WA-08-1110

January 10, 1974

Memo To: John Glynn

From: Ron C. Devitt

Subject: Cedar Hills Survey

<u>INTRODUCTION</u>: King County operates Cedar Hills sanitary landfill near Issaquah, Washington. Leachate, runoff and contaminated ground and spring water merge with a small unnamed creek and flow to Mason Creek. In the spring of 1972, a heavy slime growth developed in Mason Creek and downstream in Issaquah Creek at the Washington State Department of Fisheries hatchery at Issaquah.

Several studies were initiated to examine the problem.* The Washington State Department of Ecology undertook an independent survey to characterize the leachate and to determine the effect of the leachate on the water quality of Mason Creek.

<u>SUMMARY:</u> The flow originating from the disposal site is grossly polluted. As the distance from the fill increases, chemical, physical, and biological changes generally improve the quality of the leachate. In addition, the unnamed creek has a dilutional effect on the leachate before reaching Mason Creek.

During dry weather, the combined flow is so small that these is no direct above ground flow to Mason Creek. However, during wet weather flow, the aestethic and chemical nature of Mason Creek is significantly affected.

A similar discharge would not be permitted to such a small waterway by a Washington State industrial waste discharge permit.

The adverse effect of slime growth at the fish hatchery will be eliminated when the use of water from Issaquah Creek is discontinued and well water is employed.

*AN INVESTIGATION OF LANDFILL LEACHATE PROBLEMS AT KING COUNTY'S CEDAR HILLS SITE, John W. Mellor, University of Washington Masters thesis 1972.

CEDAR HILLS LANDFILL STUDY MUNICIPALITY OF METROPOLITAN SEATTLE, Feb. 1 thru June 21, 1972, J. T. Clark, R. J. Morrice, R. I. Matsuda, and R. S. Domenowske.



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STATION LOCATIONS

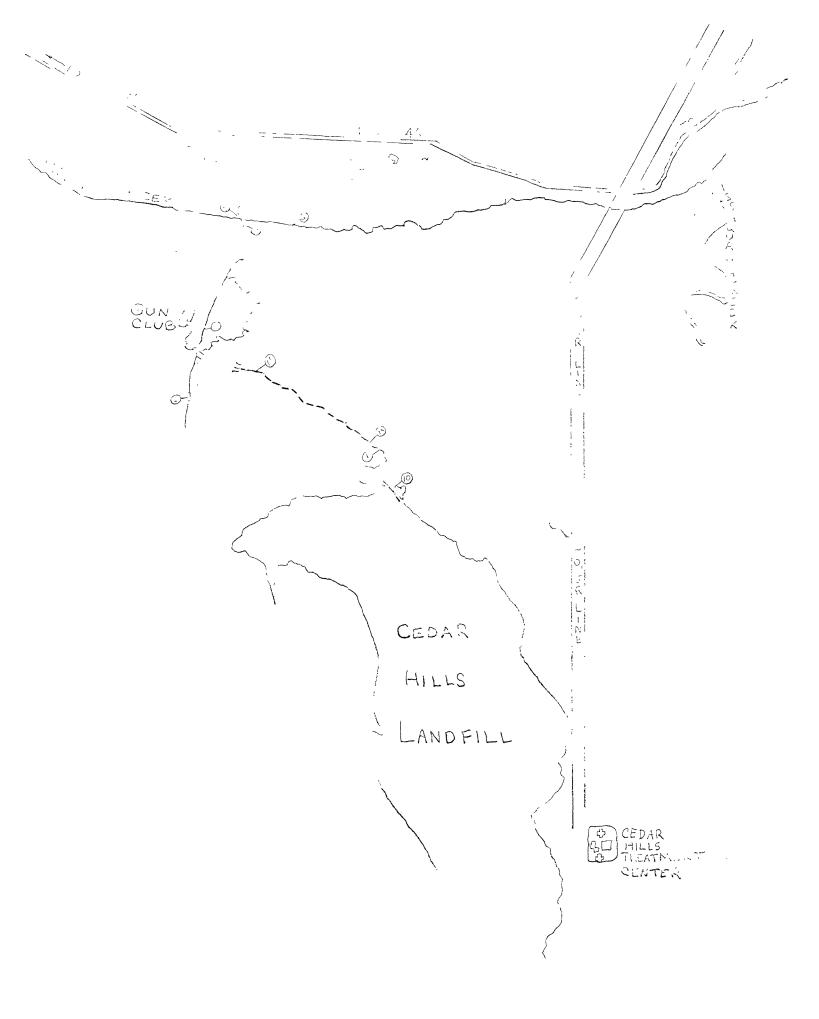
Primary stations were established at the following locations and are pinpointed on the attached map.

