



DEPARTMENT OF  
**ECOLOGY**  
State of Washington

## **PBT Trends in Lake Sediment Cores:**

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### **2019 and 2020 Results**

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# **PBT Trends in Lake Sediment Cores:**

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## **2019 and 2020 Results**

by  
Callie Mathieu

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# Abstract

The Washington State Department of Ecology has monitored persistent, bioaccumulative, and toxic (PBT) chemicals in freshwater sediment cores since 2006. This report presents results from the 2019 and 2020 sampling years of the long-term monitoring study. In 2019 we collected sediment cores from Blackmans, Sutherland, and Wildcat Lakes to reconstruct profiles of per- and polyfluoroalkyl substances (PFAS), and in 2020 from Hicks and Spanaway Lakes for analysis of hexabromocyclododecane (HBCD).

PFAS in Blackmans Lake sediments rose between the mid-1950s through the 1990s, where concentrations peaked and then remained stable. PFAS fluxes continued to rise in Blackmans Lake through 2014, and then declined slightly in the years following. PFAS concentrations and fluxes in Wildcat Lake were low but increased from the late 1940s through the late 1980s and then plateaued. No PFAS were detected in Sutherland Lake sediments in any year analyzed; PFAS accumulation in this lake is predominantly via atmospheric deposition.

HBCD increased sharply beginning in the 1980s until peaking in the most recent sediment layer at Hicks Lake and in 2009/2014 at Spanaway Lake, following the pattern of commercial use of HBCD. The temporal trends and concentrations of HBCD observed in Hicks and Spanaway Lakes were similar to those observed in other parts of the world without industrial or textile manufacturing inputs. Hicks Lake had slightly higher concentrations than typically seen in urban residential waterbodies, which may be attributed to the combination of its dense residential watershed and high organic carbon content in sediments.

We recommend continued monitoring of PFAS and HBCD as part of this long-term study, as potential reductions in lake sediments may be observed in future sampling events following major commercial phase outs over the last ten years.

# Introduction

The Washington State Department of Ecology's (Ecology's) Persistent, Bioaccumulative, and Toxic (PBT) Monitoring Program began a study in 2006 to assess environmental trends through age-dated lake sediment cores. This study carries out environmental monitoring of PBTs in Washington State's environment, as recommended in Ecology's PBT Strategy (Gallagher 2000). The state's PBT Rule (WAC 173-333) listed 24 chemicals or chemical groups, and two metals of concern, that met the criteria and laid out a plan to address them through chemical action plans. The goal of the rule was to reduce and phase out uses, releases, and exposures of PBTs in Washington state through recommendations of multi-media and cross-program measures.

The PBT Monitoring Program collects sediment cores from three lakes throughout the state each year to construct historical deposition profiles of PBTs in the environment to see whether levels in the environment are increasing or decreasing. Selected sediment core layers are analyzed for a target analyte from the PBT Rule, as well as a suite of parameters to age-date the sediment layers. On a rotating basis, target PBTs include polycyclic aromatic hydrocarbons (PAHs), per- and polyfluoroalkyl substances (PFAS), hexabromocyclododecane (HBCD), polybrominated diphenyl ethers (PBDEs), and polychlorinated biphenyls (PCBs), and mercury.

This report presents results from the 2019 and 2020 sampling years of the long-term monitoring study. In 2019, the target analyte group was PFAS, and sediment cores were collected from Blackmans, Sutherland, and Wildcat Lakes. In 2020, sediment cores were collected from Hicks and Spanaway Lakes for analysis of HBCD.

## Per- and Polyfluoroalkyl Substances (PFAS)

PFAS are a large group of chemicals containing carbon-fluorine bonds. They are used in many applications for their hydrophobic and lipophobic properties and because they are highly stable. Found in both industrial and consumer products, they are primarily used as water-, stain-, and oil-repellent coatings, as well as in aqueous film-forming foams (AFFFs) and other applications like metal plating suppressants. PFAS are a concern because they are widespread in the environment, and some compounds are highly persistent, bioaccumulative, and toxic.

This study analyzed 15 PFAS (listed in Appendix A): perfluoroalkyl acids (PFAAs) containing carbon chains length of 4-14 and two perfluoroalkane sulfonamides. While the entire class of PFAS includes thousands of individual compounds, the PFAA subgroup are terminal degradation products that are resistant to further breakdown under normal conditions and typically are the substances found in sediments. PFAAs can be described as "long-chain" or "short-chain." The long-chain group includes perfluoroalkyl carboxylates (PFCAs) with eight or more carbons and perfluoroalkyl sulfonates (PFSAs) with six or more carbons. The short-chain group includes PFCAs with seven carbons or less and PFSAs with five carbons or less.

Evidence has suggested carcinogenic potential from exposure to PFAAs, as well as high cholesterol and reproductive, developmental, and immunological effects (US EPA 2016a). Other human health concerns include increased liver enzymes, decreased vaccination response, thyroid



disorders, and pregnancy-induced hypertension and preeclampsia (US EPA 2016b). Human health concerns are primarily associated with the long-chain PFAAs, but short-chain PFAAs have shown similar effects, typically occurring at higher doses (ITRC 2023).

PFAS have been known as ubiquitous contaminants in the environment with a worldwide distribution since the early 2000s (Giesy and Kannan 2001). In Washington State, PFAS have been detected in surface water, sediments, and fish of freshwater systems (Furl and Meredith 2010; Johnson and Friese 2012; Mathieu 2013; Mathieu and McCall 2017), stormwater (Medlen 2018), and marine sediments (Long et al. 2013; Dutch et al. 2021). PFAS are released to the environment as emissions during manufacturing and the use and disposal of products containing PFAS (OECD 2013). PFAS can be transported into and through the environment via stormwater, discrete releases (such as AFFF release), wastewater treatment plant (WWTP) effluent, biosolids applications, landfill leachate, and atmospheric deposition (ITRC 2023).

PFAS have been used since the 1950s, but U.S. manufacturers largely phased out the most highly bioaccumulative substances in the 2000s (perfluorooctane sulfonate [PFOS]) and 2010s (perfluorooctanoate [PFOA] and long-chain PFAAs). In 2022, Ecology and the Washington State Department of Health (DOH) drafted a chemical action plan to address PFAS in the state (Ecology and DOH 2022). Washington state has moved to restrict PFAS in food packaging, AFFFs, cosmetics, carpets, and after-market stain and water-proofing treatments. Ecology is currently considering action on other priority products such as textiles, ski wax, non-stick cookware, cleaning products, and fire-fighter protective gear by 2025.

## Hexabromocyclododecane (HBCD)

HBCD refers to a technical mixture of brominated flame retardants, composed primarily of three isomers: alpha-, beta-, and gamma-. The largest use of HBCD was in extruded and expanded polystyrene (XPS and EPS) foam to provide thermal insulation in buildings. It has also been used as a flame retardant in furniture upholstery and other textiles, and in latex binders, adhesives, and paints (Environment Canada 2011), as well as in solder used in automobile replacement parts (US EPA 2022).

HBCD exhibits high aquatic toxicity and is a human health concern for reproductive, developmental, and neurological effects (US EPA 2010). While HBCD has been manufactured since the 1960s, concern over the persistence, bioaccumulation, and toxicity of HBCD grew in the 2000s and it was listed as a persistent organic pollutant under the Stockholm Convention in 2013. Commercial users of it in the United States began to phase it out during the 2010s. Manufacturing of HBCD in the U.S. fell from 10-50 million pounds over 2007-2011, to 1-10 million pounds per year between 2012-2015 (US EPA 2020). By 2017, it was no longer manufactured or imported domestically in the United States (US EPA 2020).

HBCD has been widely detected throughout the environment, including in air, sediment, water, and aquatic biota (Covaci et al. 2006), and freshwater fish and sediments in Washington State (Johnson and Friese 2012; Mathieu and McCall 2014; Mathieu and Wong 2016). HBCD can enter the environment through the use and disposal of products containing HBCD (US EPA

2010) and can be transported long distances through the atmosphere (Covaci et al. 2006). Diffuse particulate releases to soil can occur during construction and demolition of XPS- or EPS-insulated buildings, and particulates containing HBCD can be transferred to air or stormwater runoff and through WWTP effluent or landfill emissions (US EPA 2010).

