INTER-OFFICE MEMORANDUM

To: F. J. Haydel and R. L. Alboucq - Mead

A. H. Scott - Mead

R. L. Humphrey - 677 OB

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R. O. Gunderson - 664 OB

J. B. Todd - 676 OB

Environmental Control - KC 762

Date:

May 2, 1978

From:

W. B. Eastman

At:

Mead

Subject:

See Attached

Report

Attached please find completed report on the Phase I investigation of, "Effect of Waste Disposal Practices on Ground-Water Quality at Kaiser Aluminum & Chemical Corporation, Mead, Washington", April 14, 1978.

Phase II of this investigation has been initiated.

If you have any questions regarding this report, please feel free to contact me.

WBE:eb

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ON GROUND-WATER QUALITY

At

KAISER ALUMINUM & CHEMICAL CORPORATION Mead, Washington

December, 1977

SUMMARY

This report addresses the solid waste disposal practices of Kaiser Aluminum & Chemical Company, Mead, in relation to the possible effect on the aquifer underlying the plant site.

Two principal questions are involved:

- 1. Are pollutants entering the aquifer? ye but not quelified
- 2. If so, are the concentrations sufficient to pose as potential health hazard?

This study shows evidence that several potentially hazardous constituents, present in the disposal sites, have migrated into the soil column and are present in the aquifer. The true concentrations in the soil columns are not evaluated by the methods employed.

One water sample from one well (Production Well 3) showed cyanide concentration of 0.06 mg/l. Numerous subsequent samples showed cyanide concentration of 0.01 or less.

** Micro-gram My

All other water quality data from the plant production wells meets all U.S.P.H.S. and E.P.A. criteria. Water from the uppermost part of the same aquifer, as penetrated by two test wells, show fluoride above the recommended limits. The sulfate ion concentration approaches these limits. Compared to earlier analyses, sulfate and nitrate concentrations have definitely increased, however, this increase may be from an overall increase throughout the aquifer. Concentrations of other constituents may have also increased, but no comparative data is available.

The variation in concentration of all of these pollutants as seen among Test Hole 1 and 2, the plant production wells and the south plant wells may be due to:

- a) variation in solubility, dilution and dispersion
- b) anisotropy in the aquifer
- c) the geographic position of these points in relation to the ground water gradient.

It is certainly possible, and perhaps likely that concentrations greater than those measured are present elsewhere in the aquifer. The possibility of migrations of these pollutants into the water supplies of other users such as the Town of Mead, the State Fish Hatchery, the Golf Course and other private and public water supplies is real but undefined. It is also likely that without modification of disposal practices, concentrations of these pollutants will continue to increase with time.

This report recommends methods to reduce the mobilization of pollutants, and further investigation to define the distribution of concentration of these pollutants.

INTRODUCTION

Robinson & Noble, Inc. was retained by Kaiser Aluminum to investigate the effect of solid waste disposal practices on the aquifer at their facility near Mead, Washington. In particular, there are two sites judged to be potential sources of contaminants.

The largest site is the wet scrubber sludge pit, an area approximately 10 acres in size at the northeast corner of the plant. This area was used for disposal by-products from the wet scrubbers from 1952 to 1975. It is no longer used for this purpose but is used to receive effluent from the sewage treatment plant. At the east end of the impoundment, a 10-foot thickness of the spoil materials can be observed. The by-products present in this disposal area are reported to be calcium sulfate and calcium fluoride. Fluoride and sulfate are both potential sources of contamination to ground water. Increased concentrations of calcium caused by leaching from the disposal area would not be considered particularly hazardous.

The smaller of the two sites is the pot-lining disposal area. This area continues to be used for disposal of pot brick and other scrap material. Chemical constituents in the material disposed in this area includes C, ${\rm Al}_2{\rm O}_3$, NaOH, ${\rm Na}_2{\rm CO}_3$, Fe, ${\rm Al}_2{\rm N}_2$, F and CN.

Any leachate containing cyanide or fluoride that entered the ground water system would be considered hazardous. Other leachate containing concentrations of Fe, Al, Na or other elements within this disposal area would be more of a nuisance than actually harmful to health.

GEOLOGIC-HYDROLOGIC SETTING

The disposal sites are both located immediately north of the Kaiser facility, which is situated on a broad alluvial plateau, drained by the Little Spokane River. The hydrologic gradient across the site is northwest of the Little Spokane River. There is no specific information on local definition of the gradient in the area of the plant.

Climate of this area is semi-arid, with 15 to 20 inches of precipitation per year. Below freezing temperatures prevail from December through February, which is also the period of maximum precipitation. Effective precipitation (precipitation minus evapotranspiration) is available for infiltration from October through March.

The geologic section at the plant site is represented by a succession of fluvial and fluvial-lacustrian sediments to a depth of approximately 500 feet. The upper 50 feet of these sediments are exposed to the south of the plant in the Cunningham Gravel Pit. Here, large scale features of cross-bedding and channelization can be observed. The uppermost 200 feet of the sediments are poorly to moderately permeable and highly anisotropic. The sediments below 200 feet consist of impermeable clay horizons, and highly permeable sands and gravels. These lower sands and gravels constitute the major aquifer penetrated by the plant production wells. The entire sedimentary sequence overlies granitic basement, which may be locally fractured. Figures 1, 2 and 3 show the generalized subsurface geology.

DESCRIPTIVE LOG

			COMPLETE IN A MANAGEMENT OF THE PROPERTY OF TH
[:i	-3.2.		GROUND SURFACE SILT, SAND AND GRAVEL GRAY WITH ORGANIC MATTER
		9FT.	SAND AND GRAVEL, SILTY TO 3 FEET - BROWN 3 FEET TO 9 FEET
	0000		COARSE SAND AND GRAVEL
		32FT.	SAND, SILT AND FINE GRAVEL LIGHT BROWN (VERY SILTY AT 22.5 FEET TO 23 FEET
		33FT.	COBBLES AND SILT
		4IFT.	SAND, GRAVEL AND SILT
		48FT	FINE TO MEDIUM SAND AND SILT
			MEDIUM TO COARSE SAND AND SILT - CEMENTED CLAY (PLASTIC) AND SAND BROWN
			FINE TO MEDIUM SAND AND SILT
	(## 3		SILT, CLAY AND SAND, YELLOW BROWN DARKER WITH DEPTH
	ā ing sā		FINE TO COARSE SAND AND SILT
	0000000	70F T.	
		BAET	
0	0000000	78FT. 80FT.	
		OOF I.	
	: : : : : : :		
ļ.,	: . : :		
\[\frac{1}{2}\]		OFT	FINE TO MEDIUM SAND, SILT TAN (CLEANER AND DIRTIER STREAKS AS SHOWN)
-			
[II5FT.	FINE TO COARSE SAND AND SILT, LIGHT BROWN
		117F 1. 119F T.	FINE TO MEDIUM SAND AND SILT, BROWN CEMENTED CLAY AND SILT, BROWN PLASTIC
		123FT.	MEDIUM SAND AND SILT, CLEANER AT 123 FEET FINE TO COARSE SAND AND SILT, CEMENTED
		128FT.	SILT, FINE TO MEDIUM SAND, CEMENTED MEDIUM TO COARSE SAND - CLEAN
نسا	بنبنين	134FT.	FINE TO COARSE SAND AND SILT
STATIC WATER LEVEL 144.6 FT		140FT.	FINE TO COARSE SAND AND SOME FINE GRAVEL
		155FT.	MEDIUM TO COARSE SAND AND SILT, WITH FINE GRAVEL (CLEANER AND SILTY ZONES AS SHOWN)
<u> </u>	7:7:.7:	158.5F	T. FINE TO MEDIUM SAND, AND SILT FLUID
TC	OTAL DEPT	TH 158.	5 FEET

KAISER MEAD PROJECT NO. 177-4
TEST HOLE 1

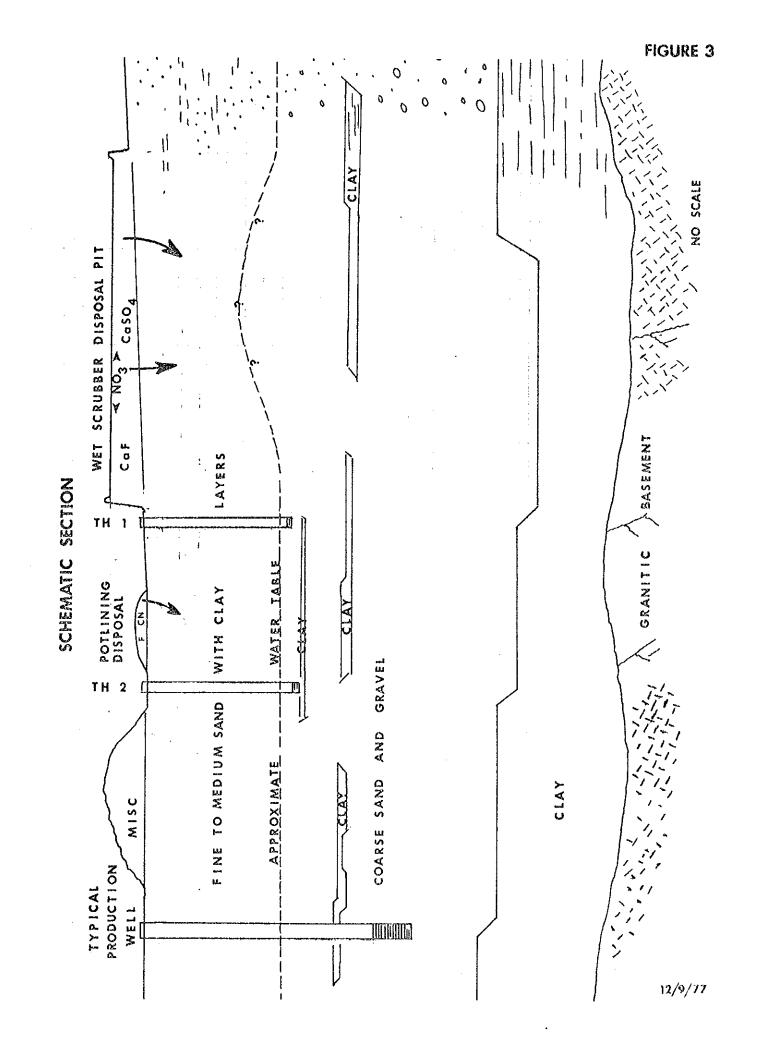
Level

Total Depth

163 ft.

TEST HOLE #2 Ground Surface fine gravel 10 Coarse sand and fine gravel - black-brown 17.5 Silt, medium to coarse sand - tan Medium to coarse sand - black-tan 26 Medium sand with streaks of clay 28 Medium to coarse sand, and granules - clean Silt, fine sand, and small pebbles රාංග ව ≥ 35 (cobbles at 35 feet) Fine to medium sand and silt - gray-tan 46 Fine to medium sand, gray-tan - clean Silt, sand cemented, light brown 59 Silt, sand, clay cemented 63 Medium to coarse sand, silt and fine gravel 67 Silt, fine to coarse sand & fine gravel - gray **خ**ن∹∃70 Silt and gravel (pebbles) tan Fine to coarse sand, fine gravel silt increasing toward 86 feet 186 90 Silt, fine to coarse sand and gravel - gray |94.5|Medium to coarse sand and pebbles - gray 102 Silt, fine to medium sand, cemented - tan 106 Medium to coarse sand - gray 109.5 Silt and sand, fine to medium - tan-gray 7111.5 Fine to medium sand, silt & clay, tan-gray, cemented 119 122 Fine to medium sand and silt - tan-gray Silt, fine to medium sand Silt, fine to medium sand w/clay & pebbles, cemented 130 133 133 Medium sand and silt cemented - light brown Static Water Silt, fine to medium sand and fine gravel 139.7 ft. Medium sand and silt - gray-brown Silt, medium to coarse sand - dark gray Medium to coarse sand, green, clean, water-bearing Medium to coarse sand and silt, water-bearing fluid $\begin{array}{c} 141 \\ 143 \end{array}$ 146 (some cemented layers) 155 157 Clay, silt gray-blue 160 Clay, silt, with sand 163 ft. Clay, gray - plastic Clay, silt gray-blue - plastic Clay, silt, with sand

> Kaiser Mead Project No. 177-4 HOBINBON & NOBLE, INCORPORATED GROUND WATER GEOLOGISTS



The regional water table, below which all materials are saturated, is about 140 feet below land surface. As noted later, there may be very localized perched water tables at shallower depths.

Under natural conditions, excess precipitation percolates through the unsaturated sedimentary materials until it reaches the water table. In passing through the unsaturated soils, the water will tend to dissolve all available chemical constituents be they "natural" or "artificial" and transport them to the water table. The route of the percolating waters is complicated by the anisotropy of the soils. Once the water reaches the water table it moves down-gradient as laminar flow and mixes with the underflow of the aquifer system. This flow, too, may be highly complex as determined by aquifer anisotropy.

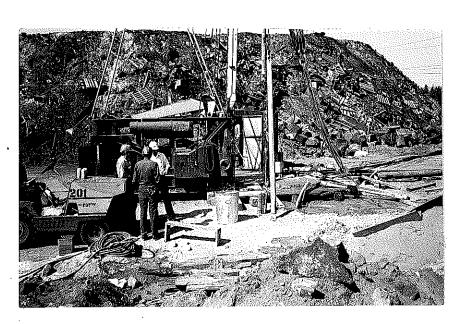
The quantity of excess precipitation determines the natural recharge to this type of aquifer. The sewage effluent placed on the abandoned wet-scrubber sludge pit constitutes a definite source of artificial recharge which will form a "mound" on the water table. Natural recharge may occur for only 7 months per year and probably total about 9 inches per year. The artificial recharge is continuous and may constitute as much as 250 inches per year as averaged over the 10-acre site. Thus, the artificial recharge may be 20 to 30 times greater than the natural recharge.

