



DEPARTMENT OF  
**ECOLOGY**  
State of Washington

## **Standard Operating Procedure EAP032, Version 2.2**

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### **Collection and Analysis of Conductivity Samples**

April 2022

Publication 22-03-209

[Recertified May 2021]

## Purpose of this Document

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The Washington State Department of Ecology develops Standard Operating Procedures (SOPs) to document agency practices related to sampling, field and laboratory analysis, and other aspects of the agency's technical operations.

## Publication Information

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This SOP is available on the Department of Ecology's website at <https://apps.ecology.wa.gov/publications/SummaryPages/2103206.html>.

Ecology's Activity Tracker Code for this SOP is 18-003.

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**Standard Operating Procedures for the Collection and Analysis of Conductivity Samples,  
EAP032, Version 2.2**

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*SIGNATURES AVAILABLE UPON REQUEST*

This SOP is a harmonized version combining SOPs EAP032 and EAP010, which were both SOPs for conducting sampling and analysis.

The Washington State Department of Ecology’s (Ecology’s) Standard Operating Procedures (SOPs) are adapted from published methods, or developed by in-house technical and administrative experts. Their primary purpose is for internal Ecology use, although sampling and administrative SOPs may have a wider utility. Our SOPs do not supplant official published methods. Distribution of these SOPs does not constitute an endorsement of a particular procedure or method.

Any reference to specific equipment, manufacturer, or supplies is for descriptive purposes only and does not constitute an endorsement of a particular product or service by the author or by Ecology.

Although Ecology follows the SOP in most instances, there may be instances in which Ecology uses an alternative methodology, procedure, or process.

## SOP Revision History

Revision date	Revision number	Summary of changes	Sections	Reviser(s)
4/9/2007	1.1	Editorial; formatting	All	Bill Ward
4/17/2007		Comments	All	Dave Hallock
4/27/2007	1.2	Edits based on comments	All	Bill Ward
5/15/2007		Editorial	All	Bill Kammin
5/18/2007	1.3	Edits based on comments	All	Bill Ward
6/7/2007		Editorial	All	Casey Clishe
6/7/2007	1.3	Edits based on comments	All	Bill Ward
6/13/2007	1.3	Final edits (removed draft, added EAP #)	All	Bill Kammin
6/14/2007	1.3	Attachment incorporation		Bill Kammin
10/14/2010	1.4	Minor revisions	5 , 8	Bill Ward
11/21/2010	2.0	Harmonized with SOP EAP010	All	Bill Ward; Nuri Mathieu
2/2/2011	2.1	Incorporated reviewer comments	All	Bill Ward; Nuri Mathieu
4/10/2013		Added new chemical awareness and waste disposal language. Attached MSDS sheet.	5, 9	Bill Ward
5/13/2014	2.2	Minor editorial and procedure updates	4, 5, 6	Bill Ward; Nuri Mathieu
8/11/2014	2.2	Footer, final edits	All	Diana Olegre
8/11/2014	2.2	Approved	All	Bill Kammin
5-12-2021	2.2	Incorporated Accessibility standards	All	Joan LeTourneau
5-12-2021	2.2	Approved	All	Arati Kaza

## **1.0 Purpose and Scope**

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- 1.1 This document is the Environmental Assessment Program (EAP), Environmental Monitoring and Trends Section, Freshwater Monitoring Unit (FMU), Standard Operating Procedure (SOP) for the field collection and analysis of conductivity samples. In addition, this method covers the ‘Direct In-Situ Method’ (Section 6.2.4) used in wadeable streams for Total Maximum Daily Load and other special studies.
- 1.2 The procedures cover meter calibration, sample collection, sample measurement, and quality assurance/quality control. The conductivity meter sample measurement methods follow those described in Standard Methods 2510 B (APHA, 2005).

## **2.0 Applicability**

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- 2.1 This SOP is intended for freshwater monitoring. Marine waters measurements or samples should be taken as specified in a Quality Assurance project Plan.