- STATION #1: Combined leachate and spring water in ravine as it emerges
 from ground.
- STATION #1B: Flow from 1A after passing downstream.
- STATION #2: Control to be compared to 1A, spring water and runoff from an uncontaminated area.
- STATION #3: Three combined flows from 1B and 2 in field at gun club.
- STATION #3A: Flow from station #3 ten yards above confluence with Mason Creek.
- STATION #4: Control for Mason Creek; Mason Creek at bridge about ten yards above confluence with flow from 3A.
- STATION #5: Mason Creek at bridge about 100 yards below confluence with flow from ##.3a,
- In addition, two secondary stations were sampled sporatically.
- STATION #9: Leachate and runoff originating in the south and east area of the fill at the powerline road.
- STATION #10: Runoff in leachate ditch (about 30 yards above Station 1A).

<u>GENERAL DRAINAGE</u>: Contaminated ground water emerges in a spring and combines with surface runoff and leachate at station #1. It proceeds down the ravine and merges with the uncontaminated water from Station #2. This combined flow enters Mason Creek at Station #3A.

<u>SAMPLING PROCEDURES:</u> From 10-31-72 to 5-29-73, six sets of grab samples were taken from the drainage from the landfill site from the control drainage and Mason Creek.

The samples were iced and analyzed by DOE water quality lab in Tumwater. Appropriate preservation was added to COD, nutrients, and iron samples. Dissolved oxygen samples were fixed in the field and determination was made by the azide modification of the Winkler method.



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UNITS OF MEASUREMENT:

pH: pH units Turbidity: Jackson Turbidity Units Conductivity: µmhos/cm @ 25°C. NO₃-N: ppm filtered NO₂-N: ppm filtered O-PO4-P: ppm filtered Total alkalinity: ppm as CaCO₃ Hardness: ppm as MgCo₃ + CaCO₃ Temperature: degrees Centigrade

All remaining results are report as parts per million.

DISCUSSION OF DATA:

Values obtained at the leachate-effected stations vary significantly. The amount of snowmelt and precipitation were the main factors in determining the amount of runoff. The more runoff, the more adversely the water quality was affected. Specifically on 10-31-72, the flow was so low that there was no above ground discharge to Mason Creek. Between Station #3 and #3A, about 10 yards below the gun club road, the flow (<1 cfs) was entering a pool and joining either ground water and/or underground creek. It would have been desirous to dye this subsurface flow in an attempt to determine if it did reach Mason Creek, but flow levels were not observed to be as low again. On subsequent dates, there was always a surface flow preventing the effective use of dye.

In contrast, it had been raining for three days preceeding the sampling on 4-18-73. Definite changes in water quality are due to the leachate.

Assuming that the differences between Station #2 and Station #1A are due to the leachate, a definite trend can be observed by comparing the mean values of selected parameters:

	Control (#2)	<u>Contaminated (#1A)</u>
pH (range) turbidity conductivity COD Iron Total Solids Total Non Vol. Solids Total Suspended Solids Total Suspended Non Vol. Alkalinity Chlorides Calcium Magnesium	7.0 - 7.8 1 88 10 <.1 67 31 4	$\begin{array}{r} 6.8 - 6.5 \\ 173 \\ 1290 \\ 790 \\ 79 \\ 1144 \\ 620 \\ 124 \\ 98 \\ 494 \\ 89 \\ 161 \end{array}$
Hardness	17	32 532

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The maximum values at Station #1A occurred during periods of high runoff.

Another significant observation is the general improvement in water quality due to natural purification of the combined leachate and runoff by travelling from Station #1A to 1B.

	<u>Station #1A</u>	<u>Station #1B*</u>
pH (range) turbidity conductivity COD Iron Total Solids Total Suspended Solids Total Non Vol. Solids Total Suspended Non Vol. Alkalinity Chlorides Calcium Magnesium	6.8 - 6.5 173 1290 790 79 1144 620 124 Solids 98 494 89 161 32	$\begin{array}{r} 8.1 - 7.1 \\ 85 \\ 865 \\ 432 \\ 35 \\ 676 \\ 371 \\ 50 \\ 38 \\ 338 \\ 64 \\ 122 \\ 24 \end{array}$
Hardness	532	404

*Data from 4-18-73 not included in calculation because of increased flow before sampling.

