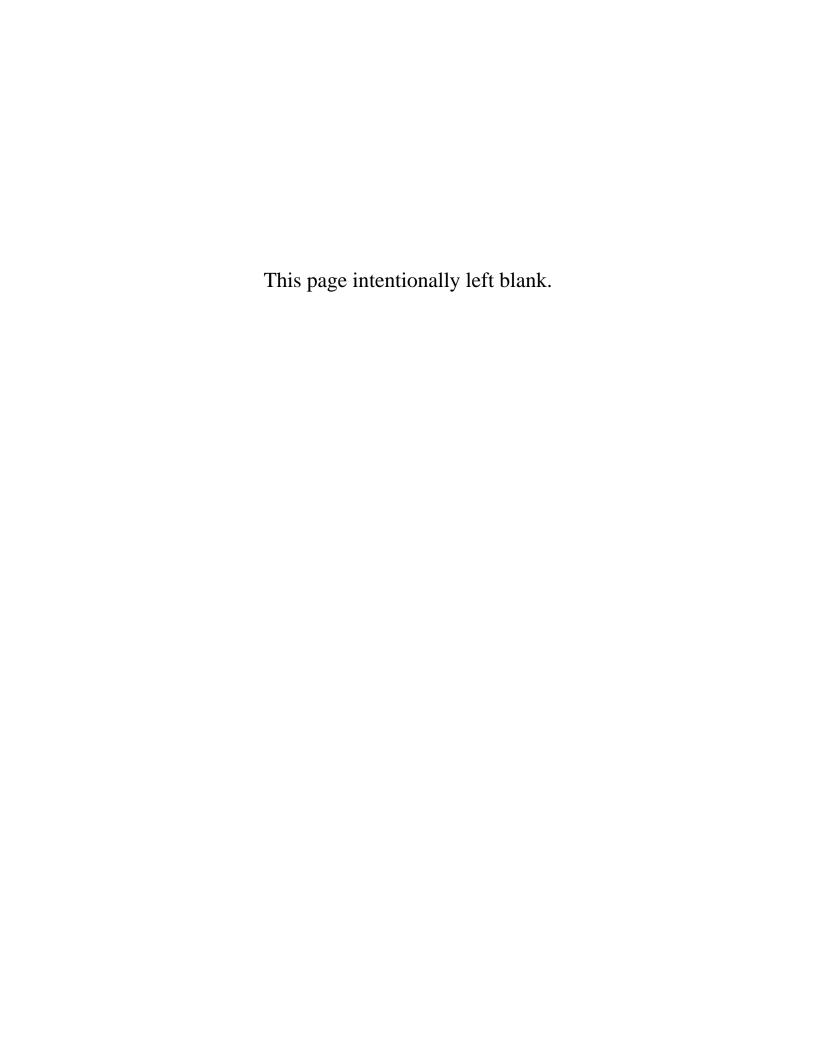
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1. POLLUTANT	2.	MARK'	x,				EFFLUENT				4. UI	VITS	5. IN 1	TAKE (optio	mat)
AND CAS NUMBER	A TEST	D, BE- C	L BE-	a, MAXIMUM D	AILY VALUE	b. MAXIMUM 3	O DAY VALUE	c.LONG TERM	AVRG. VALUE	d NO OF	a. CONCEN-	b. MASS	a. LONG AVERAG	STERM E VALUE	b, NO. OF
(if available)	RE- QUIR- EQ	D, BE- (LIEVEOLI PRU- SENT	AB" SENT	(1) CONCENTRATION	(2) MASS	(I)	(2) MASS	(i) CONCENTRATION	(z) MASS	YSES	TRATION	· · · · · · · · · · · · · · · · · · ·	(I) CONCEN- TRATION	(2) MABS	YSES
GC/MS FRACTION	- PES	STICIDE	S (cor	rtinued)											
17P, Heptachlor Epoxide (1024-57-3)				ND						1	and the state of t				
18P. PCB-1242 (53469-21-9)		,		ND						1			0	_1140-77-7-7-7-1440-7-7-7-7-7-7-7-7-7-7-7-7-7-7-7-7-7-7-	
19P. PCB-1254 (11097-69-1)				ND		·		\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \		1					
20P. PCB-1221 (11104-28-2)				ND						1		JALIAN STRUFT I SPORTES TO SE TO MINISTER AND THE SECOND THE SECOND SECO	*		
21P. PCB-1232 (111) 6-5)				ND						1					
22P. PCB-1248 (12672-29-6)				ND						11					_ r,g spare, a seeing punit block PASSEA
23P, PCB-1260 (11096-82-5)			-	ND		14 ct				1					
24P. PCB-1016 (12674-11-2)				ND						1					
25P, Toxaphene (8001-35-2)				ND						1					

EPA Form 3510-2C (Rev. 4-84)