## **3.0 Definitions**

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- 3.1 Conductivity – A measure of the ability of water to carry an electrical current. It is dependent upon the concentrations and types of dissolved ions and the water temperature. In general, a greater concentration of ions in the water will lead to a larger conductivity value.
- 3.2 EAP – Environmental Assessment Program.
- 3.3 Ecology – Washington State Department of Ecology.
- 3.4 EIM – Environmental Information Management System. A searchable database developed and maintained by the Washington State Department of Ecology.
- 3.5 Field Logbook – A weather resistant logbook containing “Rite in the Rain” ® writing paper used to document any and all field activities, sample data, methods and observations for each and all collection sites.
- 3.6  $\mu\text{mhos}$  – micro mhos ( $\text{mho} = 1/\text{ohm} = 1 \text{ Siemens}$ ) per centimeter
- 3.7 MQOs – Measurement Quality Objectives
- 3.8 MSDS – Material Safety Data Sheets provides both workers and emergency personnel with the proper procedures for handling or working with a particular substance. MSDSs include information such as physical data (e.g., melting point, boiling point, flash point), toxicity, health effects, first aid, reactivity, storage, disposal, protective equipment and spill/leak procedures.
- 3.9 Run – Scheduled sampling day(s).

## **4.0 Personnel Qualifications/Responsibilities**

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- 4.1 Field operations require training specified in EAP's Field Safety Manual (Ecology, 2012). The required trainings include: First Aid, CPR, and Defensive Driving.
- 4.2 Boat operations require that staff meet specific training requirements as described in EAP's Field Safety Manual, such as an EAP Boating Course and an approved Boating Safety Course.

- 4.3 Because the procedure requires the use of hazardous materials, training is required as per the Ecology Chemical Hygiene Plan and Hazardous Material Handling Plan (Section 1) (WA State Department of Ecology 2011), which includes Laboratory Safety Orientation, Job-Specific Orientation and Chemical Safety Procedures. The Standard Operating Procedures in Section 16 of the Chemical Hygiene Plan and Hazardous Material Handling Plan for handling chemicals must also be followed.

## 5.0 Equipment and Supplies

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- 5.1 Bridge Sampler (based on the design in Figure 4500-0:1 of the 20th Edition of Standard Methods).
- 5.2 Sampling ropes (1 @ 10 ft., 1 @ 35 ft. and 2 @ 55 ft.)
- 5.3 Extension pole with a bottle clamp.
- 5.4 Field Logbook or Field Data Report Form  
(See attachment A for form)
- 5.5 Conductivity meter with a non-linear function and a 4-cell probe
- 5.6 100  $\mu$ mhos/cm conductivity standard (See attachment C for MSDS)  
Note: This standard is not considered a health hazard.
- 5.7 Deionized water (DI water)
- 5.8 Meter Calibration Log Form (see attachment B)



## 6.0 Summary of Procedure

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- 6.1 Meter Calibration.
- Note: Always store the meter, probe, standard, and filled DI squirt overnight in a heated room. Also, calibrate probe using a standard that is above 15°C.
- 6.1.1 Soak the conductivity probe in a container of tap water for at least 2 hours (overnight is better) before calibrating it.
- 6.1.2 Clean the conductivity probe cells with a cotton swab.
- 6.1.3 Empty and refill the dedicated conductivity calibration bottles with fresh 100  $\mu$ mhos/cm conductivity standard at the beginning of each week.
- 6.1.4 Hold the cable end of the probe, rinse the other end of the probe with DI water, and flick the water off the probe with a quick downward jerking movement a few times.
- Note: Avoid striking the probe on the sink edge or counter.
- 6.1.5 Set the probe into fresh 100  $\mu$ mhos/cm conductivity standard.
- Note: The conductivity standard is easily contaminated. Keep it tightly capped and avoid diluting it with DI or stream sample water. Also, the accuracy of the freshly opened standard lasts for up to 5 days.

- 6.1.6 Check the meter settings to ensure the meter reads in the non-linear function (nLF) mode for temperature compensation and the reference temperature setting is 25°C<sup>1</sup>.
- 6.1.7 Calibrate the meter according to the meter instruction manual.
- 6.1.8 Record the conductivity standard concentration, the meter ID number, the initial and final cell constants, and any other required information on the Meter Calibration Log Form (see Attachment B for form).
- 6.1.9 Store the conductivity probe in DI water or stream sample water while collecting samples. (Do not store with the pH probe).
- 6.2 Sample Collection.
  - 6.2.1 Bridge Sampler method. This method is typically used to collect stream samples from a bridge or from the stream bank through the use of a rope.
    - 6.2.1.1 Rinse a dedicated 1 L pH and conductivity grab sample bottle (marked with red or green permanent ink) with DI water and secure it in the Bridge Sampler.
    - 6.2.1.2 Put on a high-visibility safety vest and carry the needed sampling gear to a well-mixed sampling location where a representative stream sample may be collected.
    - 6.2.1.3 Attach the sampling rope to the Bridge Sampler, remove the bottle cap, and set the cap aside where it will remain clean.
    - 6.2.1.4 Carefully lower the Bridge Sampler to the water surface, taking care to not dislodge any bridge debris onto it. Allow the bottom of the sampler to touch the water surface, and then raise the sampler off the water for a few moments to allow any debris from the bottom of the sampler to drop off and float away. Then rapidly allow it to submerge about 0.5 meters.