TEST DRILLING PROCEDURES

To examine the possibility that chemical constituents of the waste disposal area might be entering the local aquifer, it was decided that two test holes should be drilled. Test Hole 1 was located near the northwest corner of the wet scrubber sludge pit. Test Hole 2 was located on the northwest side of the pot lining disposal area, between the pot lining and the brick disposal sites.

Holman Drilling Company of Spokane was selected to perform the drilling. Cable tool drilling equipment was used on both test holes. The photos show the rig at TH-2.





Construction was begun at each site by placing a $10\text{-foot}\ x$ $10\text{-foot}\ crib$ to a depth of 10 feet below ground surface. Eightinch casing was then advanced to a depth of 30 feet. Drilling was completed with 6-inch casing in both test holes.

To insure that accurate samples were obtained, a number of unusual drilling techniques were employed. To protect the samples from possible contamination the contractor was not allowed to use any water in the drilling operation. Drilling tools and materials were also cleaned and protected to prevent surficial contamination. Continuous samples were taken with a 5-inch drive-core sampler throughout the full depth of drilling both test holes. Casing was driven with drill stem and jars as the hole was advanced by sample removal. This sampling method has also been used by other researchers on a ground water pollution project in the Spokane Valley.

Sample quality was excellent and recovery was near 100%. Sieve analysis of selected samples are included in Appendix I. The contractor is to be commended for a very professional performance, under unusual drilling circumstances. Drilling was begun on July 20, 1977 and completed on August 23, 1977.

On completion of each test hole, a 5-foot length of Johnson, 6-inch telescope 304ss 10 slot (0.010 inches) well screen was installed. The screens were fitted with a neoprene packer top and closed bottom. The exposed interval in both TH-1 and -2 was from 153 feet to 157 feet below ground surface.

After the screens were exposed, one hundred bailers of water were removed. Each bailer contained approximately $3\frac{1}{2}$ gallons,

thus, about 350 gallons was removed from each test hole. Chemical analyses were made of the 1st, 10th and 50th bailers in TH-1 and from the 1st, 10th, 50th and 100th bailer in TH-2. All water samples were collected in plastic containers.

Both test holes were completed by filling the space between the cribbing and the 8-inch casing, and grouting the annulus between the 6-inch and 8-inch casings as the 8-inch casing was withdrawn. A positive surface seal results, and the 6-inch cased test hole remains.

TESTING AND ANALYSIS PROCEDURES

As each core sample was brought from the hole, it was removed from the core barrel and representative portions were sealed in plastic bags. The samples were then taken immediately to the Kaiser Mead Laboratory for analysis. The formation samples were analyzed by selecting 10 grams of sample and mixing with 500 ml of distilled water. This mixture was brought to the threshold of boiling, to dissolve the soluble constituents and was then tested by various methods shown in Appendix II.

Although this method is essentially arbitrary, it is well suited to the purpose of identifying soluble chemical constituents which are present in the soil columns. It is recognized that the measured concentrations probably are not the true ionic concentrations, but rather an indication of their presence.

Analysis of the core samples included:

- A. Sieve analysis of selected samples (TH-1).
- B. Moisture content of samples at approximately 3-foot intervals (TH-1 and -2).
- C. Chemical analysis of samples at approximately3-foot intervals (TH-1 and -2).

For:

- 1. pH
- 2. Sodium

- 3. Calcium
- 4. Fluoride
- 5. Sulfate
- 6. Nitrate
- 7. Cyanide *

Two sand samples collected from the Cunningham Sand and Gravel Company pit south of the pit were also analyzed.

Water samples bailed from TH-1 and -2, and pumped from Plant Production Wells 1, 2, 3, 5 and 6, and the two south plant wells were analyzed chemically for Items 1 through 7 as shown above. Water samples were submitted to Kaiser Center for Technology analysis, and the Spokane County Engineers' office for comparison of results. A comparison of these results is shown in Appendix III.

Complete data of the Mead analyses are included in Appendix II (envelope). K.C.F.T. data for cyanide analyses follows cyanide section (Page 19).

A sample of the 10th bailer from TH-1 was submitted to Bennett Chemical Laboratories for analysis. Their initial analyses were discovered to be in error and were subsequently checked and corrected. These results and methodology are included in Appendix III. Bennett's evaluation of cyanide is judged to be artificially high because of the methods employed.

^{*} Because the concentrations of cyanide were below this limit of detectability of the method, the Mead cyanide data is in error. The K.C.F.T. cyanide data is considered reliable and is used as the basis for all recommendations and conclusions regarding cyanide in this report. The only inconsistency in the K.C.F.T. data in the analysis of TH-1, 10th bailer of .011mg/ml. This analysis was reported and confirmed by K.C.F.T. The source of the inconsistency may have been extraneous, and is therefore not considered significant.

COMPLICATING FACTORS

Several other factors should be recognized before attempting an evaluation of the chemical analyses.

- 1. The drilling and physical character of the samples in TH-1 and -2 reveal a number of relatively impermeable zones in the section. These silty and clayey layers significantly retard the downward infiltration of water as shown by the relatively high moisture content at these horizons. Mineralogy of the sediment tests is primarily granitic in character. However, in some horizons basaltic rock types predominate. In Test Hole 1, caliche was observed in the samples from a depth of approximately 20 feet to 40 feet. Caliche is a deposit of calcium carbonate deposited as a coating on particles in the capillary zone in arid and semi-arid regions.
- 2. Discharge from the domestic sewage treatment plant, discharges over the wet scrubber disposal pit after aeration. This discharge is approximately 185,000 gpd and therefore represents a potential infiltration of over 67 million gallons per year. The effect of such infiltration is of significance in analysis of nitrate. The writer observed the full discharge of effluent to be infiltrating into the sludge pit.
- 3. The temperature of the water bailed from TH-1 was 55°F and 67°F from TH-2. Ground-water temperature is normally about the same as the mean annual temperature. Water

Supply Bulletin 27 reports the mean annual temperature at 50°F, and the average ground-water temperature at 54°F, or 4°F above the mean annual temperature. The temperature in TH-1 is therefore near normal, but the temperature in TH-2 is obviously anamolous. The 67°F was verified with several thermometers and in repeated bailers pulled from the hole. The higher temperature may result from thermal influence of the baking pits.

EVALUATION OF RESULTS

The following discussion considers the results of each laboratory determination. Analytical determination of soil samples and bailed water from Test Holes 1 and 2 are considered, as well as soil samples from Cunningham Pit, and water samples from Production Wells 1, 2, 3, 5 and 6, and South Plant well — east and west.

Moisture Content

Moisture content in TH-1 and -2 is relatively low (less than 5%) throughout most of the section, except for a number of higher zones which correlate with the more impermeable horizons within the section. These impermeable zones retard the migration of infiltrating water and thus create wetter or perched aquifers. TH-1 shows a number of relatively wet zones, whereas TH-2 reveals only two such zones at 52 and 53 feet and 125 to 130 feet. Since the lithologic logs are very similar, the presence of a greater number of moist zones in TH-1 may be a result of greater infiltration from the sewage effluent infiltration over the wet scrubber disposal area near TH-1.

The moisture content below the water tables is 14% to 22% in both test holes, and thus parallels the void ratio in that part of the aquifer. The moisture content of the samples from the Cunningham Pit showed only 0 and 1.8%. This low moisture content is normal for samples from an exposed face during warm weather.

In the soil-sample analyses there is a suggestion of an inverse relationship between moisture content and the concentration of the various tested elements. Apparently, these concentrations are reduced and mobilized in the high moisture to the next lower stratum.

Hydrogen-Ion Concentration (pH)

Analysis of pH for the soil samples fall generally in the range of 7.6 to 7.8. There is some suggestion that the less impermeable zones have a slightly higher pH than the more permeable zones. The lowest soil pH (6.80 and 6.98) was observed in the near-surface samples in TH-2, and is probably a result of extraneous low pH material deposited at that site.

The pH of all water samples analyzed is in the range of 7.6 to 8.12.

Sodium

The high sodium content apparent in the upper 45 feet of TH-2 results from the proximity of available sodium. In TH-2, sodium content decreases rather gradually with depth.

In Test Hole 1, sodium values in soil samples are much more irregular. Analyses show the lowest values below the capillary zone (140 feet) and from 13 feet to 28 feet, and 49 feet to 61 feet. Sodium values below the capillary zone in TH-1 are twice as high as in TH-2. Thus, more sodium has reached the water table near TH-1 than near TH-2, even though the higher concentration source of sodium is closer to TH-2. Greater irregularity

of sodium concentration in section TH-1 also indicates more sodium in transit at that location.

Sodium should be highly mobile where solute is available, unless base-exchanged or clay-absorbed. Thus, sewage effluent again is suspect as the transporter for sodium.

Values for sodium in the samples collected at Cunningham Pit are below surface values at the test sites -- indicating decreased surficial contamination at the pit.

Concentration of sodium in the water is higher in TH-1 than in TH-2. Concentration of sodium in the Production Wells 1, 2, 3, 5 and 6 is less than 1/5th of the concentration for water from TH-2. Concentration of sodium in the South Plant Wells is even lower (approximately 4 mg/l in South Plant Wells vs. 5.7 mg/l average in Wells 1, 2, 3, 5 and 6.) This suggests that sodium infiltration either has not penetrated the main aquifer or is diluted by underflow in the lower and more permeable part of the aquifer.

Calcium

Analyses of calcium concentration in soil samples from TH-1 and -2 does not readily correlate with any other data such as moisture content or lithology. The calcium ion is a major cation (+) of these analyses and may result from the presence of caliche.

Caliche is normally present only in the capillary zone and was obvious in TH-1 to a depth of 40 feet, but it may be present throughout the section. Samples from the Cunningham Pit show calcium concentrations in the range of those from TH-1 and -2, which also related to the presence of caliche.

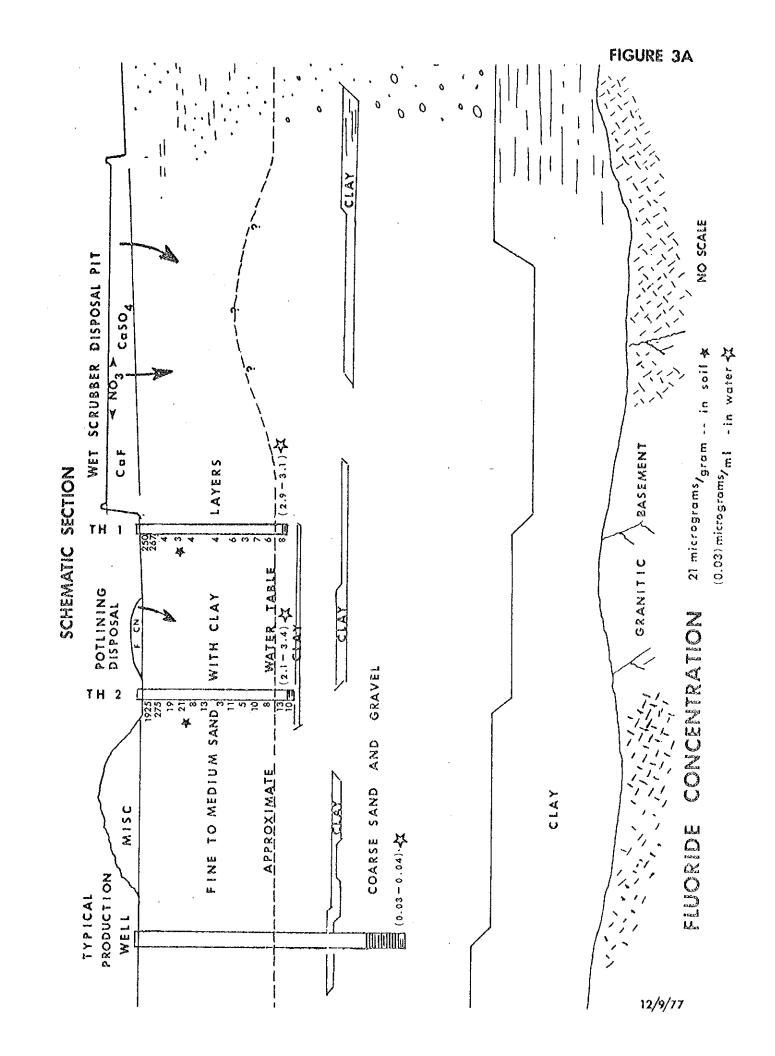
Concentration of calcium in the water bailed from TH-1 is approximately two times as high as the concentration in the Production Wells. Again indicating lack of communication or dilution in the deeper part of the aquifer.

Fluoride

Fluoride concentration in the sediment columns is high in both test holes, particularly near the surface. There is also a suggestion of higher concentration at the less permeable horizons in TH-1. Concentration in the column is greater at the top and bottom particularly in TH-1, indicating a rather rapid migration of high fluoride leachate.

Fluoride concentration in the bailed water from TH-1 and -2 exceeds the E.P.A. limit. Concentration in the production wells, however, is within the allowable limit and again illustrates dilution, lack of direct connection, or a migration pattern away from the production wells.

Washington State Bulletin No.27 reports fluoride in Production Well 6 (26/43-16F2) at .0 mg/l in 1959. Analyses of the present study indicate .03 microgram/ml (mg/l) Kaiser Mead, and .04 microgram/ml (mg/l) Kaiser Center For Technology. This difference from 1959 is within the range attributable to the more accurate analytical methods of this report, and does not necessarily reflect an increase in fluoride concentration. Available analyses from other wells in the aquifer show concentrations of up to 0.2 mg/l fluoride. Therefore, at the Mead site, fluoride migration either has not penetrated the productive part of the aquifer, or concentrations are being effectively reduced by dilution from underflow.



Sulfate

Concentration of sulfate in water bailed from TH-1 and -2 approaches the U.S.P.H.S. limit of 250 mg/l. Concentration of sulfate in water pumped from the plant production wells is less than 1/4th the concentration of water bailed from the test holes. However, concentration of sulfate in plant production wells has increased from earlier analyses.