PAGE V-9

APPENDIX B

National Dioxin Study Data



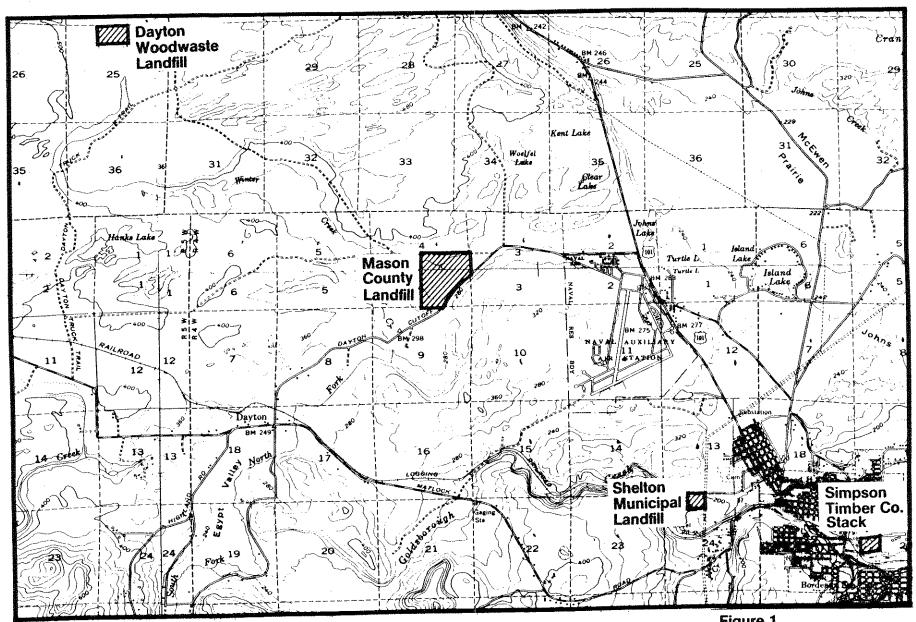
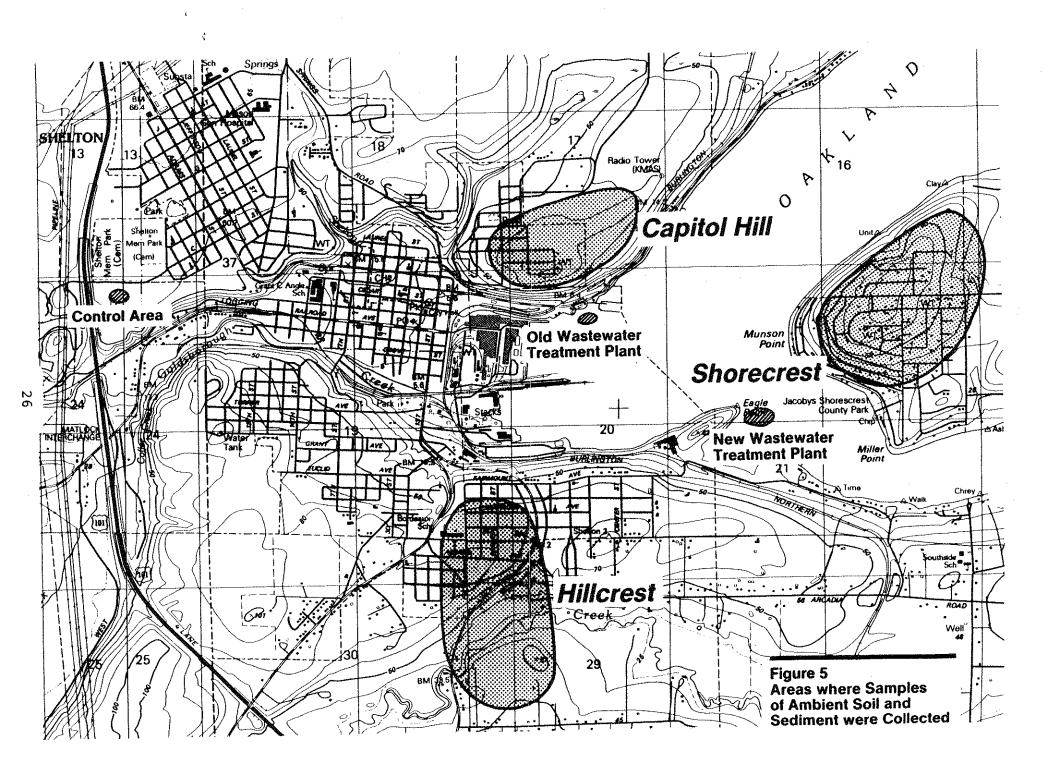