Note: This minimizes the sampling of surface film and any debris from the bottom of the sampler.
    - 6.2.1.5 Retrieve the sampler taking care not to dislodge bridge debris onto it.
    - 6.2.1.6 Replace the bottle cap.
    - 6.2.1.7 Return to the van with all the sampling gear.
  - 6.2.2 Hand Dip Method. This method is typically used to collect samples within reach of the water surface (when standing in or near the stream or lake, or from small boat).
    - 6.2.2.1 Move to a well-mixed location such as the deepest part of the active channel or another location where a representative sample may be collected.

Note: Do not contaminate the sample location by wading upstream of it or collect a sample from an eddy that had been waded.

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<sup>1</sup> The nLF setting works best for measuring low ionic strength natural waters and the temperature reference setting adjusts results to equivalent conductivities at 25°C to facilitate data comparison.

6.2.2.2 Hold the base of the pH and conductivity grab sample bottle with one hand, and remove the bottle cap. Then invert the bottle, reach upstream, and plunge the bottle into the water about 15 cm (6 inches), and tip it up toward the water surface. Allow the bottle to fill and then take it out of the water.

Note: If sampling still water or from a boat, then plunge the bottle opening into the water, and move it upstream or away from the entry location while tipping it upright to avoid contamination.

6.2.3 Extension Pole Method. This method is typically used to reach a more representative or undisturbed sample location from the stream bank, lake shore, or slow-moving stream.

6.2.3.1 Secure the pH and conductivity grab sample bottle in the extension pole clamp.

6.2.3.2 Move to a location where a representative sample may be reached with the pole.

6.2.3.3 Remove the cap from the bottle and place it where contamination will be avoided.

6.2.3.4 Position the bottle just over the desired sample location.

6.2.3.5 Invert the bottle over the desired sample location and in one quick motion plunge the mouth of the bottle about 15 cm (6 inches) and then tip it toward the water surface. Allow the bottle to fill, remove it from the water, replace the cap, and return to vehicle.

6.2.4 Direct In-Situ Method. This method is primarily used in Total Maximum Daily Load and other special studies to measure conductivity in shallow wadeable streams and occasionally used to measure it in larger streams and lakes.

6.2.4.1 Freshwater measurements should be taken from an undisturbed representative location, such as the thalweg of a stream (deepest and fastest portion), in the middle of the water column (at least several inches above the stream bed and below the surface), and, for rivers and streams, where water is visibly flowing in the downstream direction.

6.2.4.2 Rinse the conductivity probe directly in the waterbody with a swirling motion, turn on the meter, and then allow it to stabilize. Record the result following section 6.3.4 of this SOP.

6.3 Sample Measurement Procedure.

6.3.1 Direct conductivity grab sample bottle measurement.

6.3.1.1 Insert probe directly into conductivity grab sample bottle.

6.3.1.2 Turn on the conductivity meter and allow it a few minutes to indicate a stable result.

6.3.1.3 Record the conductivity result on the Field Data Report Form or Field Logbook.

Note: The meter displays results to the nearest tenth and the result needs to be rounded to the nearest whole number. If the tenths digit  $> 0.5$ , round up; if it is  $< .5$ , round down; and if it is  $=$  to 0.5 round to the nearest even number. For example, a result of 103.5  $\mu\text{mhos}$  rounds to 104  $\mu\text{mhos}$  and a result of 62.5  $\mu\text{mhos}$  rounds to 62  $\mu\text{mhos}$ .

6.3.2 Conductivity cup measurement.



- 6.3.2.1 Rinse the conductivity measurement cup and probe with DI or sample water.
- 6.3.2.2 Remove the cap to the conductivity grab sample bottle and gently over fill the conductivity measurement cup with the sample water.
- 6.3.2.3 Turn on the conductivity meter, and allow it a few minutes to indicate a stable result.
- 6.3.2.4 Record the conductivity result following section 6.3.1.3.
- 6.4 End of Day QC Procedures.
- 6.4.1 Check the calibration of the conductivity meter by rinsing the probe with DI water, shake the water off the probe, and set the probe into the 100 µmhos/cm conductivity standard (or another standard that is within the expected range of measured field results). Record a stable result on the Field Data Report Form or Field Logbook. If the result is not within 5 µmhos/cm of the standard, then troubleshoot the meter<sup>2</sup> and if possible re-measure all of the samples.
- 6.4.2 If troubleshooting the meter steps do not work, then review the troubleshooting section in the meter instruction manual to fix the problem. If the problem cannot be fixed, then turn the defective equipment into the Operation Center Technician along with a completed Equipment Problem Report Form.
- 6.5 End of Day or Run Procedures.
- 6.5.1 If the meter and probe will be used the next day, store the probe in stream sample or tap water.
- 6.5.2 If the overnight air temperatures will be below 45 °F, move the meter, probe stored in DI or stream sample water, and conductivity standard into a heated room (hotel room, regional lab, or operation center).
- 6.5.3 If the meter and probe will not be used the next day, then rinse the probe with DI water and store the meter and dry probe in a heated room.