Plant Production Well	Current SO ₄ Concentration			Earlier Concentration	Date	
1	19.3 mi	crogram	/ml = mg/l	8.8 mg/l	6/8/42	
2	40.8	11	= "	12.0 "	8/3/42	
3	18.0	H	# H	9.5 "	8/15/42	
6	22.2	l 1	= II	13.	10/22/59	

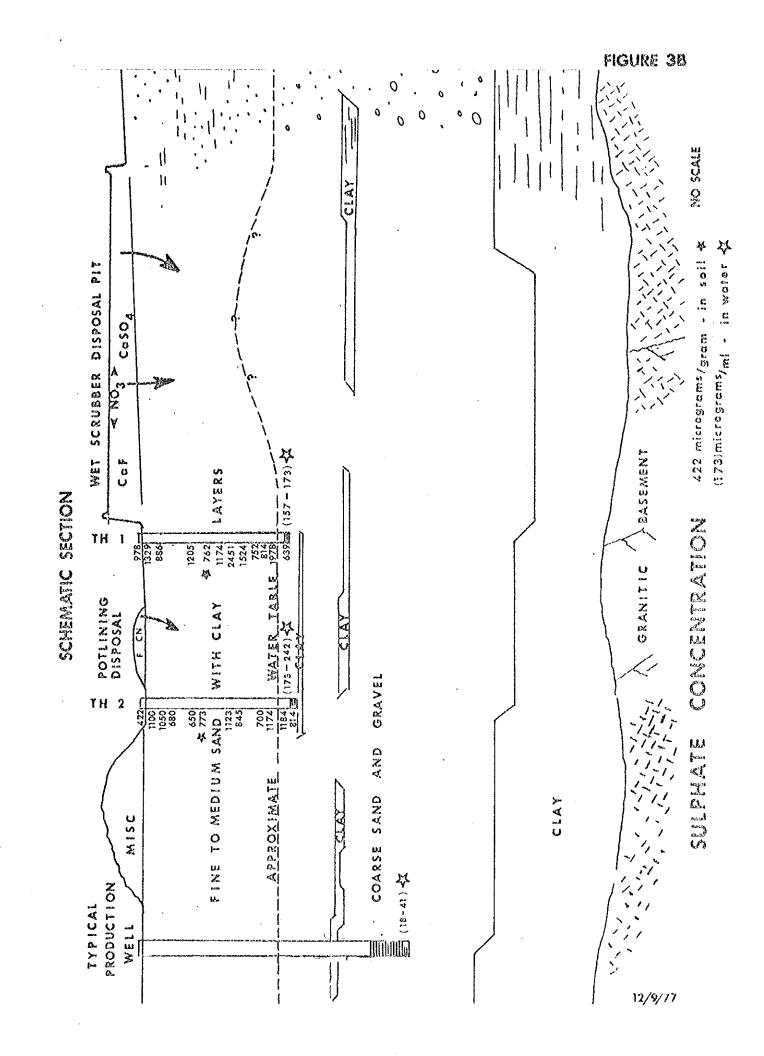
This data clearly indicates that sulfate is entering the aquifer.

Background data on water quality throughout the Spokane aquifer is very limited, but there appears to have been an overall increase in sulfate ion concentration.

Nitrate

Concentration of nitrate in both soil columns is relatively high and nearly uniformly distributed in TH-1. Concentrations are higher in TH-2, probably indicating less infiltration and greater shallow accumulation.

Concentration of nitrate in the bailed samples is two times that in the production well water. This variation may be the result of causes similar to those discussed under fluoride above.



Concentrations of nitrate in the plant production wells apparently has increased since 1959. Limited chemical analysis from other wells tapping the Spokane aquifer indicate widely fluctuating nitrate concentrations and an overall increase from earlier data.

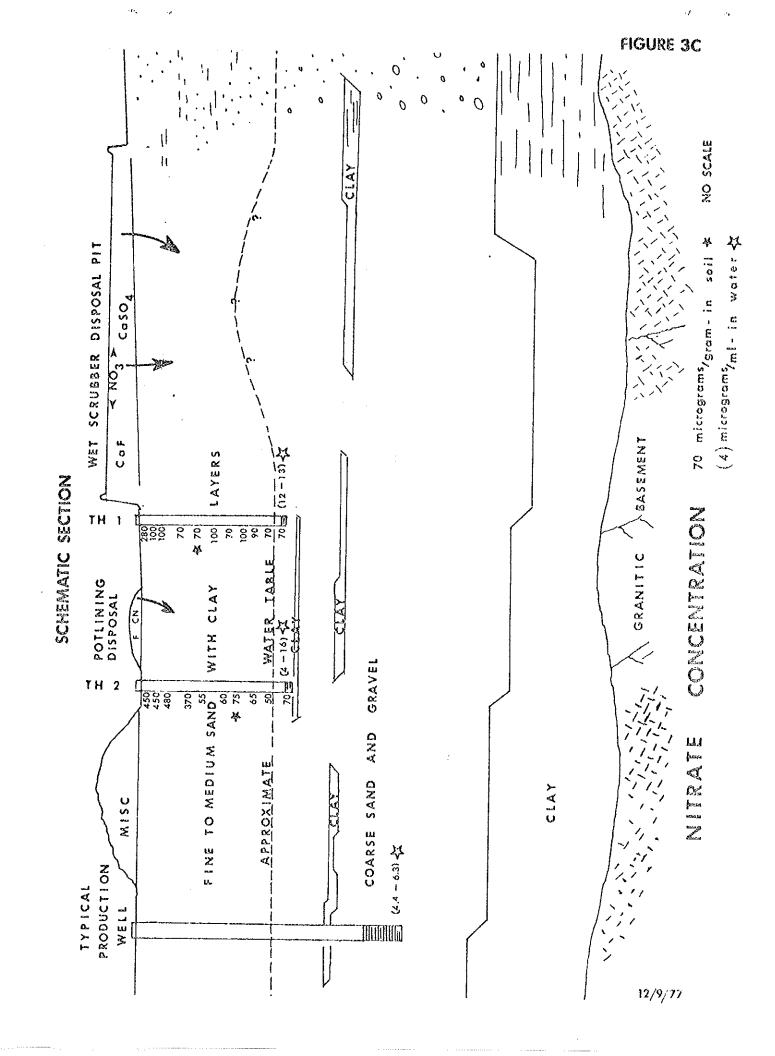
The presence of abnormal nitrate is generally taken as an indication of contamination. Spokane County is currently investigating the overall water quality in the aquifer. Considerable effort is being focused on nitrate, so that possible sources of contamination such as fertilizers and septic effluent in the East Valley can be identified and evaluated.

Cyanide

*

K.C.F.T. analysis show 0.06 mg/ml in water pumped from Production Well 3, and 0.007 mg/ml in water bailed from Test Hole 2. These concentrations indicate a possible migration of cyanide from the disposal area to the water table. *Additional analyses of water from Production Well 3 showed decreased concentrations of from 0.001 to 0.004. The decrease in cyanide concentration may be due to dilution from pumping, or due to the instability of cyanide at such low concentration.

^{*} Production Well 3 is located very near the disposal area. Such low cyanide concentration could be from direct migration of the pollutant down the exterior of the well casing.



INTER-OFFICE MEMORANDUM

To: F. J. Haydel and R. L. Alboucq - Mead

A. H. Scott - Mead

R. L. Humphrey - 677 OB

TIC - Library - CFT

R. O. Gunderson - 664 OB

J. B. Todd - 676 QB

Environmental Control - KC 762

Date:

June 13, 1978

From:

W. B. Eastman

410 1000 19

At:

Mead

Subject:

Corrections on Page 19 -Phase I Investigation of, "Effect of Waste Disposal

Practices on Ground-Water Quality at Kaiser Aluminum & Chemical

Corporation, Mead,

Washington"

On your copy of subject report, please change grams per gram concentration of soil sample to micrograms per gram, and grams per milliliter of water to micrograms per milliliter of water. Thanks. = PPM

Reported as:

Should be:

Analysis - CN g/g (soil), g/ml (water)

Analysis - CN μg/g (soil), μg/ml (water) Enclosed are the following 28 soil samples, and 16 water samples:

	•	•	•	
Soil Samples			, Analysis	CN
Test W	ell Mead Lab. No.	Depth	ug/g (soil), 4	g/ml (water)
	1517	9.5 - 10.0		T Analysis
1	1523	22.5 - 23.0	0.02	1 Millians
	1535	33.0 - 33.5	0.17	apang dagi danamating timber de staten a de et de et a de et et
	1549	49.5 - 50.0	0.04	gg magay gangan dankayan kababi Simbo dalkada Adad
	1557	68.0 - 69.0	0.04	
-1	1561	83.5 - 84.5	0.02	gygyngana ingantiyaga akki Tipinaanki Pili v Ti
- - -	1567	88.5 - 89.0	0.02	THE RESIDENCE OF THE PARTY OF T
1	1582	116.0 - 117.0	0.03	
$\frac{1}{1}$	1584	124.0 - 125.0	0.03	Appellunch der seinen der Gertrer Gertrer der der Seit Wir mit der
1	1606	136.0 - 137.0	0.01	
1	1607	140.0 - 141.0	0.04	
1	1619	155.0 - 156.5	0.02	
2	1694	4.0 - 4.5	0.69	
2	1703	40.0 - 41.0	0.01	
2 2 2 2 2 2 2	1706	52.0 - 53.0	0.04	
2	1709	64.0 - 65.0	0.18	Manhatan Camerina and Camerina
2	1716	88.0 - 89.0	0.04	
2	1719	100.5 - 101.5	0.05	
2	1743	112.5 - 113.0	0.11	
2	1746	125.0 - 126.0	0.01	
-2	1747	129.0 - 130.0	0.02	and the second s
2	1748	133.0 - 134.0	0.02	
2	1751	144.0 - 145.0	0.01	
2	1752	148.0 - 149.0	0.02	
ВТ	ank Soil Sample #1		Less Than	0.01
	ank Soil Sample #2		11 11	0.01
	iled Pot Lining Samp		4,470	
<u>Fa</u>	iled Pot Lining Samp		12,035	
ater Sample	<u>s</u>			
Pr	oduction Well #1		Less Than	0.001 277
Pro	oduction Well #2		()	0.001
		plicate)		0.064
		plicate)		0.060
	oduction Well #5		Less Than	0.001
Pr	oduction Well #6		11 11	0.001
So	uth Plant E. Product		1) (1	0.001
	uth Plant W. Product	ion Well		0.001
	-1 1st Bailer			0.010
	-1 50th Bailer		Less Than	0.001
	-2 1st Bailer		11 11 11	0.001
าน	-2 1st Railor from S	croon		0.009

0.009 0.007 0.007 0.007

0.001

Less Than

TH-2 1st Bailer from Screen
TH-2 10th Bailer
TH-2 50th Bailer
TH-2 100th Bailer

Distilled Water

CONCLUSIONS

- 1. The water table at the site is about 140 feet below ground surface. Ground water does not directly contact the material in the disposal areas.
- 2. Water from natural and unnatural sources has been sufficient to mobilize some objectionable chemicals into the "dry" sediments above the water table and into part of aquifer itself.
- 3. Specifically, the test results indicate that fluoride, sulfate, cyanide and nitrate have all entered the aquifer. However, none of the concentrations in the production wells are above the allowable maximum limits of the E.P.A. or U.S.P.H.S.
- 4. Comparison of earlier data from plant production wells, with data from this study, show that sulfate and nitrate concentrations have increased. However, similar increases have been observed in the aquifer in other parts of the valley.
- 5. Fluoride concentration in bailed water from TH-1 and -2 exceeds the E.P.A. limit of 2.4 mg/l. However, concentration in the deeper production wells averages less than 0.04 mg/l. Non-hazardous pollutants of sodium and calcium have also entered the aquifer.
- 6. The concentration of pollutants is generally greater in the upper part of the aquifer than in the lower horizon tapped by the plant production wells. The lower concentrations in

the production wells may be in part due to dilution, isolation by an impermeable zone of unknown continuity, and by the direction of ground water flow.

- 7. High ground temperatures, caused by heat from the baking pits, could possibly influence the solubility of some of the chemical constituents of the leachate and soil.
- 8. A plume of leachate around the wet scrubber disposal pit, caused by discharge of sewage effluent over the pit is likely to enhance the infiltration of pollutants.
- 9. Determination of the shape, size and direction of flow of such leachate plumes are beyond the scope of this study.

 However, we can recognize that the sediments in and overlying the aquifer are highly anisotropic, and therefore that leachate flow in these sediments would be similarly irregular.
- 10. While Test Hole sites 1 and 2 were chosen to be down-gradient from the two most obvious sources of contamination, it is probably optimistic to assume that the two test hole sites represent the highest concentration of pollutants in the area.
- 11. Further studies are required to identify the bulk quantity of pollutants, the dilution potential direction of leachate movement throughout the system.

RECOMMENDATIONS

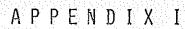
- I. Stop disposing of sewage effluent in the wet scrubber disposal pit.
- II. Evaluate the benefit of grading and covering the disposal areas with an impermeable material such as bentonite.
- III. Initiate further studies to further evaluate the impact of the contaminants on the aquifer. The goals of the study should be to:
 - A. Quantify the amount of leachate that may be entering the aquifer system.
 - B. Determine the degree of connection between the upper and lower aquifer horizons.
 - C. Determine the amount of underflow in the aquifer system.
 - D. Define and describe the plume of influence, and the migration pattern of contaminants, subject to limitations of test wells drilled.
 - E. Evaluate and compare ground-water quality of other wells in the aquifer.
 - F. Determine the relationship between the soil chemistry and the leachate and its effect on leachate mobility.

To complete the above described study a number of activities will be required. Among these are:

- Drill four to six additional test wells.
- Install a temporary water level recorder in Test Hole 2.
- W 3. Install a low capacity (1 gpm) pump in TH-1 and eventually in Test Hole 2.
 - 4. Coordinate additional drilling with geochemical
- Research of existing water quality information, where as well as collection and analysis of in the same in the aquifer -- with particular emphasis on down-gradient points.