Figure 1 Simpson Timber Company Waste Disposal Locations



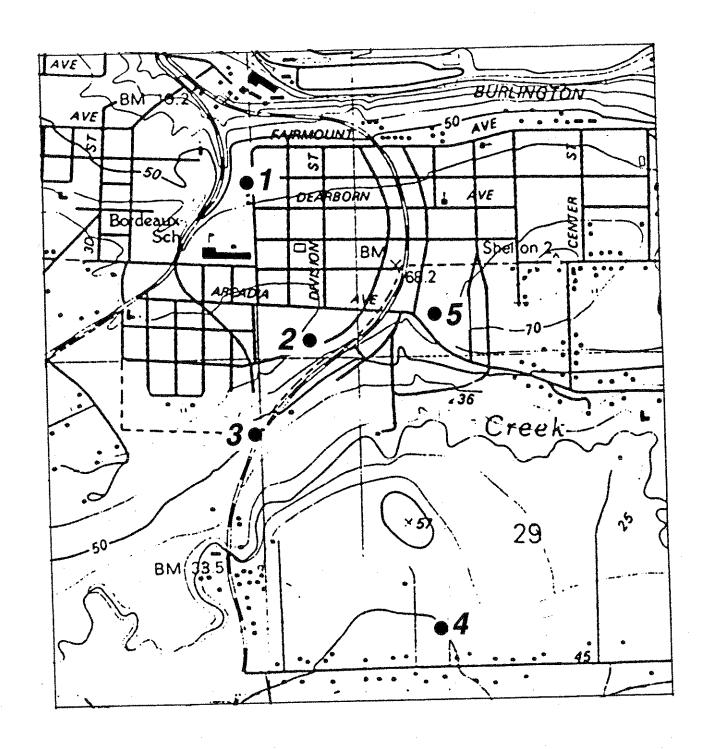


Figure 6 Locations Sampled Hillcrest



Figure 7 Locations Sampled Capitol Hill

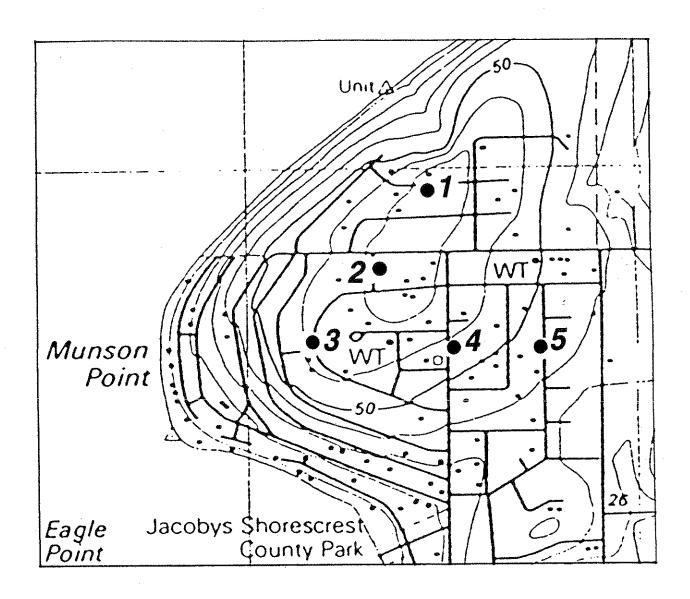


Figure 8
Locations Sampled
Shorecrest

TABLE 1 REPORTED CONCENTRATIONS (PARTS PER BILLION) OF CHLORINATED DIOXINS AND FURANS 'IN SAMPLES COLLECTED FROM STC FACILITIES AT SHELTON COMPARED WITH OTHER COMMON SOURCES

· · · · · · · · · · · · · · · · · · ·	New		High	Low	High	EPA Results From High Pressure	Wood Stoves		
		Low	Pressure	Pressure	Pressure	Boiler b	(Western		
	Powerplant	Pressure	Boiler	Boiler	Boiler	Baghouse	(u.s.)	Firepl	
	Cyclone	· ·	Baghouse	Composite	Composite	(Range)	(Range)	25-Year Old	12-Year Old
Homologue or Isomer	Residue	Boiler Baghouse	pagnouse	000000000					
Tetrachiorinated Furans		***	9.2	7.6	0.16	135 - 170			
(Total)	0.43	118 10.5 ^c		0.67	ND	4.5 - 6.5			
2,3,7,8 TCDF	0.078	10.5	0.58	0.67	HD	11.5			
Tetrachlorinated Dioxina									
(Total)	0.36	90	186	13.3	0.66	195 - 211			0.1
	ND	4.2	ND	0.16	MD	0.8 - 1.2	0.0064 - 0.02		0.1
2,3,7,8 TCDD									
Pentachlorinated Furans				2.4	ND	71 -77			
(Total)	0.076	62	3.4			7.			
1,2,3,7,8 PCDF	ND	7.2	0.37	. 0.34	ND				
2,3,4,7,8 PCDF	ND	6.9	0.33	0.22	ND				
Pentachlorinated Dioxins									ND - 0.89
	0.24	161	226	9.9	1.7	236 - 273	0.093 - 9.22	3.7 - 25.0	- U.07
(Total)	ND	14.0	0.61	0.34	ND				
1,2,3,7,8 PCDD	ND	-,,-							
Hexachiorinated Furans									
(Total)	ND	26.7	2.3	1.0	ND	44 - 59			
1,2,3,4,7,8 Hexa CDF	ND	3.3	0.25	ND	ND				
1,2,3,6,7,8 Hexa CDF	ND	4.0	0.18	ND	ND				
1,2,3,7,8,9 Hexa CDF	NÐ	ND	ND	ND	ND				
2,3,4,6,7,8 Hexa CDF	ND	2.6	ND	ND	ND			*	
2,3,4,0,7,0 nexa CDF									
Hexachlorinated Dioxina						202 34.0			
(Total)	0.83	169	190	8.0	1.3	283 - 348			
1,2,3,4,7,8 Hexa CDD	ND	10.8	4.6	0.38	ND				
1,2,3,6,7,8 Hexa CDD	ND	9.0	3.0	0.25	ND				
1,2,3,7,8,9 Hexa CDDD									
, , , , ,				-		-			
Description of Present									
Heptachlorianted Furans	ND ·	4.8	1.4	0.21	ND	24 - 38			
(Total)		2,8	0.34	0.16	ND				
1,2,3,4,6,7,8 Hepta CDF	ND	ND	ND	MD	ND				
1,2,3,4,7,8,9 Hepta CDF	, MD	un	No						
Heptachiorinated Dioxina						252 135			
(Total)	0.12	57.1	34.6	2.2	0.54	252 - 410			
1,2,3,4,6,7,8 Hepta CDD	ND	31.5	21.0	1.0	ND				
The sale and the modern one									
Octachlorinated Furans				ND	ND	2.3 - 3.0			•
(Total)	ND	0.51	ND	ND.	nu	* 5.0			
Octachlorinated Dioxins									
	0.27	14.1	5.5	0.78	0,49	44 - 72			
(Total)	V+2.								

 a_{ND} = not detected. $b_{From Table 5-22}$ of Keller, et al. (1986).