# Methods

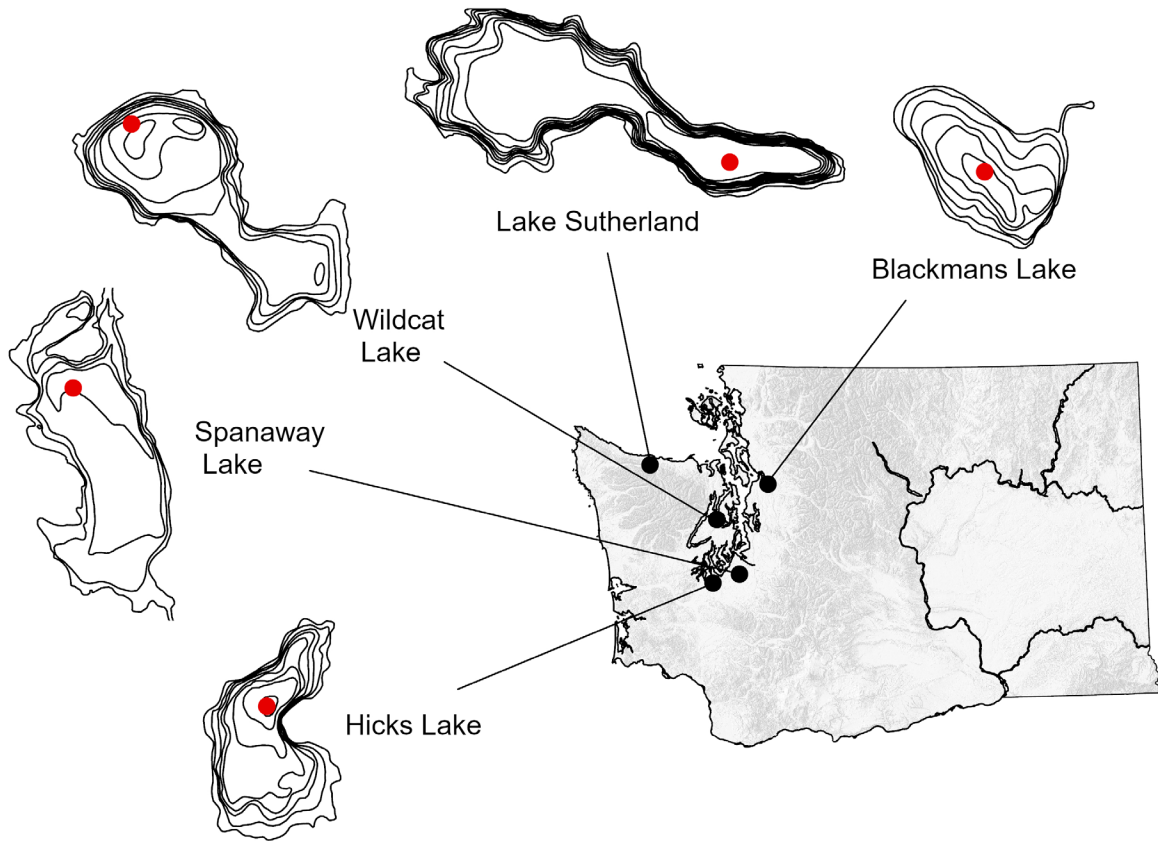
This study was carried out following the Quality Assurance Project Plan (QAPP) (Mathieu 2016), QAPP addendums (Mathieu 2019; Mathieu and Bednarek 2020), and standard operating procedure EAP038 *Collection of Freshwater Sediment Core Samples Using a Box or KB Corer* (Mathieu 2023). Brief descriptions of the project methods are given below.

## Study Locations

Study locations were selected using criteria outlined in the QAPP and described in the QAPP addendums. Figure 1 displays their locations and Table 1 gives relevant physical characteristics. More detailed descriptions of the lakes and their watersheds are provided in Appendix B.

In 2019, sediment cores were collected from our target locations: Blackmans Lake, Lake Sutherland, and Wildcat Lake. Blackmans Lake was selected to capture PFAS concentrations and fluxes in a lake with urban and residential development. Wildcat Lake was chosen based on its proximity to potential AFFF release sites at a nearby military base and a small private airport, although the lake is not hydrologically connected to known AFFF releases. Other potential sources at Wildcat Lake include residential septic systems. Lake Sutherland served as a reference lake for the 2019 sampling year, as it lies in a primarily forested, undeveloped watershed near the edge of Olympic National Forest. Aside from residential sources along the shoreline (such as septic systems) and a state highway near the northern shore, PFAS sources are likely dominated by atmospheric deposition at Sutherland Lake.

In 2020, sediment cores were collected from Hicks Lake and Spanaway Lake. Field staff were unsuccessful in retrieving cores from two out of three target study locations in 2020. Lake Kapowsin was a target listed in the QAPP but contained too much woody debris within the sediments for a proper core and boat launch construction at Leland Lake prevented access to the lake. Spanaway Lake was used as an alternate study location in 2020, as we had permits for the site and it met criteria outlined in the QAPP. We did not collect a core from a third location, due to permit limitations and field conditions. Hicks Lake reflects HBCD inputs in a densely residential watershed and Spanaway Lake represents a mixed-use urban and residential watershed.



**Figure 1. Map of 2019 and 2020 study locations.**

**Table 1. Physical descriptions of the study locations.**

| Collection Year | Waterbody       | County    | Elevation (ft) | Max Depth (ft) | Mean Depth (ft) | Lake Area (ac) | Watershed Area (ac) | WA: LA |
|-----------------|-----------------|-----------|----------------|----------------|-----------------|----------------|---------------------|--------|
| 2019            | Blackmans Lake  | Snohomish | 140            | 29             | 14              | 57             | 518                 | 9      |
| 2019            | Lake Sutherland | Clallam   | 525            | 86             | 57              | 370            | 5,107               | 14     |
| 2019            | Wildcat Lake    | Kitsap    | 377            | 33             | 18              | 120            | 1,600               | 13     |
| 2020            | Hicks Lake      | Thurston  | 162            | 35             | 18              | 160            | 1,152               | 7      |
| 2020            | Spanaway Lake   | Pierce    | 320            | 28             | 16              | 280            | 10,880              | 39     |

ac = acre; WA:LA = watershed area to lake area ratio

## Field Methods

Sediment cores were collected from deep, flat areas of the study lakes—as estimated from lake bathymetric maps and confirmed in the field with a depth sounder. Field staff used a Wildco stainless steel box corer fitted with a 13 cm x 13 cm x 50 cm acrylic liner to collect the core. Sediment cores were retained for subsampling if the desired depth was reached (approximately 35-45 cm) and the sediment-water interface was intact. Field staff siphoned overlying water off the top of the core and then used an extruding device and table to section the core into 1 cm intervals. The top 2 cm of the core were sectioned together to provide enough material for analysis where the water content is greatest, resulting in limited dry weight of sediment.

The box corer and all equipment in contact with the sediments were decontaminated prior to use following the procedures outlined in the QAPP. Field staff wore clean nitrile gloves for sample handling. Samples were kept in coolers in the field and chain of custody was kept throughout the study.

## Sample Preparation and Analysis

Staff processed the sediment core samples by homogenizing 1 cm intervals (2 cm for the top of the core) in stainless steel bowls with stainless steel spoons until the sediments were uniform in color and texture. Subsections of the sediment interval were then placed into jars, stored frozen, and submitted to the laboratory for analyses. Sediment samples for PFAS and HBCD were centrifuged, and overlying water was decanted prior to freezing.

Samples were analyzed for the parameters and by the methods outlined in Table 2. Manchester Environmental Laboratory (MEL) conducted analyses of PFAS, total lead, and total organic carbon. Contract laboratories analyzed HBCD (SGS AXYS Analytical Services Ltd.) and radiochemistry (Eurofins TestAmerica). Sample results were reported as dry weight.

**Table 2. Analytical Methods.**

| Parameter         | Sampling Year | Sample Prep Method         | Analytical Method      | Instrument                               | Laboratory                      |
|-------------------|---------------|----------------------------|------------------------|--|---------------------------------|
| PFAS              | 2019          | AOAC2007.01 modified       | EPA SW8321B modified   | LC-MS/MS isotopic dilution               | MEL                             |
| HBCD              | 2020          | Laboratory-specific method | MLA-070 Rev 02         | LC-MS/MS isotopic dilution               | SGS AXYS Analytical, Sydney BC  |
| T-Pb              | 2019, 2020    | EPA 6020                   | EPA 6020               | ICP-MS                                   | MEL                             |
| TOC               | 2019, 2020    | SM5310B (preacidified)     | SM5310B (preacidified) | TOC-V <sub>C<sub>SH</sub></sub> analyzer | MEL                             |
| <sup>210</sup> Po | 2019, 2020    | Alpha spectroscopy         | HASL-300 A-01-R        | Alpha spectroscopy                       | Eurofins TestAmerica, St. Louis |
| <sup>226</sup> Ra | 2019, 2020    | Gamma spectroscopy         | HASL-300 A-01-R        | Alpha spectroscopy                       | Eurofins TestAmerica, St. Louis |

T-Pb = total lead; TOC = total organic carbon; <sup>210</sup>Po = 210 polonium isotope; <sup>226</sup>Ra = 226 radium isotope

## Data Quality

All data for this project were accepted and deemed usable as qualified. For the 2019 samples, all measurement quality objectives (MQOs) were met with the following exceptions:

- **Polonium isotope (<sup>210</sup>Po) analysis:** One out of three lab duplicate pairs had a relative percent difference (RPD) greater than 20%. The RPD was 22% for this pair and the associated field sample was qualified “J”—indicating an estimated value. The source sample was used for age-dating analysis as qualified, and this exceedance is unlikely to affect the results.
- **PFAS analysis:** Several analytes had matrix spike and surrogate recoveries higher than MQOs. This did not affect results because the analytes were undetected in the associated samples and no qualifications were made. Likewise, perfluorotridecanoate (PFT<sub>r</sub>DA) had a laboratory control sample (LCS) recovery greater than MQOs but was not detected in the samples and no qualifiers were added. It was determined, high recoveries were due to a combination of matrix effects and the low solubility of the analytes in the solvent mix.

MQOs for the 2020 samples were met except for a high LCS recovery for <sup>226</sup>Ra. Affected results were qualified “J” as estimates. The <sup>226</sup>Ra results are ancillary data used to support interpretation of the sediment cores and this exceedance is not likely to affect the results of this study.

## Data Processing

All cores were age-dated using  $^{210}\text{Pb}$  activities and the constant rate of supply (CRS) model (Appleby and Oldfield 1978).  $^{210}\text{Po}$  was measured in our samples via alpha spectroscopy and assumed to be in equilibrium with  $^{210}\text{Pb}$  and is referred to as  $^{210}\text{Pb}$  from here on out. Supported  $^{210}\text{Pb}$  values were calculated as an average activity at deep intervals where there was no further apparent decline. Three intervals throughout the core were analyzed for  $^{226}\text{Ra}$  activities by alpha spectroscopy, and those values helped confirm the supported  $^{210}\text{Pb}$  values. Percent solids were used to calculate dry mass and an assumed sediment density of  $2.7 \text{ g/cm}^3$  was used based on other Washington State coring studies (Paulson 2004).