## **7.0 Records Management**

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- 7.1 All hardcopy documentation of the data, such as completed Field Logbook and Field Data Report Forms are kept and maintained by the project lead. These documents are organized in binders or in expanding files. After about six years, hardcopies are boxed and moved to EAP archives.
- 7.2 Data collected for Ecology's Ambient River and Stream Monitoring Program will be entered into an Access<sup>®</sup>-based database, reviewed and verified following the Quality Control and Quality Assurance procedures (see 8.1 below), uploaded into EIM, and posted on our web page [http://www.ecy.wa.gov/programs/eap/fw\\_riv/rv\\_main.html](http://www.ecy.wa.gov/programs/eap/fw_riv/rv_main.html) .

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<sup>2</sup> Check to make sure that the meter is in the non-linear function (nLF) and the temperature coefficient is 25. Then verify the meter calibration using an unopened conductivity standard (the standard is easily contaminated). If the meter needed to be changed to the non-linear function, temperature reference 25, or be recalibrated, then re-measure all of the samples. If on a run, then re-measure using the TSS sample.

7.3 Data collected for Total Maximum Daily Load or special project studies will be reviewed, verified, and stored based on the QAPP for the project.

## **8.0 Quality Control and Quality Assurance Section**

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### 8.1 Freshwater Ambient Monitoring Program

8.1.1 The data QA program for field sampling consists of three parts: (1) adherence to the SOP procedures for sample/data collection and periodic evaluation of sampling personnel, (2) consistent instrument calibration methods and schedules, and (3) the collection of one field quality control (QC) sample during each sampling run, either a duplicate field sample or a true process field blank sample. Our QA program is described in detail in

<https://apps.ecology.wa.gov/publications/summarypages/0303200.html>

(Hallock and Ehinger, 2003).

8.1.2 Duplicate (Sequential) Field Sample. This sample is collected approximately 15-20 minutes after the initial sample was collected. This sample represents the total variability due to short-term, in-stream dynamics, and sample collection and measurement.

8.1.3 True Process Field Blank Sample. This sample is subject to the sample site collection and processing conditions. The expected value for this analyzed result is less than 3  $\mu$ S (micro Siemens).

8.1.3.1 Return to the sample site with the cleaned pH and conductivity grab sample bottle and go through the normal sample-collection procedure, but do not immerse the uncapped bottle or Bridge sampler. Cap the bottle, return to the van, and fill the container with the Lab provided DI water.

8.1.3.2 Fill the conductivity measurement cup with water from the pH and conductivity grab sample bottle, allow the conductivity probe to stabilize, and record the measurement.

8.1.4 A two-tiered system is used to evaluate data quality of individual results based on field QC. The first tier consists of an evaluation of the data. Results exceeding preset limits are flagged. The second tier QC evaluation is a manual review of the data flagged in the first tier. Data are then coded from 1 through 9 (1 = data meets all QA requirements, 9 = data are unusable). Criteria for assigning codes are discussed in more detail in Hallock and Ehinger (2003). We do not routinely use or distribute data with quality codes greater than 4.

### 8.2 Total Maximum Daily Load Monitoring Program

8.2.1 The TMDL data QA program for field sampling consists of two parts: (1) adherence to the SOP procedures for sample/data collection and periodic evaluation of sampling personnel and (2) the collection of a field quality control (QC) measurement for ten percent of the samples or measurements collected for a given study.

8.2.2 The field QC measurement is taken as a replicate field measurement. This consists of moving several feet upstream of the initial measurement location and repeating the measurement procedure. This sample represents the total variability due to in-situ

measurement. Alternately, if available, a second precalibrated meter (of the same model) can be used to take a measurement in the same location as the initial measurement.

- 8.2.3 QA/QC procedures will be addressed more thoroughly on a project-by-project basis, as stated in the QAPP for the project.