Sum of 2,3,7,8- plus 1,2,5,9- plus 2,3,4,8-TCDF.

Table 2

PRELIMINARY[®] REPORTED CONCENTRATIONS (PARTS PER BILLION) OF CHLORINATED DIOXINS AND FURANS
IN SAMPLES OF AMBIENT SOILS, SEDIMENT, AND SEWAGE SLUDGE FROM THE SHELTON AREA

Homologue or Isomer	C3 Performance Sample	C4 Field Blank	C5 Capitol Hill	C6 City Landfill	C7 New STP Sediment	C8 Old STP Sediment	C9 Control Area	C9 Control Area (duplicate)	C10 Hillcrest	C11 Shorecrest
Tetrachlorinated Furans (total) 2,3,7,8-TCDF	0.053	ND	0.59	16.7	ND	ND	ND	ND	ND	ND
	ND	ND	ND	1.1	ND	ND	ND	ND	ND	ND
Tetrachlorinated Dioxins (total) 2,3,7,8-TCDD	15.4	ND	0.28	59.8	ND	0.11	ND	ND	ND	ND
	8.0	ND	ND	0.17	ND	ND	ND	ND	ND	ND
Pentachlorinated Furans (total) 1,2,3,7,8-PDCF 2,3,4,7,8-PCDF	ND ND ND	ND ND ND	ND ND ND	9.8 1.4 1.4	ND ND ND	0.050 ND ND	ND ND	ND ND ND	ND ND	ND ND ND
Pentachlorinated Dioxins (total) 1,2,3,7,8-PCDD	5.2	0.80	0.34	155	0.046	0.80	ND	ND	nd	0.088
	0.42	0.23	ND	1.4	ND	ND	ND	ND	nd	ND
Hexachlorinated Furans (total) 1,2,3,4,7,8-Hexa-CDF 1,2,3,6,7,8-Hexa-CDF 1,2,3,7,8,9-Hexa-CDF 2,3,4,6,7,8-Hexa-CDF	1.1 ND ND ND ND	ND ND ND ND ND	0.16 ND ND ND ND	5.5 0.93 1.1 ND 1.3	0.097 ND ND ND ND	0.67 ND ND ND ND	ND ND ND ND ND	ND ND ND ND	ND ND ND ND	ND ND ND ND
Hexachlorinated Dioxins (total) 1,2,3,4,7,8-Hexa-CDD 1,2,3,6,7,8-Hexa-CDD 1,2,3,7,8,9-Hexa-CDD	181	5.6	0.30	152	0.65	0.98	0.086	ND	ND	0.12
	0.43	0.28	ND	2.7	ND	ND	ND	ND	ND	ND
	0.53	0.37	ND	4.8	ND	ND	ND	ND	ND	ND
	1.7	0.89	ND	2.8	ND	ND	ND	ND	ND	ND
Heptachlorinated Furans (total) 1,2,3,4,6,7,8-Hepta-CDF 1,2,3,4,7,8,9-Hepta-CDF	ND	ND	0.81	4.9	0.40	4.3	0.12	ND	ND	ND
	ND	ND	0.20	2.5	0.094	0.68	0.023	ND	ND	ND
	ND	ND	ND	0.49	ND	ND	ND	ND	ND	ND
Heptachlorinated Dioxins (total) 1,2,3,4,6,7,8-Hepta-CDD	315	14.4	1.9	50.2	0.62	4.3	0.33	0.23	ND	0.36
	142	6.3	1.0	26.6	0.25	2.2	0.14	0.086	ND	0.17
Octachlorinated Furans (total)	0.30	ND	0.80	0,97	ND	6.2	ND	ND	ND	ND
Octachlorinated Dioxins (total)	178	86.8	6.4	19.3	1.4	13.8	0.85	0.56	ND	1.1