# Results and Discussion

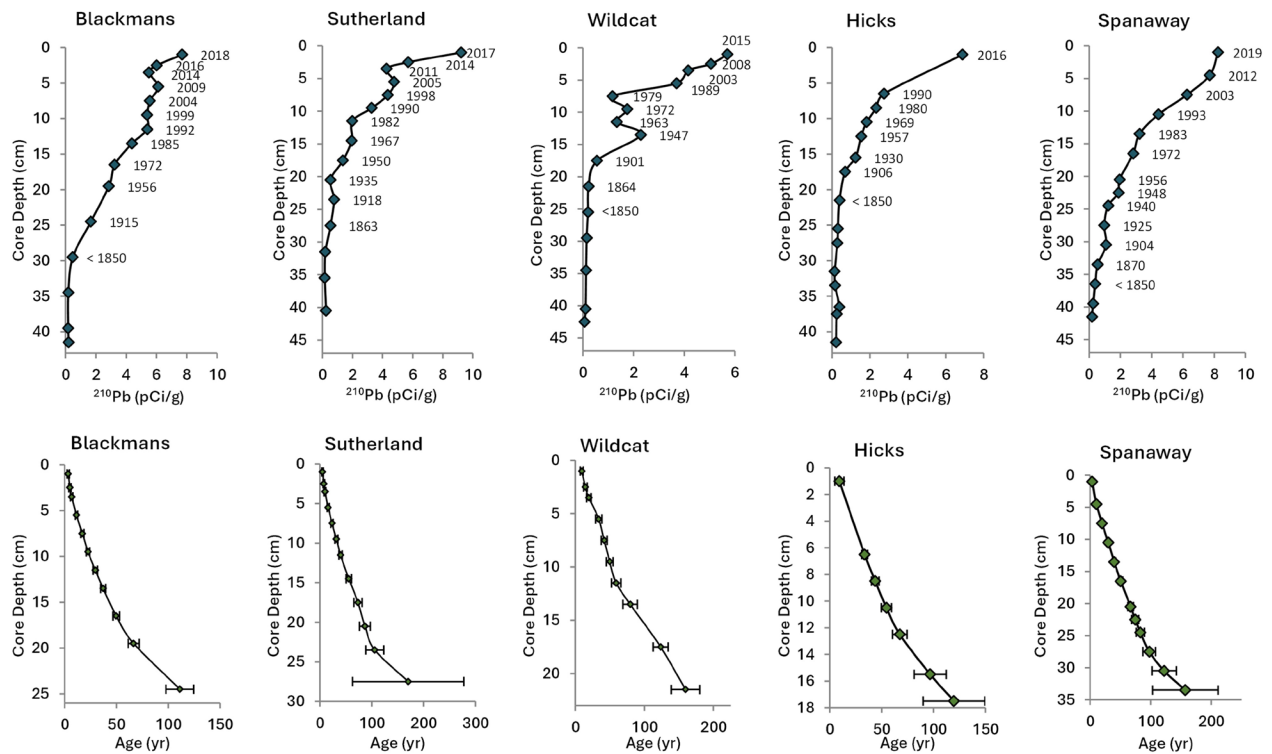
## Core Dating

Figure 2 displays the estimated dates and associated dating uncertainties for the sediment cores. Average supported  $^{210}\text{Pb}$  activities used in the cores ranged from 0.11-0.23 pCi/g. Average  $^{226}\text{Ra}$  activities from three intervals selected throughout the cores (upper, middle, and lower) agreed well with the supported  $^{210}\text{Pb}$  values: 0.18-0.26 pCi/g.

Yearly unsupported  $^{210}\text{Pb}$  fluxes estimated from the core dating models were in the range of 0.10-0.24 pCi/cm<sup>2</sup>/yr. These values were used to calculate a “focus factor” by dividing the core estimates by fluxes calculated from lake-specific precipitation values and atmospheric  $^{210}\text{Pb}$  deposition measured in Washington State (Lamborg et al. 2013). Focus factors are applied to contaminant fluxes throughout this report to correct for the focusing of fine-grained sediments to coring sites or the transport of sediments away from the coring sites. The use of focus-corrected contaminant fluxes allows us to directly compare results among lakes. Focus factors were 0.78, 0.81, 0.67, 0.38, and 1.1 for Blackmans, Sutherland, Wildcat, Hicks, and Spanaway Lakes, respectively.

A decrease in  $^{210}\text{Pb}$  activities occurring at sediment depths of 7-12 cm in the Wildcat Lake core is likely a result of post-depositional mixing. This can occur from bioturbation, water movement, or a slumping of older sediment on top of newer sediment. The model used to date the core is fairly robust for mixing of this magnitude, with estimates of errors of 2 years occurring with a mixing zone spanning 10 years (Appleby 2001).



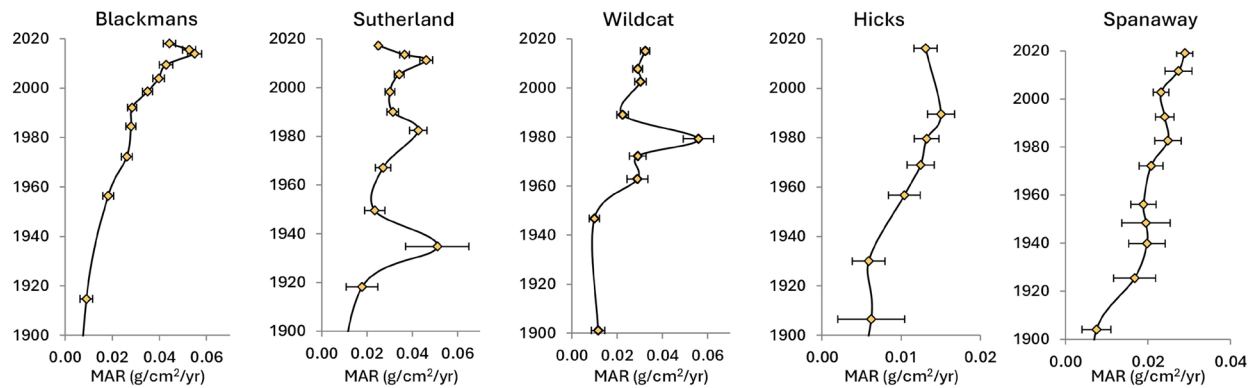


**Figure 2.**  $^{210}\text{Pb}$  Activities plotted against sediment core depth (upper row) and modeled age of sediment core depth with associated uncertainties (bottom row).

## Sediment Accumulation Rates and TOC

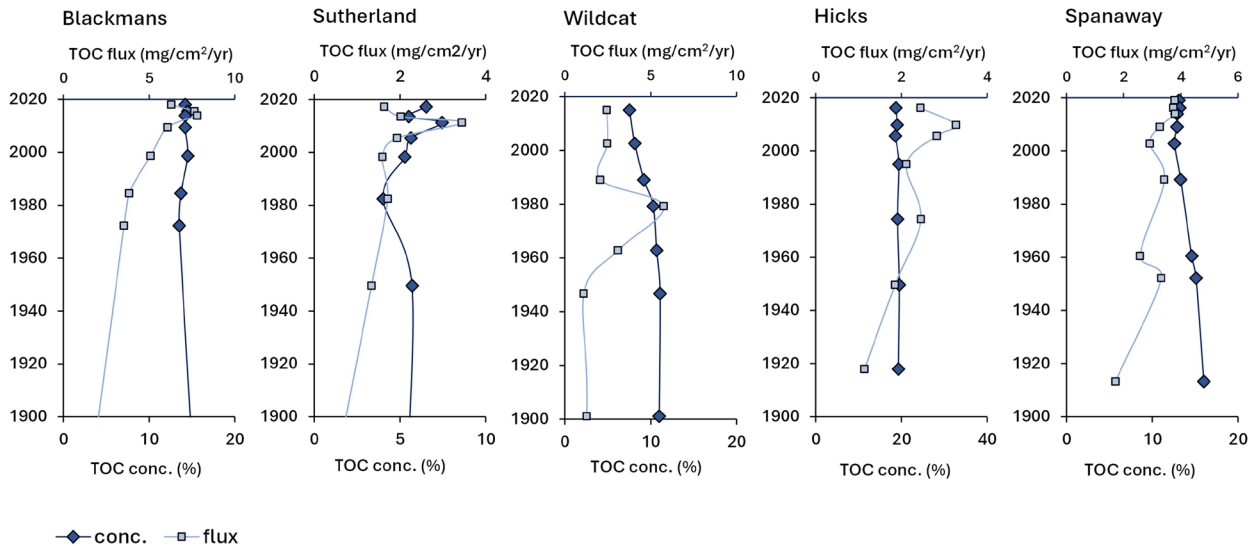
Figure 3 presents the sediment mass accumulation rates (MARs) estimated for the 2019 and 2020 lake sediment cores. Rates were calculated using the CRS model for all cores. Sediment MARs were generally low across the lakes, with average MARs ranging from 0.10-0.03 g/cm<sup>2</sup>/yr.

Sediment accumulation rates began increasing slightly between the early 1900s and 1920s for most sites. The rate of sediment accumulation increased between the 1960s and the top of the core for Blackmans, Hicks, and Spanaway Lakes, likely due to increased watershed development during those years. Sutherland and Wildcat Lakes saw periods of increased sediment accumulation at different points in time: 1930s for Sutherland and 1970s in the Wildcat Lake core. However, the sudden increase in sediment MAR at Wildcat Lake in the 1970s is likely due to the post-depositional mixing discussed in the previous section and should be interpreted with caution.



**Figure 3. 2019 and 2020 profiles of estimated sediment mass accumulation rates (MAR).**

Total organic carbon (TOC) profiles in the 2019 and 2020 sediment cores are shown in Figure 4. Concentrations of TOC changed little over the length of the core at Blackmans, Hicks, and Spanaway Lakes, ranging from 13-20%. Wildcat Lake also had stable TOC, though concentrations declined between 1960 and the top of the core (range = 7.5-11%). Lake Sutherland TOC remained within a range of 4.0-6.5%, with an uptick in organic carbon occurring around 2014.