## **9.0 Safety**

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- 9.1 Safety is the primary concern when collecting samples. Since most sample sites are located on highway bridges, road and pass conditions should always be checked before departure (especially in winter). If roadside hazards, weather, accidents, construction, etc. make sample collection dangerous, then skip that station. Note the reason on the Field Data Report Form, and notify your supervisor of the hazard when you return to the office. If the hazard is a permanent condition, relocation of the station may be necessary. Review Ecology's Safety Program Manual (Ecology, 2010) periodically to assist with these safety determinations.
- 9.2 Waste disposal. Rinse the used conductivity standard down the drain with water to reduce any impact on the wastewater treatment system.

## **10.0 References**

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- 10.1 APHA, 2005. Standard Methods for the Examination of Water and Wastewater, 21st Edition. Joint publication of the American Public Health Association, American Water Works Association, and Water Environment Federation.  
[www.standardmethods.org/](http://www.standardmethods.org/)
- 10.2 Ecology, 2011. Chemical hygiene plan and hazardous materials management plan. Washington State Department of Ecology. Olympia, WA.  
<http://aww.ecology/services/es/Safety/ChemicalHygiene.pdf>
- 10.3 Ecology, 2012. Environmental Assessment Program Safety Manual. Washington State Department of Ecology. Olympia, WA.
- 10.4 Hallock, D. and W. Ehinger, 2003. Quality Assurance Monitoring Plan: Stream Ambient Water Quality Monitoring. Washington State Department of Ecology, Olympia, WA. 27 pp. Publication 03-03-200.  
<https://apps.ecology.wa.gov/publications/summarypages/0303200.html>



# ATTACHMENT B

Meter Calibration Log Form Run: \_\_\_\_\_ Date: \_\_\_\_\_ Sampler: \_\_\_\_\_

Meter ID Numbers		Pre-Run Calibration			
Cond.		(a) Thermometer (°C)		Van pressure (Pre-cal.)	
pH		(b) Thermister (°C)		Lab pressure	
pH probe		Correction (a minus b)		Adjusted?	Y/N
Temp.		Was that corr. expected?	Y/N		

## DAY 1

Cond Meter	Standard:	Initial Meter Reading:	Initial Cell Const:	Final Cell Const:
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<b>pH Meter</b>				
Slope <sup>a</sup> :		Time:		
mv @ pH 4/10:		Buffer Temp:		
mv @ pH 7:		Response Time <sup>b</sup> :		
	True pH	Meter pH	Time	Recal?
QA Check #1		c		Y / N
QA Check #2		c		Y / N
QA Check #3		c		Y / N

pH of buffers at various temperatures					
Temp	pH7	pH9	Temp	pH7	pH9
10	7.01	9.27	20	6.98	9.19
11	7.01	9.26	21	6.98	9.19
12	7.00	9.25	22	6.97	9.18
13	7.00	9.25	23	6.97	9.17
14	7.00	9.24	24	6.97	9.16
15	7.00	9.23	25	6.97	9.16
16	6.99	9.22	26	6.96	9.15
17	6.99	9.22	27	6.96	9.14
18	6.99	9.21	28	6.96	9.13
19	6.98	9.20	29	6.95	9.13

Standard:	Meter Reading:	<sup>d</sup>	Cell Const:	Time:
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Standard:	Initial Meter Reading:	Initial Cell Const:	Final Cell Const:
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<b>pH Meter</b>				
Slope <sup>a</sup> :		Time:		
mv @ pH 4/10:		Buffer Temp:		
mv @ pH 7:		Response Time <sup>b</sup> :		
	True pH	Meter pH	Time	Recal?
QA Check #1		c		Y / N
QA Check #2		c		Y / N
QA Check #3		c		Y / N

**Footnotes:**  
<sup>a</sup> If <90%, buffers, probe, or cable may be bad.  
<sup>b</sup> If > 45 seconds try 1) cleaning probe, 2) changing cable, 3) new probe  
<sup>c</sup> If meter pH is >± 0.05 units, recalibrate; if > ± 0.10 units, recalibrate, re-read sample, & "J" data since last calibration.  
<sup>d</sup> If meter conductivity is >± 5µs/cm, recalibrate, re-read sample, & "J" data since last calibration.