Respectfully submitted,

ROBINSON & NOBLE, INC. Ground Water Geologists



The Tyler Standard Screen Scale

Cumulative Logarithmic Diagram of Screen Analysis on Sample of Test Well #1 Soil Name 17' - 18' Depth (Lab. #1623, Sample #27) WEIGHT 10 18.85 7 20 φ w 9 9 9 9 ဓ္က SCREEN SCALE RATIO 1.414 Duplicate Samples Average Openings Per Cent Per Cent Per Cent Sample U.S. Sample Tyler Sample Per Cent Cumulative Per Cent Comulative Per Cent Cumulative Milli-Weights Weights Weights Inches Mesh No. Weights Weights Weights meters 26.67 1,050 .742 18.85 13.33 .525 9.423 .371 6.680 .263 5.7 2,0 2.0 3.8 5.7 5.7 2.0 3.8 3.8 4,699 .185 3.327 .131 6 14.7 9.0 9.0 8.5 8.5 10.5 8,8 -8.8 12.6 2,362 .093 8 .065 12 1.651 10 29.6 29.51 42. 27.9 42.6 28.0 $\overline{31.1}$ 31.141.6 1.168 .046 16 .0328 20 20 .833 28.7 28.6 71.2 27.9 27.9 69.5 28.3 28.2 70.4 .589 .0232 28 30 .417 .0164 35 40 11.8 11.8 83.0 12.412.4 81.9 12.1 12.182,4 .295 .0116 50 48 .208 .0082 65 70 8.7 8.7 90.6 8.9 8,9 91.4 9.2 9.2 92.2 .147 .0058 100 100 .104 .0041 150 140 4.5 4.5 96.7 4.5 4.5 95.14.5 4.5 95.9 .074 .0029 200 200 4.2 5.0 4.2 100.1 3.4 100.1 5.0 3.4<u> 100.1</u> ,0029 200 200 .074 100.3 100.1 <u>100.0</u> 100.1 100.1 100.2 Totals, THE W. S. TYLER COMPANY, MENTOR, OHIO 44080, U.S.A

The Tyler Standard Screen Scale

Cumulative Logarithmic Diagram of Screen Analysis on Sample of Test Well #1 Soil Name 36' - 37' Depth (Lab, #1624, Sample #51) **80** 70 60 CHNI CUMULATIVE PER 20 10 ်**ဝဝ** ၁၀ မ္က OPENING SCREEN SCALE RATIO 1.414 Duplicate Samples Average Per Cent Per Cent Per Cent u.s. Sample Sample Tyler Cumulative Curnulative Per Cent Cumulativa Per Cent Per Cent Milli-Hesh Weights Weights Weights Inches Weights Weights Weights meters 26.67 1.050 18.85 .742 13.33 .525 9,423 .371 6.680 .263 1.5 1.5 0.7 1.5 0.74.699 .185 3.327 .131 2.4 3.9 2.4 2.4 2.4 3.1 2,4 3,5 2.362 .093 1,651 .065 10 12 9.8 9,8 13.7 9,2 12.7 8.6 8.6 11.7 9.2 1.168 ,046 14 16 .0328 20 2:0 .833 23.6 28.5 23.6 37.3 28.4 40.0 26.Q 26.0 .0232 38.7.589 28 30 .417 .0164 35 40 72.8 34.3 34.4 34.0 33,9 74.0 $\overline{34.2}$ 34.2 71.7 .295 .0116 60 .208 .0082 65 70 11.6 11.6 83.3 15.0 14.9 88.9 13.2 13.2 _86<u>.</u>1 .0058 100 100 .147 150 140 .0041 .104 7.3 7.0 95.9 93,4 <u>7.6</u> 7.6 90.9 7,0 .074 .0029 200 200 9.0 9.0 99.9 100.06.6.100.0.0029 200 <u>4,1</u> <u>4.1</u> 6.6 .074 99.8 99.9 100.3 100.0 100.0 100.0 W. S. TYLER COMPANY, MENTOR, OHIO 44060, U.S.A.

The Tyler Standard Screen Scale

Form to Please mention ac-

Test Well #1 Soll

THE W. S. TYLER COMPANY, MENTOR, OHIO 44060 U.S.A.

Cumulative Logarithmic Diagram of Screen Analysis on Sample of... Name 41.5' - 42.5' (Lab. #1625, Sample #57) 90 トゴのゴゴダ CENT CUMULATIVE PER 10 18.85% 20 2000 ည္စြဲစာ ျ ထ 900 8 OPENING Samples Duplicate Average SCREEN SCALE RATIO 1.414 Openings Per Cent Per Cent Per Cent U.S. Tyler Sample Sample Sample Cumulative Per Cent Cumulative Cumulative Milli-Weights Weights Weights Inches Mesh No. Weights Weights meters 26.67 1,050 18.85 .742 13.33 ,525 .371 9.423 6.680 ,263 4.699 .185 3.327 .131 6 0.4 0.3 0.4 0.3 0.3 0.4 0.42.362 .093 8 8 1.651 .065 10 12 2.8 3,2 3,0 3,0 3.3 2.9 2.9 3.2 ,046 2.8 1.168 16 .833 .0328 20 20 19.1 22.6 19.022.3 19.3 19.3 19_1 19.222,4 .689 .0232 28 30 .417 .0164 35 40 31.5 31.6 53.9 32.2 32.3 54.9 31.8 31.9 54.4 .295 .0116 50 48 .208 .0082 66 70 .70.0 70.9 16.016.1 70.4 .147 .0058 100 100 16.116.1 16.0 16.11.104 .0041 150 140 10,3 10,3 80.3 81.2 10.3 $10.\overline{3}$ 3) [8 <u>10.3</u> 10.3 .074 .0029 200 200 .074 .0029 200 200 100.0 18.799.919.219.2 100.C .19.718.7 99.9 99.8 100.0 Totals, 99.7 100.0

A cross check was made to validate the sodium and calcium standards used in the test well project. This was done by independently recalculating the amount of dried NaCl and dried CaCO3 to make the 1000 $\mu g/ml$ stock solutions, remaking the solutions, aliquoting as necessary to make the standard solutions and cross checking each of the new standard solutions with the old standards used for analysis in the test well projects. These results are shown below.

Na	(Units)	Na μg/ml	Ca	(Units)	Ca µg/m1
01d 500	New 500	1.0	<u>01d</u> 500	<u>New</u> 500	10
235	245	0.5	247	249	5
60	53	0.1	45	57	1

A further correlation on the Ca method is shown below.

microspanos

	Conc. Ca µg/m1	microgram for Mili
Tap H ₂ 0	22.8	10-6 8/1 = 3 p. L.
#1 Well	21.4	•
#2 Well	22.5	10-38/10h
#3 Well	22.7	= mg/l
#5 Well	23.5	
#6 West Well South Plant	24.5	
#7 East Well South Plant	22.5	

Test Well Kaiser 26/43 16 D2 when tested prior to 1952 showed 24 parts per million calcium (Ca) which agrees well with our data on the Kaiser-Mead well water. The investigation "Record of Wells, Water Levels, and Quality of Ground Water in the Spokane Valley, Spokane County, Washington", by Weigle and Mundorff, 1952, identifies 26/43 16 D2 as Kaiser Well #1 as numbered above.

Cyanide standards were run at various times during the course of the project. These standards were made by freshly preparing a stock solution with a calculated amount of KCN, diluting this to a standard solution and pipeting varying quantities of the standard solution to 25 ml volumetrics.

The results of these standard runs agree well with the original curve data as replicated several times over the past 3 years.

Nitrate is a more complex situation for validation than most of the methods employed in this project. The method employed was a selective ion electrode method which seems prone to some drifting problems. Nitrate can also be determined by an ultraviolet method but the method is prone to severe interference by organic material which is normally present in soil. Brucine indicator is on order and has not been received so we cannot yet employ the alternate Brucine spectrophotometric method. Duplicate samples were sent to Spokane County Laboratories and to our Research Center in Pleasanton, California, and returned results agree - 20% which may be as good as the available methodology.

I believe the analytical results on the project as a whole are accurate and represent the samples as received in the Chemical Laboratory.

INTER-OFFICE MEMORANDUM

W. B. Eastman TO

Mead AΤ

R. J. Schlager COPIES TO

R. L. Alboucq

DATE October 27, 1977

F. G. Doolittle J. Dorling PROM

Mead AΥ

subject Method Confirmation Checks and Standards Verification

REFERRING TO

The fluoride in well water and the fluoride in the soil leachate were run on the Technicon Autoanalyzer using the same method that we use for analyzing the low fluoride tape sample leachate. The 3 standards used for these analyses were the regular .25 μ g/ml, .50 μ g/ml and .75 μ g/ml which are used to set up the 0.00 μ g/ml to 1.00 μ g/ml curve. These standards were remade at least once during the period that the fluoride analyses were performed and these standards cross checked with each other very closely.

The sulfate (SO₄) method was confirmed by submitting 3 unknowns to the These unknowns were prepared by taking varying aliquots from a standard copper sulfate solution prepared by the chemist. The results are shown in the table below.

ml Cupric Sulfate Standard Solution	SO ₄ Present (gms)	SO4 Reported(gms)	Δ Grams	% Difference
5	.0679	.0690	.0011	1.6
10	.1358	.1523	.0165	12.1
20	.2716	.2784	.0068	2.5

We feel these results confirm the accuracy of the gravimetric method used to determine the sulfate in the soil samples and the well water.

APPENDIX II

The Tyler Standard Screen Scale

Form No 1.6 Please mention ab when ordering

5.7

104.9

THE W. S. TYLER COMPANY, MENTOR, OHIO 44060, U.S.A

99.9

99.9

5,4 100,0

99.9

Cumulative Logarithmic Diagram of Screen Analysis on Sample of Test Well #1 Soil Name 151.5' - 152! Interval (Lab. #1630, Sample #173) Date 90 RETAINED WEIGHT アビアゴ CUMULATIVE 10 886 ဇ္တ OPENING SCREEN SCALE RATIO 1.414 Duplicate Samples Average Openings Per Cent Per Cent Per Cent Sample U.S. Sample Sample Tyler Per Cent Cumulative Per Cent Cumulative Per Cent Milli-Mesh Ho. Weights Weights Weights Inches Weights Weights Weights meters 26.67 1.050 18.85 .742 13.33 ,525 9.423 .371 6.680 .263 4.699 .185 3.327 .131 6 0.1 0.1 0.10.10.1 0.12.362 .093 1.651 .065 10 12 1.2 0.9 0.9 0.91.1 1.11,168 .046 14 16 .0328 50 .833 20 24.3 25.2 26.0 24.7 25.8 28.6 25.126.3 23.3 .689 30 .0232 28 ,0164 35 40 .417 52.3 78.6 50.0 52.1 77.3 54.8 52, 78.0 59.6 ,295 .0116 48 60 .208 ,0082 65 70 12.2 12.4 90.0 11.7 89.5 11.8 89.8 13.0 11.4 .147 .0058 100 100 .0041 150 140 .104 94.8 5.0 94.6 5.5 4.8 4.5 4.7 94.2 .074 .0029 200 200

5.8

113.9

.074

.0029

200

200

Totals,

99.9

99.9

5,5

95.9

Please mention al-

The Tyler Standard Screen Scale
Cumulative Logarithmic Diagram of Screen Analysis on Sample of Test Well #1 Soll Interval (Lab. #1629, Sample #165) 80 CUMULATIVE PER CENT WEIGHT RETAINED 70 50 10 588583 7 ဗ္ဗ φ SCALE OPENING

scr	REEN SCALE	ERATIO 1.	414	Duplicate		Samples		Average				
Oper	ings	Tyler	U.S.	Sample		Per Cent	Sample		Per Cent	Sample		Per Cent
Milli- meters	Inches	Mesh	No.	Weights	Per Cent	Cumulative Weights	Weights	Per Cent	Cumulative Weights	Weights	Per Cent	Complative Weights
 26.67	1.050											1
 18.85	.742		Į									
 13.33	.525								ļ. <u>——</u>			<u>; </u>
 9.423	.371		Ì									1
6.680	.263	3		0.0	0 7	0 7	1 0	0.0	0.0		0.8	
4.699	.185	4	4	0.8	0.7	0.7	1.0	0.9	0.9	0.9	0.0	0.8
 3.327	,131	6	6	2.I	1 0	2.5	2.4	2.2	3.1	2.3	2.0	2.8
2.362	.093	8	8		1.8	2.3		2.2	7,1		2.0	
 1.651	.065	10	12					40.0	16.0		10.	,
1.168	.046	14	16	14.0	12.3	14.8	14.1	_12.9_	16.0	14.1	12.0	15.4
 .833	.0328	20	20	20.0	25.0	49.8	37.1	34.0	50.0	38.4	34.5	49.9
 .589	.0232	28	30	39.8	35.0	49.0	2/11	34.0	30,0	30.4	34.3	
 .417	.0164	35	40	25.9	22.8	72.6	24.3	22.3	72.3	25.1	22.5	$\frac{1}{72.4}$
 .295	,0116	48	50	25.7	1210	12.0	2.713	2213	, =		ر <u>د ۲۲ ـــــــــــــــــــــــــــــــــ</u>	72.4
 .208	.0082	65	70	10.0	8.8	81.4	10.2	9.4	81.7	10.1		81.5
.147	,0058	100	100	10.0			2012			2017		
.104	,0041 ,0029	150 200	200	10.7	0 /	90.8	10.6	9.7	91.4	10.6	9.6	101.1
 .074	,0029	200	200	10.7	-9.4	99.9.	0 3	8.5	99.9	9.8	8.8	. ,
 .074	,0028	200		10.4			9.0				<u> </u>	
 			Totals,	1.13.7	99.9	<u> </u>	109.0	99.9	<u> </u>	111.3	99.9	i