**Figure 4. 2019 and 2020 Profiles of TOC concentrations (dark blue diamonds) and fluxes (light blue squares).**

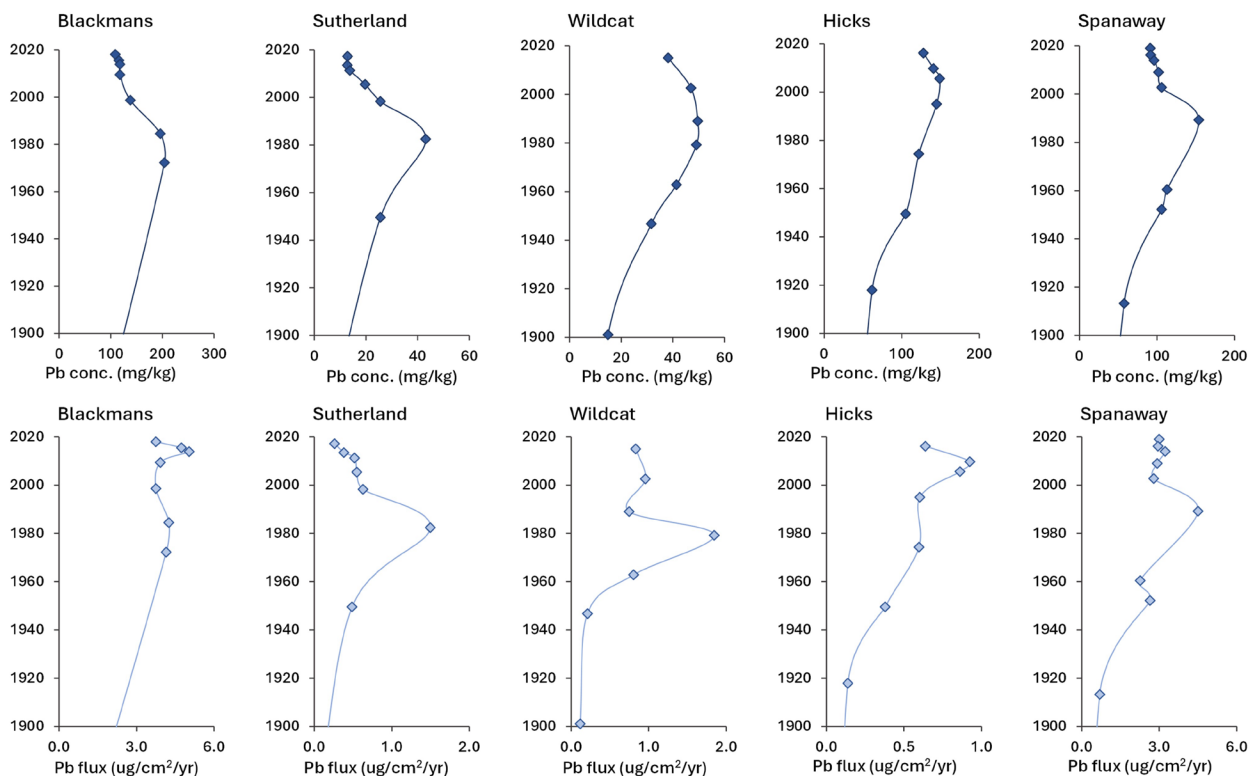
## Lead Profiles

Total lead was measured in ten intervals of each sediment core. Figure 5 displays profiles of total lead concentrations and fluxes for the 2019 and 2020 cores.

Sediment lead accumulation began increasing in the first half of the 20<sup>th</sup> century at all lakes. Lead concentrations and fluxes peaked around the 1980s at Sutherland, Wildcat, and Spanaway Lakes and then declined over the next several decades. Lead concentrations peaked in 1972 at Blackmans Lake and then declined steadily between the 1980s and 2019, but fluxes remained relatively unchanged after the 1970s apart from a spike of increased lead fluxes around 2014. Hicks Lake sediments showed increasing lead concentrations and fluxes throughout the 1900s, with peaks in the 2000s (2006 and 2010, respectively) and a slight decrease in the top-most sediment layer.

Background concentrations of lead were captured in the pre-1850 sediments of all cores (1.6-4.6 mg/kg). Lead concentrations and fluxes remained low in Sutherland and Wildcat Lake sediments throughout the core. Modern lead concentrations in Sutherland and Wildcat averaged 15 and 43 mg/kg, after peaks of 43 and 50 mg/kg, respectively. In contrast, Blackmans, Hicks, and Spanaway Lakes contained sediments with somewhat elevated lead levels, reflective of their locations in developed watersheds. Modern (post-2000) lead concentrations in these lakes ranged from 97-115 mg/kg. Peak concentrations in Hicks, Spanaway, and Blackman Lakes were 149, 154, and 204 mg/kg, respectively.

Lead fluxes were generally low in the Sutherland, Wildcat, and Hicks cores, with modern flux ranges of 0.4-0.9  $\mu\text{g}/\text{cm}^2/\text{yr}$  and peak fluxes of 0.9-1.8  $\mu\text{g}/\text{cm}^2/\text{yr}$ . Fluxes in Blackmans and Spanaway Lakes were higher, with modern fluxes of 3.0-4.4  $\mu\text{g}/\text{cm}^2/\text{yr}$  and peaks of 4.5-5.0  $\mu\text{g}/\text{cm}^2/\text{yr}$ .



**Figure 5. 2019 and 2020 profiles of total lead concentrations (top row) and total lead fluxes (bottom row).**

## Per- and Polyfluoroalkyl Substances (PFAS)

Sediment cores collected from Blackmans, Sutherland, and Wildcat Lakes were analyzed for PFAS in 2019. Fifteen PFAS were measured in 12 sediment intervals of each core. Appendix A gives a complete list of PFAS analyzed and their reporting limits. Total PFAS (T-PFAS) concentrations were calculated as the sum of all detected PFAS, including “J” qualified values.

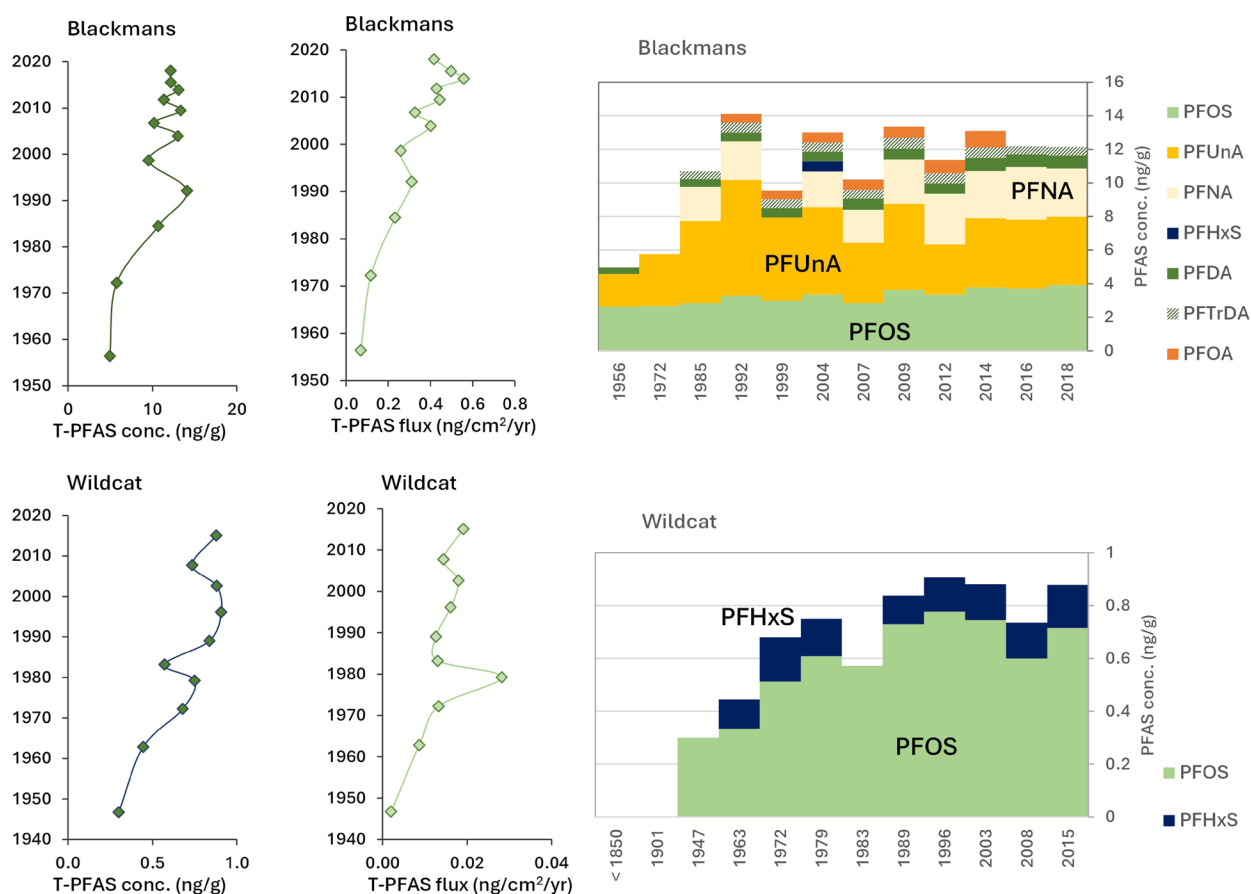
### T-PFAS profiles

At least one PFAS was detected in all sediments from Blackmans and Wildcat Lakes deposited after the mid-1940s. No PFAS were detected in sediments dated prior to that in the Wildcat Lake core, and we did not test any sediments older than that at Blackmans Lake. Lake Sutherland sediments did not contain PFAS at or above the method detection limits of 0.1-0.5 ng/g for any interval analyzed.