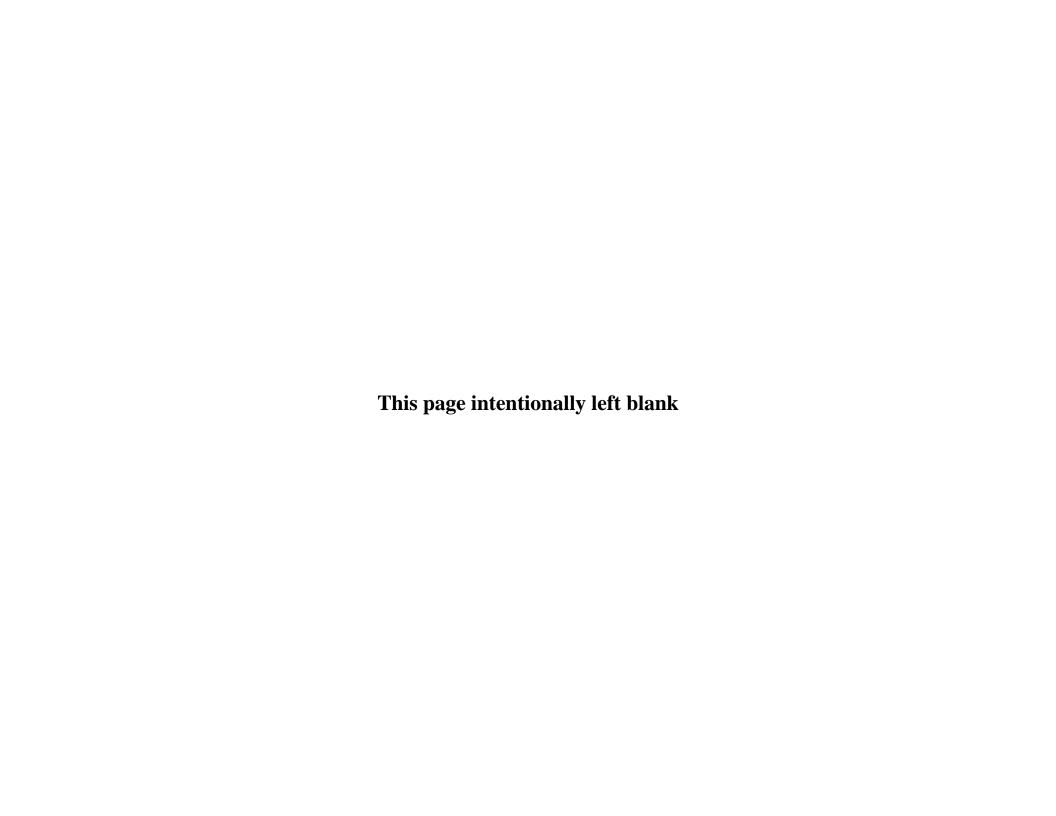
^aSee Figures 1 through 4 for location of sampling areas.

b_{EPA No. V58WQ02K1.}

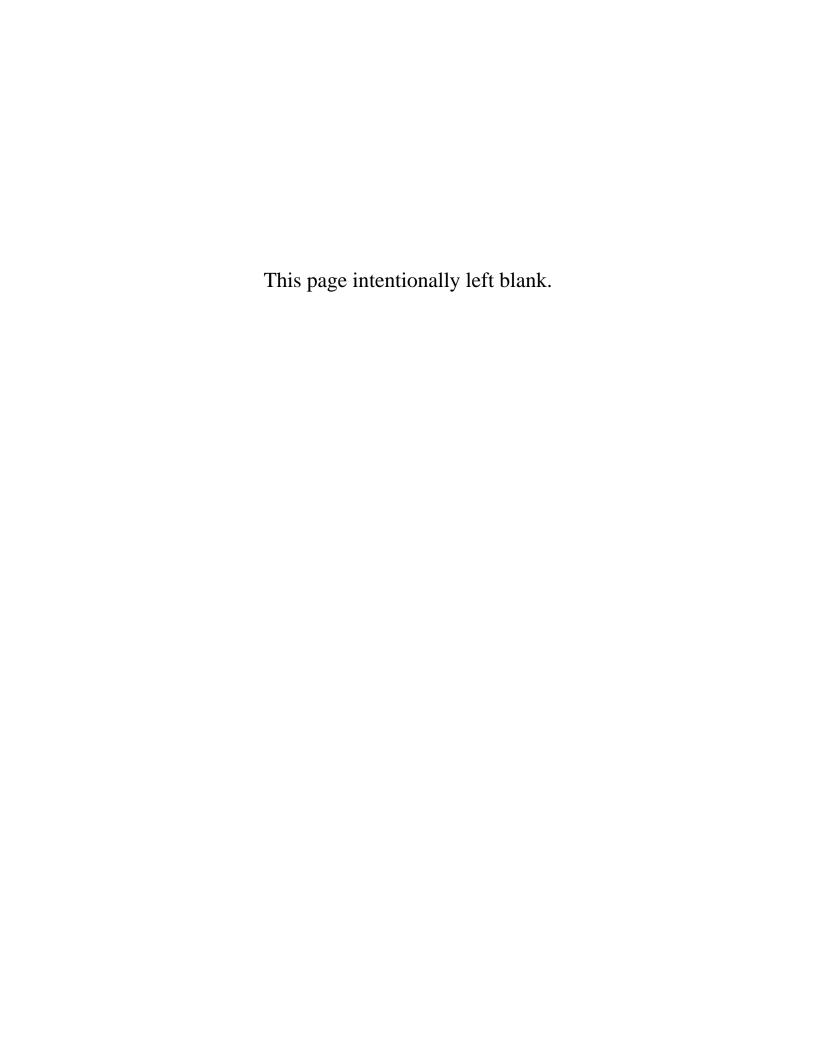
^CEPA No. E19AD99R4. se5685/050/1

d_{ND} = not detected.

epreliminary pending completion of EPA quality review.