The most drastic effect on the water quality of Mason Creek was reported on 4-18-73.

Washington State water quality criteria include Implementation and Enforcement Plan for Water Quality Regulations Department of Ecology 1970:

- 1. Turbidity shall not exceed 5 JTU over natural conditions. The natural conditions (Sta. #4) was 7 JTU; downstream (Sta. #5) the turbidity of Mason Creek was 30 JTU, or an increase of 23 JTU.
- 2. pH shall be within the range of 6.5 to 8.5 with an induced variation of less than 0.25 units. Station #4 was 6.9, Station #5 was 6.6 or an induced variation of 0.3.
- 3. Aesthetic values shall not be impaired . . . which offend the senses of sight, smell, touch, or taste. By this definition the aesthetics of Mason Creek were impaired.
- 4. Deleterious material concentrations shall be below those of public health significance, . . . or which may adversely affect any water use.

The characteristic uses of Class A water include domestic, industrial, and agricultural water supply.

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Water quality criteria¹ say that concentrations of 0.3 mg/l and 0.1 mg/l should not be deleterious to the uses of water for domestic and industrial water supplies, respectively. It is also reported that all of the waters in the United States which support good fish populations have iron concentrations below 0.7 mg/l. The concentrations of iron at Station #5 was 6.8 ppm.

Metro's survey documented that Mason Creek was substandard for total coliform although this fact was not due to effects of the leachate.

BIOLOGICAL SAMPLING:

To evaluate the effect of the leachate on biological colonization, artificial substrates were placed at Station #4 and #5 on December 5, 1972.

The substrate was similar to that developed by $Britt^2$; a 12"x12"x3" concrete anchor was poured inside a plywood frame. A redwood lattice was attached to the anchor. A modificiation was made in an attempt to quantilatively evaluate slime growth development. A 9-1/2"x4"x1/8" piece of transite was attached to a corner of the wooden lattice.

Although slime growth developed on some of the vegetation at Station #5, it failed to establish on either the wood or transite.

Invertebrates did colonize the substrates. On 5-29-73, the specimens were collected, preserved in ethanol, identified, and enumerated at the water quality laboratory in Tumwater.

		<u>Station #4</u>	<u>Station #5</u>
Tubifex Chironomids Leeches Fingernail Clams Tabanidae		31 13 8 4 1	268 140 1 1 0
	Total	57	410

The results are reported below:

As indicators of water quality, neither population would typify a "clean water" situation. The increased numbers of tubifex ("sludge worms") and chironomids ("blood worms") and decreased diversity at Station #5 would

- 1. Water Quality Criteria, 2nd Edition State Water Quality Control Board, Sacramento, California, 1963.
- New Methods of Collecting Bottom Fauna from Shoals or Rubble Bottoms of Lake and Streams, Ecology 36(3): 524-525 Britt, N.W. 1955.

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tend to indicate an enrichment of the creek not demonstrated by chemical sampling. Conditions were present which promoted the growth of "polluted water" macroorganisms.

The difference between the populations at Station #4 and #5 are considered to be due to the leachate. The water velocity and depth were similar; the substrate was identical. The development of these populations over a period of six months is a significant method of evaluating biological conditions in situ.

RCD:1t 4/1

Ceo	dar Hills	Leachat	e and	Mason	Cree	k -	Decemb	er 5,	1972	
Station:	1	٦Α	1 B	2	3	3 A	4	5	9	10
Parameter										
pH Conductivity COD NO ₃ -N NO ₂ -N NH ₃ -N	6.6 410 62 ND	6.8 890 520 1.7	7.5 850 450	7.5 75 8 ND		7.4 65 8 ND	6.7 82 ND	6.8 85 ND	6.8 240 110 ND	
Total Kjeldahl-N O-PO ₄ -P Total PhosP	ND	2.2	1.8	ND		ND	.04	.04	ND	
Total Solids *										
Total Non Vol. Soli Total Suspended Sol Total Sus N.Vol Sol Total Alkalinity	ids									
Calcium Magnesium Chlorides Iron Sulfates Hardness Turbidity	40 8.2 32 18 ND 130 20	116 28 77 60 ND 400 20	106 24 77 24 ND 360 20	2.9 2.7 29 0.1 3 18 1		2.8 2.3 25 0.1 6 15 3	3.0 2.8 27 0.4 5 17 5	2.7 2.5 27 0.3 6 16 5	31 9.0 38 3.6 3 110 6	