Figure 6 displays the dated profiles of sediment T-PFAS for Blackmans and Wildcat Lakes. Blackmans Lake PFAS concentrations and fluxes rose between the mid-1950s through the 1990s, where concentrations peaked and then varied but remained within a stable range of 9-13 ng/g (modern average = 12 ng/g). T-PFAS fluxes continued to rise between the 1990s and 2014, and then declined slightly in the years following. Modern T-PFAS fluxes to Blackmans Lake ranged from 0.40-0.56 ng/cm<sup>2</sup>/yr, with an average of 0.44 ng/cm<sup>2</sup>/yr. Sediments in Blackmans Lake were high in total organic carbon compared to other sites (14-15%), which may explain the

higher than expected PFAS concentrations. The PFAS detected tend to sorb onto organic carbon (Higgins and Luthy 2006; Ahrens et al. 2009) and positive correlations have been reported between PFAS and organic carbon in sediments (Mussabek et al. 2019).

Wildcat Lake sediments contained low levels of PFAS. T-PFAS concentrations generally increased from the late 1940s through the late 1980s and then remained within a narrow range of 0.7-0.9 ng/g through 2019. The modern T-PFAS concentration was 0.83 ng/g. T-PFAS fluxes followed a similar pattern. An influx of PFAS occurring in the 1970s can be seen in the graph, though this is likely due to mixing in the core and may not represent an actual flux increase. Fluxes appeared to continue to increase slightly between the 1990s and 2019, though values remained low throughout the core. Modern fluxes ranged from 0.014-0.019 ng/cm<sup>2</sup>/yr (average of 0.18 ng/cm<sup>2</sup>/yr).



**Figure 6. T-PFAS concentrations, T-PFAS fluxes, and individual PFAS profiles in Blackmans Lake (top row) and Wildcat Lake (bottom row) sediment cores.**

## Individual PFAS profiles

In Blackmans Lake, PFOS, perfluoroundecanoate (PFUnA), and perfluorononanoate (PFNA) were the dominant PFAS in sediments. PFOS and PFUnA were present since the 1950s, with PFUnA peaking in concentration in the early 1990s at 6.9 ng/g and ranging from 2.0-5.2 ng/g throughout the other intervals. PFOS remained steady throughout the core at concentrations of 2.6-3.9 ng/g, making up between 25-50% of the total. The relative abundance of PFUnA compared to PFOS is higher than typically observed in sediments. PFUnA is found as a breakdown product from carpets, stain-proofing for textiles, some AFFFs, and food packaging (Ogunsuyi et al. 2023). The source of PFUnA to Blackmans Lake is unclear.

PFNA first appeared in the 1980s in the Blackmans Lake core and was generally present around 3.0 ng/g from then until the present and contributed around 20% of the total PFAS concentration. Perfluorodecanoate (PFDA) and PFTrDA were also detected frequently in the core samples beginning in the 1980s, at relatively low concentrations (less than 1 ng/g). PFOA was detected at low levels (less than 1 ng/g), and only between 1992-2014.

The total PFAS burden in Wildcat Lake sediments was almost completely made up of PFOS. PFOS accounted for 70-100% of the total PFAS concentration between the 1950s through the top of the core. However, concentrations of PFOS were still low, at less than 1.0 ng/g. Perfluorohexane sulfonate (PFHxS) was the only other compound detected and first appeared in the 1960s. Concentrations of PFHxS were below 0.2 ng/g. The presence of PFHxS co-occurring with PFOS in sediments has been suggested as an indicator of AFFF sources (D'Agostino and Mabury 2017; MacInnis et al. 2019).

Out of the 15 PFAS analyzed in the sediment core samples, only the seven discussed above were detected. Perfluoroalkyl carboxylates (PFCAs) with carbon chain lengths of 9 or more and PFSAAs tend to sorb to sediments and are well preserved in sediments, making them appropriate for tracking temporal trends through sediment cores (Codling et al. 2018). The other PFAS analyzed, such as short chain PFCAs, PFOA, and precursor compounds are more water-soluble and generally are not detected in sediments in non-point source areas. This limitation was recognized in the QAPP for this study, with the awareness that environmental trends of short chain compounds would likely not be captured in sediment cores.

## Comparison to other studies

### *Trends*

The Blackmans and Wildcat Lake cores show similar temporal trends seen in other areas of North America. PFAS levels also increased from the 1950s-1970s through the early to mid-2000s in sediment cores collected from the Great Lakes (Yeung et al. 2013; Codling et al. 2014; Guo et al. 2016; Codling et al. 2018), Canadian Arctic lakes (MacInnis et al. 2019), and the Puget Sound (Strivens et al. 2021). Previous Washington lake sediment cores have also showed consistent increases over this period, with maximum concentrations occurring in the top layer (2010) (Mathieu 2013). Whereas, in sediment cores collected in Europe, PFAS concentrations peaked earlier, in the late 20<sup>th</sup> century (Langberg et al. 2021; Mourier et al. 2019).

## **Concentrations and fluxes**

PFAS concentrations in Sutherland and Wildcat Lakes (non-detects and less than 1 ng/g, respectively) were low and comparable to levels observed in remote or background areas. Balgooyen et al. (2022) reported a background value of 0.21 ng/g for PFAS in bed sediment of tributaries in Wisconsin. Sediment cores collected from remote lakes in the Great Lakes area contained similar peak T-PFAS (sum of PFAAs) concentrations (1.7-3.7 ng/g) and fluxes (0.017-0.021 ng/cm<sup>2</sup>/yr) to Wildcat Lake (0.9 ng/g and 0.028 ng/cm<sup>2</sup>/yr) (Guo et al. 2016). Codling et al. (2018) reported a range of T-PFAS in surface sediments collected from the Great Lakes of 1.5-4.6 ng/g (Codling et al. 2018). Compared to previous sediment cores in Washington, Wildcat Lake had very similar concentrations and fluxes to Lake Stevens in Snohomish County, which is surprising given the higher urban density surrounding Lake Stevens compared to the more forested Wildcat Lake (Mathieu 2013).

Blackmans Lake PFAS levels were comparable to those recorded in West Medical Lake in Spokane County (especially fluxes). West Medical Lake receives WWTP discharges and has a long water residence time, which is likely why PFAS levels are so high in the lake. Both lakes are eutrophic, though sediment accumulation rates were much higher for West Medical Lake. The Blackmans Lake surface sediment PFAS concentration and flux of 12 ng/g and 0.42 ng/cm<sup>2</sup>/yr were also comparable to a Lake Ontario core profile, which reached a peak of 13 ng/g and 0.41 ng/cm<sup>2</sup>/yr in the top sediment layer (Guo et al. 2016). The authors noted that sources to Lake Ontario include diffuse inputs from a large population and commercial/industrial base, similar to residential and commercial land uses surrounding Blackmans Lake.

While Blackmans Lake contained high PFAS concentrations, they were not as high as levels seen in areas of point-source PFAS contamination—like fire training areas and manufacturing releases. Mussabek et al. (2019) collected a sediment core near an AFFF-impacted site and found max concentrations of 76 ng/g. In France, a sediment core collected downstream of a fluoropolymer plant reached a peak T-PFAS concentration of 51 ng/g in the 1990s (Mourier et al. 2019), four times higher than peak levels found in Blackmans Lake.

# Hexabromocyclododecane (HBCD) Profiles

Three HBCD isomers (alpha-, beta-, and gamma-) were measured in nine sediment intervals from each core collected at Hicks Lake and Spanaway Lake. Total HBCD concentrations given below are sums of all three isomers. Total values include only detected results and include “J” qualified values.

## T-HBCD profiles

Figure 7 displays the HBCD profiles for Hicks and Spanaway Lakes. HBCDs were consistently detected in both lake sediment cores. Profiles from the two lakes showed sediment T-HBCD concentrations and fluxes were low until increasing in the 1980s. This is consistent with the beginning of the use of HBCD in insulation boards (UNEP 2010) which led to dramatic increases in the environment.

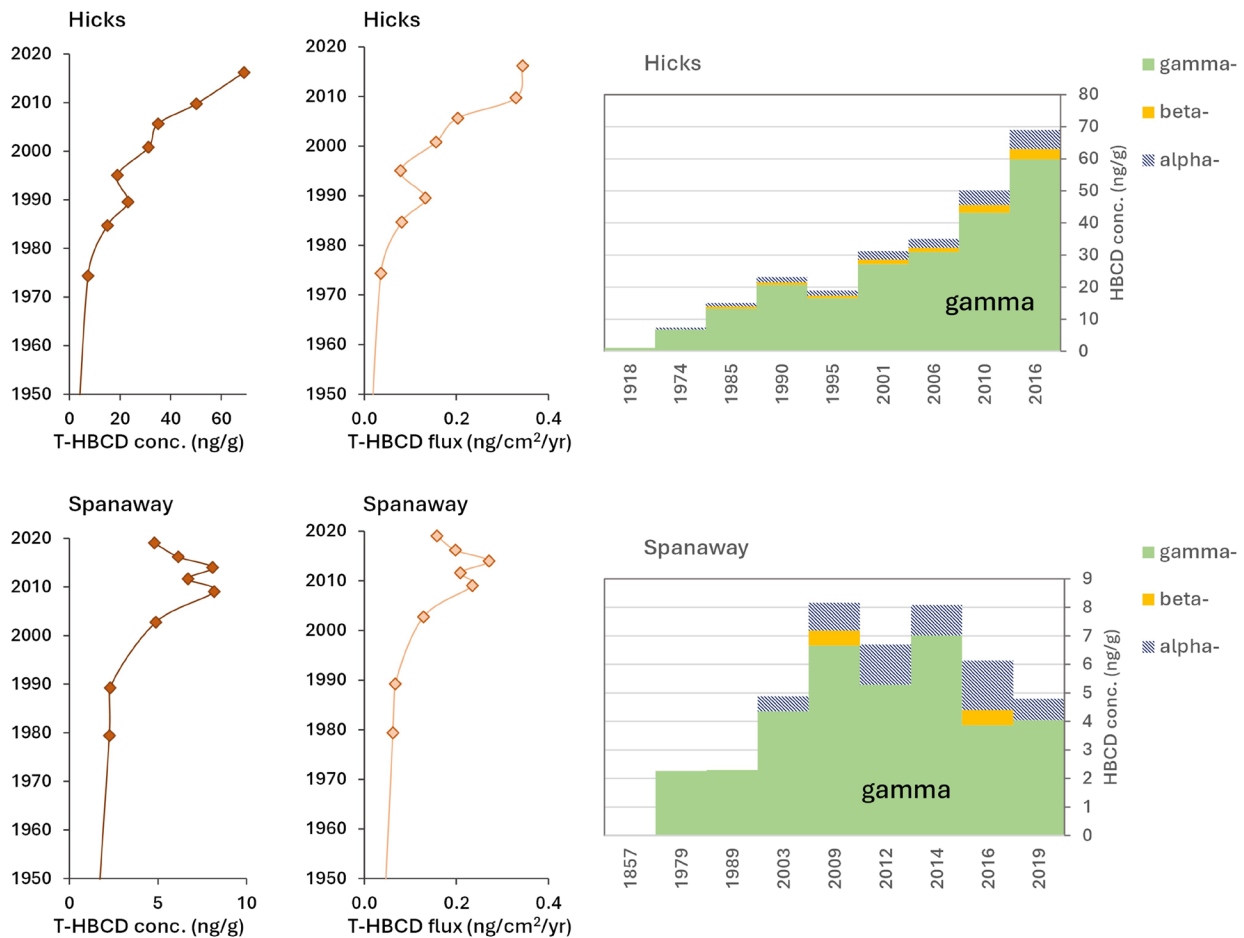
Hicks Lake T-HBCD concentrations and fluxes continued to increase through the top of the core, with the maximum occurring in the most-recent sediments, at 69 ng/g and 0.34 ng/cm<sup>2</sup>/yr. HBCD was detected at a very low concentration in a sediment layer dated 1918, which may be due to pore-water migration from the upper sediments since HBCD wasn't used commercially until the 1960s (UNEP 2010).