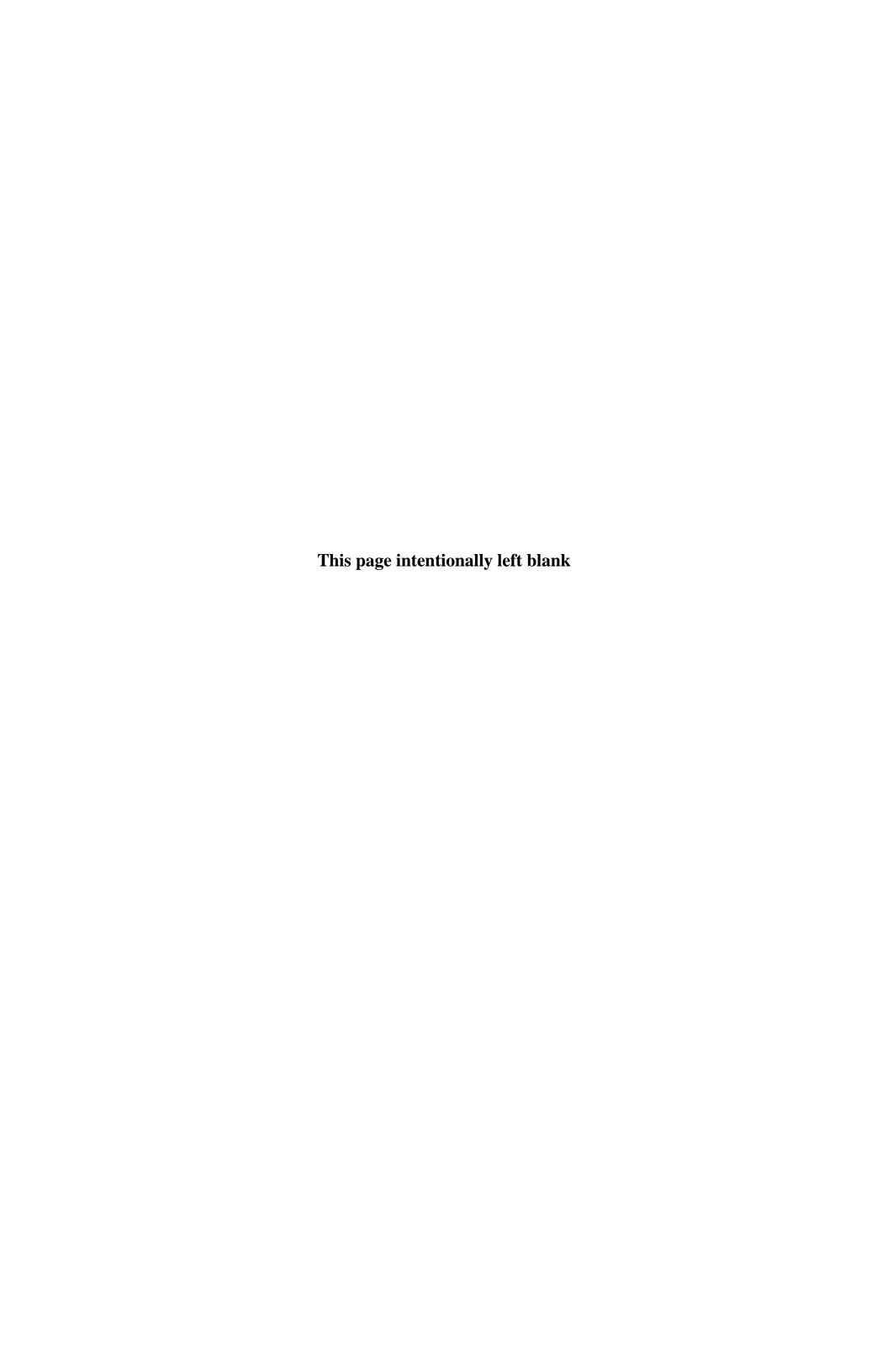


2004 Survey of Discharge Locations along Oakland Bay and Hammersley Inlet