* Insufficient sample

Cedar H	ills	Leachat	e and	Mason	Creek -	Januar	y 2,	1973	
Station:	1	1 A	1 B	2	3 3A	4	5	9	10
Parameter									
pH Condu ctivity COD NO ₃ -N NO ₃ -N	6.5 260 31	6.7 406 360	7.1 490 270	7.0 63 12	7.3 140 51	59	6.6 73 12	7.0 150 54	
NO2-N NH3-N Total Kjeldahl-N O-PO4-P Total PhosP	.30 .54	2.40 2.70	2.46 3.00		.28 .40		.06 .24	.10 .30	
Total Solids	140	528	419	49	71	40	50	126	
Total Non Vol. Solids Total Suspended Solids Total Sus N.Vol Solids Total Alkalinity	120 2 2 100	328 104 71 240	231 48 40 190	38 1 1 14	4 2 6 6 4 0	3 1	19 4 3 41	88 4 2 54	
Calcium Magnesium Chlorides Iron Sulfates	23 5.3 15 16	74 16 40 35	64 11 34 25	2.1 1.8 9 ND	10 3.6 12 2.5	1.7 9	10	15 4.7 2.8	
Hardness	79	250	200	13	40	13	1000 ang	56	

-- Insufficient sample ND None Detected

Cedar Hills Leachate and Mason Creek - March 8, 1973

Station:	1	1A	1 B	2	3	3A	4	5	9	10
Parameter										
pH Conductivity COD NO ₃ -N NO ₂ -N NH ₃ -N Total Kjeldahl-N O-PO ₄ -P Total PhosP	6.5 385 110 .010 .01 .01 .02 .01 .05	6.5 955 631 .010 .02 .15 .16 .01 .05	7.5 820 490 .030 ND .12 .13 .01 .05	7.3 90 8 .038 ND .01 .01 .01 .05	7.1 180 59 1.63 ND .01 .01 .01 .05	7.1 190 59 1.21 ND C 01 .01 .01 .05	6.5 84 16 1.18 ND .01 .01 .01 .05	6.7 105 12 .68 ND .01 .01 .01 .05		6.5 2500 2490 .77 .01 .65 .75 .01 .05
Total Solids	254	814	. 620	67	117	5130	69	78		2570
Total Non Vol. Solids Total Suspended Solids Total Sus N.Vol Solids Total Alkalinity	162 39 -26 150	430 74 54 384	318 37 25 294	51 4 20	100 2 0 58	5100 10 2 60	60 2 0 24	60 5 2 34		1090 180 140 860
Calcium Magnesium Chlorides Iron Sulfates Hardness	46 9.6 21 25 155	150 26 58 75 480	135 25 54 50 440	2.7 2.0 18 0.1 15	17 4.5 15 2.4 60	17 4.5 15 1.7 60	5.7 1.9 11 0.6 22	7.6 2.4 9.7 0.6 29		440 76 165 160 1400