The Tyler Standard Screen Scale

Cumulative Logarithmic Diagram of Screen Analysis on Sample of Test Well #1 Soil Name 112.5' - 113.5' Interval (Lab. #1628, Sample #134) ate. CENT CUMULATIVE 20 10 ø 88888 5 ဓ္က 60.00 9.4237 OPENING SCREEN SCALE RATIO 1.414 Duplicate Samples Average Openings Per Cent Per Cent Per Cent Tyler U.S. Sample Sample Sample Per Cent Per Cent Milli-Cumulative Cumulative Per Cest Cumulatia: Weights Weights Mesh No. Weights Inches Weights Weights Weights meters 26.67 1.050 18.85 .742 13.33 .525 9,423 .371 6.680 .263 4.699 .185 4 3.327 .131 6 6 0.50.4 0.4 $\overline{0}$, $\overline{3}$ 0.2 0.2 2.362 .093 8 8 1.651 .065 12 5.3 4.7 4.7 5.5 4.5 4,9 5.4 4.6 1.168 .046 14 16 .833 .0328 20 20 29.2 33,9 30.1 33.2 36.7 35.0 29.7 34.934,0 .689 .0232 28 30 .0164 40 .417 35 32.5 67.1

<u> 36.9</u>

14.1

10.4

13.7

113.6

<u> 12.4</u>

9.2

12,1

100.1

.295

.208

.147

.104

.074

.074

.0116

.0082

.0058

.0041

.0029

.0029

48

65

100

150

200

200

60

70

100

140

200

200

Totals,

66.4

78.8

88.0

1,00,1

39.2

15.2

12.0

13.0

122.1

32.1

12.4

9.8

10..6

99.9

<u>79,5</u>

89,3

99.9

38.6

14.6

1.3-4

117.8j

32.3 66.7

<u>12.4' 79.2</u>

11.3 100 0

9 5

100.0

The Tyler Standard Screen Scale

Form No. 1-6 Please mention atowhen ordering

Cumulative Logarithmic Diagram of Screen Analysis on Sample of Test Well #1 Sot1 - 98' Interval (Lab. #1627, Sample #118)Date 100% CUMULATIVE PER CENT WEIGHT RETAINED 10 5p @ ~ 9 2 8 8 8 20 8 မ္က w ហ -0 0 F -886 SCALE Duplicate SCREEN SCALE RATIO 1.414 Samples Average Openings Per Cent Per Cent Tyler U.S. Sample Sample Sample Per Cent Milli-Cumulative Per Cent Cumulative Cammative Mesh No. Weights Welghts Inches Weights Weights Weights Weights 26.67 1.050 18.85 .742 13.33 .626 9.423 .371 6.680 .263 3 4,699 .185 3.327 .131 6 G 0.1 0.1 0.10.1 0.1 0.12.362 E80, 8 8 1.651 .065 10 12 6.5 6.7 6.5 7.5 1.168 .046 7.5 7.6 7.0 7.0 14 18 .833 .0328 20 20 30.4 36.9 30.0 $\overline{30.2}$ 37.8 30.3 30. .589 ,0232 37.4 28 30 31.4 .417 ,0164 35 40 .295 .0116 48 50 30.9 67.8 31.4 31.669.431.7 31.931.3 68.6.208 .0082 65 70 13,4 <u>13.0</u> 80.8 10.9 11.0 80.4 12.1 12.0 80.6 .147 .0058 100 100 .104 .0041 150 140 10.9 10.6 91.7 15.5 15.6 96.0 13.2 13.1.074 .0029 200 200 9.0 8.7 100.1 4.1 4.11,00.1 6.66.4 .074 .0029 200 200 Totals, 103.399 5 101.5 100.2 100.1

THE W. T TYLER COMPANY, MENTOR, OHIO 44060, U.S A

Form No. Les Please mention at whom ordering

The Tyler Standard Screen Scale
Cumulative Logarithmic Diagram of Screen Analysis on Sample of Test Well #1 Soil

59' Depth (Lab. #1626, Sample, RETAINED WEIGHT CENT CUMULATIVE PER 20 10 6 ဓ္က 20 φ រប 500 OPENING SCREEN SCALE RATIO 1.414 Duplicate Samples Average Openings Per Cent Per Cent Per Cent Tyler U.S. Sample Sample Sample Milli-Per Cent Cumulative Per Cent Cumulative Per Cent Mesh Cumulative Inches No. Weights Weights Weights meters Weights Weights Weights 26.67 1.050 18.85 .742 13.33 .525 9.423 .371 6.680 .263 3 14.1 14,3 4.699 14.3 .186 4 23.5 23.8 23.8 18.8 19.0 19.0 3.327 .131 6 6 15.0 15.2 16.216.4 30.7 38.0 15.6 2.362 .093 15.8 34.4 8 8 1.651 .065 10 12 15.7 15.9 46,6 15.8 55.0 1.168 046 16.0 14 16 15.7 16.0 <u>50,8</u>

.833

.589

.417

,295

,208

.147

.104

.074

.074

.0328

.0232

0164

.0116

.0082

.0058

.0041

.0029

.0029

20

28

35

48

65

100

150

200

200

20

30

40

50

70

100

140

200

200

Totals,

13.9

10.0

7.]

6.5

15.4

98.9

14.1

10.1

7.2

6.6

100.2

15.6|100.2

<u>60.7</u>

70.8

78.0

84.6

THE W. S. TYLER COMPANY, MENTOR, OHIO 44080, U.S.A.

12.9

9.6

6.3

14.2

13.0

9.7

6.4

5 8₁

14.4

98.8 100.1 |

63.8

73.5

79.9

85.8

12.0

9.2

5,6

5.1

13.2

100.1

67.0

76.2

81.8

86.9

100.1

<u> 11.9</u>

9.1

5.5

5.0

13.0

98.8

A P P E N D I X

WELL CORING LOG

Date: 2-1-77

Well <u>#1</u> Tharpe Lake

__Depth vs. Concentration

Moisture (Percent by Weight)

Lab. No.	Depth Feet	Moisture 5 10 15 20
1512	0.05	1.4 xxx
1513	2.0 - 2.5	3.6 xxxxxxx
1514	4.0 - 4.5	3.5 xxxxxxx
1515	6.0 - 6.5	3.7 xxxxxxxx
1516	8.0 - 8.5	4.1 xxxxxxxxx
1517	9.5 - 10.0	3.6 xxxxxxxx
1518	12.0 - 12.5	3.3 xxxxxxxx
1519	13.5 - 14.5	3.3 xxxxxxxx
1520	16.5 - 17.5	3.0 xxxxxxx
1521	18.5 - 19.0	3.1 xxxxxxx
1522	20.0 - 21.5	3.3 xxxxxxxx
1523	22:5 - 23.0	10.3 xxxxxxxxxxxxxxxxxxxxx
1524	24.0 - 25.0	8.4 xxxxxxxxxxxxxxxxx
1525	26.0 - 27.0	2.8 xxxxxxx
1526	27.0 - 28.0	3.0 xxxxxxx
1535	33.0 - 33.5	2.7 xxxxxx
1536	36.0 - 37.0	3.7 xxxxxxxxx
1537	39:0 - 39:5	5.7 xxxxxxxxxxxx
1538	42.5 - 43.5	3.5 xxxxxxxx
1539	44.5 - 45.5	4.6 xxxxxxxxxx
1549	49.5 - 50	13.8 xxxxxxxxxxxxxxxxxxxxxxxx
1550	53.0 - 53.5	4.8 xxxxxxxxxx
1551	56.0 - 57.0	2.8 xxxxxxx
1552	60.0 - 61.0	11.6 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1553	64.0 - 65.0	
1557		
1558	68.0 - 69.0 72.0 - 73.0	3.6 XXXXXXXX 4.3 XXXXXXXXX
1559	76.0 - 77.0	
1560	80.0 - 80.5	
1561	83.5 - 84.5	10.0 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1567 1568	88.5 - 89.0	2.7 xxxxxx
	92.0 - 93.0	3.3 kxxxxxxx
1569	95.0 - 96.0	3.9 XXXXXXXX
1570	100.0 - 101.0	4.3 xxxxxxxxx
1571	103.0 - 104.0	4.6 xxxxxxxxx
1580	108.0 - 109.0	4.3 xxxxxxxxx
1581	112.0 - 113.0	5.8 xxxxxxxxxxx
1582	116.0 - 117.0	16.1 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1583	120.0 - 121.0	7.6 XXXXXXXXXXXXXXXXX
1584	124.0 - 125.0	12.6 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1604	128.0 - 129.0	6.7 KXXXXXXXXXXXXXX
1605	132.0 - 133.0	6.5 XXXXXXXXXXXXXX
1606	136.0 - 137.0	3.2 XXXXXXXX
1607	140.0 - 141.0	3.4 XXXXXXX
1608	144.0 - 145.0	8.9 XXXXXXXXXXXXXXXXXX
1616 1617	147.0 - 148.0	12.6
1617	150.0 - 151.5 152.0 - 153.0	14.8 KXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX
1619	155.0 - 156.5	
1620	157.0 - 158.5	17.0 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1020	15/.0 - 150.5	17.0 200000000000000000000000000000000000
1640	Soil Blank #1	1.8 Blank soil samples were taken from the Cunningham Sand
1641	Soil Blank #2	0.0 and Gravel Co. sand pit south of the South Plant
	. w v we reserve to the	

ELL CORING LOG

Depth vs. Concentration

Date: 2-1-77

Well #1____Tharpe Lake

oH of soil

Lab. No.	Depth Feet	_pii	
1512 1513 1514 1515 1517 1518 1519 1520 1522 1523 1524 1525 1536 1537 1538 1539 1550 1561 1569 1570 1581 1582 1583 1583 1580 1560 1570 1581 1582 1583 1583 1583 1583 1583 1583 1583 1583	0.05 2.0 - 2.5 4.0 - 4.5 6.0 - 6.5 8.0 - 8.5 9.5 - 10.0 12.0 - 12.5 13.5 - 14.5 16.5 - 17.5 18.5 - 19.0 20.0 - 21.5 22.5 - 23.0 24.0 - 25.0 26.0 - 27.0 27.0 - 28.0 33.0 - 33.5 36.0 - 37.0 39.0 - 39.5 42.5 - 43.5 44.5 - 45.5 49.5 - 50 53.0 - 61.0 64.0 - 65.0 68.0 - 69.0 72.0 - 73.0 76.0 - 77.0 80.0 - 80.5 83.5 - 84.5 88.5 - 89.0 92.0 - 93.0 95.0 - 96.0 100.0 - 101.0 103.0 - 104.0 108.0 - 109.0 112.0 - 113.0 116.0 - 117.0 120.0 - 121.0 124.0 - 125.0 128.0 - 129.0 132.0 - 133.0 136.0 - 137.0 140.0 - 141.0 144.0 - 145.0 147.0 - 148.0 150.0 - 151.5 152.0 - 153.0 155.0 - 156.5 157.0 - 158.5 Soil Blank #1	7.37 7.62 7.37 7.50 7.70 7.71 7.72 7.72 7.72 7.72 7.72 7.72 7.72	Blank soil samples were taken from the Cunningham
1641	Soil Blank #2	7.36	Sand and Gravel Co. sand pit south of the South Plant

Lab. No.	Identification	pH of Water
1634	Test Well #1 1st Bail	7.74
1636	Test Well #1 10th Bail	7.59
1637	Test Well #1 50th Bail	7.66
	North Plant Production Well #	7.76
	North Plant Production Well #	7.96
	North Plant Production Well #	7.99
	North Plant Production Well #	8.04
	North Plant Production Well #	8.01
	South Plant Production Well E	
	South Plant Production Well W	lest 8.12

WELL CORING LOG

Date: 2-1-77

Well #1 Tharpe Lake

—Depth vs. Concentration

Na

		Socium in water
Lab. No.	<u>Identification</u>	(Conc. Expressed in Micrograms per ml)
1634	Test Well #1 lst Bail .	48
1636	Test Well #1 10th Bail	42
1637	Test Well #1 50th Bail	45
	North Plant Production Well #1	6.2
	North Plant Production Well #2	6.3
	North Plant Production Well #3	4.2
	North Plant Production Well #5	5.7
	North Plant Production Well #6	6.3
	South Plant Production East	4.1
	South Plant Production West	4.2

Well #1 Tharpe Lake

WELL CORING LOG

Depth vs. Concentration

Date:__2-1-77

Lab. No.	Identification		(Conc. Expressed in Micrograms per ml)
1640 1641	Soil Blank #1 Soil Blank #2	770 350	Blank soil samples were taken from the Cunningham Sand and Gravel Co. sand pit south of the South Plant.
1512	0.05	830	

o. Identification	(Conc. Expressed in Micrograms per ml)
Test Well #1 1st Bail	41
Test Well #1 10th Bail	59
Test Well #1 50th Bail	43
North Plant Production Well #1	21.4
North Plant Production Well #2	22.5
North Plant Production Well #3	22.7
North Plant Production Well #5	24.3
North Plant Production Well #6	23.5
South Plant Production Well East	22.5
South Plant Production Well West	24.5
	Test Well #1 1st Bail Test Well #1 10th Bail Test Well #1 50th Bail North Plant Production Well #1 North Plant Production Well #2 North Plant Production Well #3 North Plant Production Well #5 North Plant Production Well #6 South Plant Production Well #6

WELL CORING LOG
Depth vs. Concentration

Date:___2-1-77

-Well # 1 Tharpe Lake

FLUORIDE (Conc. Expressed in Micrograms per gram of as-received soil)