In Spanaway Lake, T-HBCD peaked between 2009 and 2014, at around 8 ng/g in concentration and 0.2 ng/cm<sup>2</sup>/yr in fluxes, and then declined between 2014 and the top of the core (2019). This reduction in sediment HBCD concentrations and fluxes is consistent with U.S. manufacturers phasing out use of it during the 2010s. The top sediment layer had a T-HBCD concentration of 4.8 ng/g and a flux of 0.14 ng/cm<sup>2</sup>/yr.

## Individual HBCD profiles

The gamma- isomer was the dominant HBCD in all sediments analyzed, making up 86-90% of the total HBCD burden, on average across both cores. Alpha-HBCD contributed around 7-9% of the total in the Hicks Lake samples and between 10-28% in most Spanaway Lake samples. Beta-HBCD contributed the least in both cores, averaging 2-3% of the T-HBCD concentration. This isomeric make-up is close to that of commercial mixtures which are comprised of gamma- (75-89%), alpha- (10-13%), and beta- (1-12%) (Covaci et al. 2006; Marvin et al. 2011).





**Figure 7. T-HBCD concentrations, T-HBCD fluxes, and individual HBCD profiles in Hicks Lake (top row) and Spanaway Lake (bottom row) sediment cores.**

## Comparison to other studies

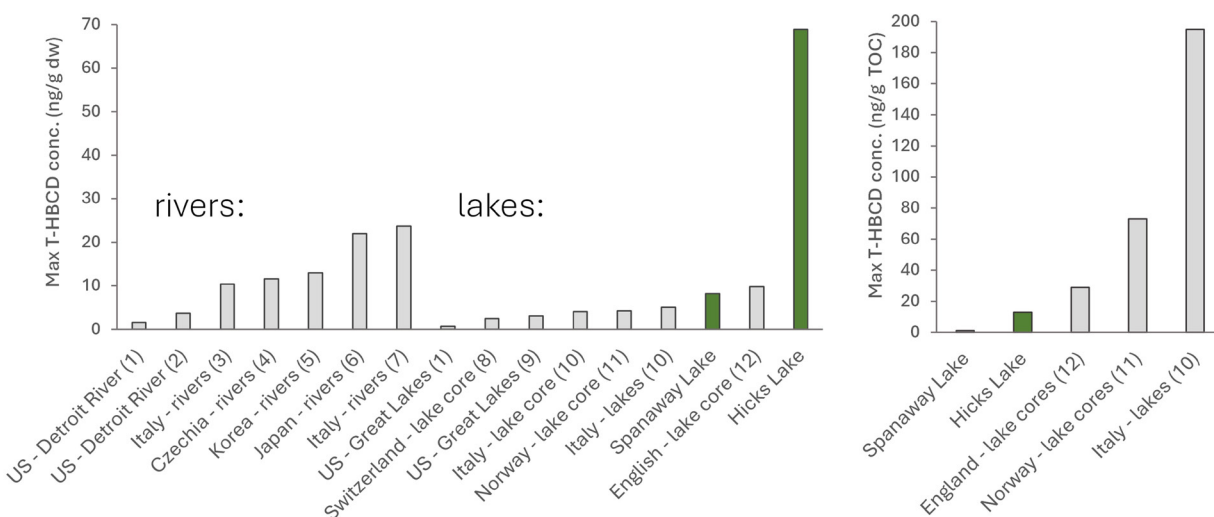
### *Trends*

Increases in HBCD levels starting in the 1980s-1990s and rising consistently through the top of the core has also been demonstrated with sediment cores collected in Australia, Europe, and Asia (Drage et al. 2015; Kohler et al. 2008; Al-Odaini et al. 2015). This reflects the growing commercial use of it beginning in the 1980s and 1990s and increases in market share following the phase out of other brominated flame retardants. Two studies in Europe reported only detecting HBCD in the top sediments of a core, suggesting recent use of it in the watershed (Poma et al. 2014; Evenset et al. 2007). Previous Washington State sediment cores also showed this pattern of HBCD appearing in the 1980s and increasing through the top of the core (Mathieu and McCall 2014).

Sediment cores collected in several lakes of England showed peak HBCD inputs occurring at various points in the 1980s, 1990s, and 2000s, with declines from peak levels at all cores except for the most urban site (Yang et al. 2016).

## Concentrations and fluxes

Peak T-HBCD concentrations observed in this study are comparable to freshwater sediments reported from other areas with diffuse or mixed (non-industrial) sources (Figure 8). Hicks Lake was on the high end of the range of maximum non-industrial HBCD values reported in the literature, but 1-2 orders of magnitude below values reported from areas with known inputs from textile manufacturing (La Guardia et al. 2012; Oh et al. 2014). The high levels in Hicks Lake sediments are likely due to dense residential development in the watershed and high organic carbon content of the sediments. Organic carbon-normalized HBCD concentrations in Hicks Lake were similar to or lower than organic carbon-normalized values reported in lakes from Europe (Yang et al. 2016; Poma et al. 2014; Evenset et al. 2007).



**Figure 8. Maximum T-HBCD concentrations reported in freshwater sediments from diffuse or mixed sources on a ng/g dry weight basis (left pane) and organic carbon basis (right pane).**

References: (1) Letcher et al. 2015; (2) Marvin et al. 2006; (3) Luigi et al. 2015; (4) Stiborova et al. 2017; (5) Jeong et al. 2014; (6) Oh et al. 2015; (7) Poma et al. 2014; (8) Kohler et al. 2008; (9) Yang et al. 2012; (10) Poma et al. 2014; (11) Evenset et al. 2007; (12) Yang et al. 2016.

In comparison to previous Washington sediment cores, Spanaway Lake HBCD concentrations were similar to those observed in Cavanaugh Lake (Skagit County) and Kitsap Lake (Kitsap County) (Mathieu and McCall 2014). Both Cavanaugh and Kitsap are Puget Sound lowland lakes with similar potential sources to Spanaway. Hicks Lake contained higher concentrations than any other Washington sediment core, but fluxes were similar to or lower than other lakes.

# Conclusions

This report presents results from the 2019 and 2020 sampling years of a long-term monitoring study to evaluate PBT trends in lake sediments of Washington State. The following conclusions were made:

## Per- and polyfluoroalkyl substances

- PFAS concentrations and fluxes in Blackmans Lake sediments rose between the mid-1950s through the 1990s, where post-1990 concentrations had a stable trend, despite some variability. PFAS fluxes continued to rise in Blackmans Lake between the 1990s and 2014, and then declined slightly in the years following.
- PFAS concentrations in Wildcat Lake were low, but generally increased from the late 1940s through the late 1980s and then remained stable through 2019. PFAS fluxes followed a similar pattern.
- No PFAS were detected in Sutherland Lake sediments in any year analyzed. This is a mostly remote and undeveloped waterbody where the major sources expected were atmospheric deposition.
- Sutherland and Wildcat Lake PFAS concentrations (non-detects and less than 1 ng/g, respectively) would be considered low compared to other areas. Blackmans Lake contained higher PFAS concentrations, but not at levels typical of highly contaminated waterbodies, like near fire training areas and manufacturing facilities.

## Hexabromocyclododecane

- Sediment core profiles showed HBCD concentrations and fluxes increased sharply in the 1980s at both Hicks and Spanaway Lakes, until peaking in the most recent sediment layer at Hicks and in 2009/2014 at Spanaway Lake. This pattern follows trends in commercial use of HBCD.
- Sediments from both lakes had an isomeric pattern similar to commercial mixtures, with abundance in the order of  $-\gamma > -\alpha > -\beta$ .
- The temporal trends and concentrations of HBCD observed in Hicks and Spanaway Lakes were similar to those observed in other parts of the world lacking industrial or textile manufacturing inputs. Hicks Lake had higher concentrations than typically seen in urban residential waterbodies, which may be attributed to the combination of its dense residential watershed and high organic carbon content in sediments.

# Recommendations

Results of this 2019 and 2020 study support the following recommendations.