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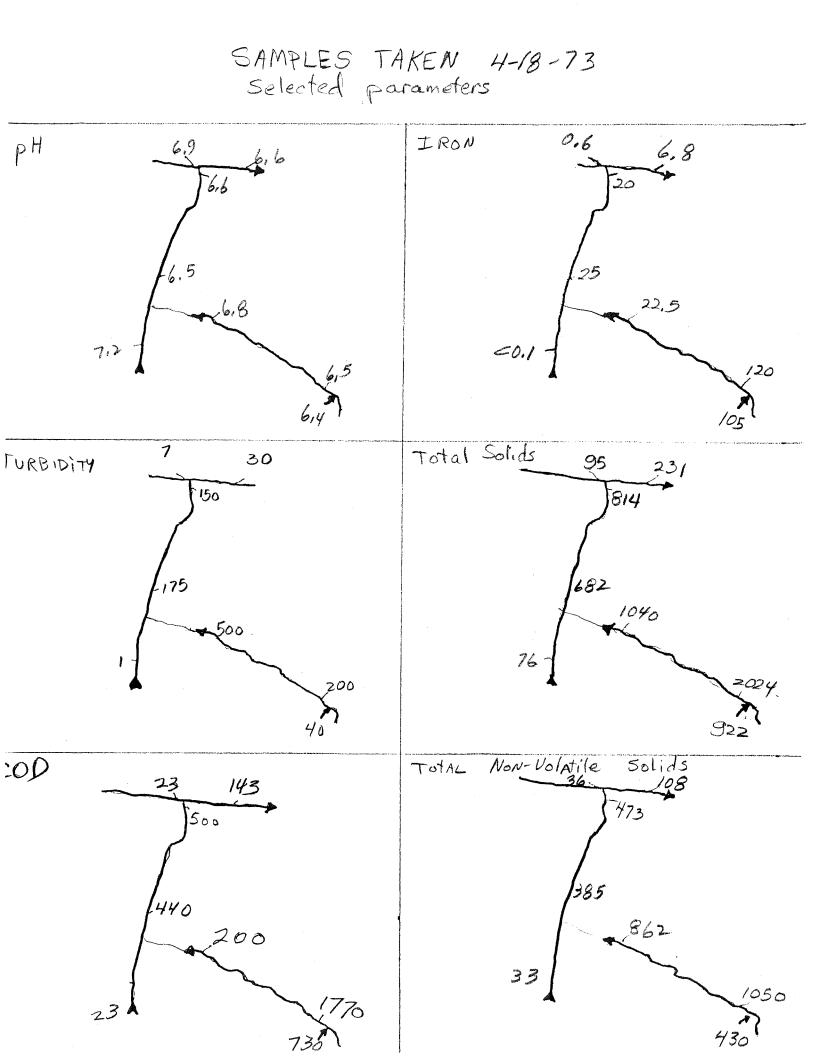
Cedar	Hills	Leacha	te and	Mason	Cree	k - /	April	18, 1	973	
Station:	1	14	18	2	3	3A	4	5	9	10
Parameter										
pH Conductivity COD NO ₃ -N NO ₂ -N NH ₃ -N Total Kjeldahl-N	6.4 1050 730 ND .04	6.5 1900 1770 ND .04	6.8 270 200 .03 .01	7.2 83 23 1.07 .01	6.5 520 440 .39 .01	6.6 590 500 .35 .01	6.9 85 23 .35 .01	6.6 230 143 .43 .01	6.9 310 60 .03 .01	
0-PO ₄ -P Total PhosP	.02 .12	.02 .28	.02 .76	ND .20	ND .32	ND .24	ND .24	ND .24	ND .16	
Total Solids	922	2024	.1040	76	682	814	95	231	210	
Total Non Vol. Solids Total Suspended Solid Total Sus N.Vol Solid Total Alkalinity		1050 184 156 690	862 932 781 60	33 1 0 24	385 239 209 160	473 180 152 180	36 4 1 28	108 20 20 74	105 20 4 100	
Calcium Magnesium Chlorides Iron Sulfates Hardness Turbidity(JTU)	150 26 51 105 ≺ 5 480 40	250 50 120 120 < 5 830 200	24 8.5 16 27.5 <5 95 500	2.6 2.2 7 4 0.1 4 5 16 1	75 13.6 33 25 4 5 245 175	90 15.6 37 20 ≮5 290 150	2.7 2.1 8 0.6 <5 15 7	35 6.3 15 6.8 9 115 30	40 8.2 22 4.6 6 135 20	

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Cedar Hills Leachate and Mason Creek - May 29, 1973 Station: 3 1 1A 1 B 2 3A 4 5 9 10 Parameter pН 6.4 6.6 8.1 7.8 6.7 7.0 7.3 Conductivity 2300 120 160 1400 1300 160 98 COD 772 650 517 4 24 12 8 $NO_3 - N$.03 .36 .24 1.22 .02 1.04 .10 NOZ-N ND ND ND ND ND ND ND NH3-N .24 .46 1.02 .14 ND .64 .10 Total Kjeldahl-N 1.26 .02 1.12 .94 .14 .26 .10 0-P04-P .10 .02 .06 .04 .02 .02 .02 Total Phos.-P .12 .14 .08 .05 .04 .06 .06 1208 Total Solids 1158 . 990 75 208 122 58 Total Non Vol. Solids 591 673 565 0 144 68 13 Total Suspended Solids Total Sus N.Vol Solids 129 95 135 7 66 9 10 90 111 0 3 50 101 3 Total Alkalinity 570 620 530 30 62 66 100 Calcium 210 215 185 3 17 18 11 36 39 37 2.8 Magnesium 5 4.8 2.3 Chlorides 94 105 92 12 17 21 16 Iron 115 105 40 ND 6.5 2.6 0.5 Sulfates 670 700 19 Hardness 615 63 65 37

ND = None Detected



4-18-73 CONTINUED