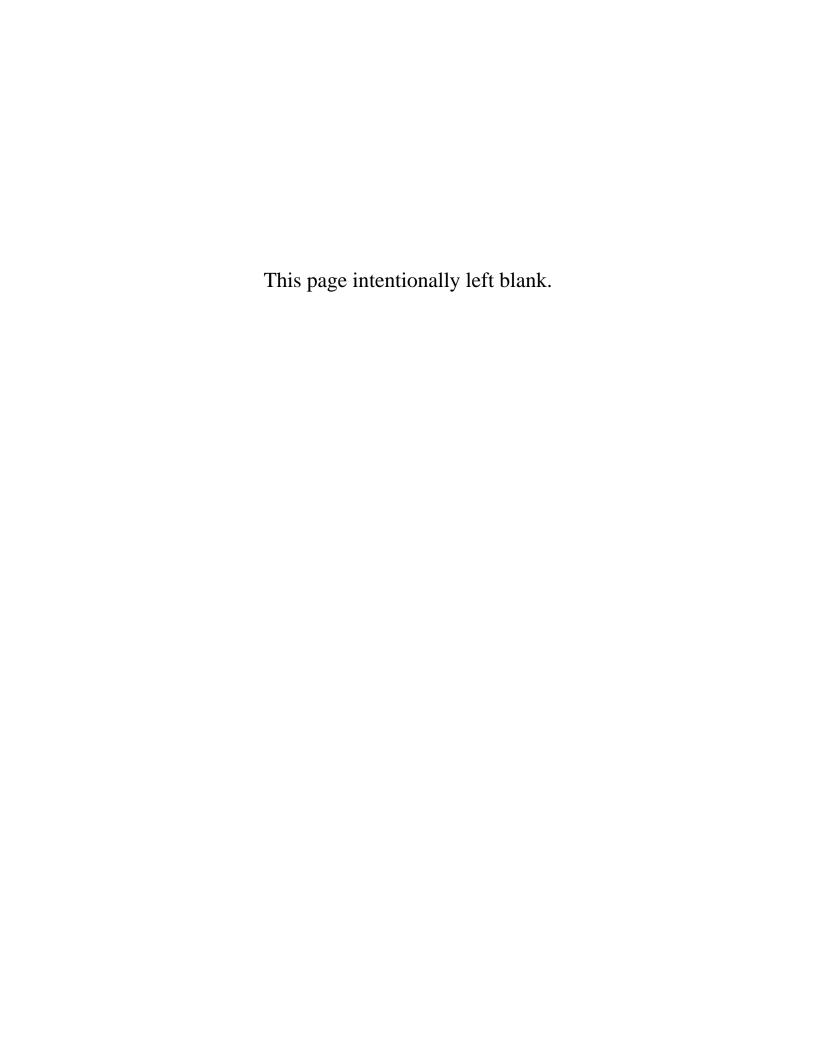


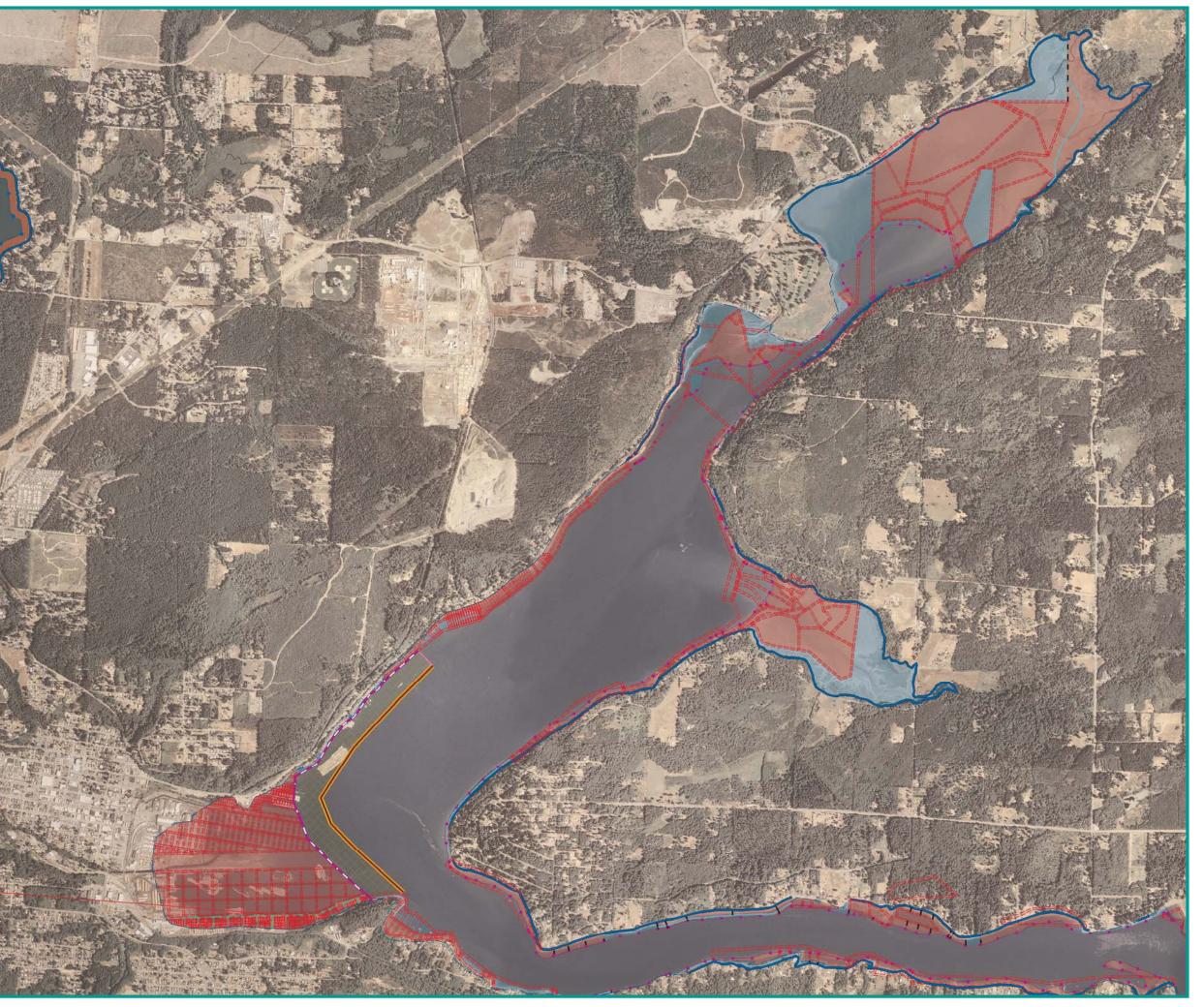
2004 Survey of discharge locations along Oakland Bay and Hammersley Inlet.