Standard:	Meter Reading:	<sup>d</sup>	Cell Const:	Time:
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DAY 3							
<b>Cond Meter</b>	Standard:		Initial Meter Reading:		Initial Cell Const:		Final Cell Const:
<b>pH Meter</b>					<u>Comments:</u>		
Slope <sup>a</sup> :			Time:				
mv @ pH 4/10:			Buffer Temp:				
mv @ pH 7:			Response Time <sup>b</sup> :				
	True pH	Meter pH	Time	Recal?			
QA Check #1		c		Y / N			
QA Check #2		c		Y / N			
QA Check #3		c		Y / N			
<b>Cond Meter</b>	Standard:		Meter Reading:	<sup>d</sup>	Cell Const:		Time:

DAY 4							
<b>Cond Meter</b>	Standard:		Initial Meter Reading:		Initial Cell Const:		Final Cell Const:
<b>pH Meter</b>					<u>Comments:</u>		
Slope <sup>a</sup> :			Time:				
mv @ pH 4/10:			Buffer Temp:				
mv @ pH 7:			Response Time <sup>b</sup> :				
	True pH	Meter pH	Time	Recal?			
QA Check #1		c		Y / N			
QA Check #2		c		Y / N			
QA Check #3		c		Y / N			
<b>Cond Meter</b>	Standard:		Meter Reading:	<sup>d</sup>	Cell Const:		Time:

MSDS

CONDUCTIVITY STANDARDS, 5 - 17,000 Micromhos/cm (0.0005 - 1.7 mS/cm) at 25°C; SODIUM CHLORIDE CONDUCTIVITY CONTROL STANDARDS

## ATTACHMENT C

### Material Safety Data Sheet

#### Section 1: Chemical Product and Company Identification

<b>Catalog Number:</b> 2234.075, 2235.09, 2236, 2236.02, 2236.03, 2236.04, 2236.05, 2236.06, 2236.07, 2236.08, 2236.09, 2236.10, 2236.11, 2236.12, 2236.14, 2236.141, 2236.15, 2236.19, 2236.21, 2236.22, 2236.24, 2236.25, 2236.26, 2236.31, 2236.35, 2236.47, 2236.49, 2236.51, 2236.52, 2236.53, 2236.54, 2236.58, 2236.60, 2236.65, 2236.65C, 2236.82, 2236.X, 2236.Y, 2237, 2238, 2238.22, 2239, 2239.34, 2240, 2240.42, 2240.45, 2240.47, 2241, 2242, 2242.3, 2243, 2243.12, 2244, 2244.20, 2244.25, 2244.37, 2244.39, 2244.50, 2245, 2245.60, 2245.70, 2245.8, 2245.90, 2246, 2246.10, 2246.12, 2246.13, 2246.14, 2246.17, 2246.70, C-865, CX-1101, CX-1102, R2236015, R2236035, R2236270, R2236330, R2236400, R2236415, R2236480, R2236515, R2236527, R2238940, R2242350, R2242726, R2242900, R2243005, R2243140, R2243150, R2243180, R2244230, R2245830, R2245850, R2246005, R2246115, R2246141, R2246155, R2246160	
<b>Product Identity:</b> CONDUCTIVITY STANDARDS, 5 - 17,000 Micromhos/cm (0.0005 - 1.7 mS/cm) at 25°C; SODIUM CHLORIDE CONDUCTIVITY CONTROL STANDARDS	
<b>Manufacturer's Name:</b> RICCA CHEMICAL COMPANY LLC	<b>Emergency Contact(24 hr) -- CHEMTREC®</b> Domestic: 800-424-9300 International: 703-527-3887
<b>CAGE Code:</b> 4TCW6, 0V553, 4XZQ2	
<b>Address:</b> 448 West Fork Dr Arlington, TX 76012	<b>Telephone Number For Information:</b> 817-461-5601
<b>Date Prepared:</b> 6/16/98	<b>Revision: 34</b> Last Revised: 04/24/2012 Date Printed: 04/08/2013 12:57:21 pm

#### Section 2. Composition/Information on Ingredients

Component	CAS Registry #	Concentration	ACGIH TLV	OSHA PEL
Sodium Chloride	7647-14-5	< 1% (w/v)	Not Available Not Available	Not Available Not Available
Water, Deionized	7732-18-5	Balance	Not Available Not Available	Not Available Not Available

#### Section 3: Hazard Identification

**Emergency Overview:** Does not present any significant health hazards. May cause irritation. Wash areas of contact with water.