- Continue monitoring trends of PFAS and HBCD in the environment as part of the long-term sediment core program. With major phase outs of both chemical groups occurring over the last ten years, potential reductions in lake sediments might be observed in future sampling events.
- Future PFAS analyses should use U.S. Environmental Protection Agency (US EPA) Method 1633 to achieve lower detection limits. Reporting limits for this study were too high to capture low-level signals from the background lake site. Greater sensitivity will allow for detections in lakes where atmospheric deposition is likely the only or primary source.

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# Glossary, Acronyms, and Abbreviations

## Glossary

**CRS Model:** The Constant Rate of Supply (CRS) Model is applied to  $^{210}\text{Pb}$  measurements in sediment core samples to estimate dates and varying sedimentation rates (Appleby and Oldfield 1978). The model works by measuring the difference in supported and unsupported  $^{210}\text{Pb}$  in sediment horizons, and the relation of that difference to the inventory of unsupported  $^{210}\text{Pb}$  of the whole core. Using the known half-life (22.3 years) of  $^{210}\text{Pb}$  and the amount of the unsupported isotope, the rate of sedimentation and the date of formation can be calculated for approximately the last 150 years.

**Flux:** Flux is an estimated rate of net deposition of a contaminant to the lake. Flux rates normalize the variance involved with interpreting dry weight concentrations under varying sedimentation rates. Contaminant flux rates were calculated as the product of the sediment mass accumulation rate and dry weight contaminant concentration.

**Focus Factor:** A focus factor corrects for the focusing of fine-grained sediments to the coring location or the transport of sediments away from coring sites. Sediment cores for this study are often collected in the deepest part of the lake because fine-grained sediments preferentially deposit in these areas.

**Supported  $^{210}\text{Pb}$ :** Supported  $^{210}\text{Pb}$  is represented by the small amount of precursor gas  $^{222}\text{Rn}$  (radon) that is captured in soils. Supported  $^{210}\text{Pb}$  in this study was estimated as the average  $^{210}\text{Pb}$  value at deep intervals where it appeared to no longer decline.

**Unsupported  $^{210}\text{Pb}$ :** Unsupported  $^{210}\text{Pb}$  represents the atmospherically deposited  $^{210}\text{Pb}$  resulting from the decay of  $^{222}\text{Rn}$  that escapes into the atmosphere. Unsupported  $^{210}\text{Pb}$  in this study was estimated by subtracting supported  $^{210}\text{Pb}$  from total  $^{210}\text{Pb}$  at a given depth.

## Acronyms and Abbreviations

|           |  |
|-----------|--|
| AFFF      | aqueous film-forming foam                    |
| CRS       | constant rate of supply                      |
| Ecology   | Washington State Department of Ecology       |
| EPA       | U.S. Environmental Protection Agency         |
| EPS       | expanded polystyrene                         |
| HBCD      | hexabromocyclododecane                       |
| LCS       | laboratory control sample                    |
| MARs      | mass accumulation rates                      |
| MDL       | method detection limit                       |
| MEL       | Manchester Environmental Laboratory          |
| MQO       | measurement quality objective                |
| MS        | matrix spike                                 |
| MSD       | matrix spike duplicate                       |
| N-EtFOSAA | N-ethyl perfluorooctane sulfonamide acetate  |
| N-MeFOSAA | N-methyl perfluorooctane sulfonamide acetate |
| PAHs      | polycyclic aromatic hydrocarbons             |

|         |  |
|---------|--|
| Pb      | lead                                   |
| PBDEs   | polybrominated diphenyl ethers         |
| PBT     | persistent, bioaccumulative, and toxic |
| PCBs    | polychlorinated biphenyls              |
| PFAA    | perfluoroalkyl acid                    |
| PFAS    | per- and polyfluoroalkyl substance     |
| PFBS    | perfluorobutane sulfonate              |
| PFCA    | perfluoroalkyl carboxylic acid         |
| PFDA    | perfluorodecanoate                     |
| PFDoDA  | perfluorododecanoate                   |
| PFHpA   | perfluoroheptanoate                    |
| PFHxA   | perfluorohexanoate                     |
| PFHxS   | Perfluorohexane sulfonate              |
| PFNA    | perfluorononanoate                     |
| PFOA    | perfluorooctanoate                     |
| PFOS    | perfluorooctane sulfonate              |
| PFPeA   | perfluoropentanoate                    |
| PFSA    | perfluoroalkyl sulfonate               |
| PFTeDA  | perfluorotetradecanoate                |
| PFTTrDA | perflurotridecanoate                   |
| PFUnA   | perfluoroundecanoate                   |
| Po      | polonium                               |
| Ra      | radium                                 |
| RL      | reporting limit                        |
| Rn      | radon                                  |
| RPD     | Relative percent difference            |
| T-      | total-                                 |
| TOC     | Total organic carbon                   |
| US EPA  | U.S. Environmental Protection Agency   |
| WWTP    | Wastewater treatment plant             |
| XPS     | extruded polystyrene                   |

### Units of Measurement

|                         |  |
|-------------------------|--|
| ac                      | acre                                       |
| °C                      | degrees centigrade                         |
| cm                      | centimeter                                 |
| dw                      | dry weight                                 |
| ft                      | feet                                       |
| g                       | gram, a unit of mass                       |
| g/cm <sup>3</sup>       | grams per centimeter cubed                 |
| mg/kg                   | milligram per kilogram                     |
| ng/g                    | nanograms per gram (parts per billion)     |
| ng/cm <sup>2</sup> /yr  | nanograms per centimeter squared per year  |
| pCi/g                   | picocuries per gram                        |
| pCi/cm <sup>2</sup> /yr | picocuries per centimeter squared per year |
| µg/cm <sup>2</sup> /yr  | microgram per centimeter squared per year  |

# Appendices

## Appendix A. PFAS Analytes and Reporting Limits

Table A-1. PFAS measured and median reporting limits for this study.

| Analyte name                                 | CAS Number  | Acronym   | Median RL (ng/g dw) | Median MDL (ng/g dw) |
|--|-------------|-----------|---------------------|----------------------|
| Perfluorobutane sulfonate                    | 45187-15-3  | PFBS      | 1.8                 | 0.1                  |
| Perfluorodecanoate                           | 73829-36-4  | PFDA      | 1.8                 | 0.1                  |
| Perfluorododecanoate                         | 171978-95-3 | PFDODA    | 3.6                 | 0.1                  |
| Perfluoroheptanoate                          | 120885-29-2 | PFHpA     | 1.8                 | 0.3                  |
| Perfluorohexane sulfonate                    | 108427-53-8 | PFHxS     | 1.8                 | 0.6                  |
| Perfluorohexanoate                           | 92612-52-7  | PFHxA     | 1.8                 | 0.3                  |
| Perfluorononanoate                           | 72007-68-2  | PFNA      | 1.8                 | 0.2                  |
| Perfluorooctane sulfonate                    | 45298-90-6  | PFOS      | 1.8                 | 0.2                  |
| Perfluorooctanoate                           | 45285-51-6  | PFOA      | 1.8                 | 0.2                  |
| Perfluoropentanoate                          | 45167-47-3  | PFPeA     | 1.8                 | 0.5                  |
| Perfluorotetradecanoate                      | 365971-87-5 | PFTeDA    | 7.3                 | 0.1                  |
| Perfluorotridecanoate                        | 862374-87-6 | PFTTrDA   | 7.3                 | 0.1                  |
| Perfluoroundecanoate                         | 196859-54-8 | PFUnA     | 1.8                 | 0.1                  |
| N-ethyl perfluorooctane sulfonamide acetate  | 2991-50-6   | N-EtFOSAA | 1.8                 | 0.2                  |
| N-methyl perfluorooctane sulfonamide acetate | 2355-31-9   | NMeFOSAA  | 1.8                 | 0.2                  |

RL = reporting limit

MDL = method detection limit

## **Appendix B. Study Location Descriptions**

### ***Blackmans Lake***

Blackmans Lake is a small lake within the city of Snohomish in Snohomish County. The watershed surrounding the lake is largely made up of residences and commercial development, which replaced logging and agriculture during the 1970s-1990s (ESA 2017). Water enters the lake through Grass Bottom Creek and two small inlets and flows out of the lake through Swifty Creek to the South. Precipitation in the area averages around 44" a year. The underlying basin geology is made up of Vashon till and recessional outwash (WGS 2022).

### ***Lake Sutherland***

Lake Sutherland is a remote lake on the Olympic Peninsula (Clallam County) near the edge of Olympic National Forest. Its watershed is largely undeveloped forestland, though there are several residences along the shoreline and a state highway near the northern shore. The lake is fed via several perennial streams around the lake and flows out to Indian Creek on the Eastern shore. An average of 58" of rain falls on the lake and watershed annually. The lake is surrounded by underlying volcanic rocks and landslide deposits (WGS 2022). Lake Sutherland was formed sometime after an ice sheet retreated about 13,000 years ago by landslides that separated a large waterbody into present-day Lake Crescent and Lake Sutherland (Logan and Schuster 1991).

### ***Wildcat Lake***

Wildcat Lake is located on the Kitsap Peninsula in a mostly forested watershed of Kitsap County. Ten percent of the watershed is impervious, with most development being residential (Kitsap County 2010). This lake was chosen based on its proximity to potential AFFF releases sites at a nearby military base and a small private airport. Neither of which are hydrologically connected via surface water, but regional re-deposition of PFAS has been found to be relevant to PFAS loading to waterbodies. The average yearly precipitation for Wildcat Lake is about 60". Soils surrounding the lake include till along the south shore and advanced outwash near the outlet of Wildcat Creek (Kitsap County 2010).

### ***Hicks Lake***

Hicks Lake is a 160-acre lake in the city of Lacey, Thurston County. It is the first of four lakes interconnected by Woodland creek, which eventually discharges into Henderson Inlet. The downstream lakes are Pattison Lake, Long Lake, and Lake Lois. The shoreline consists of residential waterfront lots. Land use in the watershed is primarily dense residential, but also includes small undeveloped forest lots and some commercial properties. There is a public access point on the west side of the lake via the Washington Fish and Wildlife boat launch and Wanschers Community Park. The lake is primarily used for recreational boating, fishing, and swimming. The watershed geology consists mostly of glacial till and some alluvial sediments (WGS 2022).