APPENDIX D

DNR Aquatic Boundaries and Parcels









Aquatic Boundaries

1/4 Mile From ELT

200 Mile EEZ Line

- 6 Mile OCSLA Line

Agreement Boundary

County Boundary

Degree of Latitude

Degree of Longitude

• Extreme Low Tide

---- Harbor Line Inner

Harbor Line Outer

Harbor Line Side

Indefinite Boundary

 Limit of Navigability - Mean High Tide

Mean Low Tide Mean Lower Low Tide Mean River Level Mean Sea Level ---- Mean Tide Level — - Meander Line

Ordinary High Water Ordinary Low Water --- Other PLS Line ==== Platted Line

Two Mile City Limit Line

---- Protracted Line

City Boundary

- - · Deed Line

Abandoned Shoreline

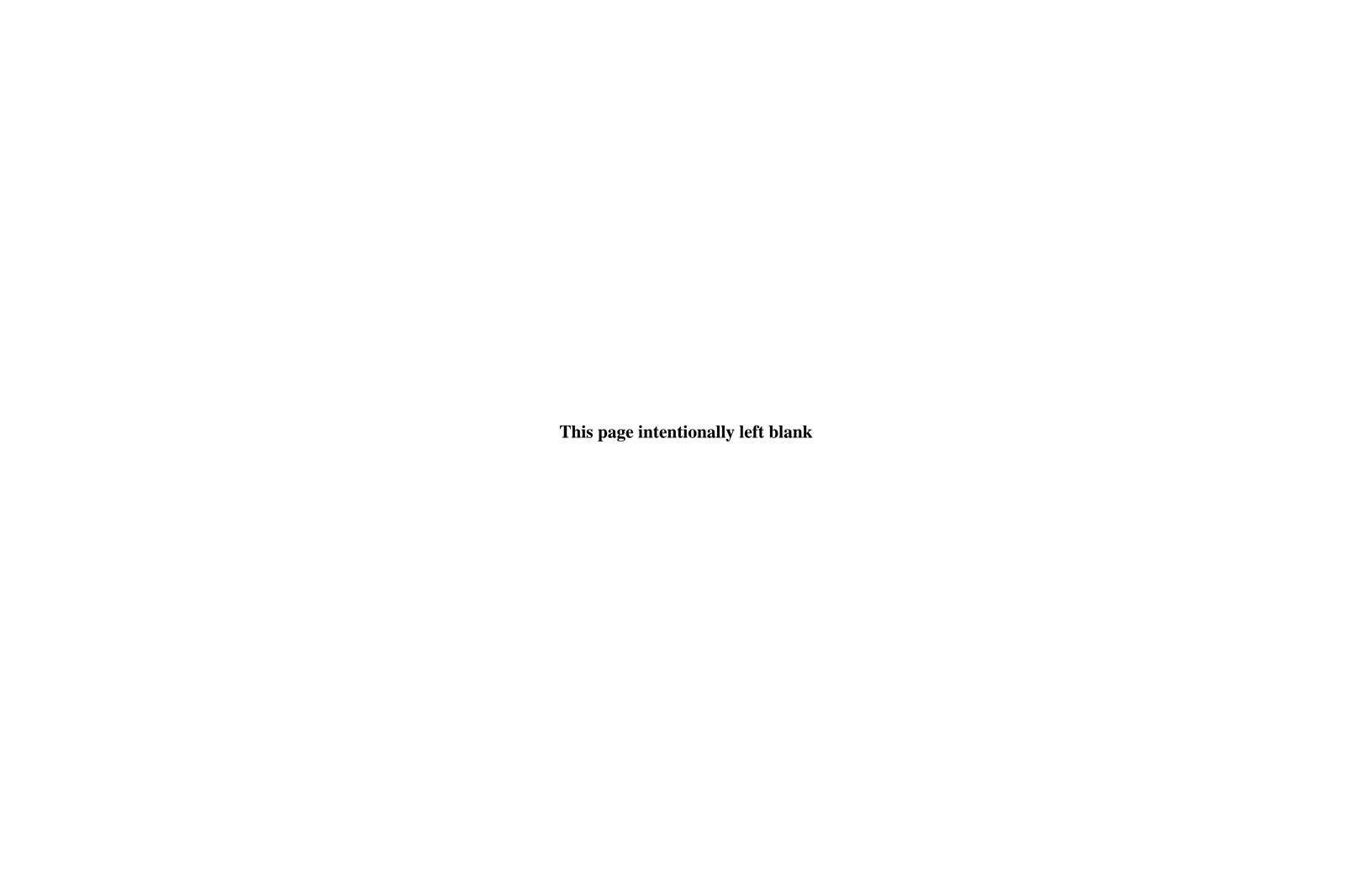
Feet 0 660 1320 2640

Washington State Plane South; NAD1983 HARN

Data Sources: Aquatic Boundaries and Parcels, DNR 2008; Orthophotos, NAIP (1 meter) 2006. 2008313tstr490

While every attempt has been made to use the best available data, this is a geographic representation of real world conditions with varying levels of uncertainty. As such it is not appropriate to use in place of an official survey or in legal proceedings





Chapman Cove Piling Lease Location Information