Lab. No.	Depth Feet Fluori	ide µg/gm
	,	
1512 1513 1514 1515 1516 1517 1518 1519 1520 1522 1523 1524 1525 1526 1536 1537 1538 1539 1549 1550 1551 1553 1553 1560 1561 1582 1583 1583 1583 1583 1583 1583 1583 1583	0.05 2.0 - 2.5 4.0 - 4.5 6.0 - 6.5 8.0 - 8.5 9.5 - 10.0 12.0 - 12.5 13.5 - 14.5 16.5 - 17.5 18.5 - 19.0 20.0 - 21.5 22.5 - 23.0 24.0 - 25.0 26.0 - 27.0 27.0 - 28.0 33.0 - 33.5 36.0 - 37.0 39.0 - 39.5 42.5 - 43.5 44.5 - 45.5 49.5 - 50 53.0 - 53.5 56.0 - 57.0 60.0 - 61.0 64.0 - 65.0 68.0 - 69.0 72.0 - 72.0 76.0 - 77.0 80.0 - 80.5 83.5 - 84.5 88.5 - 89.0 92.0 - 93.0 95.0 - 96.0 100.0 - 101.0 103.0 - 104.0 108.0 - 109.0 112.0 - 113.0 116.0 - 117.0 120.0 - 121.0 124.0 - 125.0 128.0 - 129.0 132.0 - 133.0 136.0 - 137.0 140.0 - 141.0 144.0 - 145.0 147.0 - 148.0 150.0 - 151.5 152.0 - 153.0 155.0 - 156.5	10
1620	157.0 - 158.5	8.5 kxxxxxxxxxxxxxxx
T640 1641	Soil Blank #1 Soil Blank #2	6.0 Blank soil samples were taken from the 5.5 Cunningham Sand and Gravel Co. sandpit sout of the South Plant
Lab. No. 1634 1636 1637	Identification Test Well #1 1st Bail Test Well #1 10th Bail Test Well #1 50th Bail	Fluoride in Water (Conc. Expressed in Micrograms per ml) 3.07 2.90 3.10
	North Plant Production Well North Plant Production Well North Plant Production Well North Plant Production Well North Plant Production Well	#2 0.03 #3 0.03 #5 -
	South Dlant Duadwation Nall	Foot 0.00

South Plant Production Well East South Plant Production Well West

0.03 0.04 - RFI <u>M 854</u>

WELL CORING LOG

Date: 2-1-77

Well #1 Tharpe Lake

Depth vs. Concentration

S0₄

Lab. No.	Depth Feet	SO ₄ μg/gm
1512	0.05	
1513	2.0 - 2.5	978 xxxxxxxxxxxxxxx
1514	4.0 - 4.5	1,123 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1515	6.0 - 6.5	1,638 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1516	8.0 ~ 8.5	1,360 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1517	9.5 - 10.0	1,318 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1518	12.0 - 12.5	1,329 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1519	13.5 - 14.5	577 XXXXXXXXX
1520	16.5 - 17.5	700 xxxxxxxxxx
1521	18.5 - 19.0	443 xxxxxxxx
1522	20.0 - 21.5	886 xxxxxxxxxxxxxx
1523	22.5 - 23.0	793 xxxxxxxxxxxx
1524	24.0 - 25.0	556 xxxxxxxxx
1525	26.0 - 27.0	628 xxxxxxxxxx
1526	27.0 - 28.0	474 xxxxxxx
1535	33.0 - 33.5	1,792 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1536	36.0 - 37.0	834 XXXXXXXXXXXXXXX
1537	39.0 - 39.5	968 XXXXXXXXXXXXXXXX
1538	42.5 - 43.5	1,421 xxxxxxxxxxxxxxxx
1539	44.5 - 45.5	1,318 xxxxxxxxxxxxxxxxx
1549	49.5 - 50	1,205 xxxxxxxxxxxxxxxx
1550	53.0 - 53.5	670 xxxxxxxxxx
1551	56.0 - 57.0	906 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1552	60.0 - 61.0	700 xxxxxxxxxxx
1553	64.C - 65.O	. 299 xxxxx
1557	68.0 - 69.0	762 xxxxxxxxxxxx
1558	72-0 - 73.0	319 kxxxx
1559	76.0 - 77.0	1,092 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1560	80.0 - 80.5	1,174 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1561	83.5 - 84.5	855 XXXXXXXXXXXXX
1567	88.5 - 89.0	2,451 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1568	92.0 - 93.0	1,257 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1569	95.0 - 96.0	1,185 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1570	100.0 - 101.0	680 xxxxxxxxxxx
1571	103.0 - 104.0	1,524 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1580	108.0 - 109.0	566 kxxxxxxxxx
1581	112.0 - 113.0	845 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1582	116.0 - 117.0	690 kxxxxxxxxxxx
1583	120.0 - 121.0	752 kxxxxxxxxxxx
1584	124.0 - 125.0	793 XXXXXXXXXXXX
1604	128.0 - 129.0	834 KXXXXXXXXXXXXX
1605	132.0 - 133.0	917 KXXXXXXXXXXXXX
1606	136.0 - 137.0	814 XXXXXXXXXXXXX
1607	140.0 - 141.0	1,679 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1608	144.0 - 145.0	1,978 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1616	147.0 - 148.0	72] KXXXXXXXXXX
1617	150.0 - 151.5	1,102 kxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx
1618	152.0 - 153.0	515 XXXXXXXX
1619 1620	155.0 - 156.5 157.0 - 158.5	1,576 XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX
	107.0 = (100.0	033 1000000000
1640	Soil Blank #1	1,545 Blank soil samples were taken from the Cunningha
1641	Soil Blank #2	1,390 Sand and Gravel Co. sand pit south of the South
		Plant
lab No	Identification	Sulfate in Water

	•	Sultate in Water
Lab. No.	Identification	(Conc. Expressed in Micrograms per ml
1634	Test Well #1 1st Bail	173.4
1636	Test Well #1 10th Bail	171.6
1637	Test Well #1 50th Bail	157.0
	North Plant Production Well #1	19.3
	North Plant Production Well #2	40.8
	North Plant Production Well #3	13.0
	North Plant Production Well #5	
	North Plant Production Well #6	22.2
	South Plant Production Well East	25.7
	South Plant Production Well West	33.3

Date: 2-1-77

Well # 1 Tharpe Lake

Depth vs. Concentration

NO₃

(Conc. Expressed in Micrograms per gram of as-received soil)

Lab. No.	Depth Feet	N <u>O_{3 μ}g/gm</u>		
1512	0.05		100 ×××××××××	1,000
. 1513 1514	2.0 - 2.5	140 xxxxxxxxx	xxxxxxxxx	
1515	4.0 - 4.5 6.0 - 6.5	70 xxxxxxxx 90 xxxxxxxx	XXXXX	
1516 1517	8.0 - 8.5	100 xxxxxxxxx	XXXXXXXX	
1517 1518	9.5 - 10.0 12.0 - 12.5	100 xxxxxxxx	xxxxxxx	
1519	13.5 - 14.5	90 xxxxxxxx 90 xxxxxxxx	XXXXXXX	
1520 1521	16.5 - 17.5	85 xxxxxxxx		
1521	18.5 - 19.0 20.0 - 21.5	85 xxxxxxxxx	XXXXXX	
1523	22.5 - 23.0	100 ×xxxxxxxx 85 ×xxxxxxxx		
1524 1525	24.0 - 25.0	85 xxxxxxxxx	(XXXXXX	
1526	26.0 - 27.0 27.0 - 28.0	100 xxxxxxxx	(XXXXXXX	ļ
1535	33.0 - 33.5	85 xxxxxxxx	XXXXXX	
1536 1537	36.0 - 37.0	70 xxxxxxxx		
1537	39.0 - 39.5 42.5 - 43.5	70 xxxxxxxx		
1539	44.5 - 45.5	85 xxxxxxxxx 85 xxxxxxxxx		
1549	49.5 - 50	70 xxxxxxxx		
1550 1551	53.0 - 53.5 55.0 57.0	70 xxxxxxxxx	XXXX	\
: 1552	56.0 - 57.0 60.0 - 61.0	70 XXXXXXXXX		İ
1553	64.0 - 65.0	70 XXXXXXXX 70 XXXXXXXX		
: 1557 1558	68.0 - 69.0	65 XXXXXXXXX	xxx	3
1559	72.0 - 73.0 76.0 - 77.0	55 XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX		
1560	80.0 - 80.5	47 xxxxxxxxx 100 xxxxxxxxx		
1561 1567	83.5 - 84.5	85 XXXXXXXXX	xxxxx	
1568	88.5 - 89.0 92.0 - 93.0	100 ××××××××× 90 ×××××××××		
1569	95.0 - 96.0	90 xxxxxxxxx		
1570 1571	100.0 - 101.0	70 XXXXXXXXX		
1580	103.0 - 104.0 108.0 - 109.0	60 XXXXXXXXX		
1581	112.0 - 113.0	100 ××××××××× 90 ×××××××××		•
1582 1583	116.0 - 117.0	90 XXXXXXXXXX	(XXXXXX	
1584	120.0 - 121.0 124.0 - 125.0	90 XXXXXXXXX		
1604	128.0 - 129.0	100 XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX		
1605 1606	132.0 - 133.0	85 XXXXXXXXX	(XXXXX	
1607	136.0 - 137.0 140.0 - 141.0	70 ××××××××× 70 ×××××××××		
1608	144.0 - 145.0	70 ^^^^^^		
: 1616 1617	147.0 - 148.0	75 XXXXXXXXX	xxxx	
1618	150.0 - 151.5 152.0 - 153.0	70 ××××××××× 55 ×××××××××	t .	
1619	155.0 - 156.5	70 ×××××××××		
1620	157.0 - 158.5	70 ×××××××××		j .
1640	Cost Diami, Ha		/	
1641	Soil Blank #1 Soil Blank #2	380 Blank 205 Cunni	soil samples wer	e taken from the
			ngnam Sand and Gr of the South Pla	avel Co. sand pit
		- • • • • • • • • • • • • • • • • • • •	Nitrate in Water	
<u>Lab. No.</u> 1634	Identification Test Vella "	(Conc. Expre	essed in Microgra	ms per ml)
1636	Test Well #1 1st Bail Test Well #1 10th Bail		12	
1637	Test Well #1 50th Bail	•	24	
	North Plant Production	Well #1	13 5.4	
	North Plant Production	Well #2	6.3	
	North Plant Production North Plant Production	Well #3	4.4	
	North Plant Production	weil #5 Well #6	5.6 6.3	
		+ · · · · // //	0.5	

11.0 11.7

South Plant Production Well Fast South Plant Production Well Wost

WELL CORING LOG

Date: 2-1-77

Well #2 Dump Site

Depth vs. Concentration

Moisture (Percent by Weight)

Lab. No.	Depth_Feet	Moisture	5	10	3		
1693	0.0 - 0.5	1.3 xxx		10	15	20	21
7694	4.0 - 4.5	1.1 xxx	İ			ļ	
1695	8.0 - 8.5	4.3 xxxxx	/VVVV			}	
1696	12.0 - 12.5	3.7 xxxxx	/VVV				
1697	15.5 - 16.5	4.0 xxxxx	(2222				
1698	20.0 - 21.0	4.6 xxxxx	(XXXXX				
1699	24.0 - 25.0	4.0 xxxxx	XXX		· ·		
1700	28.0 - 29.0	2.2 xxxxx			ļ		
1701	32.0 - 33.0	2.9 xxxxx	x l				
1702	36.0 - 37.0	2.6 xxxxxx				-	
1703	40.0 - 41.0	2.4 xxxxxx	į				
1704	44.0 - 45.0	2.6 xxxxxx				!	
1705	48.0 - 49.0	4.5 xxxxxx	xxxxx	ļ			
1706	52.0 - 53.0	18.7 xxxxxx	XXXXXXXXXXX	xxxxxxxxxxx			
1707	56.0 - 57.0	5.9 xxxxxx	XXXXXXXX	^^^^^	******	CXXX	
1708	60.0 - 61.0	4.2 xxxxxx	XXXX	į.			
1709	64.0 - 65.0	1.8 xxxx			İ		
1710	68.0 - 69.0	2.7 xxxxxx					
1711	72.0 - 73.0	2.8 xxxxxx	x				
1712	76.0 - 77.0	3.4 xxxxxx	xx				
1714 1715	80.0 - 81.0	4.2 xxxxxx	xxxx				
1716	84.0 - 85.0	3.3 xxxxxx				}	1
1717	88.0 - 89.0	3.1 xxxxxx					•
1718	92.0 93.0	4-1 xxxxxx	xxxx .				;
1719	96.5 - 97.0	4-1 xxxxxx	<xxx< td=""><td></td><td></td><td>ł</td><td>:</td></xxx<>			ł	:
1741	100.5 - 101.5	3.7 xxxxxx	(XX				į
1742	104.5 - 105.5	4.1 xxxxxx	(XXX	}	i		
1743	108.0 - 109.0 112.5 - 113.0	4.6 (xxxxxxx)					}
1744	116.5 - 117.0	6.1 xxxxxxx	(XXXXXXXX	1		ţ	1
1745	120.0 - 121.0	5.4 xxxxxx	(XXXXXX	ļ		İ]
1746	125.0 - 126.0	4.8 xxxxxx	XXXXX				1
1747	129.0 - 130.0	10.6 xxxxxx	XXXXXXXXXXX	XXXXXX]	
1748	133.0 - 134.0	10.1 xxxxxxx	XXXXXXXXXXX	XXXXX			ĺ
1749	137.0 - 138.0	5.2 xxxxxxx	XXXXX				
1750	141.0 - 142.0	5.8 xxxxxxx	XXXXXXX	İ		1	
1751	144.0 - 145.0	6.2 xxxxxxx	XXXXXXXX		-	-	İ
1752	148.0 - 149.0	12 9 0000000	XXXXXXXXXXX	×××××××××××	xxx		
1753	152.0 - 153.0	18 3 222222	XXXXXXXXXXX	xxxxxxxxxxx	×		ĺ
1754	156.0 - 157.0	21 5 222222	XXXXXXXXXXX	xxxxxxxxxxx	«xxxxxkxxxxx	(X	
1755	157.0 - 158.0	21.3000000	^^XXXXXXXXXXX	××××××××××	«xxxxxkxxxxx	xxxx x xxxx	
	- ,,,,,,	2110 64444	^^^^	×××××××××××	xxxxxxxkxxxxx	xxxx*xxx	
1640	Soil Blank #1	1.8 Blank	Soil cample	ne waxa taka-			
1641	Soil Blank #2	0.0 and G	ravel Co c	es were taker	I from the Cu	inningham Sar	rd
				and pit south	i or the Sout	n Plant.	