## ***Spanaway Lake***

Spanaway Lake is in Pierce County within a watershed dominated by residential, commercial, and military land. Many residences in the area use on-site septic systems, including along the lake shoreline (Hobbs et al. 2022) and as many as 4,000 are on septic systems in the watershed (Pierce County 2016). Major hydrological inputs to the lake include groundwater discharge and inflow from Coffee Creek on the southern shoreline (Pierce County 2017). Surface water drains from the lake to the north through Spanaway Creek. Average annual precipitation to the lake is 41". Spanaway Lake's watershed is composed of highly permeable glacial till and outwash soils overlying bedrock deposits, resulting in very little surface water runoff (Pierce County 2017; Johnson et al. 2011).



## Appendix B References

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## Appendix C. Supporting data

**Table C-1. Total organic carbon and radiochemistry results for the Blackmans Lake core.**

| Lake      | Sediment interval (cm depth) | TOC (%) | <sup>210</sup> Pb (pCi/g) | <sup>226</sup> Ra (pCi/g) |
|-----------|------------------------------|---------|---------------------------|---------------------------|
| Blackmans | 0–2                          | 14      | 7.69 J                    | ---                       |
| Blackmans | 2–3                          | 14      | 6.02                      | ---                       |
| Blackmans | 3–4                          | 14      | 5.50                      | ---                       |
| Blackmans | 5–6                          | 14      | 6.12                      | ---                       |
| Blackmans | 7–8                          | ---     | 5.57                      | ---                       |
| Blackmans | 9–10                         | 15      | 5.38                      | ---                       |
| Blackmans | 11–12                        | ---     | 5.40                      | 0.30                      |
| Blackmans | 13–14                        | 14      | 4.38                      | ---                       |
| Blackmans | 16–17                        | 13      | 3.24                      | ---                       |
| Blackmans | 19–20                        | ---     | 2.86                      | ---                       |
| Blackmans | 24–25                        | ---     | 1.67                      | 0.25                      |
| Blackmans | 29–30                        | 16      | 0.48                      | ---                       |
| Blackmans | 34–35                        | ---     | 0.18                      | ---                       |
| Blackmans | 39–40                        | 18      | 0.18                      | ---                       |
| Blackmans | 41–42                        | 18      | 0.20                      | 0.19                      |

--- = not analyzed in that interval; J = estimated value.

**Table C-2. Total organic carbon and radiochemistry results for the Sutherland Lake core.**

| Lake       | Sediment interval (cm depth) | TOC (%) | <sup>210</sup> Pb (pCi/g) | <sup>226</sup> Ra (pCi/g) |
|------------|------------------------------|---------|---------------------------|---------------------------|
| Sutherland | 0–2                          | 6.5     | 9.23                      | ---                       |
| Sutherland | 2–3                          | 5.5     | 5.7                       | ---                       |
| Sutherland | 3–4                          | 7.5     | 4.26                      | ---                       |
| Sutherland | 5–6                          | 5.6     | 4.77                      | ---                       |
| Sutherland | 7–8                          | 5.3     | 4.35                      | ---                       |
| Sutherland | 9–10                         | ---     | 3.28                      | 0.171                     |
| Sutherland | 11–12                        | 4.0     | 1.99                      | ---                       |
| Sutherland | 14–15                        | ---     | 1.95                      | ---                       |
| Sutherland | 17–18                        | 5.7     | 1.37                      | ---                       |
| Sutherland | 20–21                        | ---     | 0.537                     | ---                       |
| Sutherland | 23–24                        | ---     | 0.778                     | 0.26                      |
| Sutherland | 27–28                        | 5.3     | 0.54                      | ---                       |
| Sutherland | 31–32                        | ---     | 0.20                      | ---                       |
| Sutherland | 35–36                        | 7.0     | 0.16                      | ---                       |
| Sutherland | 40–41                        | 8.5     | 0.24                      | 0.21                      |

--- = not analyzed in that interval; J = estimated value.

**Table C-3. Total organic carbon and radiochemistry results for the Wildcat Lake core.**

| Lake    | Sediment interval (cm depth) | TOC (%) | <sup>210</sup> Pb (pCi/g) | <sup>226</sup> Ra (pCi/g) |
|---------|------------------------------|---------|---------------------------|---------------------------|
| Wildcat | 0–2                          | 7.5     | 5.71 J                    | ---                       |
| Wildcat | 2–3                          | ---     | 5.06                      | ---                       |
| Wildcat | 3–4                          | 8.2     | 4.16                      | ---                       |
| Wildcat | 5–6                          | 9.2     | 3.70                      | ---                       |
| Wildcat | 7–8                          | 10      | 1.17                      | ---                       |
| Wildcat | 9–10                         | ---     | 1.75                      | 0.26                      |
| Wildcat | 11–12                        | 11      | 1.34                      | ---                       |
| Wildcat | 13–14                        | 11      | 2.29                      | ---                       |
| Wildcat | 17–18                        | 11      | 0.56                      | ---                       |
| Wildcat | 21–22                        | ---     | 0.23                      | 0.16                      |
| Wildcat | 25–26                        | 11      | 0.20                      | ---                       |
| Wildcat | 29–30                        | ---     | 0.15                      | ---                       |
| Wildcat | 34–35                        | 9.9     | 0.13                      | ---                       |
| Wildcat | 40–41                        | ---     | 0.11                      | ---                       |
| Wildcat | 42–43                        | 10      | 0.07                      | 0.16 J                    |

--- = not analyzed in that interval; J = estimated value.

**Table C-4. Total organic carbon results for the Hicks Lake core.**

| Lake  | Sediment interval (cm depth) | TOC (%) |
|-------|------------------------------|---------|
| Hicks | 0–2                          | 19      |
| Hicks | 2–3                          | 19      |
| Hicks | 3–4                          | 19      |
| Hicks | 5–6                          | 19      |
| Hicks | 9–10                         | 19      |
| Hicks | 13–14                        | 20      |
| Hicks | 16–17                        | 19      |
| Hicks | 29–30                        | 18      |
| Hicks | 39–40                        | 17      |
| Hicks | 42–43                        | 16      |

**Table C-5. Radiochemistry results for the Hicks Lake core.**

| Lake  | Sediment interval (cm depth) | <sup>210</sup> Pb (pCi/g) | <sup>226</sup> Ra (pCi/g) |
|-------|------------------------------|---------------------------|---------------------------|
| Hicks | 0–2                          | 6.88                      | ---                       |
| Hicks | 6–7                          | 2.75                      | ---                       |
| Hicks | 8–9                          | 2.33                      | ---                       |
| Hicks | 10–11                        | 1.83                      | ---                       |
| Hicks | 12–13                        | 1.54                      | ---                       |
| Hicks | 14–15                        | ---                       | 0.27 J                    |
| Hicks | 15–16                        | 1.24                      | ---                       |
| Hicks | 17–18                        | 0.69                      | ---                       |
| Hicks | 21–22                        | 0.396                     | ---                       |
| Hicks | 23–24                        | ---                       | 0.25 J                    |
| Hicks | 25–26                        | 0.31                      | ---                       |
| Hicks | 27–28                        | 0.28 U                    | ---                       |
| Hicks | 31–32                        | 0.14 U                    | ---                       |
| Hicks | 33–34                        | 0.16 U                    | ---                       |
| Hicks | 36–37                        | 0.39                      | ---                       |
| Hicks | 37–38                        | 0.248                     | ---                       |
| Hicks | 38–39                        | ---                       | 0.32 J                    |
| Hicks | 39–40                        | ---                       | ---                       |
| Hicks | 41–42                        | 0.21 U                    | ---                       |

--- = not analyzed in that interval; J = estimated value.

**Table C-6. Total organic carbon results for the Spanaway Lake core.**

| Lake     | Sediment interval (cm depth) | TOC (%) |
|----------|------------------------------|---------|
| Spanaway | 0–2                          | 13      |
| Spanaway | 2–3                          | 13      |
| Spanaway | 3–4                          | 13      |
| Spanaway | 5–6                          | 13      |
| Spanaway | 7–8                          | 13      |
| Spanaway | 11–12                        | 13      |
| Spanaway | 19–20                        | 15      |
| Spanaway | 21–22                        | 15      |
| Spanaway | 29–30                        | 16      |
| Spanaway | 40–41                        | 17      |

**Table C-7. Radiochemistry results for the Spanaway Lake core.**

| Lake     | Sediment interval<br>(cm depth) | <sup>210</sup> Pb (pCi/g) | <sup>226</sup> Ra (pCi/g) |
|----------|---------------------------------|---------------------------|---------------------------|
| Spanaway | 0–2                             | 8.24                      | ---                       |
| Spanaway | 4–5                             | 7.71                      | ---                       |
| Spanaway | 7–8                             | 6.26                      | ---                       |
| Spanaway | 8–9                             | ---                       | 0.22 J                    |
| Spanaway | 10–11                           | 4.44                      | ---                       |
| Spanaway | 13–14                           | 3.23                      | ---                       |
| Spanaway | 16–17                           | 2.82                      | ---                       |
| Spanaway | 17–18                           | ---                       | 0.21 J                    |
| Spanaway | 20–21                           | 1.95                      | ---                       |
| Spanaway | 22–23                           | 1.89                      | ---                       |
| Spanaway | 24–25                           | 1.22                      | ---                       |
| Spanaway | 27–28                           | 0.98                      | ---                       |
| Spanaway | 30–31                           | 1.09                      | ---                       |
| Spanaway | 33–34                           | 0.55                      | ---                       |
| Spanaway | 36–37                           | 0.39                      | ---                       |
| Spanaway | 38–39                           | ---                       | 0.10 J                    |
| Spanaway | 39–40                           | 0.27                      | ---                       |
| Spanaway | 41–42                           | 0.20                      | ---                       |

--- = not analyzed in that interval; J = estimated value.