**Target Organs:** eyes, skin

**Eye Contact:** May cause irritation, redness, pain, and tearing.

**Inhalation:** Not likely to be hazardous by inhalation.

**Skin Contact:** May cause irritation, redness, and pain.

**Ingestion:** Large quantity may cause burning of the throat, nausea, vomiting, and abdominal pain

## MSDS

CONDUCTIVITY STANDARDS, 5 - 17,000 Micromhos/cm (0.0005 - 1.7 mS/cm) at 25°C; SODIUM CHLORIDE CONDUCTIVITY CONTROL STANDARDS

**Chronic Effects/Carcinogenicity:** None

**IARC** - No.

**NTP** - No.

**OSHA** - No.

**Reproductive Information:** Reproductive effects cited in 'Registry of Toxic Effects of Chemical Substances' for Sodium Chloride. Reproductive effects cited in 'Registry of Toxic Effects of Chemical Substances' for Sodium Chloride.

**Teratology (Birth Defect) Information:** Mutation data cited in 'Registry of Toxic Effects of Chemical Substances' for Sodium Chloride. Mutation data cited in 'Registry of Toxic Effects of Chemical Substances' for Sodium Chloride.

### Section 4: First Aid Measures - In all cases, seek qualified evaluation.

**Eye Contact:** Irrigate immediately with large quantity of water for at least 15 minutes. Call a physician if irritation develops.

**Inhalation:** Remove to fresh air. Give artificial respiration if necessary. If breathing is difficult, give oxygen.

**Skin Contact:** Flush with plenty of water for at least 15 minutes. Call a physician if irritation develops.

**Ingestion:** Dilute with water or milk. Do not induce vomiting. Call a physician if necessary.

### Section 5: Fire Fighting Measures

**Flash Point:** Not Available.

**Method Used:** Not Available.

**LFL:** Not Available.

**UFL:** Not Available.

**Extinguishing Media:** Use any means suitable for extinguishing surrounding fire.

**Fire & Explosion Hazards:** Not considered to be a fire or explosion hazard.

**Fire Fighting Instructions:** Use normal procedures/instructions.

**Fire Fighting Equipment:** Use protective clothing and breathing equipment appropriate for the surrounding fire.

### Section 6: Accidental Release Measures

Absorb with suitable material and dispose of in accordance with local regulations.

Small amounts may be flushed to the sewer with excess water.

### Section 7. Handling and Storage

As with all chemicals, wash hands thoroughly after handling. Avoid contact with eyes and skin. Protect from freezing and physical damage.

**Safety Storage Code:** General

### Section 8: Exposure Control/Personal Protection

**Engineering Controls:** No specific controls are needed. Normal room ventilation is adequate.

**Respiratory Protection:** Normal room ventilation is adequate.

**Skin Protection:** Chemical resistant gloves.

**Eye Protection:** Safety glasses or goggles.

### Section 9: Physical and Chemical Properties

**pH:** 6.7 - 7.3

**Appearance:** Clear, Colorless Liquid

**Odor:** Odorless

**Boiling Point(°C):** Approximately 100.1

**Solubility in Water:** Infinite

**Melting Point(°C):** Approximately (-6) -0

**Specific Gravity:** Approximately 1 - 1.01

**Vapor Pressure:** Not Applicable.

### Section 10: Stability and Reactivity

**Chemical Stability:** Stable under normal conditions of use and storage.

**Incompatibility:** Strong oxidizing agents, Lithium, Bromine Trifluoride.

**Hazardous Decomposition Products:** Oxides of Sodium and fumes of Chlorine may be released when heated to decomposition.

**Hazardous Polymerization:** Will not occur.



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### Section 11. Toxicological Information

LD50, Oral, Rat: (Sodium Chloride) 3000 mg/kg, details of toxic effects not reported other than lethal dose value.

LD50, Dermal, Rabbit: (Sodium Chloride) >10 gm/kg.

### Section 12. Ecological Information

**Ecotoxicological Information:** No information found.

**Chemical Fate Information:** No information found.

### Section 13. Disposal Considerations

Dilute with water, then flush to sewer if local regulations allow. If not allowed, save for recovery or recycling in an approved waste disposal facility. Always dispose of in accordance with local, state and federal regulations.

### Section 14. Transport Information

Part Numbers:

This product is not regulated.

### Section 15. Regulatory Information (Not meant to be all inclusive - selected regulation represented)

**OSHA Status:** The above items either do not contain any specifically hazardous material or the potentially hazardous material is present in such low concentration that the items do not present any immediate threat to health and safety. These items do not meet the OSHA Hazard Communication Standard (29 CFR 1910.1200) definition of a hazardous material.