. RFI <u>M 854</u>

Well #2 Dump Site

WELL CORING LOG

— Depth vs. Concentration

pH of Soil

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Date: 2-1-77

Lab. No.	Depth Feet	_pH			
1693			webbah		
1693	0.0 - 0.5 4.0 - 4.5	6.80 6.98			
1695	8.0 - 8.5	7.66			
1696	12.0 - 12.5	9.17			
1697	15.5 - 16.5	7.90			
1698	20.0 - 21.0	7.81			
1699	24.0 - 25.0	7.79			
1700 1701	28.0 - 29.0	7.95			
1701	32.0 - 33.0 36.0 - 37.0	7.92 7.83	•		
1703	40.0 - 41.0	7.03 7.77			
1704	44.0 - 45.0	7.79			
1705	48.0 - 49.0	7.76			
1706	52.0 - 53.0	7.73			
1707	56.0 - 57.0	7.69			
1708	60.0 - 61.0	7.80	•		
1709	64.0 - 65.0	7.71			
1710 1711	68.0 - 69.0	7.66			
1712	72.0 - 73.0 76.0 - 77.0	7.74 7.69			
1714	80.0 - 81.0	7.83 7.81			
1715	84.0 - 85.0	7.78	•		
1716	88.0 - 89.0	7.69			
1717	92.0 - 93.0	7.54			
1718	96.5 - 97.0	7.74			
1719	100.5 - 101.5	7.72			
1741	104.5 - 105.5	7.64			
1742 1743	108.0 - 109.0 112.5 - 113.0	7.55			
1744	116.5 - 117.0	7.67 7.68			
1745	120.0 - 121.0	7.55			
1746	125.0 - 126.0	7.63			
1747	129.0 - 130.0	7.70			
1748	133.0 - 134.0	7.67			
1749	137.0 - 138.0	7.63			
1750	141.0 - 142.0	7.67			
1751 1752	144.0 - 145.0 148.0 - 149.0	7.67		•	
1753	152.0 - 153.0	7.48 7.61			
1754	156.0 - 157.0	7.61			
1755	157.0 - 158.0	7.54			
		, , ,			
1640	Soil Blank #1	7.66	Blank soil sampl	es were take	n
1641	Soil Blank #2	7.36	Sand and Gravel	Co. sand pit	

Lab. No. 1768 1769 1770 1851 1852	Identification Test Well #2	pH of Water 7.78 7.88 7.76 7.82 7.81
	North Plant Production Well #1 North Plant Production Well #2 North Plant Production Well #3 North Plant Production Well #5 North Plant Production Well #6	7.76 7.96 7.99 8.04 8.01
	South Plant Production Well East South Plant Production Well West	7.91 8.12

Date; 2-1-77

· Well <u>#2</u> Dump Site

Depth vs. Concentration

Lab. No.	Depth Feet	Na μg/gm	7.0	700	***
1693	0.0 - 0.5		10	100	1000
1694	4.0 - 4.5	1,940 xxxxxx	·×××××××××××××××××××××××××××××××××××××	××××××××××××××××××××××××××××××××××××××	(XXXXXXXXXXXXXXXX
1695	8.0 - 8.5	, , A Y Y Y X X X X X X X	. X X X X X X X X X Y Y Y Y Y Y Y Y	YYYYYYYYYYYY	
1696	12.0 - 12.5	- * · * · · · · ! X X X X X X X	.XXXXXXXXXXXYXXXY	YYYYYVVVVVVVVVV	***************************************
1697	15.5 - 16.5	, , , , , , , , , , , ,	. X X X X X X X X X X X X X X X X X X X	<i>xxxxxxxxxxxxxx</i>	'VVVVVV 1
1698 1699	20.0 - 21.0 24.0 - 25.0	- · · · · · · · · · · · · · · · · · · ·	**************************************	V	
1700	24.0 - 25.0 28.0 - 29.0	, ~~ 1XXXXXXX	<i>.XXXXXXXXXXXXX</i> YYY	YYYYYYY VVV <i>V</i> VV	·vvvvvv l
1701	32.0 - 33.0	~~~ 1 1 2 2 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	**************	YYYYYYYYYYYYY	1271417141117
1702	36.0 - 37.0	543	**********	*****************	XXXX
1703	40.0 - 41.0	XXXXXXXI	XXXXXXXXXXXXYYYY		VVV
1704	44.0 - 45.0	••••	<i>XXXXXXXXXXXXXX</i> YYY	/ YYYYYVVVV \ VVVVV	XXX
1705 1706	48.0 - 49.0	700 1444444	XXXXXXXXXXXXXXXXXX	«xxxxxxxxxxxx	
1707	52.0 - 53.0 56.0 - 57.0	co xxxxxxx	XXXXXXXXXXXX	'	
1708	60.0 - 61.0	25 1000000	XXXXXXXXXXXX		
1709	64.0 - 65.0		×××××××××××××××××××××××××××××××××××××××		
1710	68.0 - 69.0	18 XXXXXXX	XXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX	*****	
1711	72.0 - 73.0	35 xxxxxxxx	×××××××××××××××××××××××××××××××××××××××		
1712 1714	76.0 - 77.0	35 xxxxxxx	xxxxxxxxxxxxxx		
1715	80.0 - 81.0 84.0 - 85.0		xxxxxxxxxxxxx	(
1716	88.0 - 89.0		XXXXXXXXXXX		
1717	92.0 - 93.0	27 1000000	xxxxxxxxxxxxxx xxxxxxxxxxxx		
1718	96.5 - 97.0	20	^^^^^XXXXXXXXXXXXXXXXXX		
1719	100.5 - 101.5	27 kxxxxxxx	XXXXXXXXXXXXXX	^	į
1741 1742	104.5 - 105.5 108.0 - 109.0	23 xxxxxxxx	xxxxxxxxxxxxx		
1742	112.5 - 113.0		xxxxxxxxxxxxx		
1744	116.5 - 117.0	20 1000000	XXXXXXXXXXXXXXX	xx	
1745	120.0 - 121.0	7 7 (000000000	«xxxxxxxxxxxxxxxxx	ļ	
1746	125.0 - 126.0	77 ((XXXXXXXXXX		
1747	129.0 - 130.0	10 kxxxxxxx	(XXXXXXXXX		
1748 1749	133.0 - 134.0	10 kxxxxxxx	(XXXXXXXX		
1750	137.0 - 138.0 141.0 - 142.0	18 kxxxxxxx 12 kxxxxxxx	(XXXXXXXXXXX		
1751	144.0 - 145.0	'∸ ¤xxxxxxx	XXXXXXXXXX		
1752	148.0 - 149.0	9 ****	XXXXXXXXXX XXXXXXXXX		
1753	152.0 - 153.0	5 xxxxxxxx			
1754	156.0 - 157.0		XXXXXXXXXX		
1755	157.0 - 158.0	10 KXXXXXX	xxxxxxxxx		
1640	Soil Blank #1	7 E 4 1			
1641	Soil Blank #2	154 Blank 15 and G	soil samples we	re taken from the	Cunningham Sand
	- Jane 72	io and G		it south of the So	outh Plant
Lab. No.	Identifica	tion	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	odium in Water	
1768	Test Well #2 ls	t Bail	(Conc. E	xpressed im Micro	rams per ml)
1769	Test Well #2 ls	t Sample from So	creen	35.0 25.0	
1770	lest Well #2 10	th Bail		33.0	
1851	Test Well #2 50	th Bail		32.0	
1852	Test Well #2 100	th Bail		33.0	
	North Plant Prod	uction Wall #1		<i>.</i>	
	North Plant Prod	uction Well #2		6.2 6.3	
	North Plant Prod	uction Well #3		4.2	
	North Plant Prod	uction Well #5		5.7	
	North Plant Prod	uction Well #6		6.3	
	South Plant Prod	uction Wall C+			
	South Plant Prod	uction Well East uction Well West	•	4.1	•
		OF HELL MEST	•	4.2	

WELL CORING LOG

Date: 2-1-77

· Well #2 Dump Site

Depth vs. Concentration

Lab. No. 1693 1694 1695 1696 1697 1698 1699 1700 1701 1702 1703 1704 1705 1706 1707 1708 1709 1710 1711 1712 1714 1715 1716 1717	Depth Feet 0.0 - 0.5 4.0 - 4.5 8.0 - 8.5 12.0 - 12.5 15.5 - 16.5 20.0 - 21.0 24.0 - 25.0 28.0 - 29.0 32.0 - 33.0 36.0 - 37.0 40.0 - 41.0 44.0 - 45.0 48.0 - 49.0 52.0 - 53.0 56.0 - 57.0 60.0 - 61.0 64.0 - 65.0 68.0 - 69.0 72.0 - 73.0 76.0 - 77.0 80.0 - 81.0 84.0 - 85.0 88.0 - 89.0 92.0 - 93.0	R0		
1719 1741 1742 1743 1744 1745 1746 1747 1748 1479 1750 1751 1752 1753 1754 1755	100.5 - 101.5 104.5 - 105.5 108.0 - 109.0 112.5 - 113.0 116.5 - 117.0 120.0 - 121.0 125.0 - 126.0 129.0 - 130.0 133.0 - 134.0 137.0 - 138.0 141.0 - 142.0 144.0 - 145.0 148.0 - 149.0 152.0 - 153.0 156.0 - 157.0 157.0 - 158.0	270 230 390 450 320 660 830 780 780 770 760 730 790 770		
1640 1641	Soil Blank #1 Soil Blank #2	770 350	Blank soil samples were taken from the Cunningham Sand and Gravel Co. sand pit south of the South Plant	

		<u>Calcium in Water</u>
Lab. No.	<u>Identification</u>	(Conc. Expressed in Micrograms per m
1768 1769	Test Well #2 lst Bail	48.0
1709	Test Well #2 lst Sample from Screen	41.4
1851	Test Well #2 10th Bail	41.0
1852	Test Well #2 50th Bail Test Well #2 100th Bail	40.0
1002	rest werr #2 footh Bati	39.0
	North Plant Production Well #]	21.4
	North Plant Production Well #2	22.5
	North Plant Production Well #3	22.7
	North Plant Production Well #5	24.3
	North Plant Production Well #6	23.5
	South Plant Production Well East	22.5
	South Plant Production Well West	24.5

WELL CORING LOG

Date: 2-1-77

Well#2 Dump Site

Depth vs. Concentration

FLUORIDE

Lab. No.	Depth Feet	Fl <u>uoride ug</u> /gm	10	100	1000
1693	0.0 - 0.5	1,925.0 xxxxxxxxxxx			
1694	4.0 - 4.5	/ U • U XXXXXXXXXXXXX	XXXXXXXXXXXXXXX	(XXXXX	XXXXXXXXXXX
169 5 1696	8.0 - 8.5	IZ-5 XXXXXXXXXXXX	xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx		
1697	12.0 - 12.5 15.5 - 16.5	4/5-U XXXXXXXXXXXXX	xxxxxxxxxxxx	(XXXXXXXXXXXXXX	
1698 -	15.5 - 16.5 20.0 - 21.0	OJ-J KXXXXXXXXXXX	XXXXXXXXXXXXXXX	XXXX	
1699	24.0 - 25.0	TO.U KXXXXXXXXXXX	xxxxxxx		
1700	28.0 - 29.0	23.5 xxxxxxxxxxxx	xxxxxxxx		
1701	32.0 - 33.0	26.0 xxxxxxxxxxxxxxx	«xxxxxxxx	į	
1702	36.0 - 37.0	18.5 xxxxxxxxxxxxxxxxx	XXXXX		
1703	40.0 - 41.0	18.5 xxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxxx	(XXXXX	į	
1704	44.0 - 45.0	21.0 xxxxxxxxxxx 21.5 xxxxxxxxxxx	(XXXXXXXX		
1705	48.0 - 49.0	9.0 xxxxxxxxxxxx	(XXXXXXXX		
1706	52.0 - 53.0	7.5 xxxxxxxxxxx	~~~~~		
1707	56.0 - 57.0	6.5 xxxxxxxxxxxx			
1708	60.0 - 61.0	13.0 xxxxxxxxxxxx	xxxxx		
1709	64.0 - 65.0	IU.U kxxxxxxxxxxx	(xxxxx)		
1710	68.0 - 69.0	15.0 kxxxxxxxxxxxx	xxxxxxx		
1711 1712	72.0 - 73.0	Li-U kxxxxxxxxxxxx	xxxxxk		
1714	76.0 - 77.0 80.0 - 81.0	10.5 kxxxxxxxxxxxx	xxxxx		
1715	80.0 - 81.0 84.0 - 85.0	5.0 kxxxxxxxx			
1716	88.0 - 89.0	4-0 kxxxxxx			
1717	92.0 - 93.0	4.5 xxxxxxxx 2.5 xxxx	į		
1718	96.5 - 97.0	4.0 xxxxxxx		1	
1719	100.5 - 101.5	11.0 xxxxxxxxxxx			
1741	104.5 - 105.5	18.5 xxxxxxxxxxxxx	XXXXXX		
1742	108.0 - 109.0	5.0 xxxxxxxxx	^^^		
1743	112.5 - 113.0	20.0 xxxxxxxxxxxx	xxxxxxx		
1744	116.5 - 117.0	15.0 kxxxxxxxxxxx	XXXXXXX	j	
1745	120.0 - 121.0	IU.U kxxxxxxxxxxx	xxxxxl		
1746	125.0 - 126.0	10.0 kxxxxxxxxxxx	xxxxx		
1747 1748	129.0 - 13 <u>0</u> .0	/-5 kxxxxxxxxxxxxx	xx	İ	
1748	$133.0 - 13\overline{4.0}$ $137.0 - 138.0$	7.5 kxxxxxxxxxxxx	xx		
1750	141.0 - 142.0	5.0 kxxxxxxxx		F	
1751	144.0 - 145.0	12.5 kxxxxxxxxxxx	xxxxx		
1752	148.0 - 149.0	7.5 kxxxxxxxxxxx	xx		
1753	152.0 - 153.0	12.5 xxxxxxxxxxx 5.0 xxxxxxxx	XXXXX		İ
1754	156.0 - 157.0	7.5 kxxxxxxxxxxxxx			
1755	157.0 - 158.0	10.0 <u>kxxxxxxxxxxx</u>	AX VVVVV		
1640				<u> </u>	
1641	Soil Blank #1 Soil Blank #2	6.0 Blank soi	l samples wer	e taken from the	Cunningham Sand
	JOHN DIGHT #2	5.5 and Grave	1 Co. sand pi	it south of the So	uth Plant
		-			