**TSCA Status:** All components of this solution are listed on the TSCA Inventory or are mixtures (hydrates) of items listed on the TSCA Inventory

Sara Title III:

**Section 302 Extremely Hazardous Substances:** Not Applicable.

**Section 311/312 Hazardous Categories:** No

**Section 313 Toxic Chemicals:** Not Applicable.

**California:** None Reported.

**Pennsylvania:** None Reported.

**RCRA Status:** Not Applicable.

**CERCLA Reportable Quantity:** None Reported.

**WHMIS:** Not Applicable.

### Section 16. Other Information

NFPA Ratings:

Health: 0

Flammability: 0

Reactivity: 0

Special Notice Key:N one

HMIS Ratings:

Health: 0

Flammability: 0

Reactivity: 0

Protective Equipment: B (Protective Eyewear, Gloves)

## MSDS

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Rev 1, 10-19-98: (Section 1) Specified catalog numbers; (Section 16) removed periods from NFPA and HMIS, added ® to HMIS.

Rev 2, 3-18-99: Added catalog numbers 2236.06, 2236.07, 2236.14, 2240.45, 2242, 2244.20, 2244.39, 2245.70 and 2246.17; (Section 11) specified LD50 information is for Sodium Chloride.

Rev 3, 6-14-99: (Section 1) added catalog numbers 2236.03 and 2236.11.

Rev 4, 9-20-2000: Reformatted from WordPerfect® to Microsoft Word®; (Section 1) revised emergency telephone number to CHEMTREC® 800-424-9300, added catalog numbers 2235.09, 2236.51, 2236.52, 2236.54, 2236.60, 2238.22, 2239.34, 2243.12, 2246.12 and 2246.14, Removed catalog numbers 2236.CA, 2236.CC and 2236.CO; (Section 3) Added Mutation data citing; (Section 4) removed note to physician; (Section 7) added storage code. Rev 5, 10-09-2001: Reformatted to electronic data format.

Rev 6: 12-14-2001: (Section 1) added catalog numbers 2240.47 and 2245.60. Rev 7, 02-06-2002: (Section 1) added catalog number 2236.21. Rev 8, 02-26-2002: (Section 1) added catalog number 2236.24. Rev 9, 03-05-2002: (Section 1) added catalog number 2246.10. Rev 10, 03-21-2002: (Section 1) added catalog number 2242.3. Rev 11, 03-22-2002: (Section 3) removed yellow colored solution reference. Rev 12, 10-02-2003: (Section 1) added catalog number 2236.65C. Rev 13, 01-13-2004: (Section 1) added catalog number R2242900. Rev 14, 2-9-04: (Sec 1) added cat no R2245850. Rev 15, 2-24-04: (Sec 1) added catalog number R2236035. Rev 16, 4-6-04: (Sec 1) added catalog number R2238940. Rev 17, 7-6-04: added cat nos 2234.075 and R2243180. Rev 18, 8-24-04: (Sec 1) added cat no 2236.31. Rev 19, 12-13-04: added cat no R2246141.

Rev 20, 3-31-05: (Sec 1) added cat nos. R2242350, R2243150, 2245.8, R2245830 and R 2246115. Rev 21, 4-6-05: (Sec 1) added cat. no. R2236330. Rev 22, 6-14-05: (Sec 1) added cat nos R2236480 and R2236527. Rev 23, 8-11-05: (Sec 1) added cat nos. R2243005 and R2246005. Rev 24, 3-1-06: (Sec 1) added cat nos. R2236270, R2236515, and R2244230. Rev 25, 4-18-06: (sec 1) added cat no R2243140. Rev 26, 7-5-06: (Sec 1) added cat no R2246155. Rev 27, 7-26-07: (Sec 1) added Red Bird Cat. No. C-865. Rev 28, 8-24-07: (Sec 1) added cat no 2236.22. Rev 29, 10-29-07: (Sec 1) added cat no 2244.37. Rev 30, 5-19-09: (Sec 1) added cat no 2236.141. Rev 31, 12-22-09: (Sec 1) added cat no R2236400. Rev 32, 8-5-10: (Sec 1) added Red Bird cat nos CX- 1101 and CX-1102. Rev 33, 9-13-10: (Sec 1) added cat nos R2236415 and R2242726. Rev 34, 4-24-12: (Sec 1) revised title, added cat nos R2236015 and R2246160, (Sec 9) added pH information.

When handled properly by qualified personnel, the product described herein does not present a significant health or safety hazard. Alteration of its characteristics by concentration, evaporation, addition of other substances, or other means may present hazards not specifically addressed herein and which must be evaluated by the user. The information furnished herein is believed to be accurate and represents the best data currently available to us. No warranty, expressed or implied, is made and RICCA CHEMICAL COMPANY assumes no legal responsibility or liability whatsoever resulting from its use.