Lab. No. 1768 1769 1770 1851 1852	Identification Test Well #2	Fluoride in Water (Conc. Expressed in Micrograms per ml) 2.1 3.4 3.9 2.4 2.4
	North Plant Production Well #1 North Plant Production Well #2 North Plant Production Well #3 North Plant Production Well #5 North Plant Production Well #6	0.04 0.03 0.03 - 0.03
	South Plant Production Well East South Plant Production Well West	0.03 0.04

Date: 2-1-77

Depth vs. Concentration

Well #2 Dump Site

SO₄
(Conc. Expressed in Micrograms per gram of as-received soil)

Lab. No.	Depth Feet	- <u>\$04 μg/gm</u>	1,000	10,000	
1693	0.0 0.5	422.3 XXXXX		10,000	
1694	4.0 - 4.5	679.8 XXXXX			
1695	8.0 - 8.5		xxxxxxxxxx		
1696	12.0 - 12.5	741.6 ×××××			
1697	15.5 - 16.5	267.8 XXXXX			
1698	20.0 - 21.0		xxxxxxxxxxxx		
1699	24.0 - 25.0	1,133.0 xxxxx			
1700	28.0 - 29.0	679.8 ×××××	xxxxxx		
1701	32.0 - 33.0	865.2 xxxxx			
1702	36.0 - 37.0		xxxxxxxxxx		
1703	40.0 - 41.0	824.0 xxxxx			
1704 1705	44.0 - 45.0	339.9 xxxxx			
1706	48.0 - 49.0 52.0 - 53.0	648.9 xxxxx			
1707	52.0 - 53.0 56.0 - 57.0	525.3 xxxxx			
1708	60.0 - 61.0	442.9 xxxxx			
1709	64.0 - 65.0	772.5 xxxxx	xxxxxxxx	į	
1710	68.0 - 69.0	659.2 xxxxx	×××××××××××××××××××××××××××××××××××××××		
1 <i>7</i> 11	72.0 - 73.0	1,462.6 xxxxx	××××××		
1712	76.0 - 77.0	772.5 xxxxx	~~~~~~		
1714	80.0 - 81.0	638.6 xxxxx	XXXXXXX		
1715	84.0 - 85.0	844.6 xxxxx			
1716	88.0 - 89.0	1,122.7 xxxxxx			
1717	92.0 - 93.0	659.2 xxxxx	(XXXXXX		
1718	96.5 - 97.0	957.9 xxxxx			
1719	100.5 - 101.5	344.5 xxxxx			
1741	104.5 - 105.5	453.2 xxxxxx			
1742	108.0 - 109.0	1,318.4 xxxxxx	(XXXXXXXXXXX		
1743	112.5 - 113.0	422.3 XXXXX	(XX		
7744	116.5 - 117.0	587.1 ×××××			
1745	120.0 - 121.0	700.4 XXXXX		į	
1746	125.0 - 126.0	422.3 XXXXXX			•
1747 1748	129.0 - 130.0	1,174.2 XXXXX			
1749	133.0 - 134.0	494.4 XXXXX	xxx		
1750	137.0 - 138.0 141.0 - 142.0	288.4 XXXXX		. 1	
1751	141.0 - 142.0	360.5 XXXXXX			
1752	144.0 - 145.0	875.5 XXXXXX			
1753	152.0 - 153.0	1,297.8 xxxxxx 1,184.5 xxxxxx		ł	
1754	156.0 - 157.0	556.2 XXXXXX			
1755	157.0 - 158.0		xxxxxxxxxxx		
	100.0	OTOTA KANANA	^^^^^		-
1640	Soil Blank #1	1,545 Blank	soil samples were t	aken from the Cunn	her? medoi
1641	Soil Blank #2		ravel Co. sand pit s	outh of the South	rynam sanu Plant
		·		sage of the south	· iaiic

Lab. No. 1768 1769 1770 1851 1852	Identification Test Well #2 1st Bail Test Well #2 1st Sample from Screen Test Well #2 10th Bail Test Well #2 50th Bail Test Well #2 100th Bail	Sulfate in Water (Conc. Expressed in Micrograms per ml) 241.6 172.8 177.1 175.9 182.1
· ·	North Plant Production Well #1 North Plant Production Well #2 North Plant Production Well #3 North Plant Production Well #5 North Plant Production Well #6 South Plant Production Well East South Plant Production Well West	19.3 40.3 18.0 22.2 25.7 33.3

- WELL CORING LOG .

Date: 2-1-77

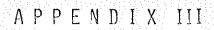
Depth vs. Concentration

Well #2 Dump Site

N03

Lab. No.	Depth Feet	NO3 μg/gm	100	1.000	
1693 1694 1695 1696 1697 1698 1699 1700 1701 1702 1703 1704 1705 1706 1707 1708 1709 1711 1712 1714 1715 1716 1717 1718 1719 1741 1742 1743 1744 1745 1746 1747 1748 1749 1750 1751 1752 1753 1754 1755	Depth Feet 0.0 - 0.5 4.0 - 4.5 8.0 - 12.5 15.5 - 16.5 20.0 - 21.0 24.0 - 25.0 28.0 - 29.0 32.0 - 33.0 36.0 - 37.0 40.0 - 41.0 44.0 - 45.0 48.0 - 49.0 52.0 - 53.0 56.0 - 67.0 60.0 - 61.0 64.0 - 65.0 68.0 - 69.0 72.0 - 73.0 76.0 - 77.0 80.0 - 81.0 84.0 - 85.0 88.0 - 89.0 92.0 - 93.0 96.5 - 97.0 100.5 - 101.5 104.5 - 105.5 108.0 - 109.0 112.5 - 113.0 116.5 - 117.0 120.0 - 121.0 125.0 - 126.0 129.0 - 130.0 133.0 - 134.0 137.0 - 138.0 141.0 - 142.0 144.0 - 145.0 152.0 - 153.0 156.0 - 157.0 157.0 - 158.0	450	CXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX	KXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXXX	
1640 1641	Soil Blank #1 Soil Blank #2	380 Blank s	soil samples were avel Co. sand pit	taken from the south of the S	Cunningham Sand South Plant

		Nit	rate in Water
L <u>ab. N</u> o	<pre>_ Identification</pre>	(Conc. Expre	ssed in Micrograms per ml)
1768	Test Well #2 lst Bail		4
1769	Test Well #2 1st Sample from	Screen	11
1 <i>77</i> 0	Test Well #2 10th Bail	,	16
1851	Test Well #2 50th Bail		16
1852	Test Well #2 100th Bail		16
			•
	North Plant Production Well #1		5.4
	North Plant Production Well #2		6.3
	North Plant Production Well #3		4.4
	North Plant Production Well #5		5.6
	North Plant Production Well #6		6.3
		•	
	South Plant Production Well Ea	st	11.0
	South Plant Production Well We		11.7
			•



BENNE , S CHEMICAL LABORATOL , INC.

ANALYTICAL CHEMISTS & ASSAYERS

901 SOUTH 9th STREET

TACOMA, WASHINGTON 98405

(206) 272-4507 or 272-7969

September 6, 1977

Mr. Jim Carr Robinson and Noble 10318 Gravelly Lake Drive, S.W. Tacoma, WA 98499

Dear Mr. Carr:

Re: Sample submitted on 8/6/77

In answer to your inquiry as to the methods used in analyzing the above sample, we would like to inform you that our procedures are taken from the "Standard Methods for the Examination of Water and Wastewater", 13th and 14th editions published by APHA-AWWA and WPCF.

Specifically, calcium, iron, and sodium were determined by atomic absorption methods; Section 311, 14th edition. The remaining ions were analyzed as follows.

Cyanide - Section 207C, 13th edition. We made this determination directly on the sample without distillation. We are aware of negative interferences due to refractory heavy metal complexes but do not know of anything that gives positive errors. Incidentally, the decimal is placed incorrectly in our report for the cyanide. Enclosed, please find our corrected copy.

Nitrate - Section 419D, 14th edition.
Sulfate - Section 427C, 14th edition.
Fluoride - Section 414B, 14th edition.
Aluminum - Section 103B, 13th edition./
Bicarbonate Alkalinity - Section 102, 13th edition, using mixed indicator.

Total Dissolved Solids - Section 148B, 13th edition.

In addition, whenever possible we check the ionic balance to confirm our accuracy. For instance in this series of tests we also determined the total hardness by titration (Section 122B, 13th edition), magnesium (atomic absorption), and chloride by titration (Section 112A, 13th edition). Then some calculations confirmed our analysis. Namely, total hardness by titration checked the calculated result and the anion equivalents check the cation equivalents. Therefore, we feel our results are reasonably accurate and reflect the values in the water.

Yours truly,

BENNETT CHEMICAL LABORATORIES, INC.

Hosen B. Prince

Galen B. Prine

8.H

GBP:hb

BENNETIS CHEMICAL LABORATORIES, INC.

ANALYTICAL CHEMISTS & ASSAYERS

901 SOUTH 9th STREET TACOMA, WASHINGTON 98405 (206) 272-4507 or 272-7969

REPORT OF ANALYSIS

September 7, 1977

Our analysis of the sample of

Water

From

Robinson and Noble Received sample on August 5, 1977

Marked:

As shown above

CORRECTED COPY

рн - 7.9

	74.3 mg/liter
Sodium	27.0 mg/liter
I.F. OU	0.23 mg/liter
Cyanide we are use as as as as as as as as as as as as as	0.032 mg/liter
Nitrate and any data and gas and and data and and data and seed and and and and and and and and and an	3.94 mg/liter
Sulfate was seen and and case are seen and and case are seen and s	158.0 mg/liter
Fluoride make and and and and and and and and and and	0.09 mg/liter
Aluminum	0.01 mg/liter*
Bicarbonate Alkalinity	178.0 mg/liter**
Total Dissolved Solids	41.1.0 mg/liter
*Less than	

Robinson and Noble
To 10318 Gravelly Lake Drive, S.W. Tacoma, wA 98499

BENNETTS CHEMICAL LABORATORIES, INC.

By Halin B Prince

INTER-OFFICE MEMORANDUM

W. B. Eastman TO

Mead

COPIES TO

October 13, 1977 DATE

F. G. Doolittle 7, Deelett PROM

Mead AΤ

subject Methods Used for Analyzing Soil and Water Samples in Test Well Project

REFERRING TO

The following analyses were performed on the well water samples and the 500 ml soil leachate

	Water	Soil
Suspended Solids pH		þlí
CN ⁻		CN
۴¯		F
NO ³		NO_3^-
so ₄ =		so ₄ =
Na		Na
Ca ⁺⁺		Ca ⁺⁺

The methods used and references to these methods (when applicable) are as follows:

- Nitrate method used was 419B from "Standard Methods for the Examination of Water and Waste Water", Fourteenth Edition.
- Cyanide Methods used were all glass distillation into NaOH В. followed by analysis using cyanide colorimetric method B pages 448-457 of "Standard Methods for the Examination of Water and Waste Water", Twelfth Edition.
- The pH of the well waters and of the soil extract solution С. was determined using an 801A orion research Digital Ion Analyzer.

- D. Sulfate was determined using method 427B "Standard Methods for the Examination of Water and Waste Water", Fourteenth Edition.
- E. Fluoride was determined using method 203 Tentative Method of Analysis for Fluoride content of the Atmosphere and Plant Tissues. "Methods of Air Sampling and Analysis" published by the APHA, 1972 edition.

This is a Technicon Autoanalyzer method and several minor modifications are used with the Mead routine analytical procedure.

- F. The suspended solids were determined using method 208D "Standard Methods for the Examination of Water and Waste Water", Fourteenth Edition.
- G. The calcium and sodium were run by Atomic Absorption (AA) methods found in the 1968 edition of "Analytical Methods for AA Spectrophotometry published by Perkin-Elmer. The methods are titled "Extractable Calcium in Soils" and "Extractable Sodium in Soils".

WELL WATER CHECK SAMPLES

	LB CN / 191	/m1		us No3/ml	īm			TE 7 27	
Sample Identification	Mead	CFT ^{2,9}	Mead	CFT ^{2,4}	CFT ² ,5	Spokane 3,6	Mead	CTT2,7	CFT 2,8
South Plant East Well	-002	<ooi< td=""><td>11.0</td><td>13.00</td><td>13.40</td><td>13.5</td><td>.03</td><td>.03</td><td><.05</td></ooi<>	11.0	13.00	13.40	13.5	.03	.03	<.05
North Plant Well #6	.003	A.001	6.3	8.20	3.25	œ 	.03	70.	7,05
Test Well #1, 10th Bail	000.	.011	24.0	11.80	12.10	ı	2.90	3.58	3.40
Distilled Water	ı	<.001	1	<.05	<.05	ł	i	۲۰°۲	<.05 0.05

Notes:

- 1) KACC Mead Lab. analysis.
- Kaiser Center for Technology analysis.
-) Spokane County Engineers' Office analysis.
- 4) Chromotropic acid, spectrophotometric.
- 5) Cd reduction, $NO_2 + NO_3$, diazo spectrophotometric.
- 6) Same as Mead NO3 method.
- Technicon Auto Analyzer.
- 8) Zr:SPADNS spectrophotometric.
- UV irradiation, distillation, pyridine-barbituric acid, Auto Analyzer.